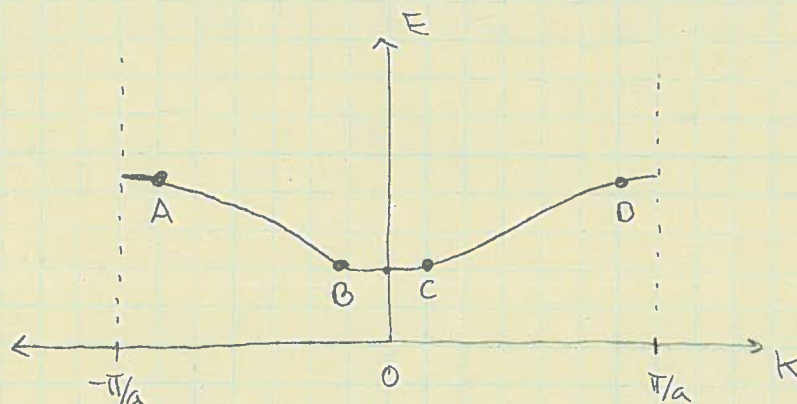


3.15



a) Effective mass $\propto \frac{1}{\frac{d^2E}{dk^2}}$

A: $\frac{d^2E}{dk^2}$ is negative $\Rightarrow m^*$ negative

B: $\frac{d^2E}{dk^2}$ is positive $\Rightarrow m^*$ positive

C: $\frac{d^2E}{dk^2}$ is positive $\Rightarrow m^*$ positive

D: $\frac{d^2E}{dk^2}$ is negative $\Rightarrow m^*$ negative

b) Velocity direction $\propto \frac{dE}{dk}$

A: $\frac{dE}{dk} < 0 \Rightarrow v$ is in $-x$ direction

B: $\frac{dE}{dk} < 0 \Rightarrow v$ is in $-x$ direction

C: $\frac{dE}{dk} > 0 \Rightarrow v$ is in $+x$ direction

D: $\frac{dE}{dk} > 0 \Rightarrow v$ is in $+x$ direction

3.16

The general expression for a parabola is

$$E(k) = C_1 k^2 + C_2$$

↑ curvature ↑ intercept = 0 in both curves

For A: $E(0.04 \text{ \AA}^{-1}) = C_1 (0.04 \text{ \AA}^{-1})^2 = 0.05 \text{ eV}$

$$\Rightarrow C_1 = 7.8125 \text{ eV} \cdot \text{\AA}^2$$

so $E(k) = 7.8125 k^2 \text{ [eV]}$

$$m^* = \frac{\hbar}{\frac{d^2 E}{dk^2}} = \frac{\hbar^2}{2 \cdot (7.8125) \text{ eV} \cdot \text{\AA}^2}$$

$$m^* = \frac{(1.054 \times 10^{-34} \text{ F} \cdot \text{s})^2}{2(7.8125) \text{ eV} \cdot \text{\AA}^2 \cdot (1.6 \times 10^{-19} \text{ F/eV}) (1 \times 10^{-10} \text{ m/\AA})^2}$$

$$= 4.44 \times 10^{-31} \text{ kg} \Rightarrow \boxed{0.488 m_0}$$

For B: $E(0.08 \text{ \AA}^{-1}) = C_1 (0.08 \text{ \AA}^{-1})^2 = 0.5 \text{ eV}$

$$\Rightarrow C_1 = 78.125 \text{ eV} \cdot \text{\AA}^2$$

so $E(k) = 78.125 \text{ eV} \cdot \text{\AA}^2 \cdot k^2 \text{ [eV]}, k \text{ in } \text{\AA}^{-1}$

$$m^* = \frac{(1.054 \times 10^{-34} \text{ F} \cdot \text{s})^2}{2(78.125 \text{ eV} \cdot \text{\AA}^2) (1.6 \times 10^{-19} \text{ F/eV}) (1 \times 10^{-10} \text{ m/\AA})^2}$$

$$= 4.44 \times 10^{-32} \text{ kg} \Rightarrow \boxed{0.0488 m_0}$$

3.20

$$E = E_0 - E_1 \cos(\kappa(k - k_0))$$

$$m^* = \frac{\hbar^2}{\frac{d^2 E}{dk^2}}$$

$$\frac{dE}{dk} = E_1 \sin(\kappa(k - k_0)) \cdot \kappa$$

$$\frac{d^2 E}{dk^2} = E_1 \cos(\kappa(k - k_0)) \cdot \kappa^2$$

$$\text{so } m^*(k) = \frac{\hbar^2}{E_1 \kappa^2 \cos(\kappa(k - k_0))}$$

m^* is a function of k , so evaluating at k_0

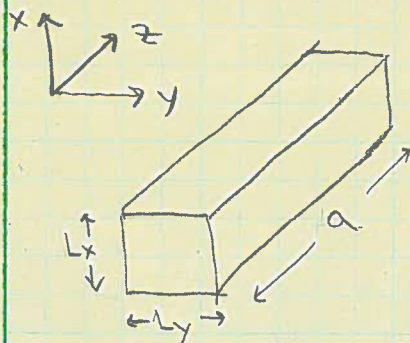
$$m^*(k_0) = \frac{\hbar^2}{E_1 \kappa^2}$$

3.25

1-D electron gas in GaAs \rightarrow two quantized dimensions

$$m_n^* = 0.067 m_0$$

$$E = \pm \frac{p^2}{2m_n^*} \Rightarrow \text{Two momentum states per energy level}$$



- L_x, L_y are quantum sized
- a is large

- electrons energies are quantized in x and y and continuous in z

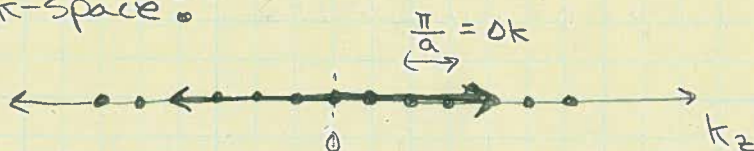
- the total energy is

$$E = E_{nx} + E_{ny} + \frac{\hbar^2 k_z^2}{2m_n^*}$$

$$E_{nx} = \frac{\pi^2 \hbar^2 n_x^2}{2m_n^* L_x^2}$$

$$E_{ny} = \frac{\pi^2 \hbar^2 n_y^2}{2m_n^* L_y^2}$$

- E_{nx} and E_{ny} are quantized
- Follow a procedure similar to the one on slide 9 of lecture 10
- in k -space:



note: we have to consider $+$ and $-$ k values since we are told there are two momentum states per energy level

$$N_s(k) = \frac{2 \cdot k_z}{(\pi/a)} \cdot 2 \cdot \frac{1}{2}$$

\uparrow spin \uparrow positive k_z only

$$N_s(k) = \frac{2ka}{\pi}$$

- translate $N_3(k)$ into $N_3(E)$ assuming free electron in z -direction

$$E = E_{nx} + E_{ny} + \frac{\hbar^2 k_z^2}{2m_n^*} \Rightarrow k_z = \frac{\sqrt{2m_n^*(E - E_{nx} - E_{ny})}}{\hbar}$$

$$\text{so } N_3(E) = \frac{2a \sqrt{2m_n^*(E - E_{nx} - E_{ny})}}{\pi \hbar}$$

- divide by unit length $= a$

$$\frac{N_3(E)}{a} = \frac{2}{\pi \hbar} \sqrt{2m_n^*(E - E_{nx} - E_{ny})}$$

$$\text{- Finally } \frac{1}{a} \frac{dN_3(E)}{dE} = \frac{1}{2} \frac{2}{\pi \hbar} \sqrt{2m_n^*} (E - E_{nx} - E_{ny})^{-\frac{1}{2}}$$

$$\boxed{\text{DOS}_{1D} = \frac{1}{\pi \hbar} (2m_n^*)^{1/2} \frac{1}{\sqrt{E - E_{nx} - E_{ny}}}}$$

states
unit length · unit E

- we can plug in some values for GaAs

$$\text{DOS}_{\text{GaAs}_{1D}} = \frac{(2m_n^*)^{1/2}}{\pi \hbar} \cdot \frac{1}{\sqrt{E - E_{nx} - E_{ny}}}$$

$$= \frac{(2(9.11 \times 10^{-31} \text{ kg})(0.067))^{1/2}}{\pi(1.054 \times 10^{-34} \text{ J} \cdot \text{s})} \frac{1}{\sqrt{E - E_{nx} - E_{ny}}}$$

$$= \frac{1.055 \times 10^{18}}{\sqrt{E - E_{nx} - E_{ny}}} \left[\text{J}^{-1} \cdot \text{m}^{-1} \right]$$

* constant has units of $\text{m}^{-1} \cdot \text{J}^{-1/2}$

* E is in J

- * Considering all energy levels

$$\boxed{\text{DOS}_{1D} = \frac{1}{\pi \hbar} (2m_n^*)^{1/2} \sum_{n_x, n_y} \frac{1}{\sqrt{E - E_{nx} - E_{ny}}}}$$

3.26

a) determine the total number ($\#/\text{cm}^3$) of energy states in Si between E_c and $E_c + 2kT$ at (i) $T = 300\text{K}$, (ii) $T = 400\text{K}$

- assuming this is "bulk" Si
- need conduction band D.O.S.

$$g_c(E) = \frac{4\pi(2m_n^*)^{3/2}}{h^3} \sqrt{E - E_c} \quad \frac{\#}{\text{unit V} \cdot \text{unit E}}$$

- to get total $\#/\text{cm}^3$ we need to integrate with respect to E

$$\frac{\#}{\text{cm}^3} = \int_{E_c}^{E_c + 2kT} g_c(E) dE = \frac{4\pi(2m_n^*)^{3/2}}{h^3} \int_{E_c}^{E_c + 2kT} (E - E_c)^{1/2}$$

$$= \frac{2}{3} \cdot \frac{4\pi(2m_n^*)^{3/2}}{h^3} (E - E_c)^{3/2} \Big|_{E_c}^{E_c + 2kT}$$

$$= \frac{8\pi(2m_n^*)^{3/2}}{3h^3} [(2kT)^{3/2}]$$

For Si: $m_n^* = 1.08 m_0$ (From Table B.4 - 0.0.s. effective mass)

$$(i) T = 300\text{K}: \frac{8\pi(2(1.08)(9.11 \times 10^{-31} \text{kg}))^{3/2}}{3(6.62 \times 10^{-34} \text{J.s})^3} \left[2(1.38 \times 10^{-23} \text{J/K})(300\text{K}) \right]^{3/2}$$

$$= 6.01 \times 10^{25} \text{m}^{-3} = \boxed{6.01 \times 10^{19} \text{cm}^{-3}}$$

$$(ii) T = 400\text{K}: \frac{8\pi(2(1.08)(9.11 \times 10^{-31} \text{kg}))^{3/2}}{3(6.62 \times 10^{-34} \text{J.s})^3} \left[2(1.38 \times 10^{-23} \text{J/K})(400\text{K}) \right]^{3/2}$$

$$= 9.25 \times 10^{25} \text{m}^{-3} = \boxed{9.25 \times 10^{19} \text{cm}^{-3}}$$

b) For GaAs $m_n^* = 0.067 m_0$

(i) Just scale the result for Si at 300K by

$$\left(\frac{m_{n, \text{GaAs}}^*}{m_{n, \text{Si}}^*} \right)^{3/2} \Rightarrow 6.01 \times 10^{19} \text{ cm}^{-3} \left(\frac{0.067}{1.08} \right)^{3/2} = \boxed{9.29 \times 10^{17} \text{ cm}^{-3}}$$

(ii) Scale result for Si at 400K by

$$\left(\frac{m_{n, \text{GaAs}}^*}{m_{n, \text{Si}}^*} \right)^{3/2} \Rightarrow 9.25 \times 10^{19} \text{ cm}^{-3} \left(\frac{0.067}{1.08} \right)^{3/2} = \boxed{1.43 \times 10^{18} \text{ cm}^{-3}}$$

3.28 a) Plot DOS in conduction band for Si
over $E_c < E < E_c + 0.4 \text{ eV}$

b) Repeat a) in valence band for $E_v - 0.4 \text{ eV} < E < E_v$

$$a) \quad g_c(E) = \frac{4\pi (2m_n^*)^{3/2}}{h^3} \sqrt{E - E_c}$$

SEE attached Matlab code and plot

$$b) \quad g_v(E) = \frac{4\pi (2m_p^*)^{3/2}}{h^3} \sqrt{E_v - E}$$

SEE attached Matlab code and plot

3.2a

```
clear all;
```

```
m0 = 9.11e-31; %kg  
mn = 1.08.*m0;  
h = 6.622e-34; %J s
```

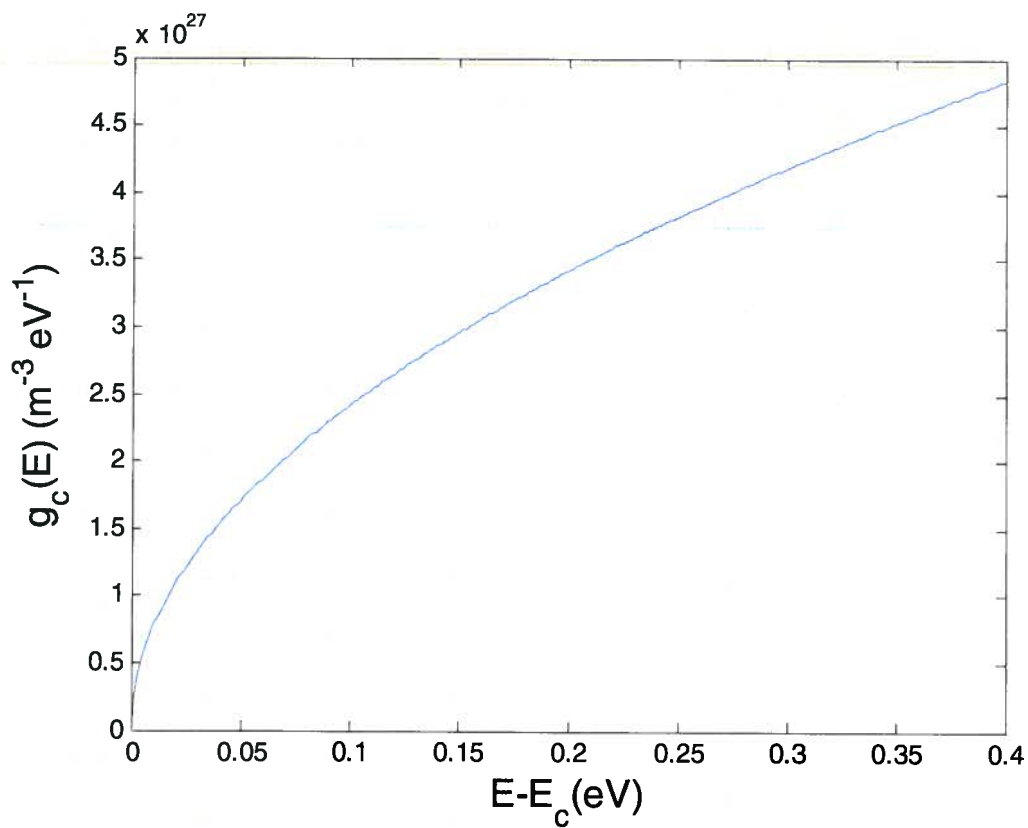
```
E_Ec = 0:.001:0.4;
```

```
gc = (1.6e-19).^(3/2).*((4.*pi.*(2.*mn).^(3/2))./(h.^3))*(E_Ec).^0.5
```

```
plot(E_Ec,gc)
```

```
xlabel('E-Ec(eV)', 'FontSize', 14)
```

```
ylabel('gc(E) (m-3 eV-1)', 'FontSize', 14)
```



```
clear all;
```

```
m0 = 9.11e-31; %kg
```

```
mp = 0.56.*m0;
```

```
h = 6.622e-34; %J s
```

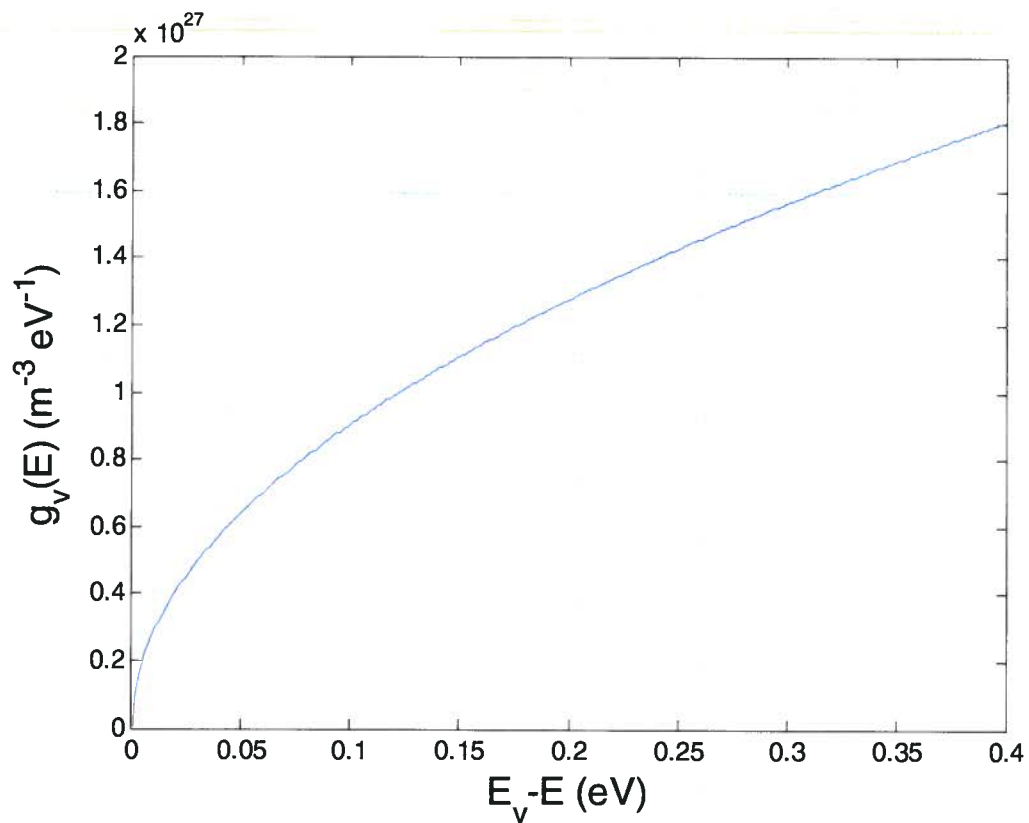
```
Ev_E = 0:.001:0.4;
```

```
gv = (1.6e-19).^(3/2).*((4.*pi.*(2.*mp).^(3/2))./(h.^3))*(Ev_E).^0.5
```

```
plot(Ev_E,gv)
```

```
xlabel('E_v-E (eV)', 'FontSize', 14)
```

```
ylabel('g_v(E) (m^-3 eV^-1)', 'FontSize', 14)
```



3.30

$$F_F(E) = \frac{1}{e^{(E-E_F)/KT} + 1}$$

$$-0.2 \leq (E-E_F) \leq 0.2 \text{ eV}$$

a) $T = 200 \text{ K}$

b) $T = 300 \text{ K}$

c) $T = 400 \text{ K}$

* SEE Attached Matlab code and plot

3.32

Determine the probability that an energy level is occupied by an electron if the state is above the Fermi level by

(a) KT

(b) $5KT$

(c) $10KT$

$$F_F(E) = \frac{1}{e^{(E-E_F)/KT} + 1}$$

(a) $E-E_F = KT \Rightarrow F_F(E) = \frac{1}{e^1 + 1} = \boxed{0.269}$

(b) $E-E_F = 5KT \Rightarrow F_F(E) = \frac{1}{e^5 + 1} = \boxed{6.69 \times 10^{-3}}$

(c) $E-E_F = 10KT \Rightarrow F_F(E) = \frac{1}{e^{10} + 1} = \boxed{4.54 \times 10^{-5}}$

3.30

```
clear all;
```

```
%T = 200; %K
```

```
%T = 300; %K
```

```
T = 400; %K
```

```
k = 8.62e-5; %eV/K
```

```
E_Ef = -0.2:.001:0.2; %eV
```

```
F = 1./(exp(E_Ef./(k.*T)) + 1);
```

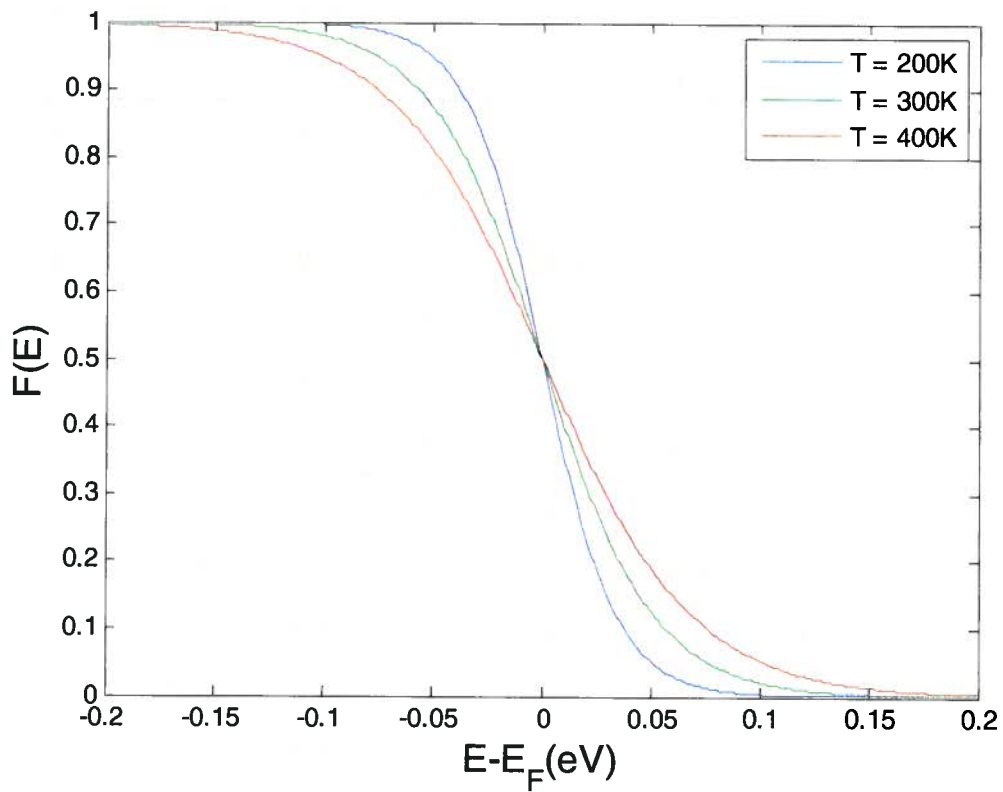
```
plot(E_Ef,F)
```

```
xlabel('E-EF(eV)', 'FontSize', 14)
```

```
ylabel('F(E)', 'FontSize', 14)
```

```
legend('T = 200K', 'T = 300K', 'T = 400K', 'Location', 'NorthEast')
```

```
hold all;
```



3.33

Determine the probability that an energy level is empty of an electron if the state is below the Fermi level by

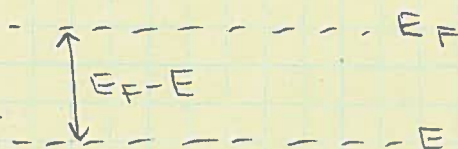
(a) kT (b) $5kT$ (c) $10kT$

For an empty state we look at

$$1 - F(E) = 1 - \frac{1}{e^{(E-E_F)/kT} + 1}$$

$$= \frac{e^{(E-E_F)/kT} + 1 - 1}{e^{(E-E_F)/kT} + 1}$$

$$= \frac{1}{e^{(E_F-E)/kT} + 1}$$



$$(a) \quad 1 - F(E) = \frac{1}{e^1 + 1} = \boxed{0.269}$$

$$(b) \quad 1 - F(E) = \frac{1}{e^5 + 1} = \boxed{6.69 \times 10^{-3}}$$

$$(c) \quad 1 - F(E) = \frac{1}{e^{10} + 1} = \boxed{4.54 \times 10^{-5}}$$

3.39

$$F_F(E) = \frac{1}{e^{(E-E_F)/KT} + 1}$$

$$F_B(E) = e^{-(E-E_F)/KT} \quad \text{Boltzmann}$$

(a) Find $(E-E_F)$ where $F_F(E)$ is within 1% of $F_B(E)$

↳ since $F_B(E)$ is always larger we write (see Ex 3.8)

$$\frac{\frac{1}{e^{(E-E_F)/KT}} - \frac{1}{e^{(E-E_F)/KT} + 1}}{1} = 0.01$$

$$\Rightarrow \frac{e^{(E-E_F)/KT} + 1}{e^{(E-E_F)/KT}} - 1 = 0.01$$

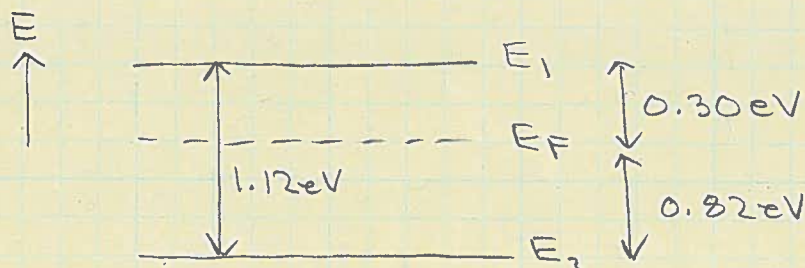
$$\frac{e^{(E-E_F)/KT} + 1 - e^{(E-E_F)/KT}}{e^{(E-E_F)/KT}} = 0.01$$

$$\rightarrow e^{-(E-E_F)/KT} = 0.01$$

$$(E-E_F) = -\ln(0.01) \cdot KT = 4.6 KT$$

$$(b) F_F(E) = \frac{1}{e^{4.6} + 1} = 0.00995 \approx 0.01$$

3.42



$$T = 300 \text{ K}$$

a) $E_1 - E_F = 0.30 \text{ eV}$ → what is probability of occupation at $E = E_1$?

→ what is probability of non-occupation at $E = E_2$?

$$F_F(E_1) = \frac{1}{e^{(E_1 - E_F)/kT} + 1}$$

$$= \frac{1}{e^{(0.3/0.0259)} + 1} = \boxed{9.32 \times 10^{-6}}$$

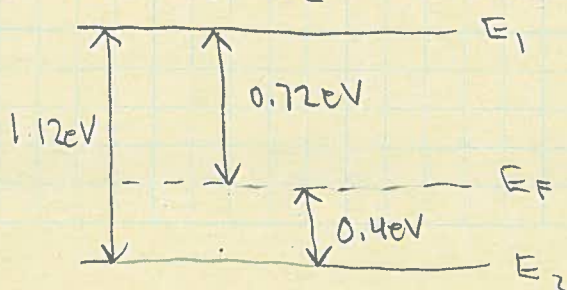
$$1 - F_F(E_2) = \frac{1}{e^{(E_F - E_2)/kT} + 1}, \quad E_F - E_2 = 1.12 - 0.3 = 0.82 \text{ eV}$$

$$= \frac{1}{e^{(0.82/0.0259)} + 1} = \boxed{1.78 \times 10^{-14}}$$

⇒ so since $E_1 \gg E_F$ the chances of occupation are low

⇒ since $E_2 \ll E_F$, the chances of E_2 being occupied (empty) are high (low)

(b) IF $E_F - E_2 = 0.4 \text{ eV}$



$$F_F(E_1) = \frac{1}{e^{(0.72/0.0259)} + 1} = \boxed{8.45 \times 10^{-12}}$$

$$1 - F_F(E_2) = \frac{1}{e^{(0.4/0.0259)} + 1} = \boxed{1.96 \times 10^{-7}}$$