Hillsborough County Fire Rescue STANDING ORDERS AND PROTOCOL

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Subject: SODIUM BICARBONATE

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Michael Lozano, Jr., M.D., HCFR Medical Director

Sodium Bicarbonate

- 1. CLASSIFICATION
 - a. Electrolyte
 - b. Buffer
- 2. ACTIONS / DESCRIPTIONS
 - a. Reacts with hydrogen ions to form water and carbon dioxide
 - b. Buffers metabolic acidosis
- 3. INDICATIONS
 - a. Intubated patient with continued long arrest interval
 - b. Tricyclic antidepressant overdose
- 4. CONTRAINDICATIONS
 - a. Metabolic or respiratory alkalosis
- 5. PRECAUTIONS
 - a. Patients with heart failure
- 6. Adverse Reactions
 - a. Metabolic Alkalosis
 - b. Increases intracellular PCO² and increases tissue acidosis.
 - Produces CO², which crosses cell membranes more rapidly than bicarbonate (potentially worsening intracellular acidosis).
 - d. Electrolyte imbalance
 - e. Tissue sloughing at injection site with extravasation
- 7. INFORMATIONAL/DISCUSSION POINTS
 - a. Adequate ventilation and CPR are the major "buffer agents" in cardiac arrest
 - b. Blood gas analysis should guide bicarbonate administration (pre-hospital).
 - c. When to use in tricyclic antidepressant overdose