

Cooling Tower: Part I

Thermal, Fluids, and Energy
Systems Lab

(ME EN 4650)

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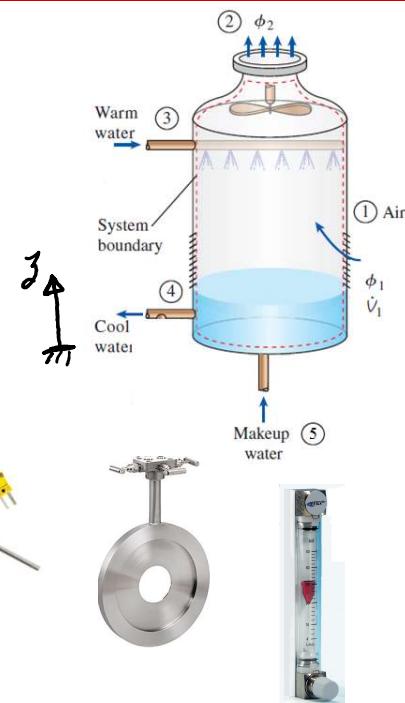
Cooling Tower Performance

What we want to know:

- $T_{db}, T_{wb}, T_w \text{ vs } \dot{f}$
- Efficiency, $\eta \text{ vs } \dot{m}_{w,in}$
- $\dot{m}_{w,out}$
- \dot{Q} , heat rejected

What we can measure:

- $T_{db}, T_{wb}, T_w \text{ vs. } \dot{f}$
(thermocouples)
- ΔP (orifice plate) → \dot{m}_a
- $\dot{m}_{w,in}$ (rotameter)



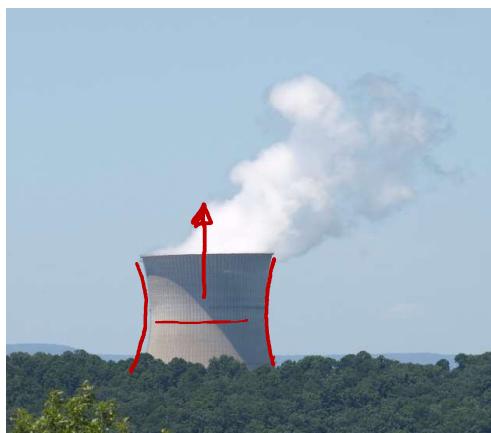
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Lecture Outline

- Background
- Experimental Setup
- Measurements and Instrumentation
- Required Figures
- Data Analysis

Hyperboloid Cooling Towers



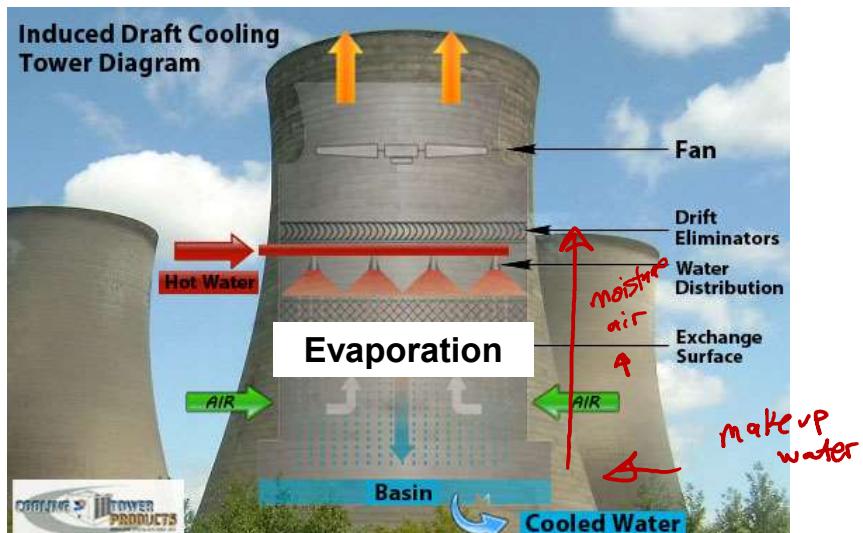
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Rectangular Cooling Towers



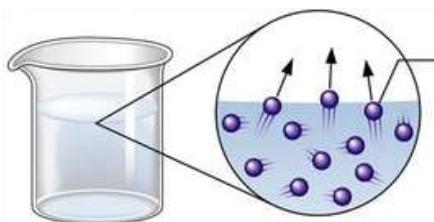
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Induced Draft Counterflow Cooling Tower



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Evaporation



The high energy molecules escape the surface.

$$\text{Evaporation criterion: } \frac{P_v}{\text{vapor pressure of water}} < \frac{P_g}{\text{saturation pressure of liquid water}}$$

@ room temp,
 $P_v = 2.4 \text{ kPa}$
 $P_g = 2.844 \text{ Pa}$

- Evaporation will ALWAYS occur at room temperature, as long as air is NOT saturated.

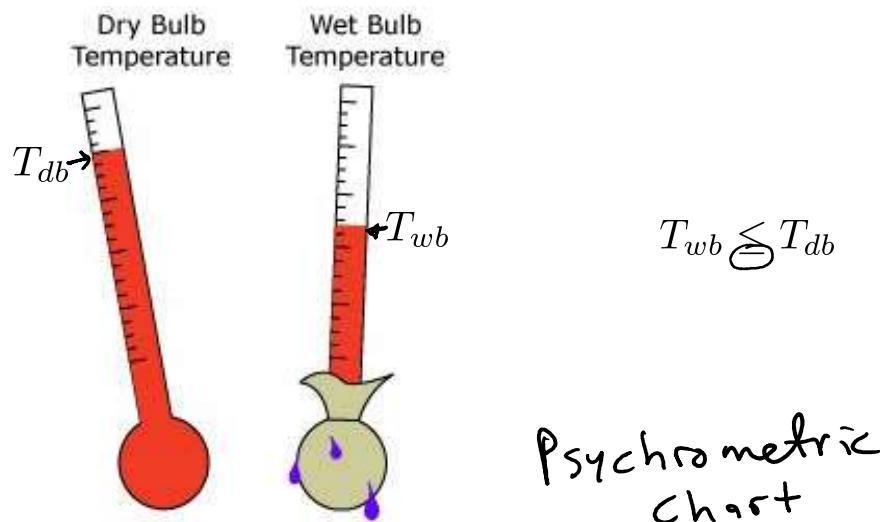
Rate of heat rejected: $Q_e = \dot{m}_e h_e$

$\uparrow \quad \uparrow$
 2256 kJ/kg
 0.4 g/s

$\approx 1 \text{ kW}$

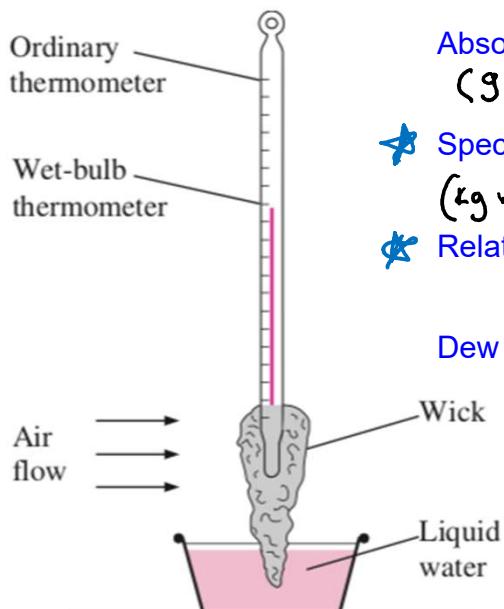
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Wet-Bulb Temperature



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Wet-Bulb Temperature Measuring Water Vapor Content in Air



Absolute Humidity: mass water vapor per (g w.v./cm³air) unit volume of air

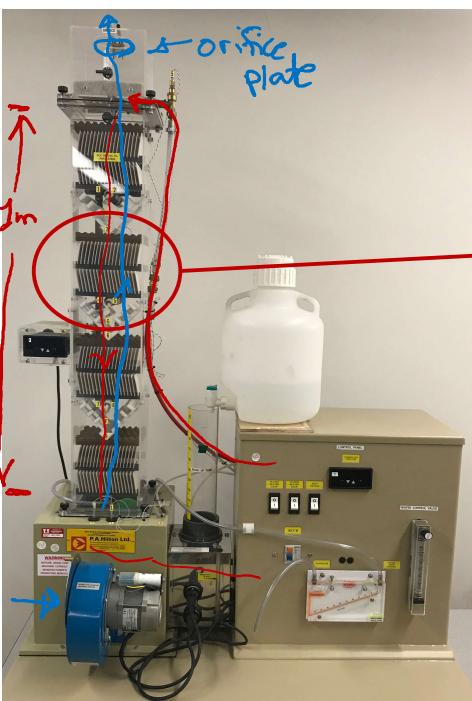
* **Specific Humidity, ω :** mass of water vapor (g w.v./kg dry air) per unit mass of dry air

* **Relative Humidity, ϕ :** absolute humidity over max allowable (%)

Dew Point Temperature: temp @ which air must be cooled to be sat. w/ w.v. (°C)

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Experimental Setup



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Measurements



Quantity	Symbol	Units	Instrument
Dry bulb temperature	T_{db}	°C	thermocouple
Wet bulb temperature	T_{wb}	°C	thermocouple with wet cloth
Water temperature	T_w	°C	thermocouple
Orifice meter pressure drop	ΔP	mm H ₂ O	inclined manometer
Inlet water flow rate	$\dot{m}_{w_{in}}$	g/s	rotameter



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Experiments

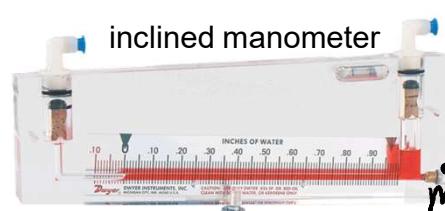
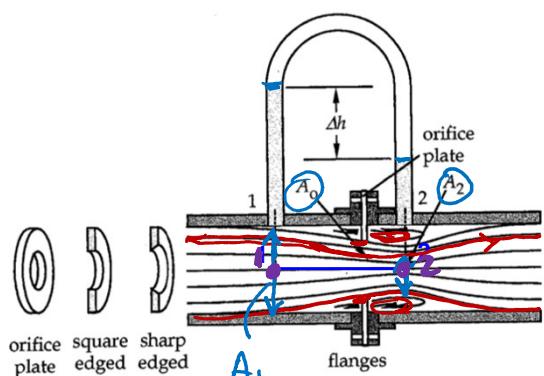


Quantity	Symbol	Units	Instrument
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Orifice meter pressure drop	ΔP	mm H ₂ O	inclined manometer
Inlet water flow rate	$\dot{m}_{w_{in}}$	g/s	rotameter

Exp	$\dot{m}_{w_{in}} \text{ (g/s)}$
1	30
2	20
3	40

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Air Flow Rate Measurement: Sharp-Edged Orifice Meter



Conservation of mass:

$$\rho A_1 V_1 = \rho A_2 V_2$$

$$V_1 = \frac{A_2}{A_1} V_2$$

Bernoulli's equation:

$$\frac{P_1}{\rho} + \frac{1}{2} V_1^2 = \frac{P_2}{\rho} + \frac{1}{2} V_2^2$$

$$\frac{P_1}{\rho} + \frac{1}{2} \left(\frac{A_2}{A_1} \right)^2 V_1^2 = \frac{P_2}{\rho} + \frac{1}{2} V_2^2$$

$$V_2 = \sqrt{\frac{(P_1 - P_2)}{\rho} \left[\frac{1}{2} - \left(\frac{A_2}{A_1} \right)^2 \right]^{-1}}$$

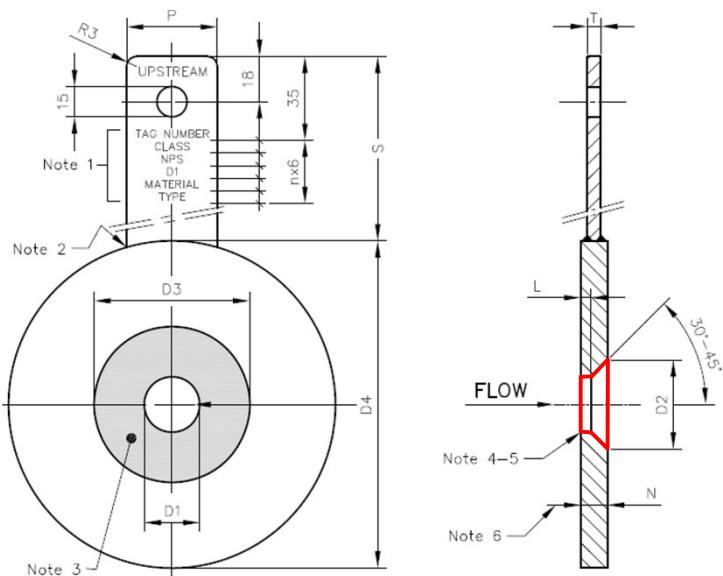
discharge coeff.

$$V_2 = \tilde{A} \sqrt{\frac{\Delta P}{\rho}}, \quad \Delta P = P_1 - P_2$$

$$\dot{m} = \rho A_2 V_2 = \rho A_2 \tilde{A} \sqrt{\frac{\Delta P}{\rho}} = C \sqrt{\rho \Delta P}$$

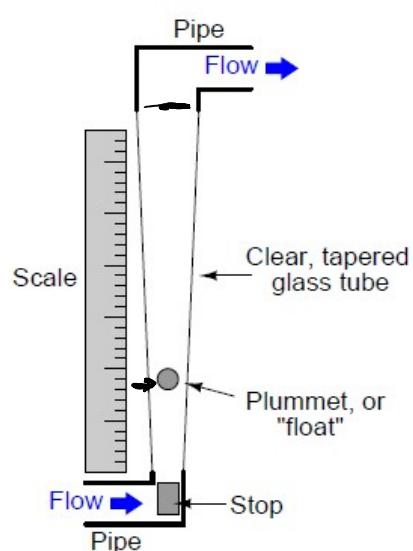
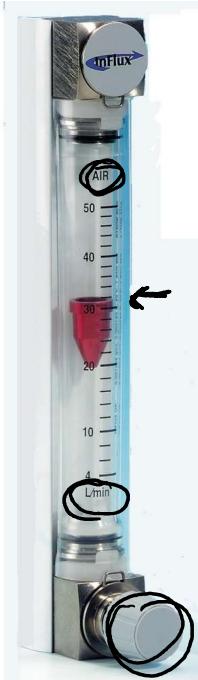
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Orifice Plate

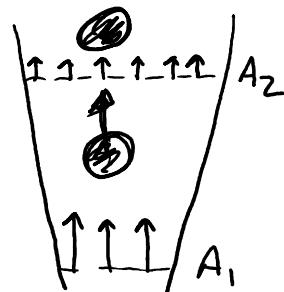


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Water Flow Rate Measurement: Rotameter

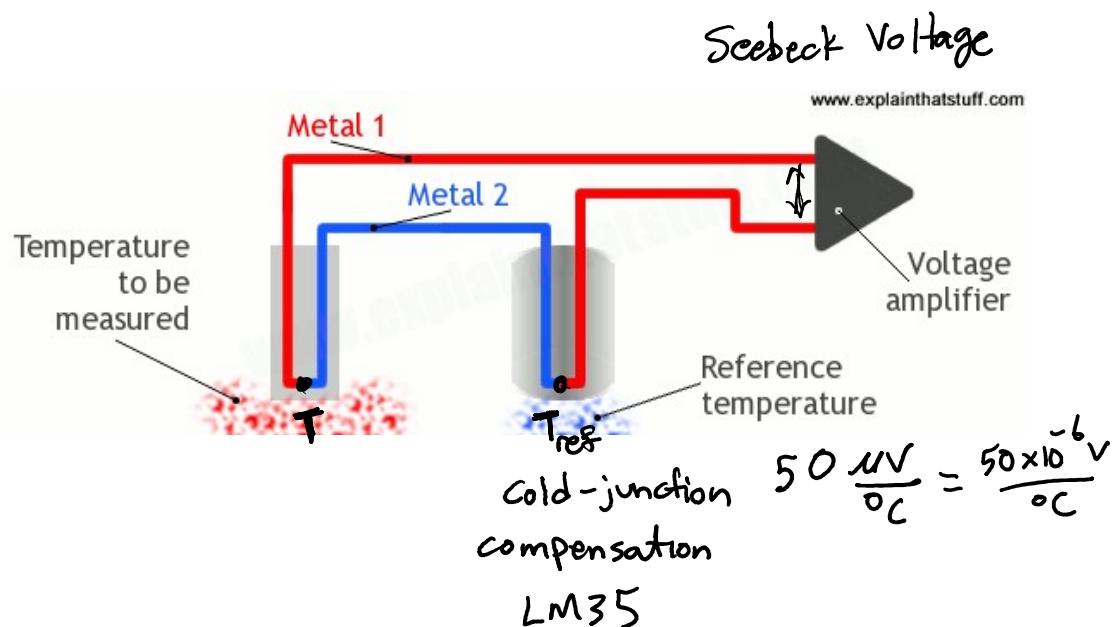


$$\frac{F_B}{\rho g} = C_D \frac{1}{2} \rho A_2 v^2$$

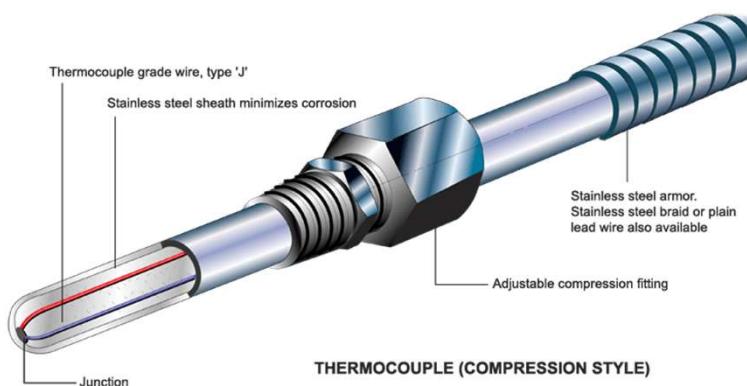


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Thermocouple: Principle of Operation



Thermocouple: Probe Type



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Thermocouple Application



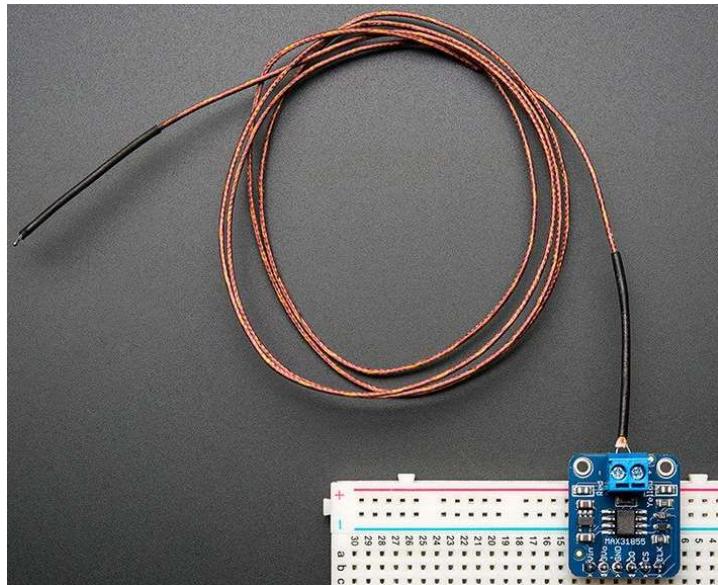
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Thermocouple: Wire Type

THERMOCOUPLE CHARACTERISTICS TABLE						
ANSI/ASTM	Symbol Single	Generic Names	Color Coding		Magnetic Yes/No	Environment (Bare Wire)
			Individual Conductor	Overall Jacket Extension Grade Wire		
T	TP TN	Copper Constantan, Nominal Composition: 55% Cu, 45% Ni	● Blue ● Red	● Blue	X X	Mild Oxidizing, Reducing. Vacuum or Inert. Good where moisture is present.
J	JP JN	Iron Constantan, Nominal Composition: 55% Cu, 45% Ni	○ White ● Red	● Black	X X	Reducing Vacuum, Inert. Limited use in oxidizing at High Temperatures. Not recommended for low temps.
E	EP EN	Chromel®, Nominal Composition: 90% Ni, 10% Cr Constantan, Nominal Composition: 55% Cu, 45% Ni	● Purple ● Red	● Purple	X X	Oxidizing or Inert. Limited use in Vacuum or Reducing.
K	KP KN	Chromel, Nominal Composition: 90% Ni, 10% Cr Alumel®, Nominal Composition: 95% Ni, 2% Mn, 2% Al	● Yellow ● Red	● Yellow	X	Clean Oxidizing and Inert. Limited use in Vacuum or Reducing
N	NP NN	Nicrosil®, Nominal Compositions: 84.6% Ni, 14.2% Cr, 1.4% Si Nisil®, Nominal Composition: 95.5% Ni, 4.4% Si, 1% Mg	● Orange ● Red	● Orange	X X	Clean Oxidizing and Inert. Limited use in Vacuum or Reducing
S	SP SN	Platinum 10% Rhodium Pure Platinum	● Black ● Red	● Green	X X	Oxidizing or Inert Atmospheres. Do not insert in metal tubes. Beware of contamination.
R	RP RN	Platinum 13% Rhodium Pure Platinum	● Black ● Red	● Green	X X	Oxidizing or Inert Atmospheres. Do not insert in metal tubes. Beware of contamination.
B	BP BN	Platinum 30% Rhodium Platinum 6% Rhodium	● Gray ● Red	● Gray	X X	Oxidizing or Inert Atmospheres. Do not insert in metal tubes. Beware of contamination.
C	P N	Tungsten 5% Rhenium Tungsten 26% Rhenium	● Green ● Red	● Red	X X	Vacuum, Inert, Hydrogen Atmospheres. Beware of Embrittlement.

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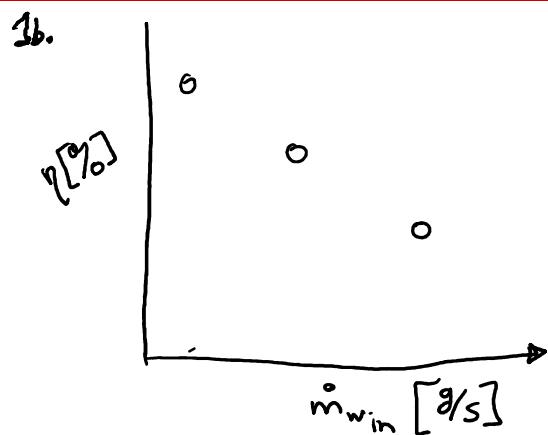
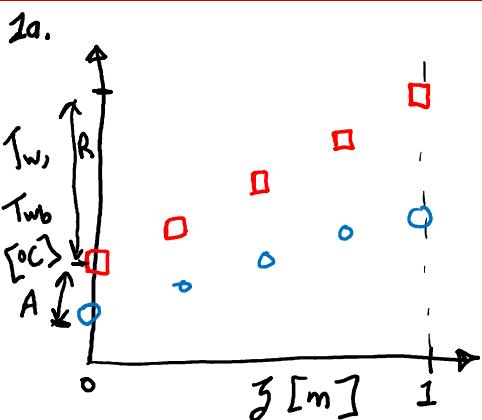
Thermocouple Breakout Board for Arduino



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Figures 1a & 1b



Range: $R = T_{w_{in}} - T_{w_{out}}$

$$\eta = \frac{R}{R+A} * 100$$

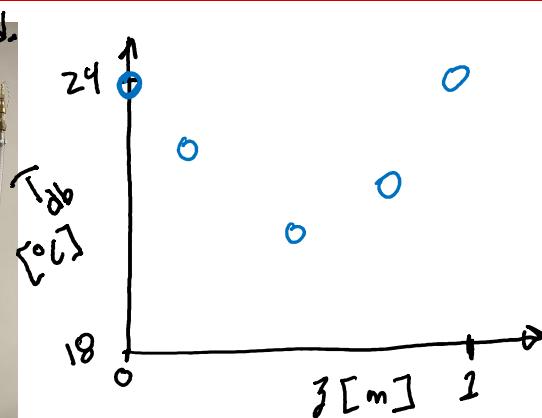
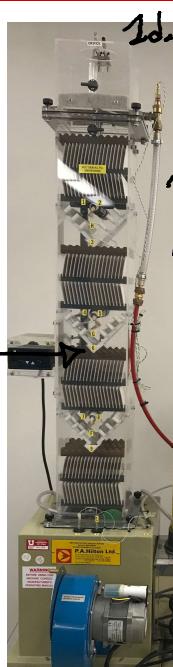
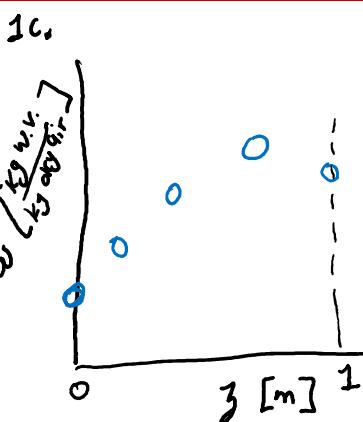
Approach: $A = T_{w_{out}} - T_{w_{in}}$

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Figures 1c & 1d

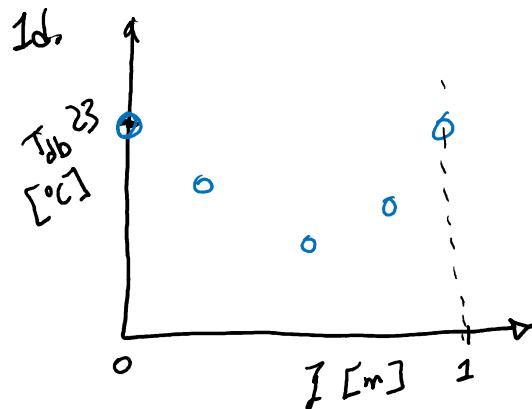
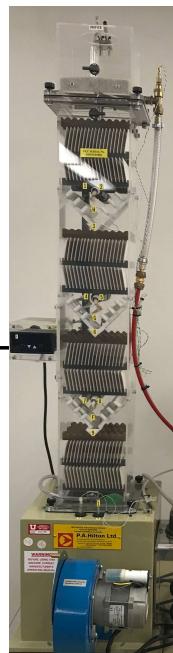
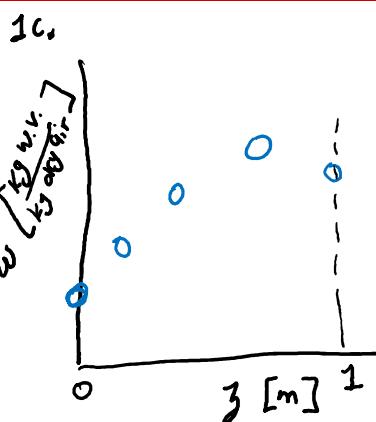


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Figures 1c & 1d

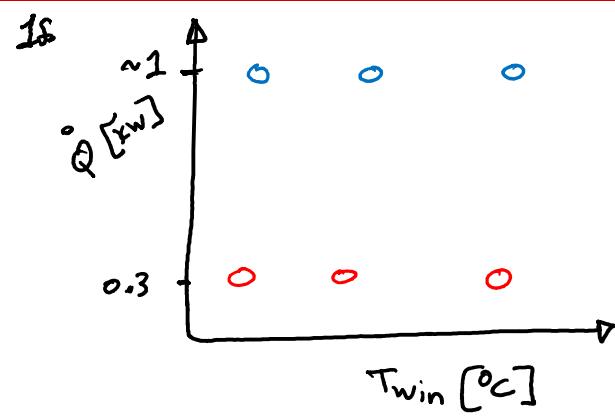
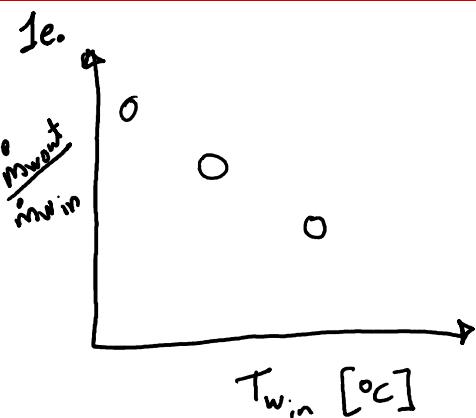


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Figures 1e & 1f



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Cooling Tower: Part II

Thermal, Fluids, and Energy
Systems Lab

(ME EN 4650)

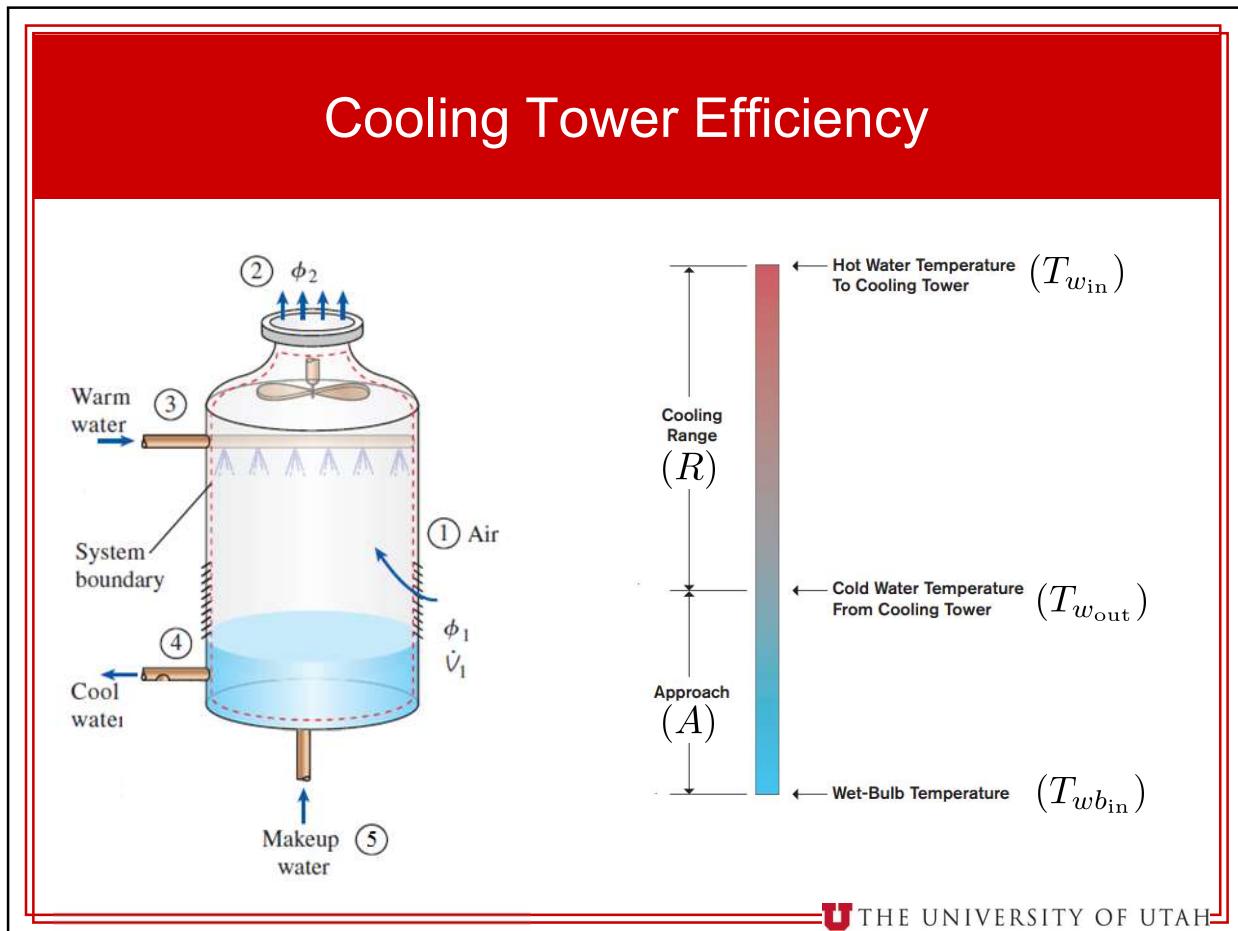
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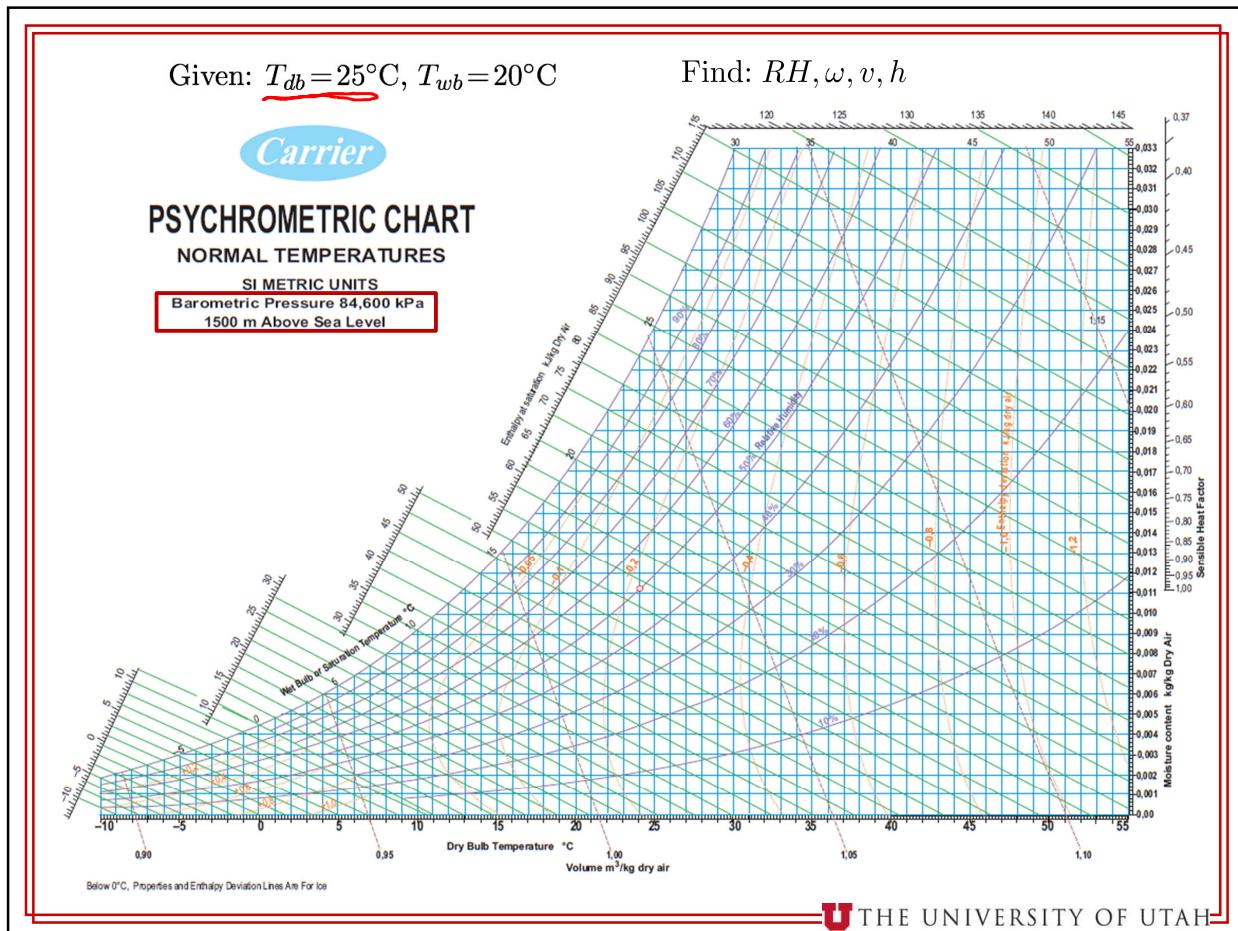


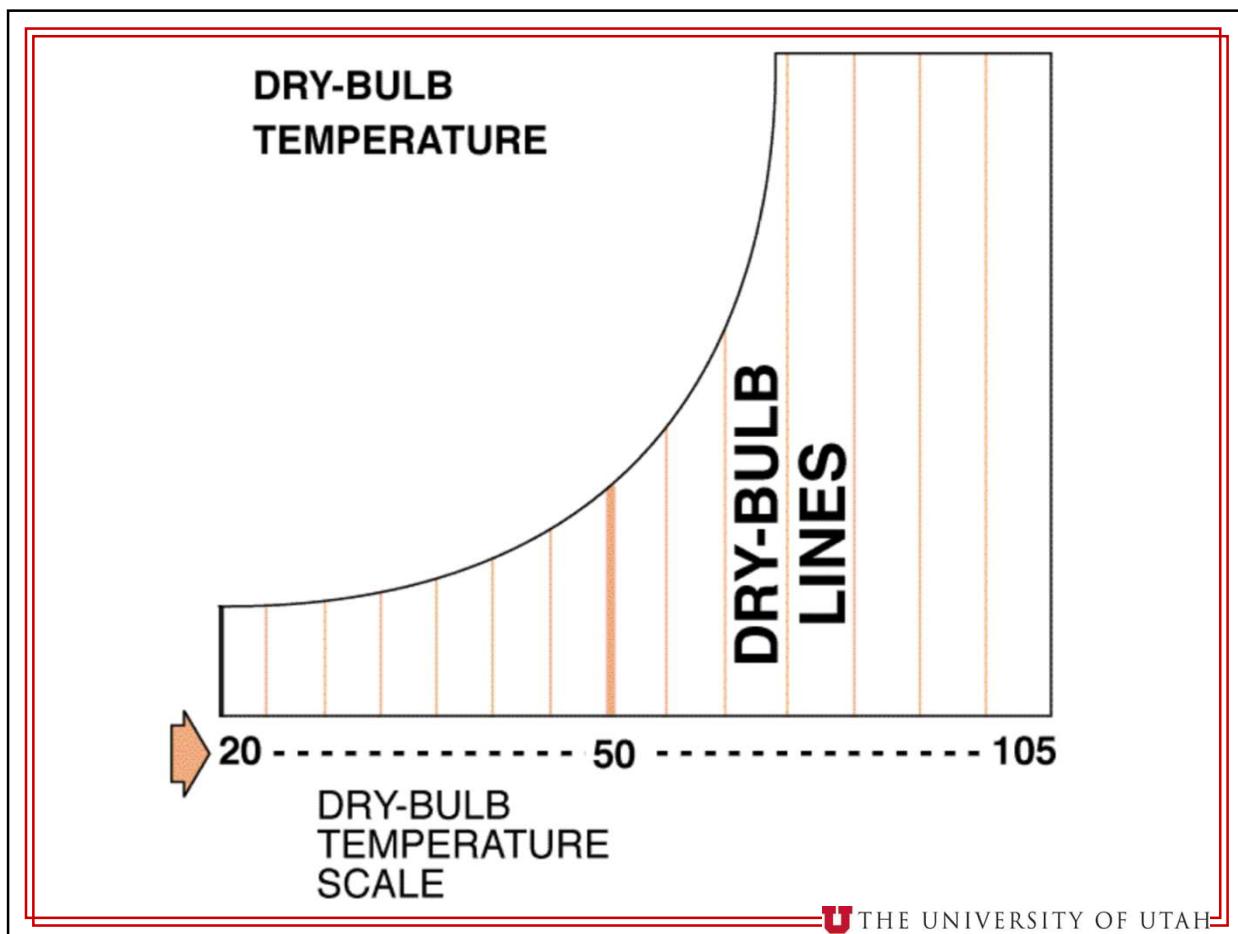
Quantities of Interest (Required Figures)

- T_w and T_{wb} vs z
 - η vs $\dot{m}_{w_{in}}$
 - ω vs z
 - T_{db} vs z
 - $\dot{m}_{w_{out}}/\dot{m}_{w_{in}}$ vs $T_{w_{in}}$
 - \dot{Q}_a and \dot{Q}_{amb} vs $T_{w_{in}}$
- $R = T_{w_{in}} - T_{w_{out}}$
 $A = T_{w_{out}} - T_{w_{in}}$
 $\eta = \frac{R}{R+A} \cdot 100$
 ← specific humidity
 "psychrometric chart"

Cooling Tower Efficiency

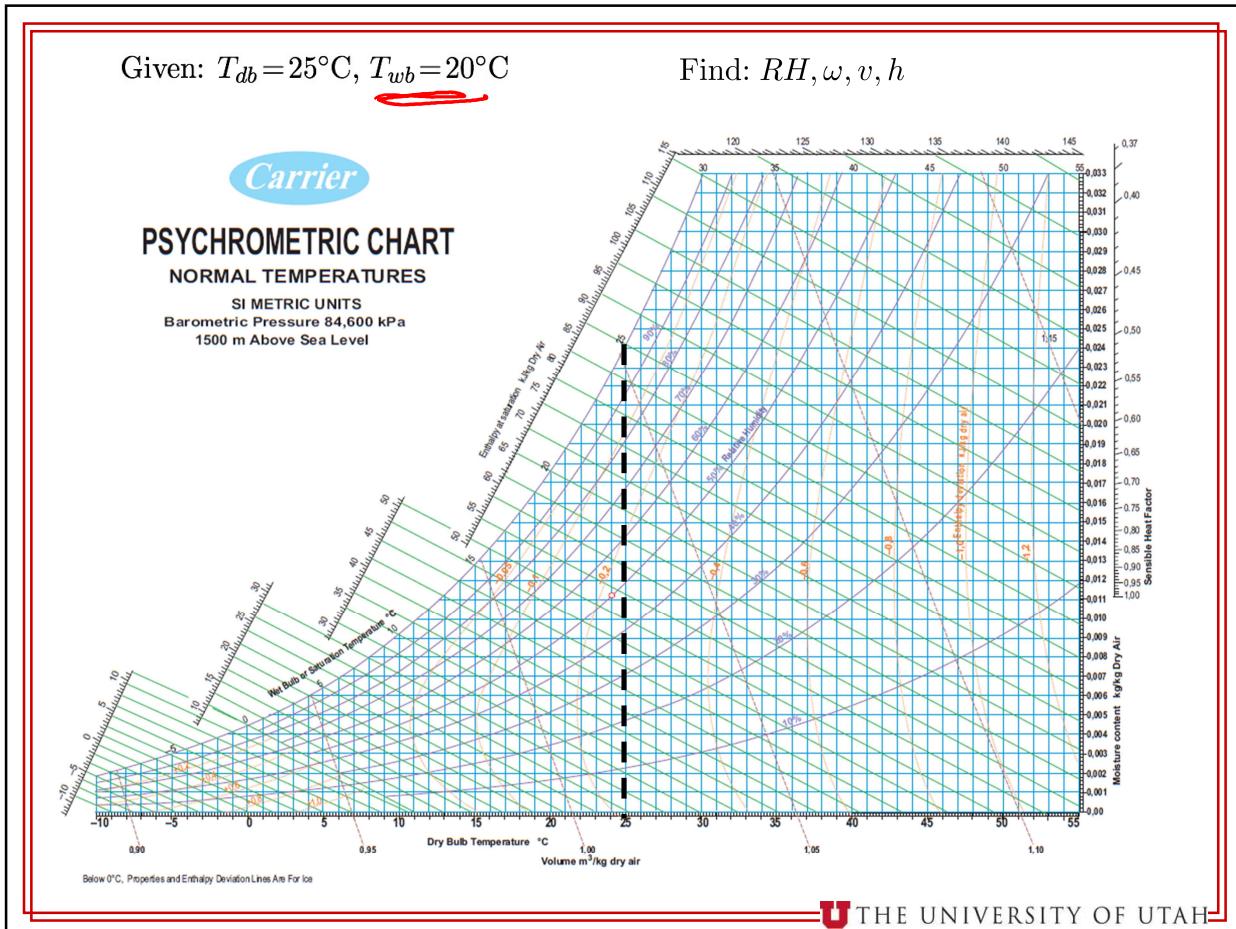


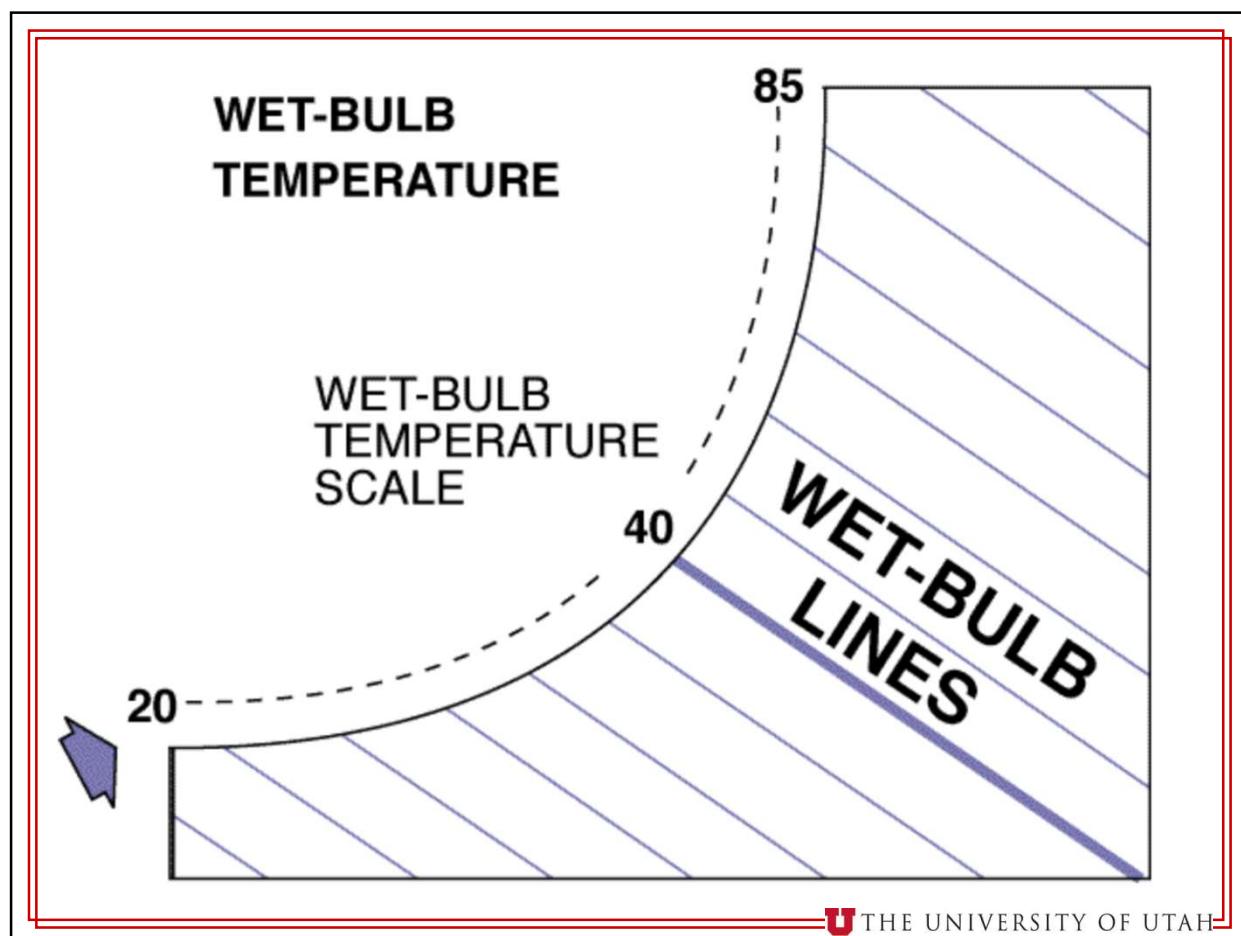


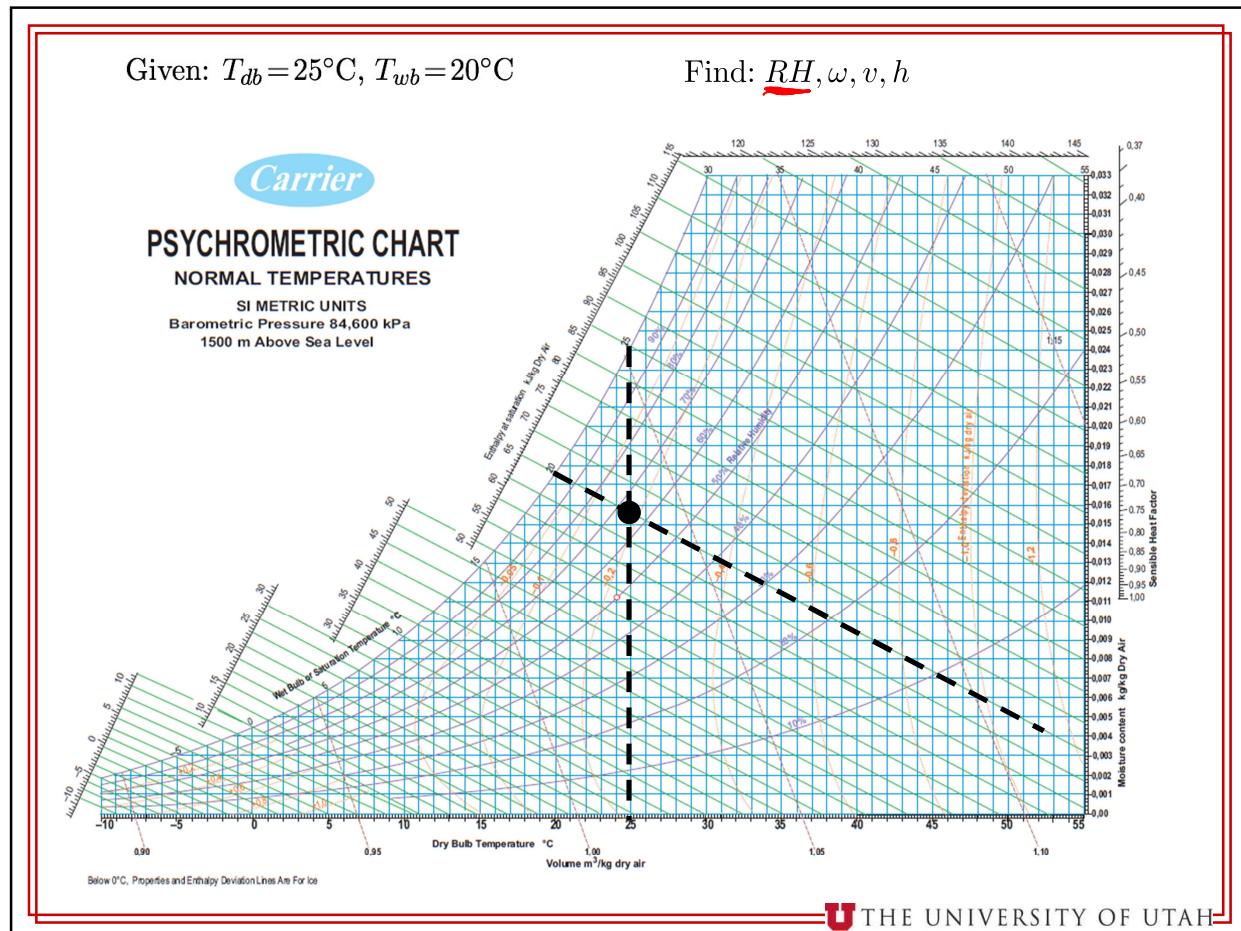


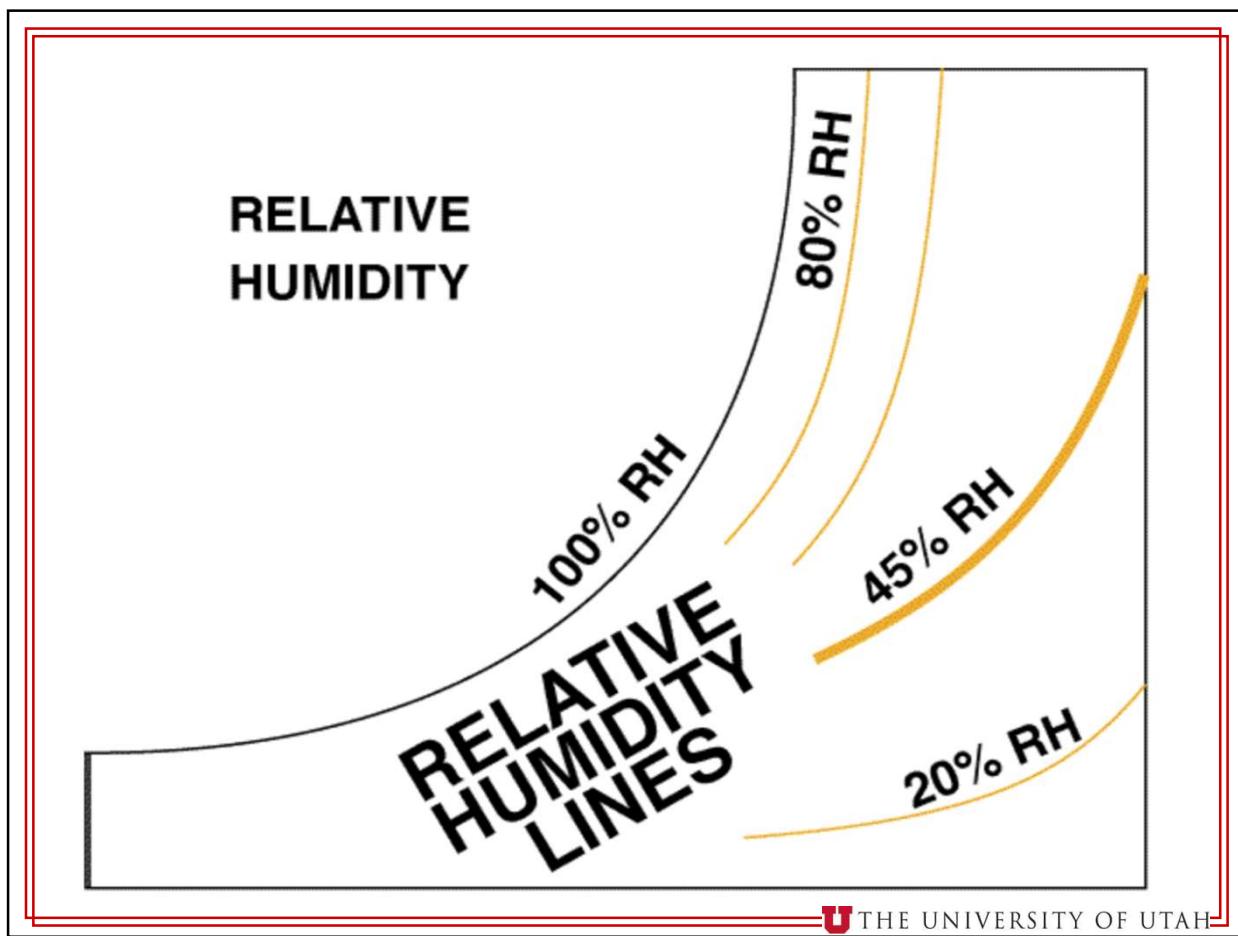
Given: $T_{db} = 25^\circ\text{C}$, $T_{wb} = 20^\circ\text{C}$

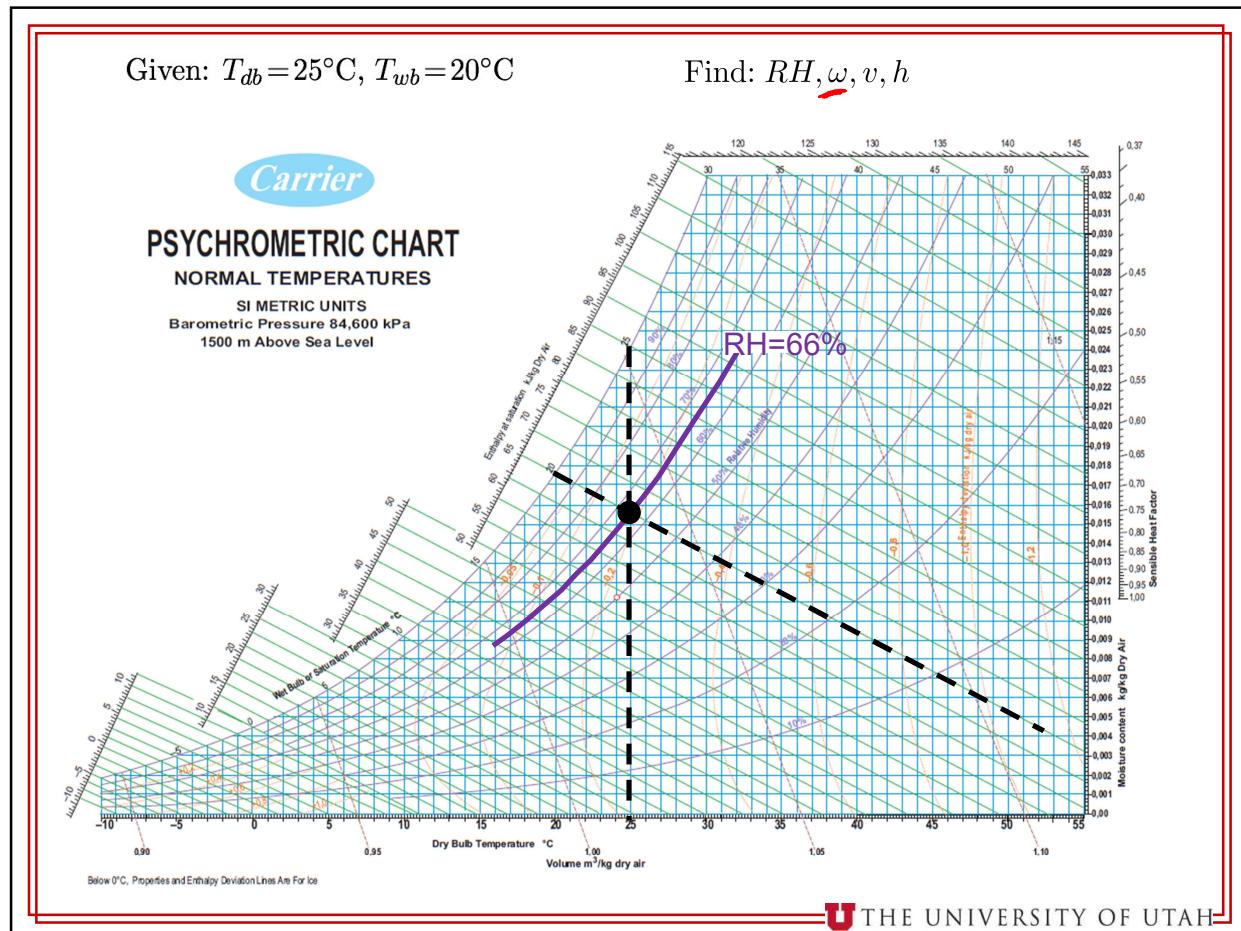
Find: RH, ω, v, h

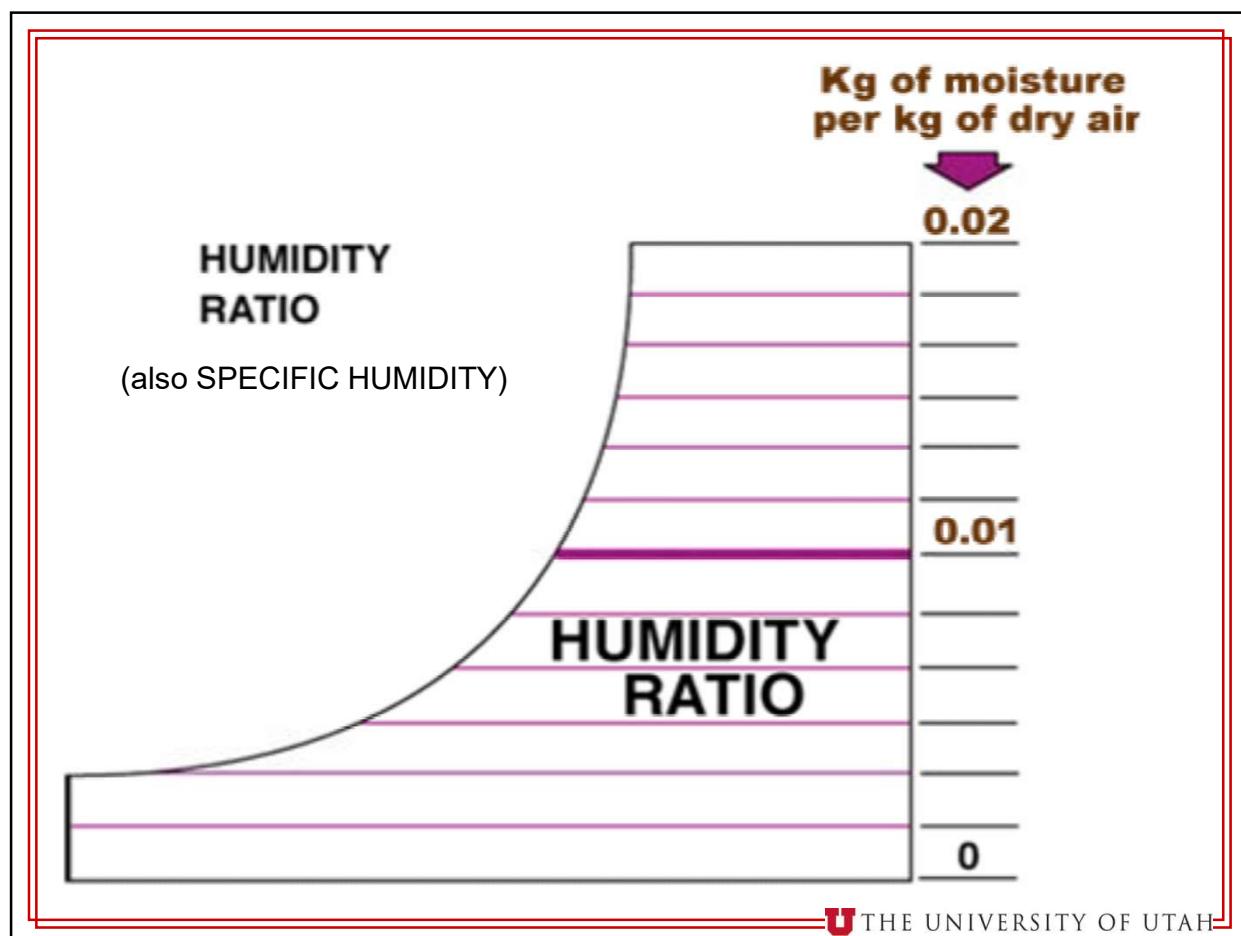


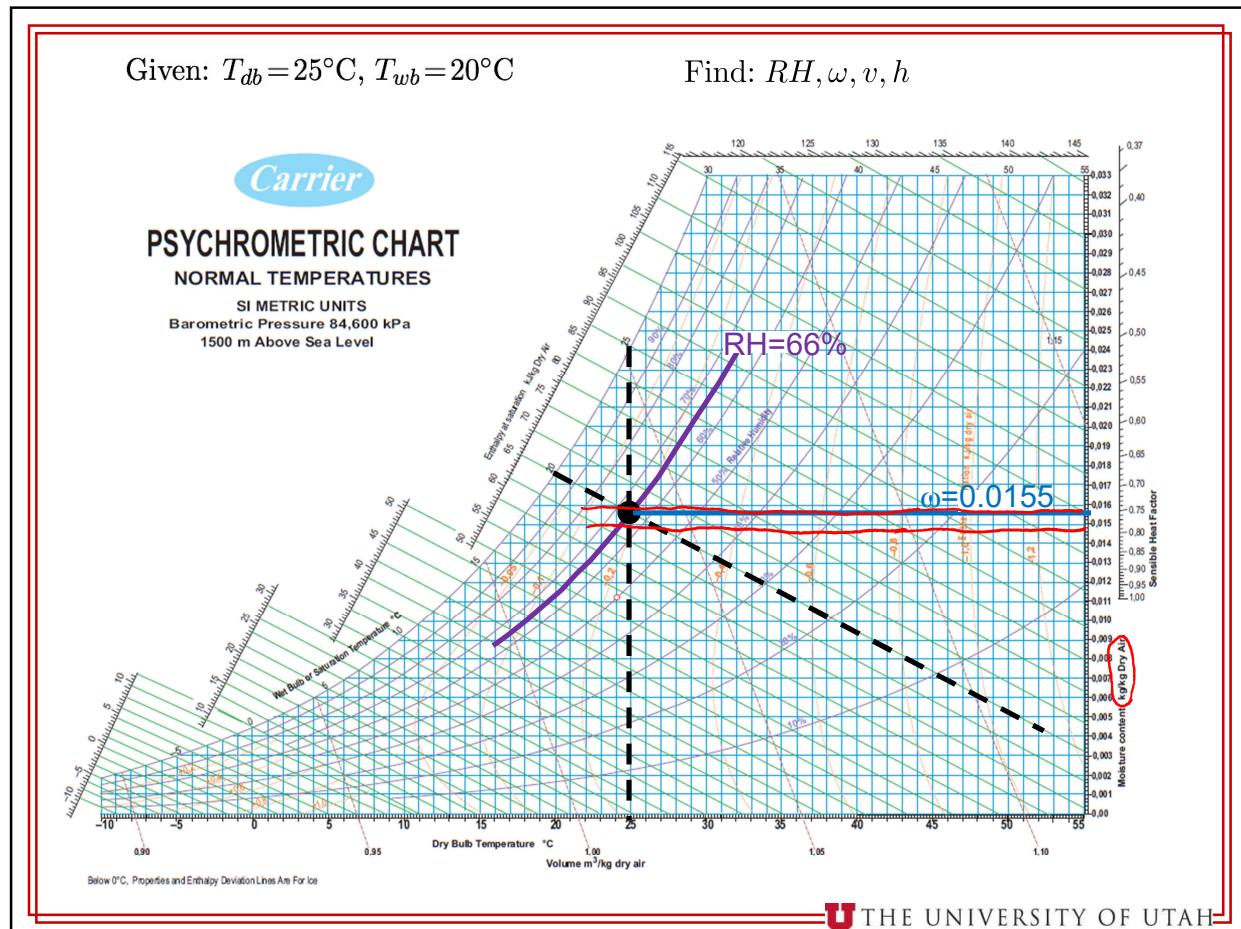


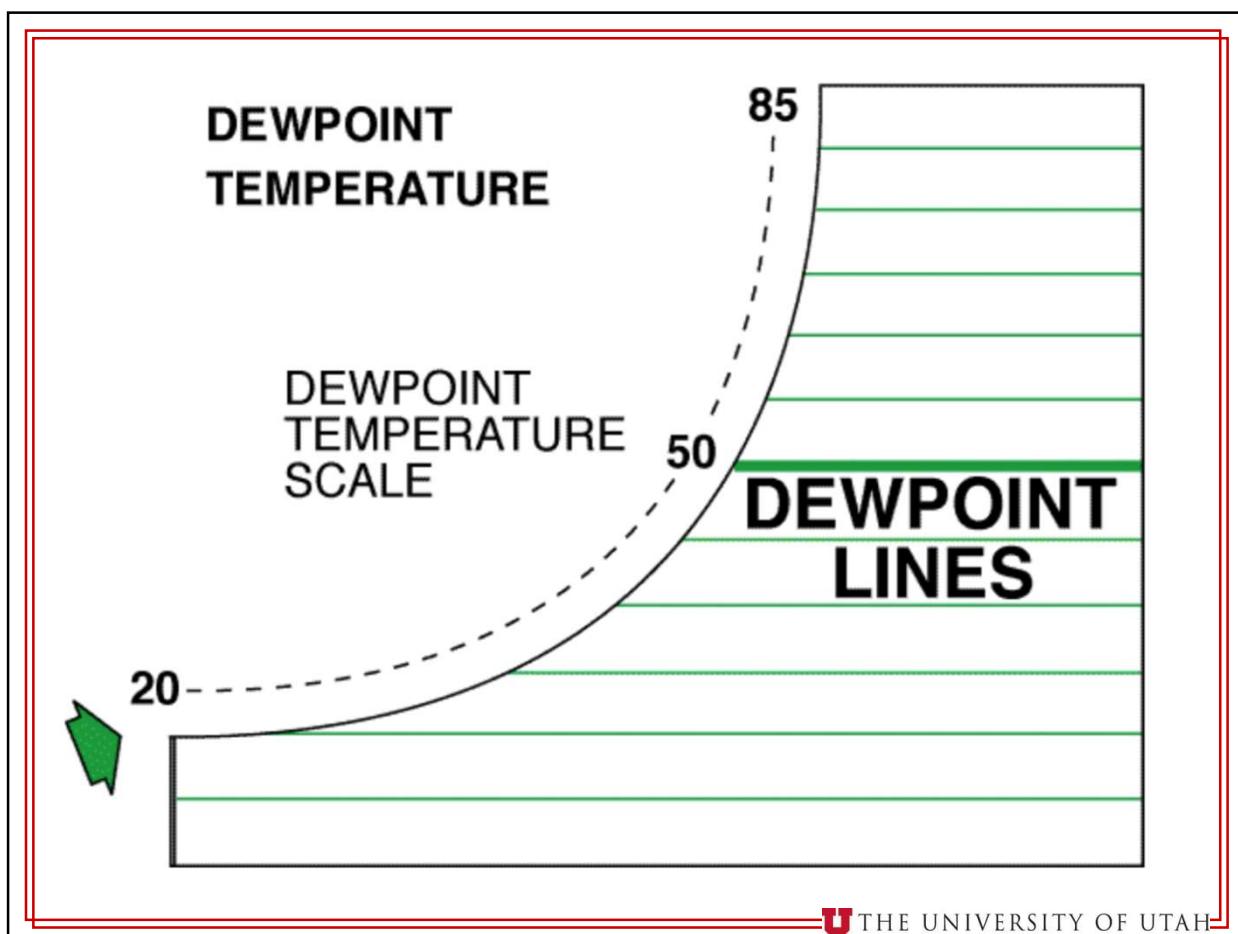


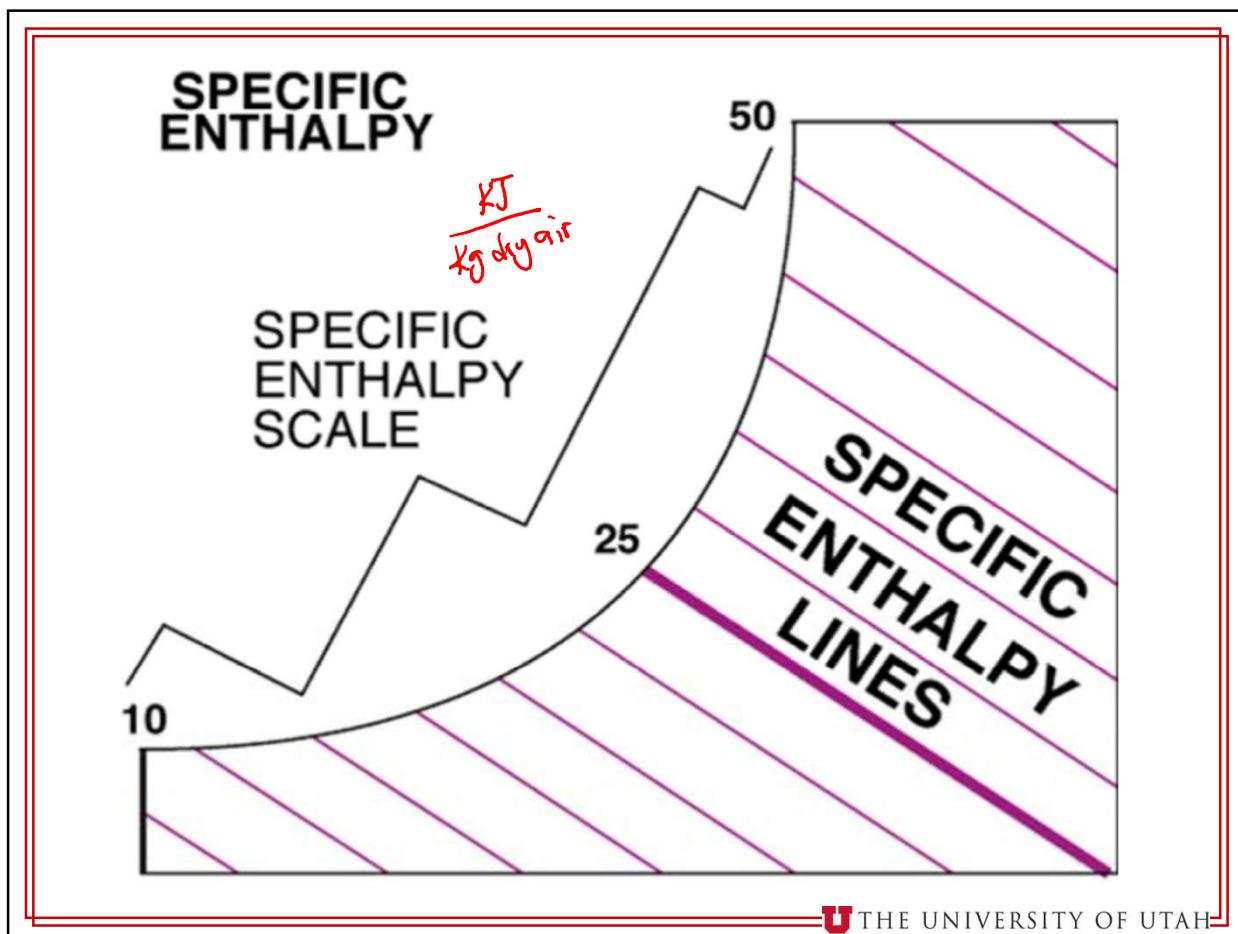


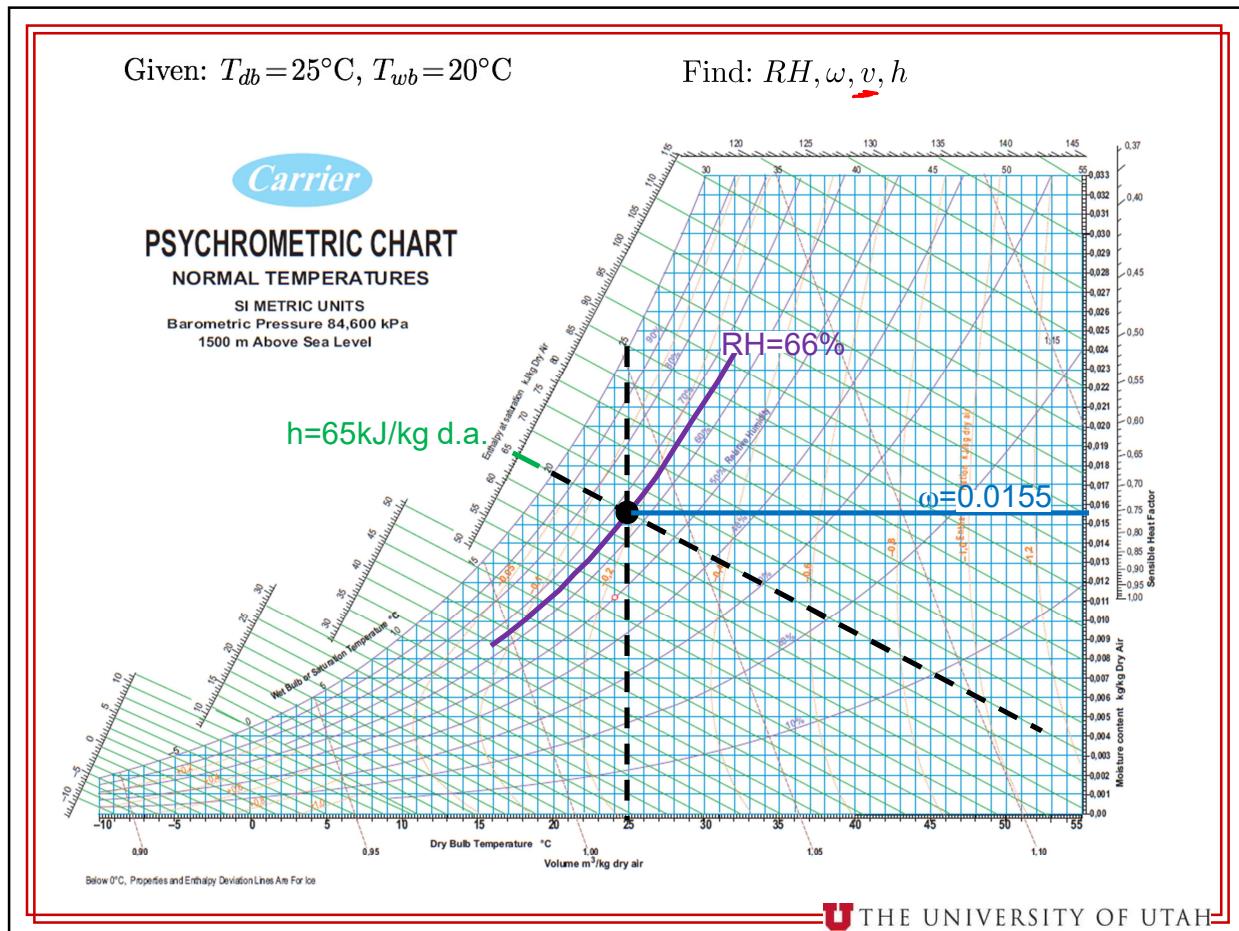


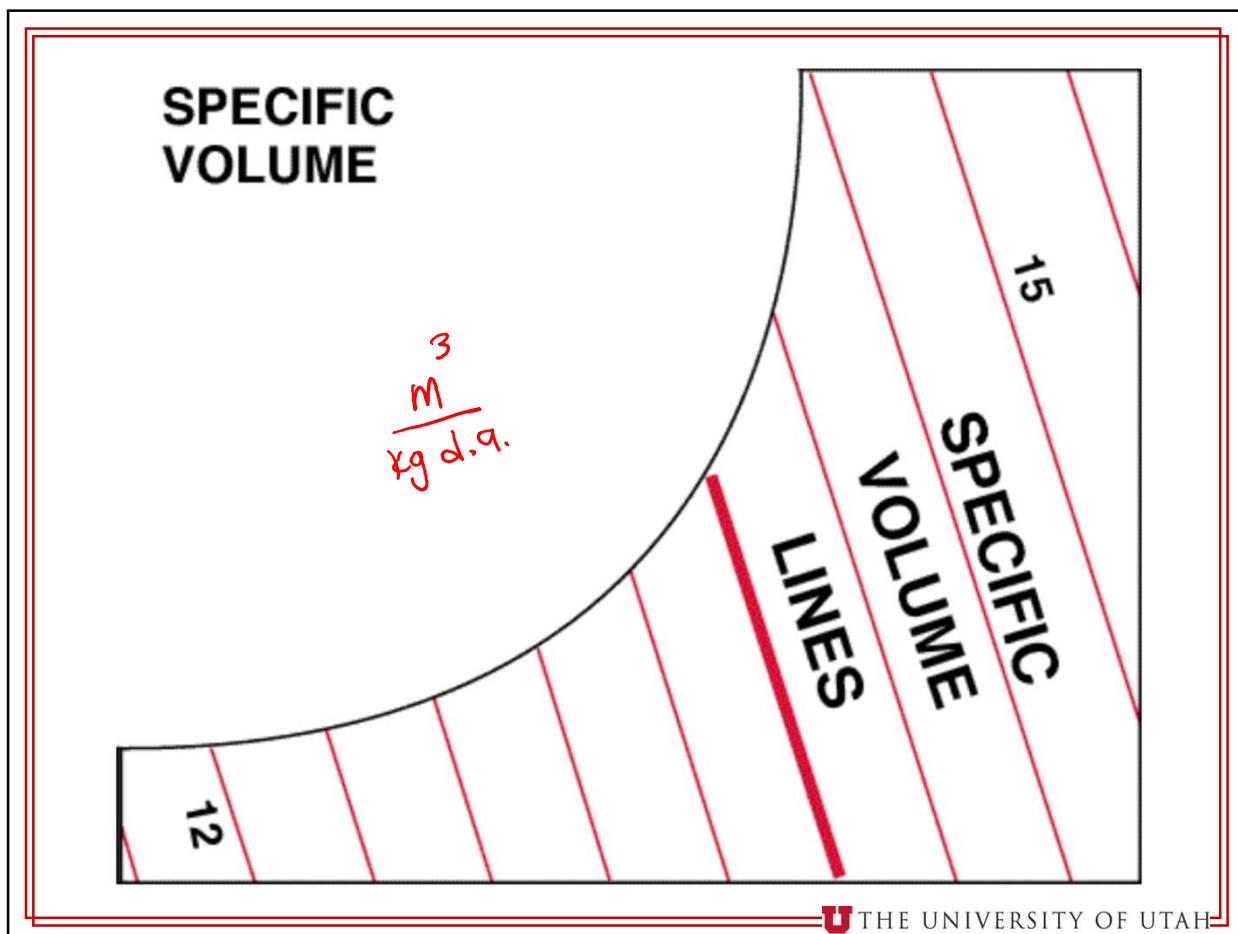


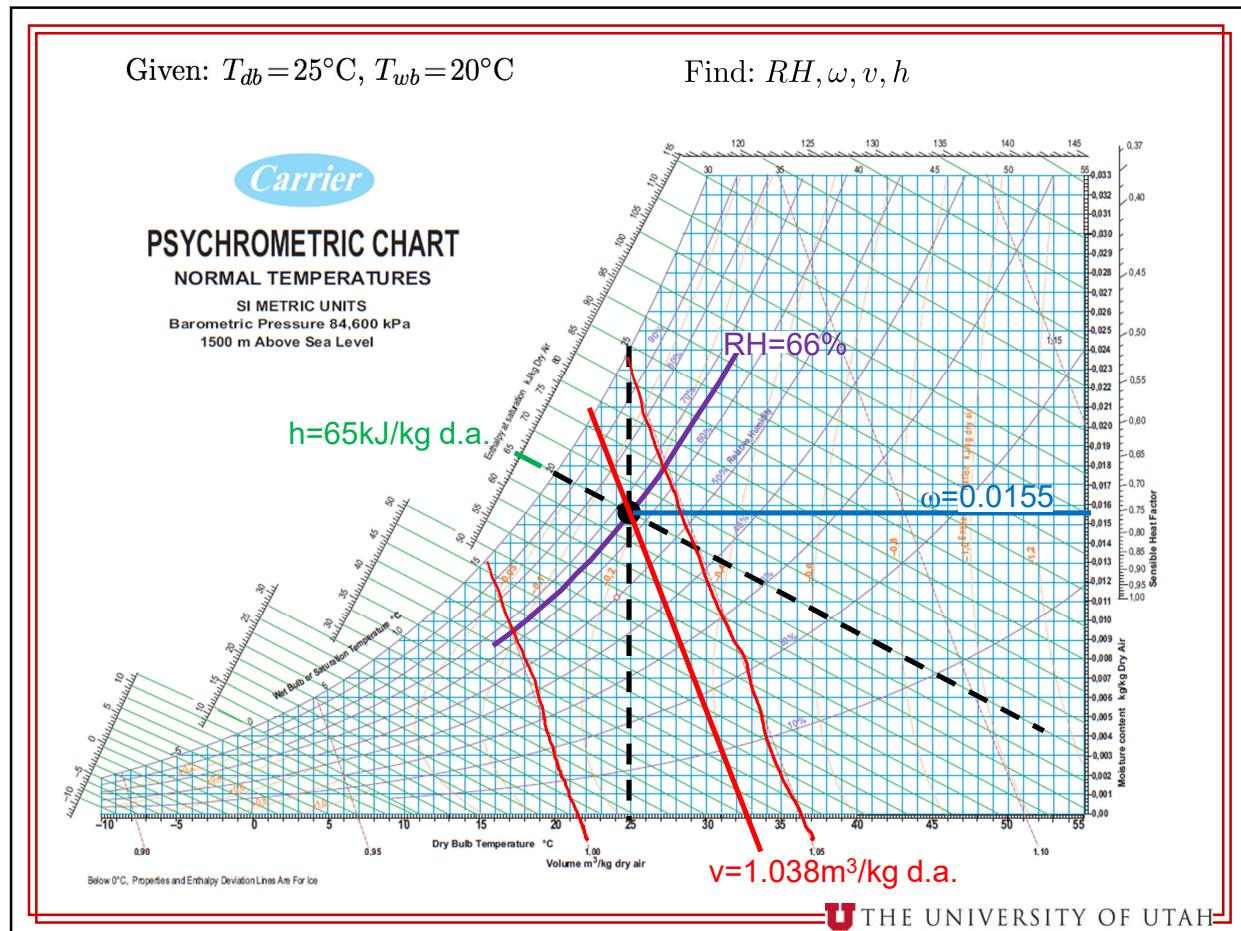












Matlab Function: “Psychrometrics”

```
function [Tdb, w, phi, h, Tdp, v, Twb] = Psychrometrics (varargin)
% function [Tdb, w, phi, h, Tdp, v, Twb] = Psychrometrics (varargin)
%
% ACCEPTED VARIABLES
% Tdb (dry bulb temperature) and Tdp(dew point temperature) in °C
% w (humidity ratio) in kg/kg of dry air
% phi (relative humidity) in %
% h (enthalpy) in J/kg of dry air
% v (specific volume) in m3/kg of dry air
% Twb (wet bulb temperature) in °C
% P (atmospheric pressure) in kPa
```

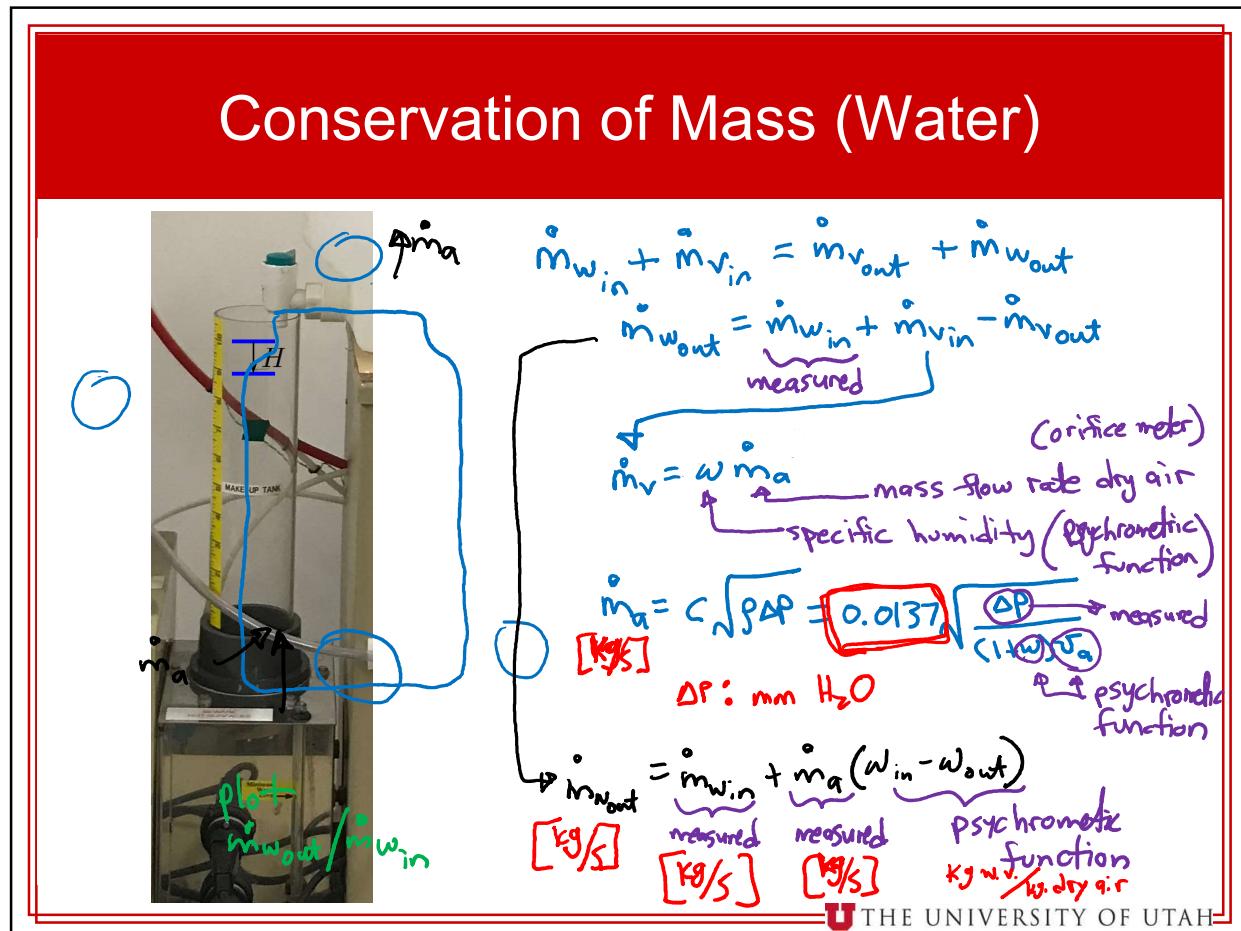
→ [Tdb, w, phi, h, Tdp, v, Twb] = Psychrometrics ('tdb', 25, 'twb', 20, 'p', 86);

°C °C kPa
 ↓ ↓ ↓
 dry wet atm.
 bulb bulb pressure

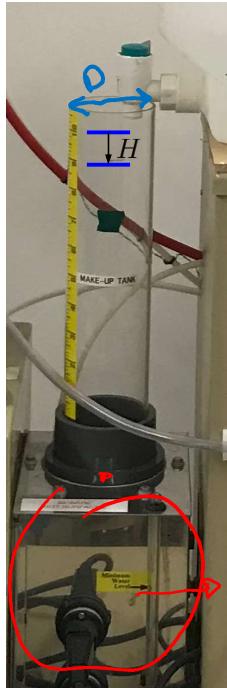
For - Loop

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Conservation of Mass (Water)



Conservation of Mass (Water)



$$\dot{m}_{\text{makeup}} = \dot{m}_{\text{win}} - \dot{m}_{\text{wout}}$$

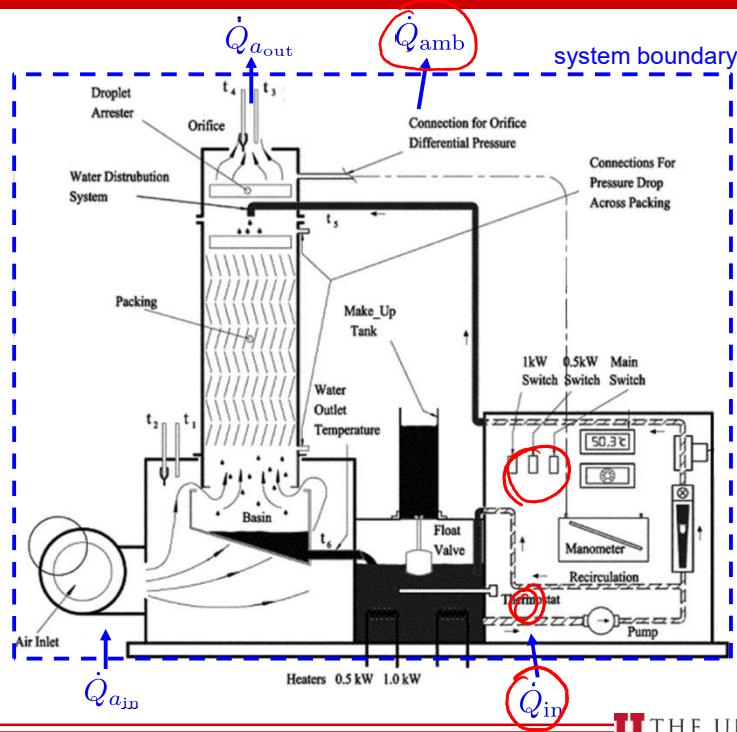
$$= \dot{m}_a (\omega_{\text{out}} - \omega_{\text{in}})$$

In lab,

$$\rightarrow \dot{m}_{\text{makeup}} \approx \frac{\pi/4 D^2 H \rho_w}{t}$$

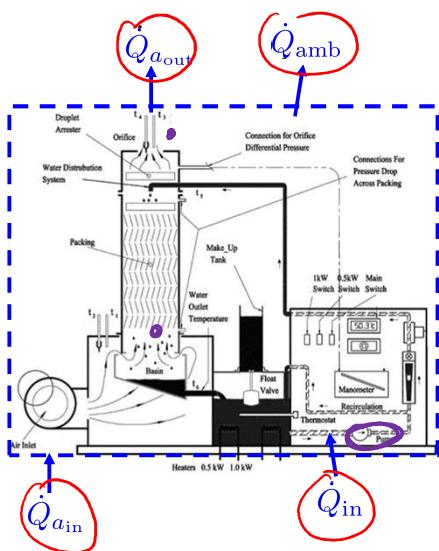
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Conservation of Energy (System)

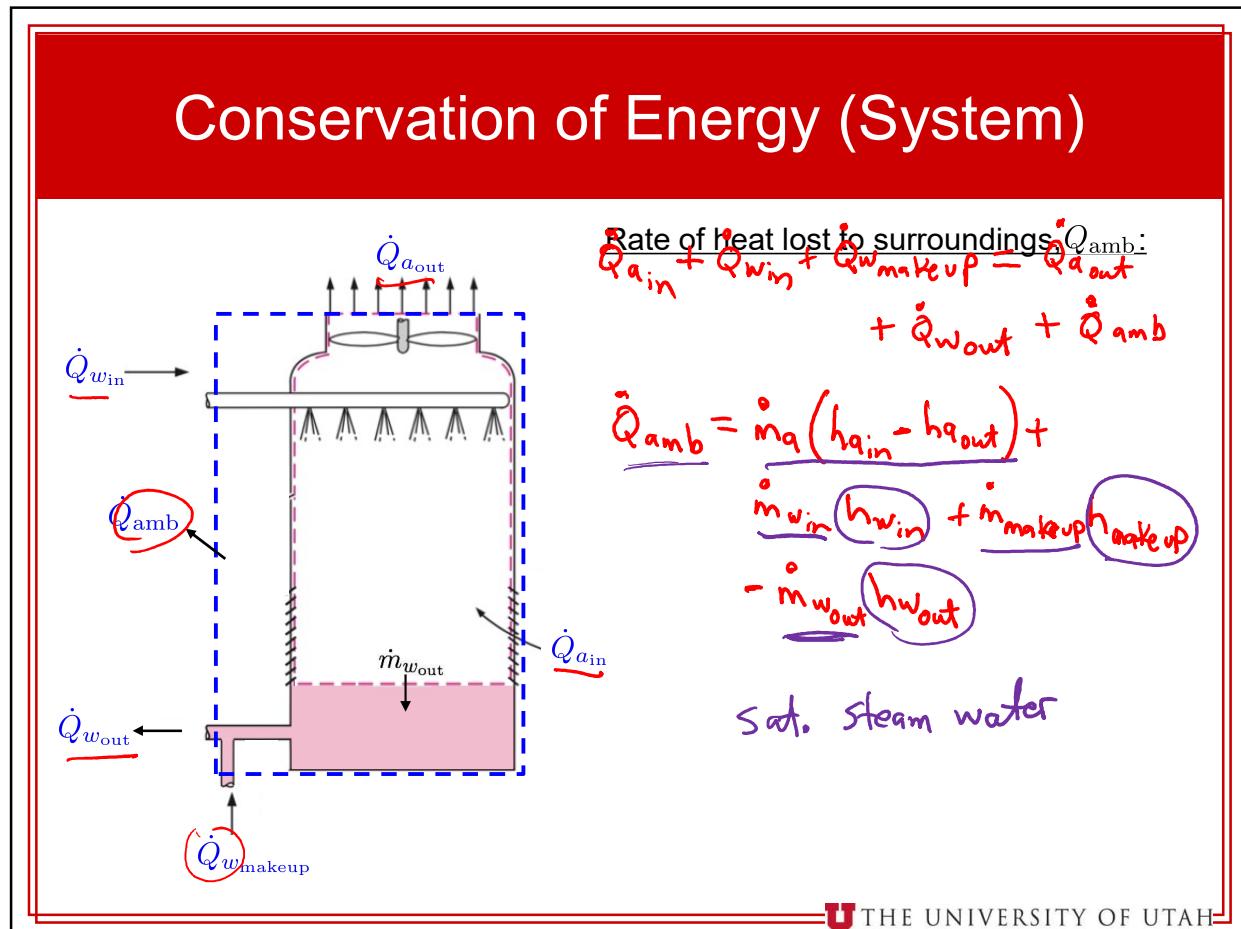


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Conservation of Energy (System)



Conservation of Energy (System)



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Questions??

Thank you for your attention!

Let me or the TAs know if you have questions

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