

$$Q = mC\Delta T \quad \text{or} \quad Q = mC(T_f - T_i)$$

Material	Specific Heat $\left(\frac{J}{kg^\circ C}\right)$
water	4,184
oxygen	918
aluminum	900
argon	520
steel	470
zinc	388
bronze	370
Platinum	133
gold	129

1. My material is water

I have a mass of 6 kg

The initial temperature is 5°C.

The final temperature is 40°C.

What is the heat energy?

Looking For	Formula	
Already Know		
Answer in a complete sentence <i>with unit</i> :		

2. My material is steel.

I have a mass of 80 kg.

My initial temperature is 26°C.

My final temperature is 28°C.

What is the heat energy?

Looking For	Formula	
Already Know		
Answer in a complete sentence <i>with unit</i> :		

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3.

My material is bronze.

I have a mass of 0.4 kg.

My initial temperature is 20°C.

My final temperature is 300°C.

What is the heat energy?

Looking For	Formula	
Already Know		
Answer in a complete sentence <i>with unit</i> :		

4.

My material is oxygen.

I have a mass of 20 kg.

My initial temperature is 20°C.

My final temperature is 35°C.

What is the heat energy?

Looking For	Formula	
Already Know		
Answer in a complete sentence <i>with unit</i> :		