

Artificial Chemistries



Autonomic Computer Systems

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Motivation

Chemical program evolution rather than design

- Chemical system quickly become complex
- Exploit emergent properties (self-organization, self-healing, robustness)
- Bottom-up approach
- From Pre-Programmed to Emergent Chemical Programs

Chemistry not only as a metaphor for **parallel** computation, but also **emergent** computation, c.f. Section 5

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1. Classification & Definition of AC

- ❑ A branch of Artificial Life (ALife), dedicated to the study of chemical processes related to life and organizations in general [1]
- ❑ AC is on the level of **molecules**, molecules as information carriers and processors. It is about chemistry, biochemistry, prebiotic evolution.
- ❑ AC can be defined by a **triple** (S, R, A) , where S is the set of all possible molecules, R is a set of collision rules representing the interaction among the molecules $s_i \in S$, and A is an algorithm describing the reaction vessel or domain and how much the rules are applied to the molecules inside the vessel. [1]

2. Motivation & Goal of AC

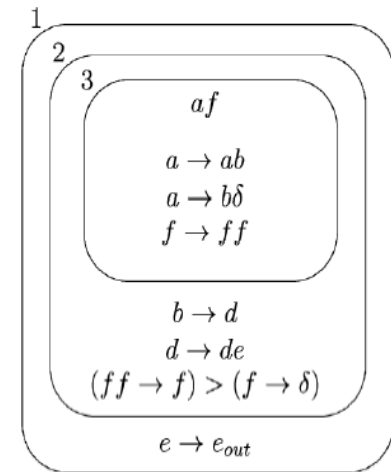
- ❑ Simulate (bio-)chemical processes, e.g. origin of life from primordial soup of molecules
- ❑ Explore alternative computation models: **chemical computing**, c.f. section 3
- ❑ Understand **emerging/self-maintaining structures**, e.g. chemical organization theory [2], c.f. sections 5, 8
- ❑ **Terminology:** *Molecular species, Multiset, (well-stirred) Vessel, Concentration, Chemical reaction*

3. Artificial Chemical Computing

- Program (= set of molecules) is executed (= molecules are consumed and produced in chemical reactions) until no more reactions can take place

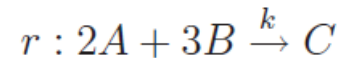
- Examples:

- Formal models {
- Computing with rules in a multiset (Gamma)
 - Computing with hierarchical multisets (Membrane Computing (P Systems))
 - > result is within molecule
 - Computing with Concentrations** (algebraic functions, also with real chemistry)
 - > result is reflected by the **concentration**



P Systems

4. Dynamics



- **Concentration levels** convey information about **dynamics**. **Computation** is used to assess changes in concentration. The Result is ready when the system reaches **equilibrium**. Concentration dynamics are describes by a system of **ODEs**:

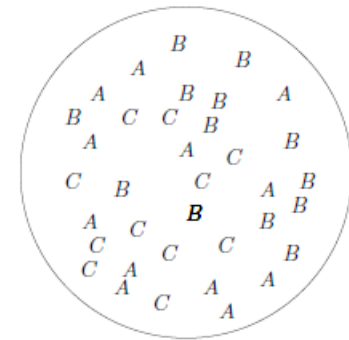
$$d\vec{x}/dt = M\vec{v}$$
- **Terminology**: *law of mass action*, $v = k[A]^2[B]^3$ $v = k \prod_{i=1}^n [S_i]^{\alpha_i}$
reaction networks, *equilibrium*: $d\vec{x}/dt = \vec{0}$
simulation algorithms: *deterministic* vs *stochastic*

4. Dynamics

Stochastic simulation:

Gillespie's Algorithm (1977)

calculates **which** reaction occurs **when**



$$|A| = 12 \quad |B| = 13 \quad |C| = 11$$

$$a_1 = c_1 |A| |B|$$
$$a_1 = 1 \cdot 12 \cdot 13 = 156$$

$$a_2 = c_2 |C|$$
$$a_2 = 10 \cdot 11 = 110$$

$$(1) \quad a_0 = a_1 + a_2 = 266$$

$$(2) \quad E[\tau] = \frac{1}{a_0} = 3.76 \cdot 10^{-3}$$

- Advantages:
 - Realistic for small numbers of molecules
 - Can easily deal with the dynamic creation of new species
- Shortcomings: not scalable to large numbers of molecules, or large numbers of possible reactions

4. Dynamics

Chemical reactions network emerge from an initial population and seem to stabilize -> emergence of **organizations** (c.f. section 8, Chemical Org. Theory)

Dynamics can be analyzed by the following methods:

- ▣ **Perturbation analysis:** equilibrium and stability (steady-state analysis)
- ▣ **Simulation algorithms:** deterministic vs stochastic

Problem: The analysis doesn't include the creation of new species (c.f. section 7, Algorithmic Chemistry)

4. Dynamics

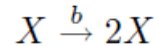
- ❑ Dynamics control which of the parallel execution paths is processed at which speed
- ❑ Under selective pressure, the growth rate of a set of molecules determines its fitness (see Evolutionary Dynamics)

5. Emergent Computation

- **Creation/Selection process** from an **evolutionary dynamics** (c.f. Section 6) perspective
- **Molecules** with **structure** (program, data) instead of simple atoms
- Reaction rules defined by the molecular structure instead of being fixed and predefined
- **Advantages:** more robust and efficient, more flexible in the face of complex and dynamic environments, do not need to foresee all contingencies, better model for complex natural processes
- **Terminology: Emergent phenomena** (*unexpected global behavior or pattern arising from the interaction of parts in a complex system*)
Self-organization (*process by which the system is able organize itself from the inside, out of a disordered state*), **Complex System**

6. Evolutionary Dynamics

Terminology:



- Replication (autocatalytic reaction) and Death (decay reaction) $X \xrightarrow{d} .$
- Selection and Fitness (=rate of reproduction)
- Growth laws $\dot{x}_i = f_i x_i^c$, for varying c
- Variation (e.g. mutations), induced external/passive or internal/active, Reproduction = Replication + Variation, quasispecies
- Interactions (Random catalytic reaction networks with Catalytic Reaction Equation:)

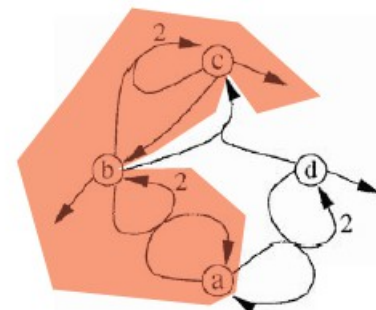
$$\frac{dx_k}{dt} = \sum_{i=1}^n \sum_{j=1}^n \alpha_{ij}^k x_i x_j - x_k \phi(t)$$

7. Algorithmic Chemistry

- A Constructive system: A set of molecules may generate **new molecule species**, which again influence the reaction dynamics
 - Considers molecule structure:
 - Molecules are λ -expressions from λ -calculus
 - All molecules represent **functions** (programs in functional programming)
 - Functions operate on functions to produce other functions: $f + g \rightarrow f + g + f(g($
- Precursor of chemical organization theory

8. Chemical Organization Theory

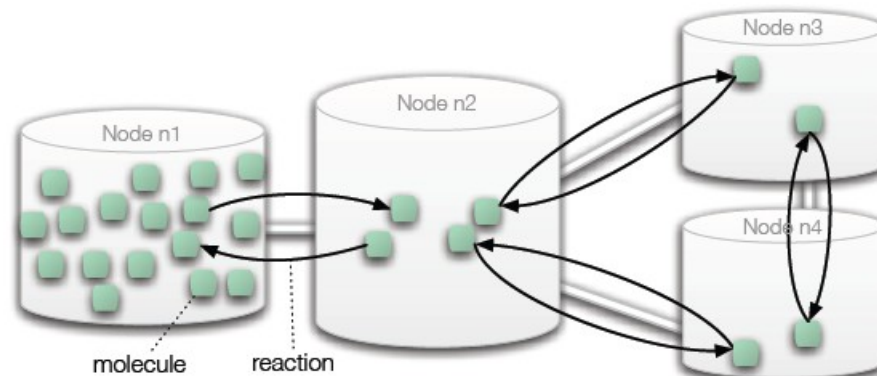
- ❑ Identify organizations (stable set of molecules) and the transition between them
- ❑ **Definition: An organization is a set of molecule species that is closed and self-maintaining**
- ❑ Chemical Organization Theory determines the stability of the set of molecule species and analyzes then dynamic fixpoints (= instances of organizations)
- ❑ The set of organizations form a lattice
- ❑ **Mutations** cause molecule species to come and go
- ❑ Down- and upward movement



9. Examples in Fraglets

□ Distributed Averaging

- Distributed calculation as an **emergent phenomena**
- Make use of the chemical equilibrium
- Reaction network is distributed among the whole network
- System strives to equilibrium where solution is presented

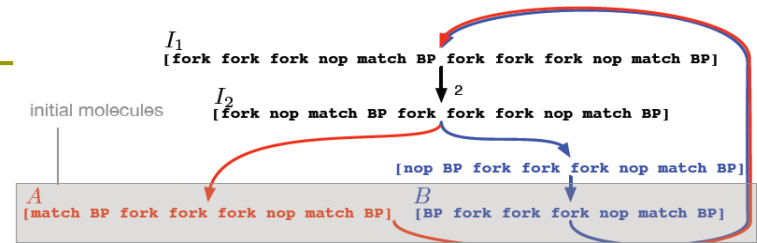


9. Examples in Fraglets

□ Self-Healing Programs

- Autocatalytic reaction
- Two molecules species form an organization
- population grows hyperbolically -> introduce limitation (total number of molecules) -> survival of the common (first program).
- The resulting program is robust to invasion from other programs and code deletion -> **inherent self-healing emerges**

Application: Self-Healing link load balancing protocol



References

- ▣ P. Dittrich, J. Ziegler, and W. Banzhaf. “Artificial Chemistries – A Review”. *Artificial Life*, 7(3):225–275, 2001.
- ▣ P. Dittrich and P. Speroni di Fenizio. “*Chemical Organization Theory*”. *Bulletin of Mathematical Biology*, 69(4):1199–1231, 2007.