


	DO	Turbidity
What is it?	Oxygen gas trapped in water (diffusion, aeration, plants in water)	Visible particles suspended in water (silt/mud, biological)
How is it measured?	<ul style="list-style-type: none"> • Sensor (calibration!) • Chemicals $\text{mg/L} \leftrightarrow \text{ppm}$	<ul style="list-style-type: none"> • Sensor (calibration) • Secchi disk  • <u>NTU</u>
Why is it important for water quality?	Many aquatic organisms do not have lungs (they can't breathe air) — they get O_2 from water	<ul style="list-style-type: none"> • Turbidity can interfere with gill function (clogs) • Turbidity can increase temperature (sunlight) • Drinking, swimming, recreation is less fun

DO and drinking water:

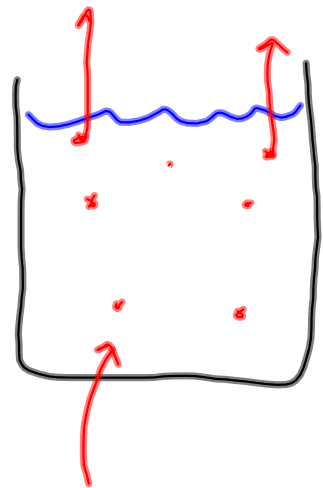
- Organisms do not get useable oxygen by drinking water

DO is critical [Some organisms have gills (or other organs) that can extract oxygen gas (DO) from water... but not by drinking it

DO is meaningless [Other organisms breathe - they get oxygen directly from air

DO and temperature:

- Warmer water can hold less DO



DO molecules

- as temperature increases, the oxygen molecules wiggle more
- the more they wiggle, the more likely they are to jump out of the water

Turbidity & rainfall:

- Turbidity is particles in water
- Rainfall causes erosion (where soil particles are carried away by water)
- Much of the water that falls as rain ends up in streams, rivers, and lakes
- SO: rainfall increases turbidity

WATER QUALITY IS RELATIVE:

- ideal values for the parameters are different!
- Some human use (drinking water, recreation)
 - Some other organism's use (animals, plants)
 - Some ecological function (like flooding, erosion, nutrient transport)
 - Historical conditions

locations	pH or turbidity	2 nd factor (ex. depth of creek)
under slabty bridge to gym		
fast water above Quad		
By big pipe by Highland		