

HW 2 - ASTR404

Daniel George - dgeorge5@illinois.edu

Q1) Sirius A & B

a) Masses

Defining constants

```
In[29]:= G = Quantity["GravitationalConstant"];  
oP = Quantity[49.94, "Years"];  
{p, θ} = Quantity[{.37921, 7.61}, "ArcSeconds"];
```

Computing distance to binary

```
In[32]:= d = UnitConvert[FormulaData["ParallaxDistance", {p → p}][[2]], "m"]  
Out[32]= 8.1371284512369 × 1016 m
```

Finding semi-major axis of reduced mass

```
In[33]:= aR = Solve[θ == A/d, A][[1, 1, 2]]  
Out[33]= 3.0021383017581 × 1012 m
```

Finding distances to center of mass

```
In[34]:= {aA, aB} = NSolve[{a1 + a2 == aR, a1/a2 == .466}, {a1, a2}][[1, All, 2]]  
Out[34]= {9.5429498541562 × 1011 m, 2.0478433163425 × 1012 m }
```

Finding total mass of system (Kepler's 3rd Law)

```
In[35]:= M = NSolve[oP^2 == 4 Pi^2 aR^3 / (G m), m][[1, 1, 2]]  
Out[35]= 6.452828528697 × 1030 kg
```

Finding individual masses

```
In[36]:= {mA, mB} = NSolve[{m1 aA == m2 aB, m1 + m2 == M}, {m1, m2}] [[1, All, 2]]
```

```
Out[36]:= { 4.4016565680061 × 1030 kg , 2.0511719606909 × 1030 kg }
```

b) Luminosities

Luminosity-magnitude relation

```
In[37]:= L = 100.4 (4.75-M) 1 L☉ ;
```

Luminosity of Sirius A

```
In[38]:= LA = L /. M → 1.36
```

```
Out[38]:= 22.698648518838 L☉
```

Luminosity of Sirius B

```
In[39]:= LB = L /. M → 8.79
```

```
Out[39]:= 0.024210290467362 L☉
```

c) Radius of Sirius B

Stefan Boltzmann Law

```
In[40]:= L == 4 Pi R^2 (1 σ) T^4 ;
```

Solving for radius of Sirius B

```
In[41]:= rB = NSolve[LB == 4 Pi R^2 (1 σ) T^4 /. T → 24790 K, R] [[2, 1, 2]]
```

```
Out[41]:= 5.867646270211 × 106 m
```

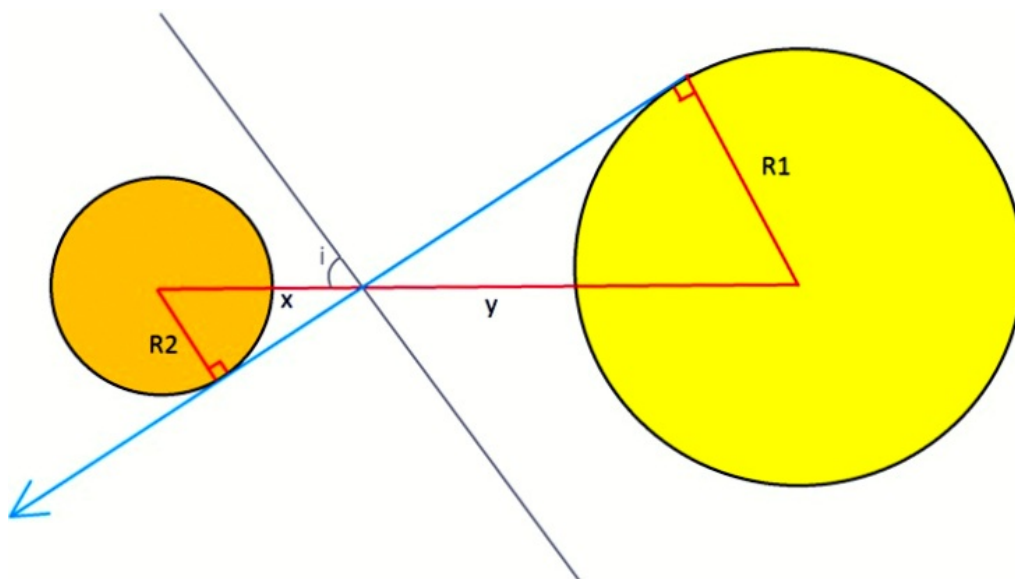
Comparison to Earth and Sun

```
In[42]:= Transpose@{{"RSirius-B", "REarth", "RSun"}, UnitConvert[
  {rB, Earth (planet) ["Radius"], Sun (star) ["Radius"]}, "km"]} // TableForm
```

Out[42]/TableForm=

$R_{\text{Sirius-B}}$	5867.646270211 km
R_{Earth}	6367.4447 km
R_{Sun}	6.955×10^5 km

Q2) Eclipsing binaries



a) Analytic solution

Solving inequalities

```
In[43]:= eqi = Reduce[
  (*Equations*)
  {Abs@Cos[i] < R1/(y) == R2/(x),
   (x + y) == a,
  (*Constraints*)
   0 ≤ i ≤ Pi, x > R2 > 0 m, y > R1 > 0 m}
  , {i, x, y}, Reals] // Simplify
```

```
Out[43]= x ==  $\frac{a R_2}{R_1 + R_2}$  && y ==  $\frac{a R_1}{R_1 + R_2}$  &&
   $\left(\frac{\pi}{2} < i < \text{ArcCos}\left[-\frac{R_1 + R_2}{a}\right] \mid \mid \left(\text{ArcCos}\left[\frac{R_1 + R_2}{a}\right] < i \&\& 2 i \leq \pi\right)\right) \&\&$ 
  0 m < R1 < a && 0 m < R2 && R1 + R2 < a && a > 0 m
```

Thus we can see from the above expression that the angle i must satisfy the following constraint:

$$\cos^{-1}\left(\frac{R_1 + R_2}{a}\right) < i < \cos^{-1}\left(-\frac{R_1 + R_2}{a}\right)$$

b) Numerical solution

Substituting given values

```
In[44]:= eqi /. {a -> 3 au, R1 -> 2 Sun(star) ["Radius"], R2 -> Sun(star) ["Radius"]} //
  Simplify // Last
```

```
Out[44]= 1.566147 < i < 1.575445
```

Q3) Boltzmann equation

Energy of nth state of hydrogen

```
In[45]:= e[n_] := -13.6/n^2 eV
```

Number of distinct states in nth shell

In[46]:= $g[n_] := 2 n^2$

Ratio of electrons in different shells

In[47]:= $eqB[i_, j_] := n[i] / n[j] == g[i] / g[j] \text{Exp}[(e[j] - e[i]) / (k T)]$

T at which $n_2 / n_1 = 1\%$

In[48]:= $\text{Solve}[1/100 == eqB[2, 1][[2]], T]$

Out[48]= $\{\{T \rightarrow 19755.79160745 \text{ K}\}\}$

T at which $n_2 / n_1 = 10\%$

In[49]:= $\text{Solve}[10/100 == eqB[2, 1][[2]], T]$

Out[49]= $\{\{T \rightarrow 32087.284631675 \text{ K}\}\}$

Q4) Ideal gas properties

a) Most probable velocity

Maxwell-Boltzmann distribution

This is proportional to the probability density function for velocities:

In[50]:= $n_v[v_] = 4 \pi v^2 n \left(\frac{m}{2 \pi k T} \right)^{3/2} \text{Exp}[-m v^2 / (2 k T)]$

Out[50]= $e^{-\frac{m v^2}{2 k T}} n \sqrt{\frac{2}{\pi}} \left(\frac{m}{k T} \right)^{3/2} v^2$

Finding extrema

We equate the derivative of the distribution function to zero and solve for the velocity.

In[51]:= $\text{vMP} = \text{Assuming}[k > 0 \ \&\& \ T > 0 \ \&\& \ m > 0 \ \&\& \ n > 0, \\ \text{Solve}[D[n_v[v], v] == 0 \ \&\& \ v > 0, v, \text{Reals}] // \text{Simplify}]$

Out[51]= $\{\{v \rightarrow \sqrt{2} \sqrt{\frac{k T}{m}}\}\}$

This value of velocity is either a maxima or minima of the distribution function since the derivative is zero at this point.

Checking whether this is a maxima

In[55]:= D[n_v[v], {v, 2}] /. vMP[[1]]

$$\text{Out[55]} = -\frac{4 n \sqrt{\frac{2}{\pi}} \left(\frac{m}{k T}\right)^{3/2}}{e}$$

The second derivative is negative at this value. Therefore this must be a maxima. Hence this is the most probable velocity.

b) RMS velocity

Definition of RMS velocity

$$v_{\text{RMS}}^2 = \frac{\int_0^\infty v^2 n_v(v) dv}{\int_0^\infty n_v(v) dv}$$

Computing value

In[53]:= vRMS = v -> Assuming[k > 0 && T > 0 && m > 0 && n > 0,
Sqrt[Integrate[v^2 n_v[v], {v, 0, Infinity}]/Integrate[n_v[v], {v, 0, Infinity}]]]

$$\text{Out[53]} = v \rightarrow \sqrt{3} \sqrt{\frac{k T}{m}}$$

c) Pressure

Formula for pressure

$$P = \frac{1}{3} \int_0^\infty m v^2 n_v(v) dv$$

Computing value

In[54]:= P -> 1/3
Integrate[m n_v[v] v^2, {v, 0, Infinity}, Assumptions -> k > 0 && T > 0 && m > 0 && n > 0]

$$\text{Out[54]} = P \rightarrow k n T$$