



FIG. 14: The diffusion scaling with the excess entropy at (a) high and (b) low temperatures for the soft repulsive shoulder system.

by a mixture of hard spheres of two different sizes. The concentration of components of such mixture is pressure and temperature dependent. As it was shown in literature (see, for example, [69]) the excess entropy scaling holds for binary mixtures too. But in the case of quasi-binary mixture since the effective concentration depends on the pressure and temperature the behavior becomes more complex. This brings to the breakdown of the scaling rules for this case.

Obviously, the systems with bounded potentials can not be approximated by hard sphere potentials too. It seems that this may be the reason of violation of Rosenfeld entropy scaling for these systems.

We thank V. V. Brazhkin and Daan Frenkel for stimulating discussions. Our special thanks to Prof. Ch. Chakravarty who attracted our attention to the problems considered here. The work was supported in part by the Russian Foundation for Basic Research (Grant No 08-02-00781).

- [1] P. Debenedetti, J. Phys.: Condens. Matter 15, R1669 (2003).
- [2] S. V. Buldyrev, G. Franzese, N. Giovambattista, G. Malescio, M. R. Sadr-Lahijany, A. Scala, A. Skibinsky, and H. E. Stanley, Physica A 304, 23 (2002).
- [3] C. A. Angell, Annu. Rev. Phys. Chem. 55, 559 (2004).
- [4] P. G. Debenedetti, Metastable Liquids: Concepts and Principles (Princeton University Press, Princeton, 1998).
- [5] V. V. Brazhkin. S. V. Buldyrev, V. N. Ryzhov, and H. E. Stanley [eds], New Kinds of Phase Transitions: Transformations in Disordered Substances [Proc. NATO Advanced Research Workshop, Volga River] (Kluwer, Dordrecht, 2002).
- [6] J. R. Errington and P. G. Debenedetti, Nature (London) 409, 18 (2001).
- [7] P.A. Netz, F.V. Starr, H.E. Stanley, and M.C. Barbosa, J. Chem. Phys. 115, 318 (2001).
- [8] O. Mishima and H. E. Stanley, Nature 396, 329 (1998).
- [9] C.A. Angell, E.D. Finch, and P. Bach, J. Chem. Phys. 65, 3063 (1976).
- [10] H. Thurn and J. Ruska, J. Non-Cryst. Solids 22, 331 (1976).
- [11] G.E. Sauer and L.B. Borst, Science **158**, 1567 (1967).
- [12] F.X. Prielmeier, E.W. Lang, R.J. Speedy, H.-D. Lüdemann, Phys. Rev. Lett. 59, 1128 (1987).
- [13] F.X. Prielmeier, E.W. Lang, R.J. Speedy, H.-D. Lüdemann, B. Bunsenges, Phys. Chem. 92, 1111 (1988).
- [14] L. Haar, J. S. Gallagher, G. S. Kell, NBS/NRC Steam Tables. Thermodynamic and Transport Properties and Computer Programs for Vapor and Liquid States of Water in SI Units, Hemisphere Publishing Co., Washington DC, 1984, pp. 271-276.
- [15] A. Scala, F. W. Starr, E. LaNave, F. Sciortino and H. E. Stanley, Nature 406, 166 (2000).
- [16] http://www.lsbu.ac.uk/water/anmlies.html
- [17] J. R. Errington, Th. M. Truskett, J. Mittal, J. Chem. Phys. 125, 244502 (2006).
- [18] J. Mittal, J. R. Errington, Th. M. Truskett J. Chem. Phys. 125, 076102 (2006).
- [19] P. C. Hemmer and G. Stell, Phys. Rev. Lett. 24, 1284(1970).
- [20] G. Stell and P. C. Hemmer, J. Chem. Phys. 56, 4274 (1972).
- [21] G. Malescio, J. Phys.: Condens. Matter 19, 07310 (2007).
- [22] E.Velasco, L. Mederos, G. Navascues, P. C. Hemmer, and G. Stell, Phys. Rev. Lett. 85, 122 (2000).
- [23] P. C. Hemmer, E. Velasko, L. Mederos, G. Navascues, and G. Stell, J. Chem. Phys. 114, 2268 (2001).
- [24] M. R. Sadr-Lahijany, A. Scala, S. V. Buldyrev and H. E. Stanley, Phys. Rev. Lett. 81, 4895 (1998).