X-ray spectroscopic characterization of As(V)-rich TI(III)-particles in a weathered TI-As-Fe-sulfide mineralization

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Thallium is a highly toxic trace metal of growing environmental concern. In Erzmatt (Swiss Jura mountains) soils were found to contain high levels of TI and As due to their formation from carbonate rock hosting a weathered hydrothermal TI-As-Fe mineralization. Although TI in this site was constrained to a limited area, it is ideally suited to study the long-term transformation of TI at contaminated sites and its speciation in soils.

In a previous study, the Erzmatt soils were found to contain TI^I-substituted jarosite and avicennite (TI^{III}₂O₃) as important secondary TI-bearing minerals [1]. The dominant soil-formed TI species was TI^I associated with illite. Elevated levels of TI^{III} were associated with soil Mn concretions, but accounted for only a minor fraction of total soil TI. Circumstantial evidence suggested that another unidentified TI^{III}-rich phase is present in the Erzmatt soils.

In continuing work, we identified and characterized the unknown TI^{III}-rich phase by examining soil and rock thin sections as well as individual particles isolated from soil and rock combining laboratory X-ray fluorescence spectrometry (XRF) and X-ray diffraction (XRD) with synchrotron-based bulk X-ray absorption spectroscopy (XAS) and micro-resolved XRF/XRD tomography.

Our results suggest that the composition of the TI-rich particles corresponds to TI_2O_3 with up to 0.25 As(V)/TI. Extended X-ray absorption fine structure (EXAFS) spectra in combination with XRD data suggest that the As^V-rich $TI^{III}_2O_3$ grains are amorphous or nanocrystalline precursors of crystalline avicennite. We speculate that these particles form by the oxidation of TI-As sulfide minerals such as lorandite (TIAsS₂) or ellisite (TI_3AsS_4) and that the As^V inhibits the crystallization of TI_2O_3 into avicennite.

Avicennite and As^V -rich Tl_2O_3 particles in the Erzmatt soils are coated with Mn-oxides that we identified as layered MnO₂, possibly in association with a tectomanganate. The MnO₂-crusts contain up to 0.2 Tl/Mn as Tl(I) and may have formed via a redox reaction between dissolved Mn²⁺ and Tl^{III}.

The results from this work provide new insights into the processes that control the release of TI from TI-bearing sulfide minerals and the sequestration of TI in soils, and are relevant with respect to the assessment of risks arising from geogenically and anthropogenically TI-contaminated sites.

REFERENCES

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