

Spark Ignition Engine

Thermal, Fluids, and Energy
Systems Lab

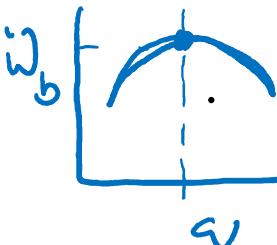
(ME EN 4650)

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University of Utah*

Spark Ignition Engine Performance

What we want to know:

- mechanical power (\dot{W}_b) vs crankshaft speed (ω)
- efficiency (η) vs. ω
- compare with ideal engine (follows Otto cycle)
- Mean effective pressure vs. ω



What we can measure:

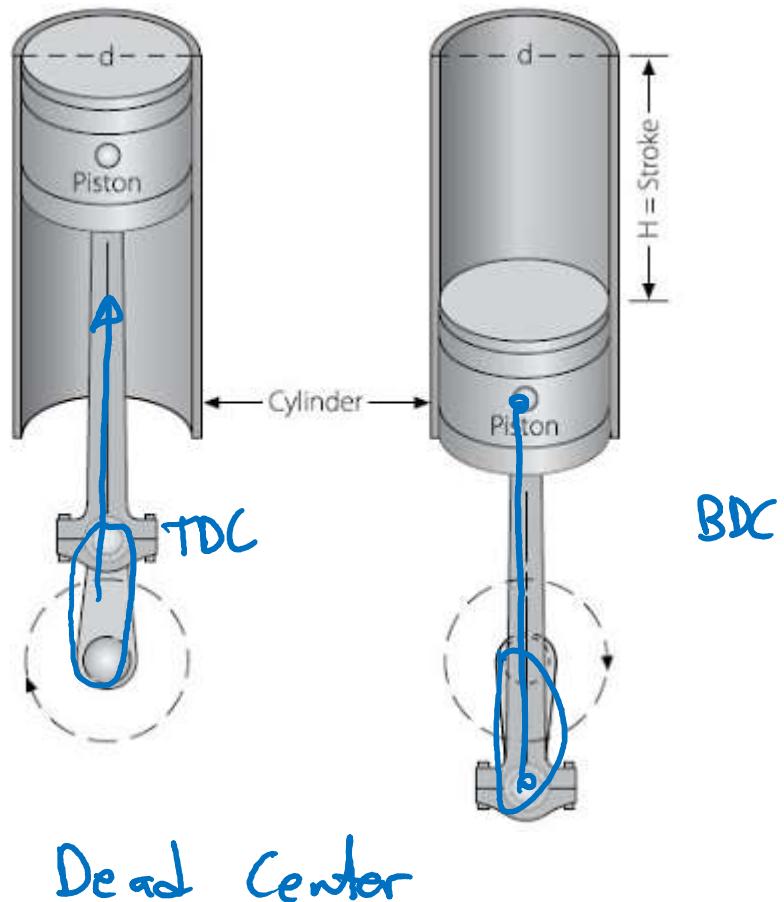
- crankshaft speed (optical encoder)
- engine torque (dynamometer)
- exhaust, air, fuel temp (thermocouples)
- inlet air flow rate (orifice meter)
- fuel flow rate (fill tube)

Bore, B	65.1 mm
Stroke, S	44.4 mm
Displacement Volume, V_d	0.148 L
Compression Ratio, r	7:1
Ignition Timing	20° before TDC
Fuel Type	91 octane gasoline

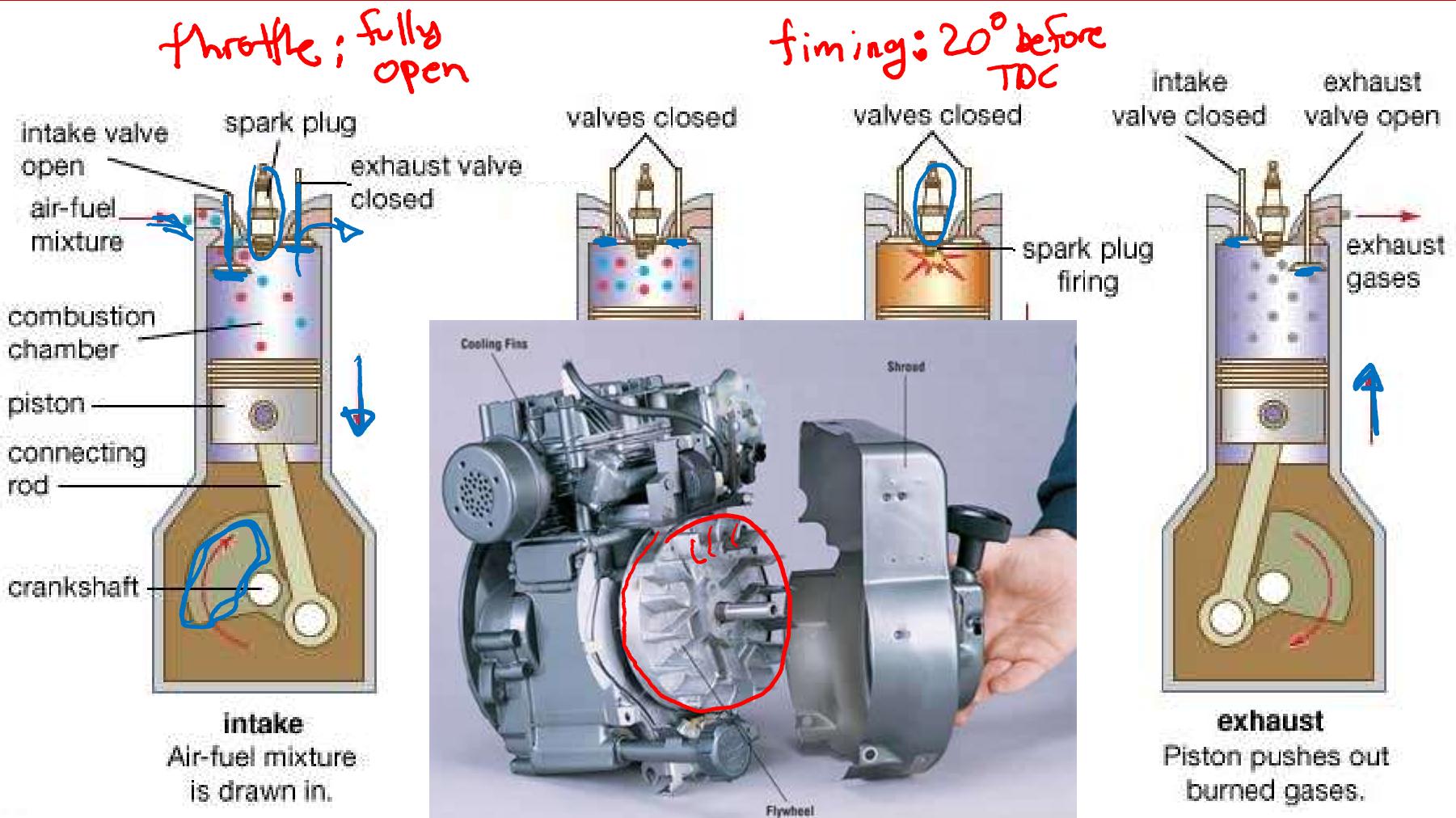
Lecture Outline

- Background
- Experimental Setup
- Measurements and Instrumentation
- Required Figures
- Data Analysis

Engine Size: Compression Ratio



Four-Stroke Engine: Physical Cycle



Measurements



Quantity	Symbol	Units	Instrument
Sampling interval	Δt	sec	stopwatch
Inlet air temperature	T_{air}	°C	thermocouple
Inlet fuel temperature	T_{fuel}	°C	thermocouple
Exhaust gas temperature	$T_{exhaust}$	°C	thermocouple
Inlet air volume flow rate	\dot{V}_{air}	L/min	orifice meter
Volume of fuel consumed	V_{fuel}	cm³	fill tube level
Rotational speed of crankshaft	ω	RPM	optical encoder
Torque of crankshaft	τ	N·m	dynamometer

Experiments



Quantity	Symbol	Units	Instrument
Sampling interval	Δt	sec	stopwatch
Inlet air temperature	T_{air}	°C	thermocouple
Inlet fuel temperature	T_{fuel}	°C	thermocouple
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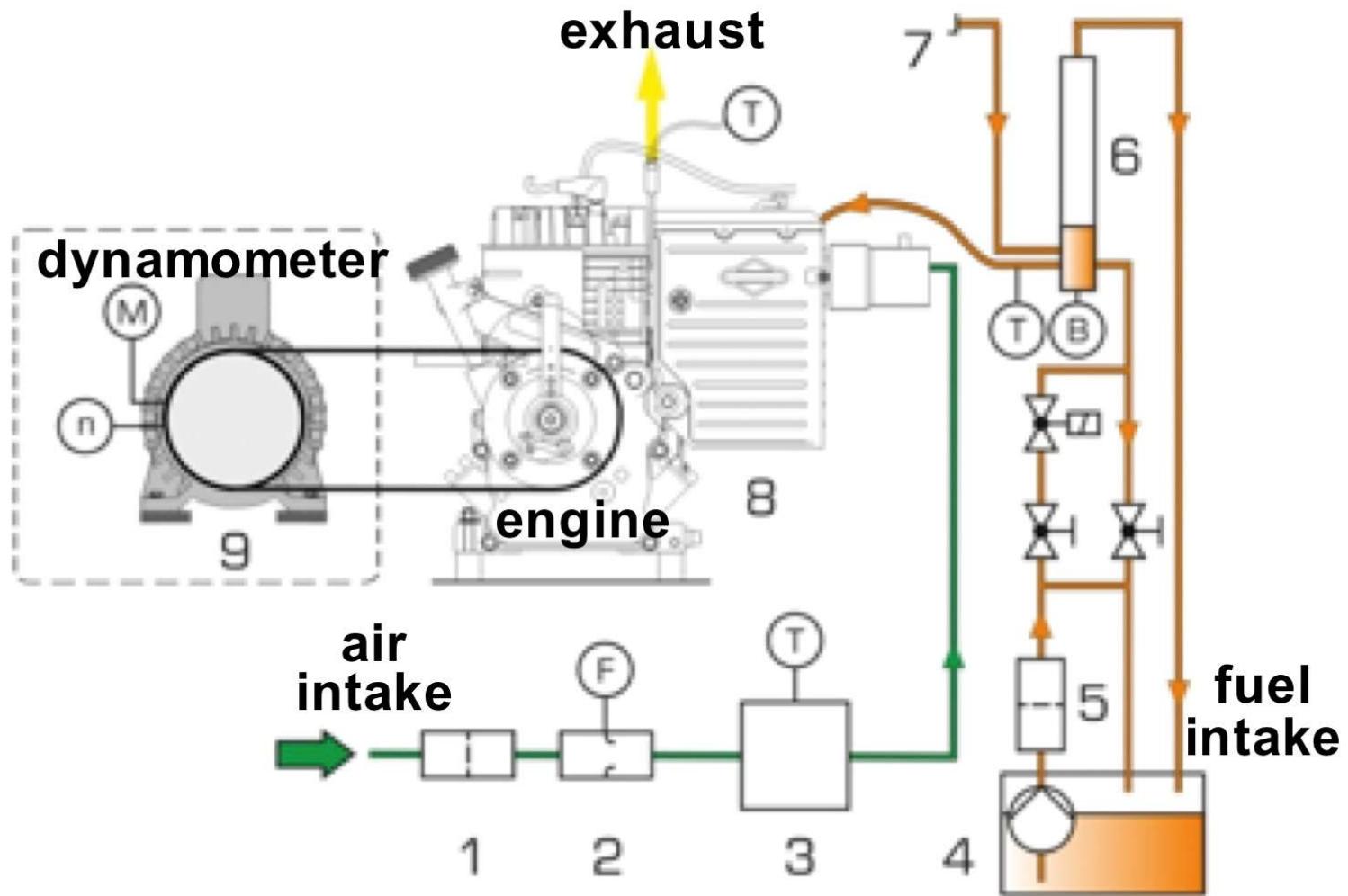
Exp #1 (Performance)

- Throttle fully open
- 5 loads
- 5 crankshaft speed

Exp #2 (Friction)

- dry run (no combustion)
- measure T & ω

Experimental Setup



Experimental Apparatus

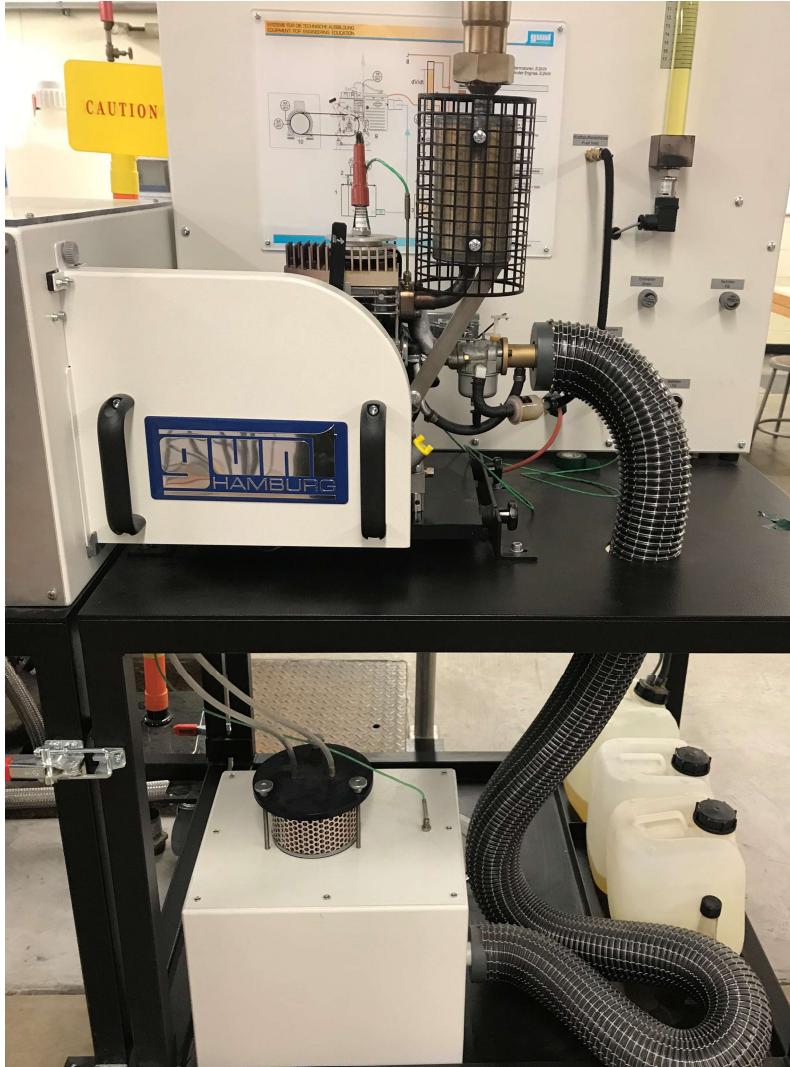


Dynamometer



Single-cylinder engine

Experimental Apparatus



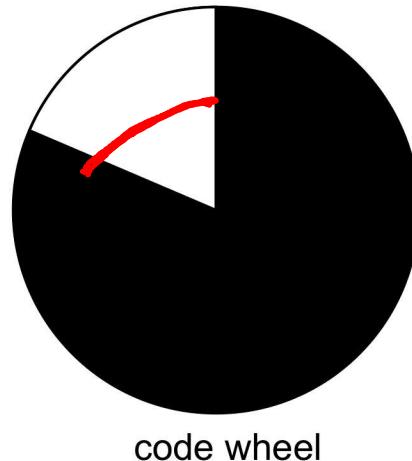
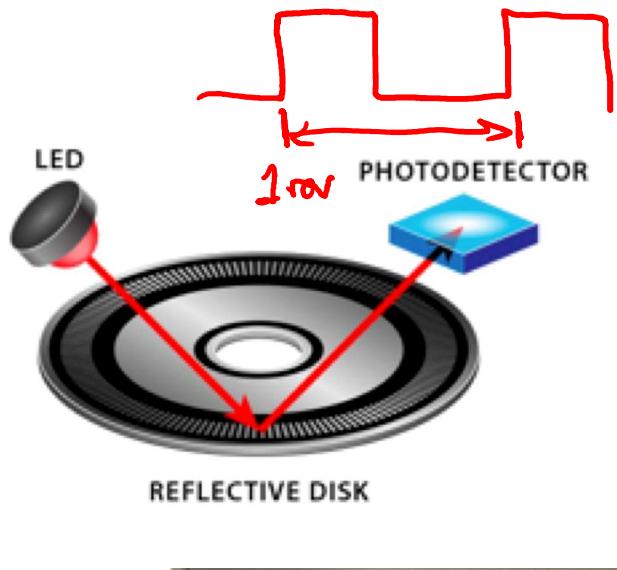
Fuel pump



Dynamometer

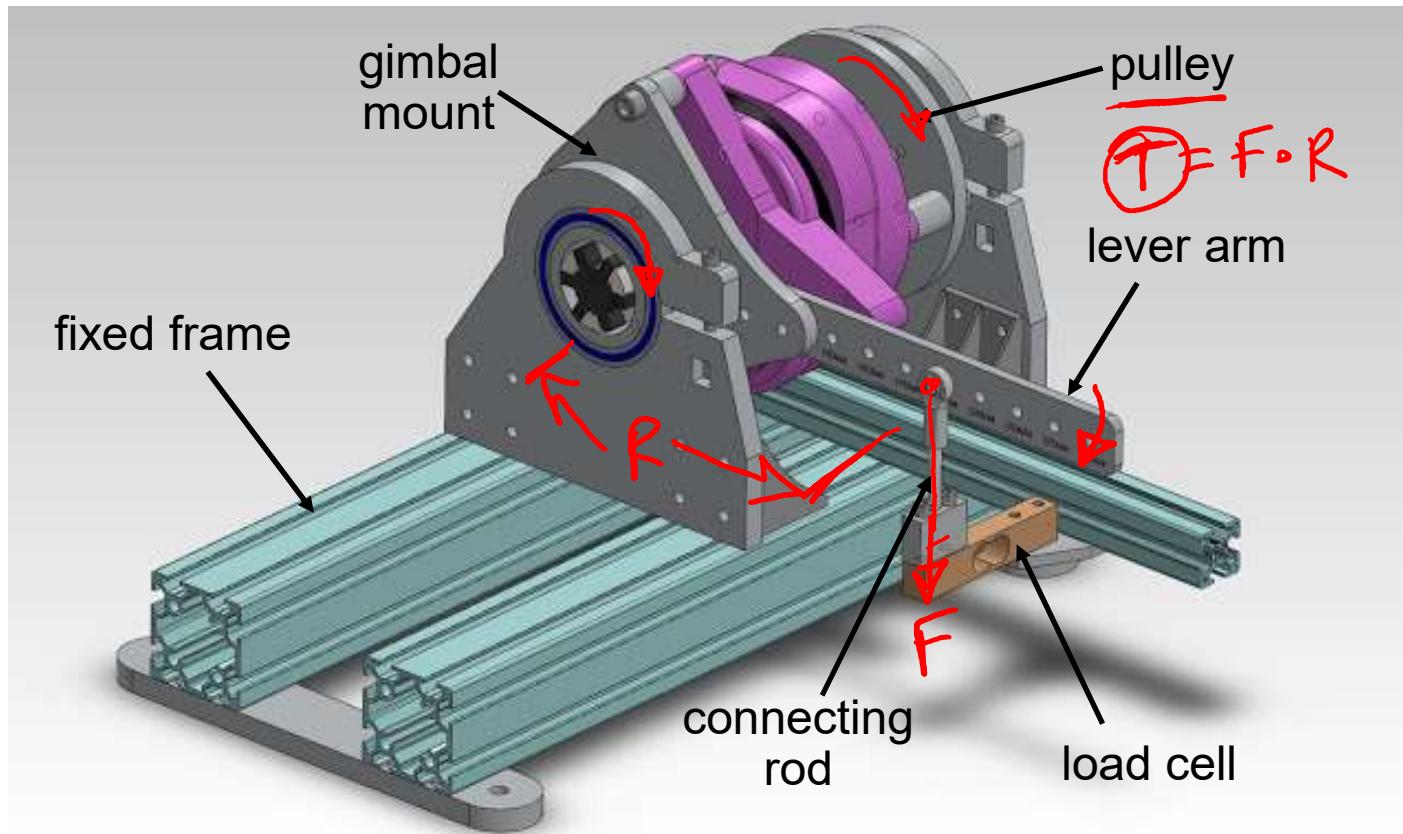


Crankshaft Speed: Optical Encoder

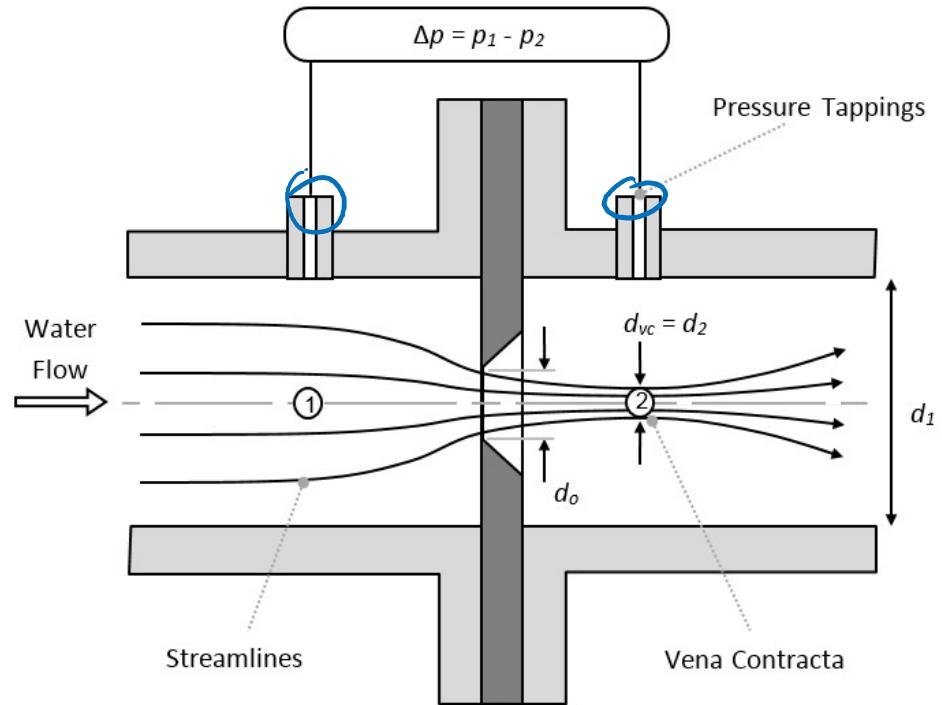
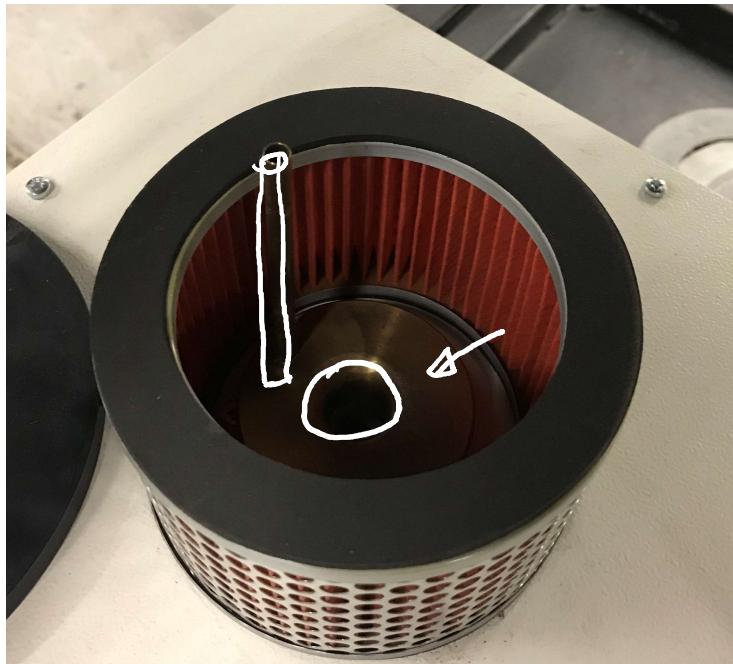


Brake Dynamometer: Measure Torque

Gimbaled Motor Mount with Lever Arm and Load Cell



Air Flow Rate: Orifice Meter



$$\dot{m}_a = \rho_a \dot{V}_a$$

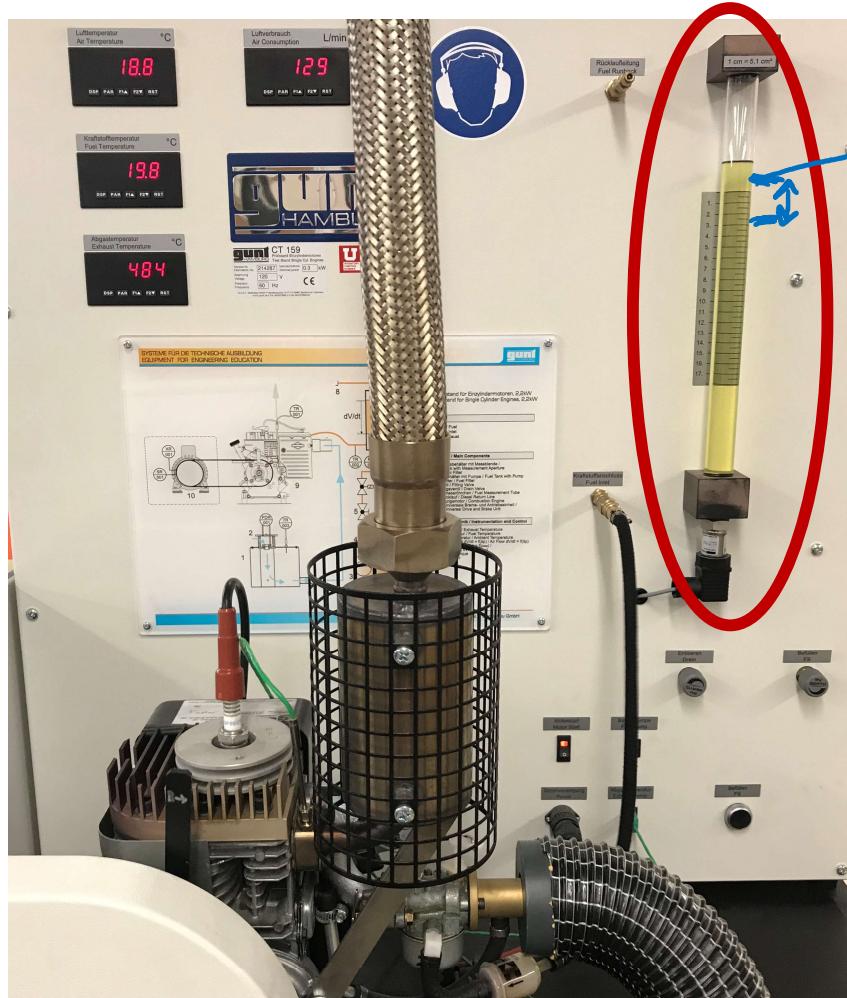
$$\rho_a = \frac{P}{RT} \quad (\text{Ideal gas law})$$

R = 287.1 J/kg·K

$$\dot{V}_a = C \sqrt{\Delta P}$$

discharge coefficient
(l/min)

Fuel Flow Rate: Fill-Tube



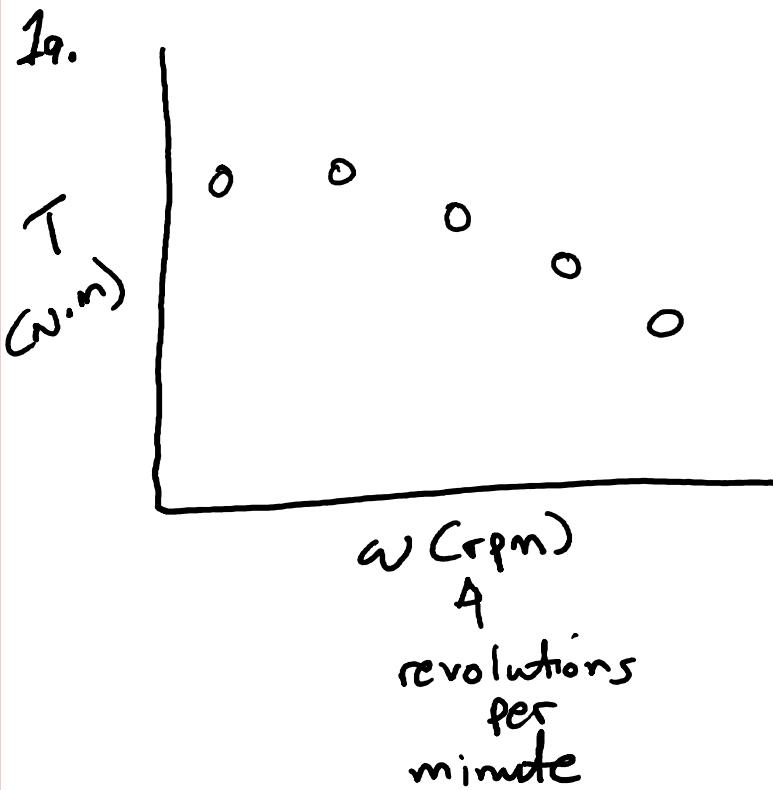
$$1\text{cm} = 5.1\text{cm}^3$$

$$\dot{V}_f = \frac{V_f}{t}$$

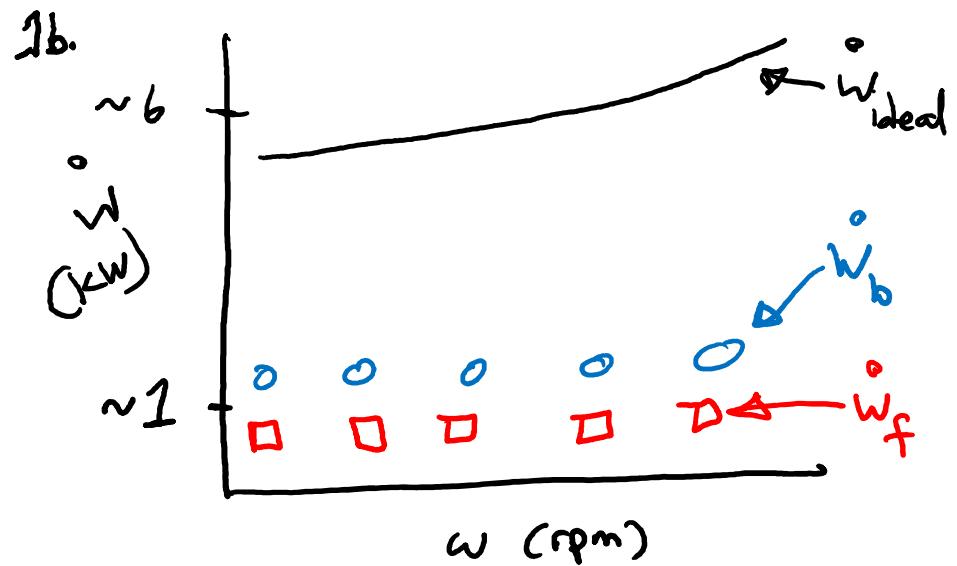
$$\dot{m}_f = \rho_f \dot{V}_f$$

q1-octane gas
 $\sim 726 \text{ kg/m}^3$

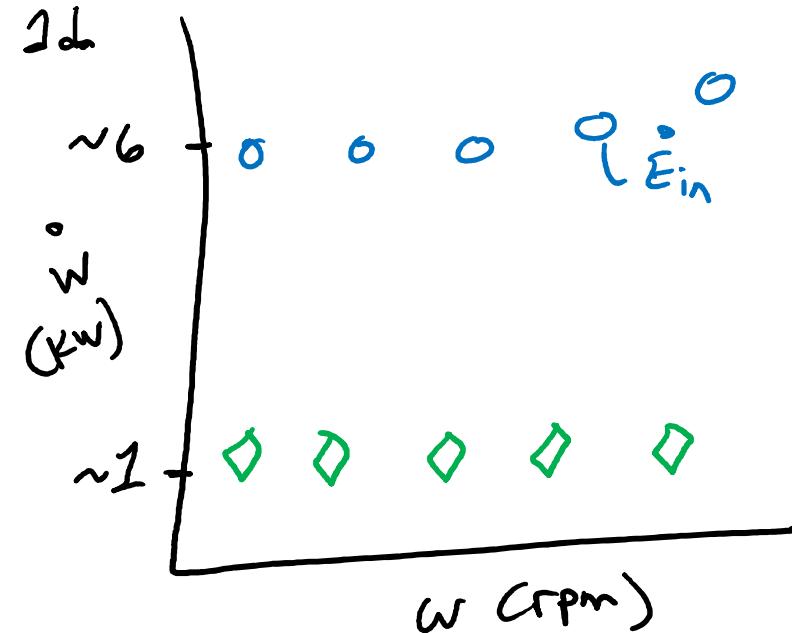
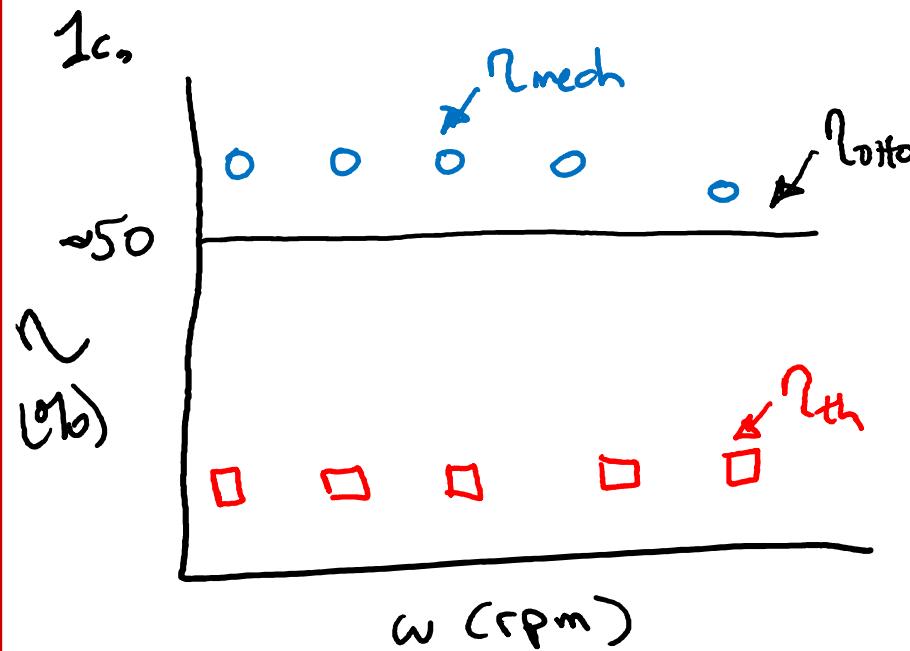
Figures 1a & 1b



torque-speed curve

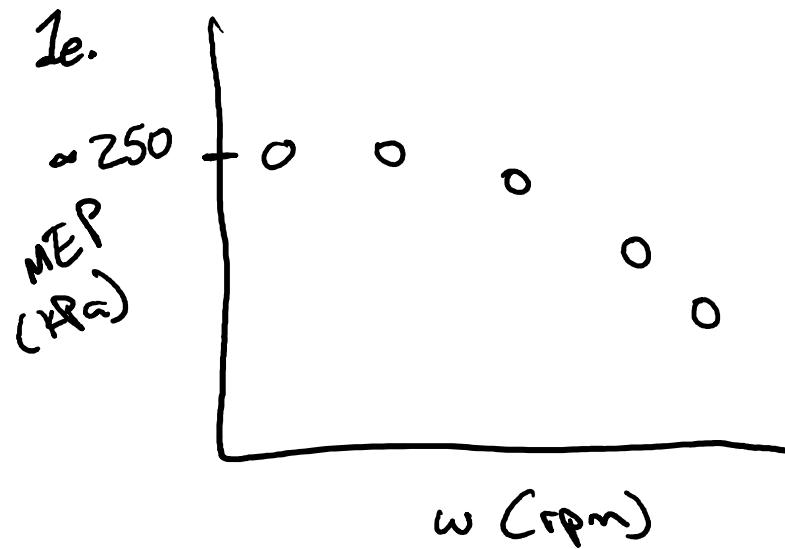


Figures 1c & 1d



\dot{E}_{in}
 \dot{w}_b
 \dot{w}_f
 \dot{Q}

Figures 1e



Spark Ignition Engine: Part II

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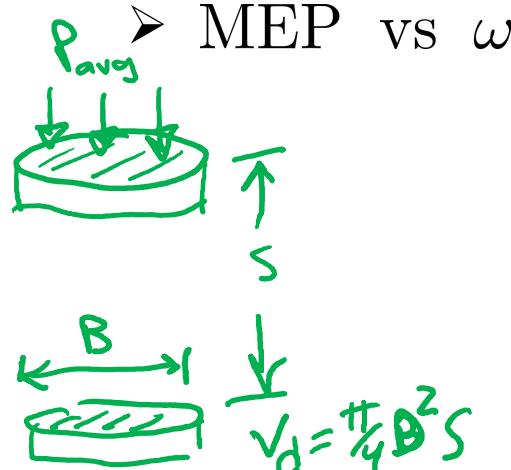
Quantities of Interest (Required Figures)

- τ vs ω (rps)
measured \rightarrow measured (optical encoder)
- $\dot{W}_b, \dot{W}_f, \dot{W}_{\text{net}}$ vs ω
- $\eta_{\text{th}}, \eta_{\text{mech}}, \eta_{\text{Otto}}$ vs ω
- $\dot{E}_{\text{in}}, \dot{W}_b, \dot{Q}, \dot{W}_f$ vs ω
- MEP vs ω

$$\dot{W}_b = \tau \omega \quad [\text{rad/s}] \quad \dot{W}_f = \tau_f \omega \quad [\text{dry run}]$$

$$\eta_{\text{mech}} = \frac{\dot{W}_b}{\dot{W}_b + \dot{W}_f} \cdot 100 \quad [\%]$$

$$\begin{aligned} \dot{E}_{\text{in}} &= \dot{W}_b + \dot{W}_f + \dot{Q} \\ Q &= \dot{E}_{\text{in}} - (\dot{W}_b + \dot{W}_f) \end{aligned}$$



$$W = (\text{MEP}) V_d \quad 0.148 \text{ L (spec)} \quad \text{circled}$$

$$\text{MEP} = W/V_d = \dot{W}_b t_c / V_d$$

$$\text{MEP} = [\tau \cdot \omega] \left[\frac{1}{4} \pi (2\pi)(z) \right] / V_d$$

$$\boxed{\text{MEP} = \frac{4\pi \tau \omega}{V_d}}$$

Thermal Efficiency

$$\eta_{th} = \dot{w}_b / \dot{E}_{in} \rightarrow \text{rate of energy into system via fuel}$$

$$\dot{E}_{in} = \dot{m}_f (LHV)$$

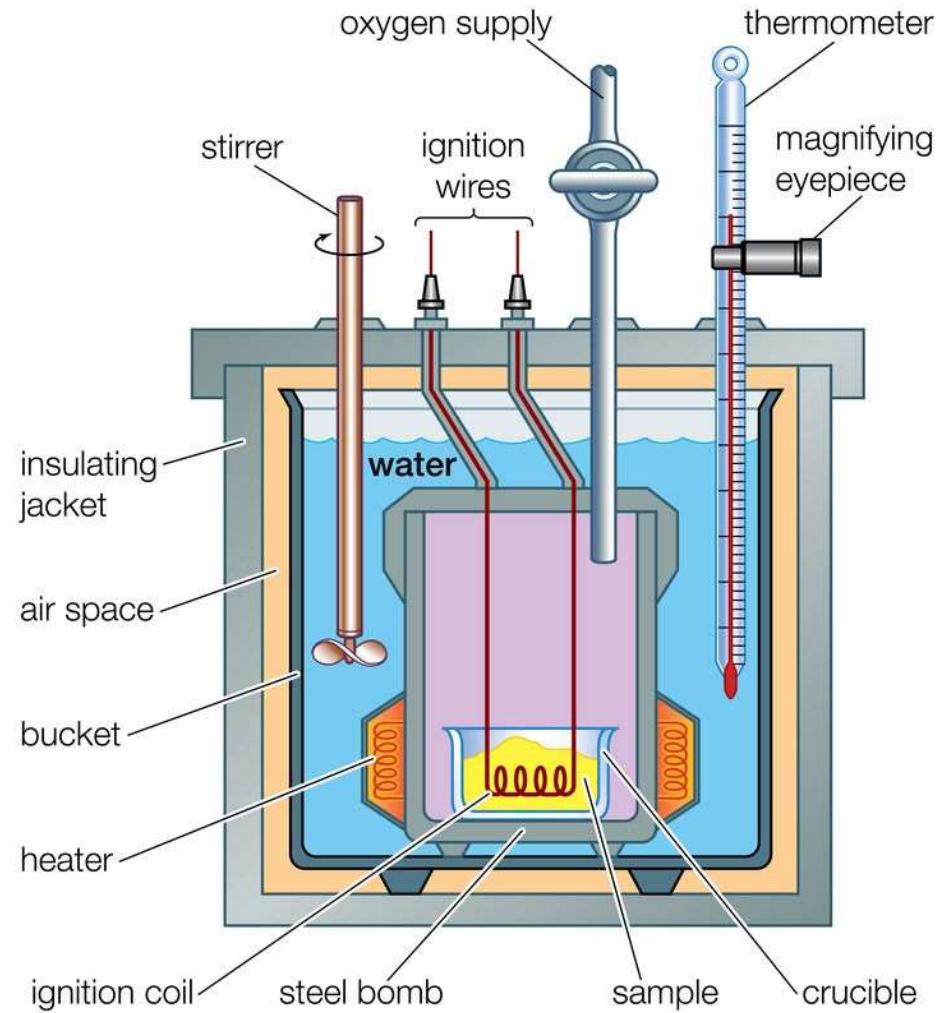
Lower heating value [MJ/kg]

\dot{m}_f \dot{V}_f
 P_f x_f \dot{x}_f / \dot{m}_f measured

property
tables

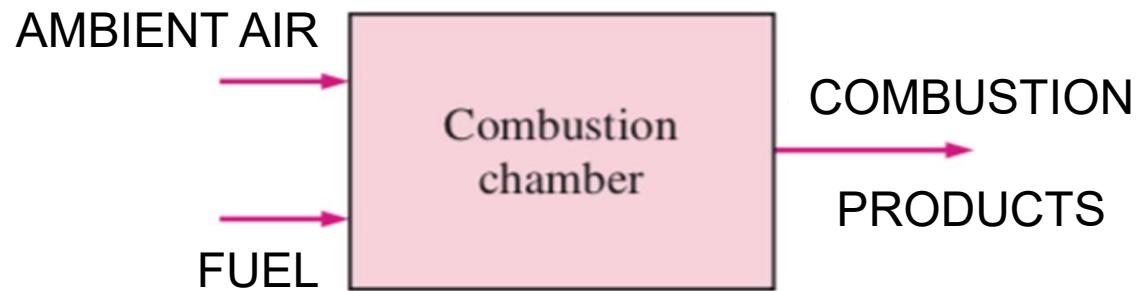
(91 octane)
gasoline

Bomb Calorimeter

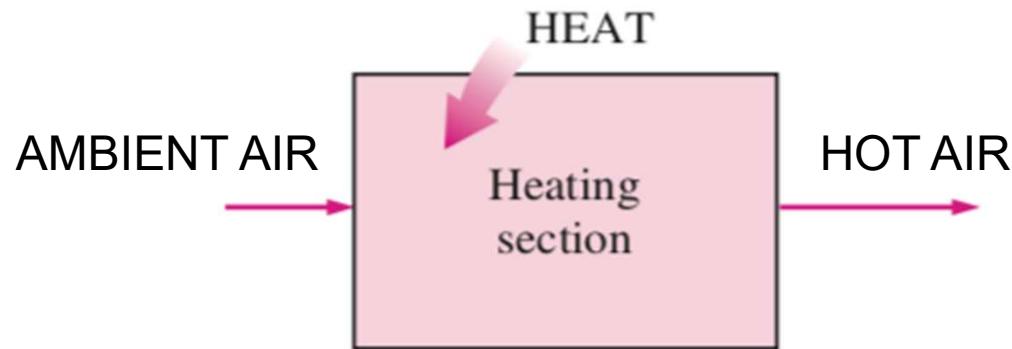


Ideal Thermodynamic Model

Actual:

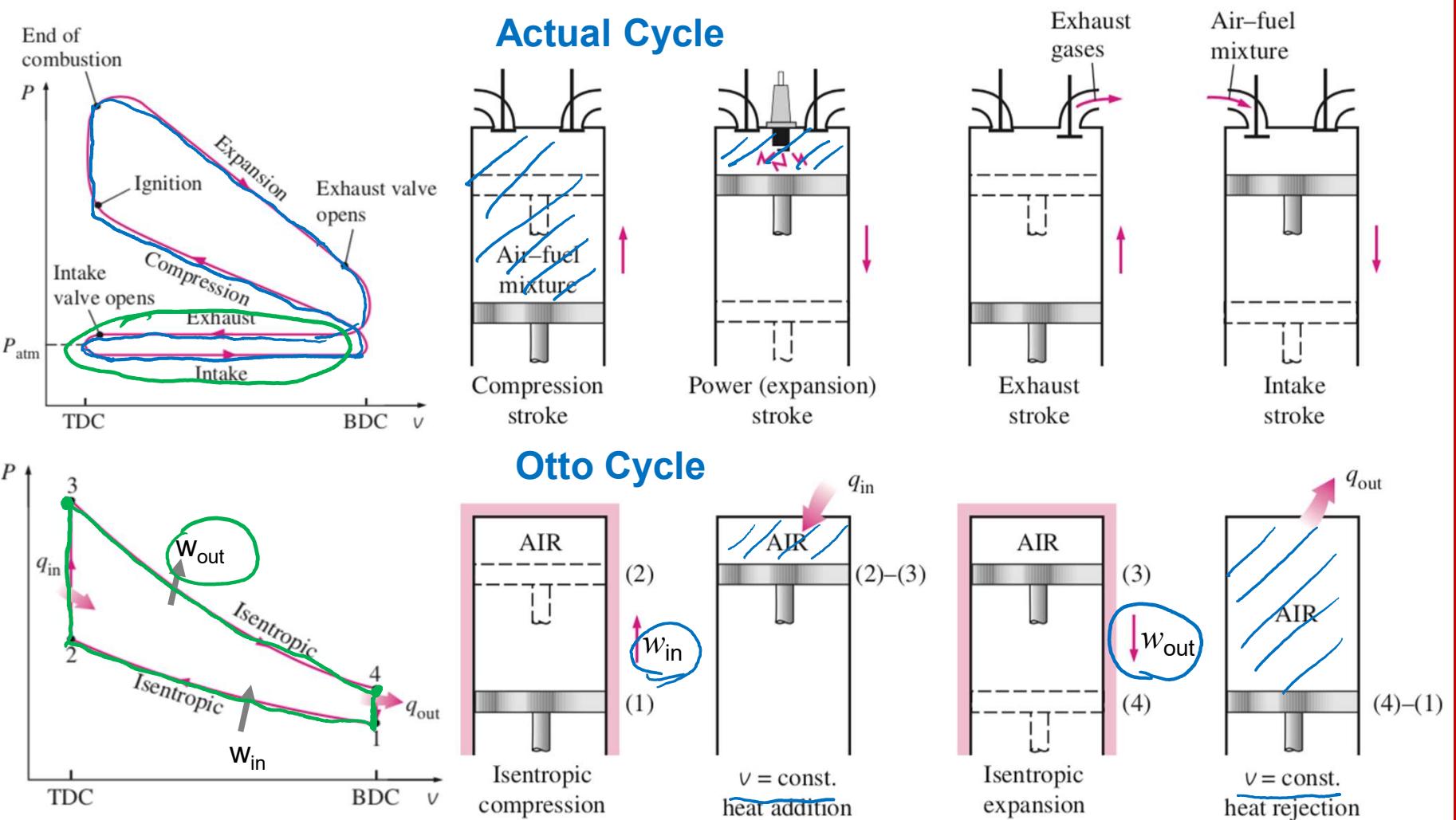


Ideal:



- air (ideal gas)
- processes are reversible
- combustion \rightarrow heat addition
- exhaust \rightarrow heat rejection

Thermodynamic Model



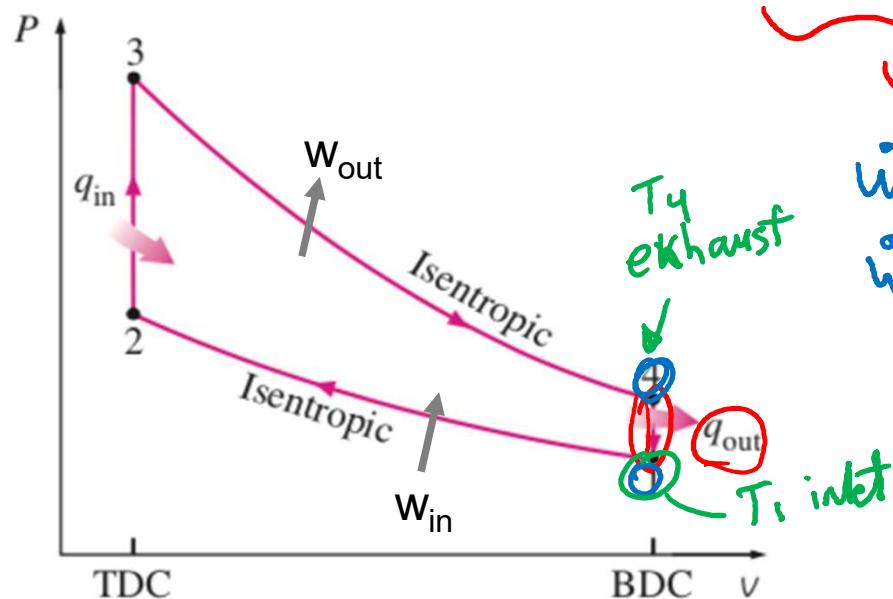
Ideal Thermal Efficiency and Power

Ideal Thermal Efficiency

$$\eta_{Otto} = 1 - \frac{1}{r^k - 1} \quad \text{where} \quad K = \frac{C_p}{C_v} = 1.4 \quad \text{standard-air}$$

↑ compression ratio

Ideal Power



$$\dot{W}_{out} - \dot{W}_{in} = \dot{Q}_{in} - \dot{Q}_{out}$$

↓ \dot{W}_{net} ↓ \dot{E}_{in} ↓ \dot{Q}_{out} ↓ \dot{Q}_{in} ↓ \dot{E}_{in} ↓ \dot{W}_{net}

constant-volume

$$\dot{W}_{net} = \dot{E}_{in} - \dot{m}_a (u_4 - u_1)$$

$$\dot{W}_{net} = \dot{E}_{in} - \dot{m}_a C_v (T_4 - T_1)$$

↑ measured ↑ measured ↑ measured ↑ measured ↑ measured ↑ measured

@ standard air ideal gas law @ 1

Questions??

Thank you for your attention!

Let me or the TAs know if you have questions