

2.5

$$\Phi_{\text{Au}} = 4.90 \text{ eV} \quad \text{For Au}$$

$$\Phi_{\text{Ce}} = 1.90 \text{ eV} \quad \text{For Ce}$$

work function $\Phi = h\nu_0$ where ν_0 is the minimum frequency to remove an electron

$$c = \lambda\nu_0 \Rightarrow \lambda_{\text{max}} = \frac{c}{\nu_0}$$

$$\nu_0 = \frac{\Phi}{h} = \frac{4.90 \text{ eV}}{4.135 \times 10^{-15} \text{ eV} \cdot \text{s}} = 1.185 \times 10^{15} \text{ s}^{-1}$$

$$\text{so } \lambda_{\text{max}} = \frac{3 \times 10^8 \text{ m/s}}{1.185 \times 10^{15} \text{ s}^{-1}} = 2.53 \times 10^{-7} \text{ m}$$

$$= 253 \text{ nm}$$

For Au

→ ultra violet

similarly for Ce:

$$\nu_0 = \frac{\Phi}{h} = \frac{1.90 \text{ eV}}{4.135 \times 10^{-15} \text{ eV} \cdot \text{s}} = 4.595 \times 10^{14} \text{ s}^{-1}$$

$$\lambda_{\text{max}} = \frac{3 \times 10^8 \text{ m/s}}{4.595 \times 10^{14} \text{ s}^{-1}} = 6.52 \times 10^{-7} \text{ m}$$

$$= 652 \text{ nm}$$

For Ce

→ red light

2.10

$$(a) \quad \lambda = 85 \text{ \AA}$$

$$p = \frac{h}{\lambda} = \frac{6.625 \times 10^{-34} \text{ J} \cdot \text{s}}{85 \times 10^{-10} \text{ m}} = 7.794 \times 10^{-26} \frac{\text{kg} \cdot \text{m}}{\text{s}}$$

$$\text{so } v = \frac{p}{m} = \frac{7.794 \times 10^{-26} \text{ kg} \cdot \text{m/s}}{9.11 \times 10^{-31} \text{ kg}} = 8.555 \times 10^4 \text{ m/s}$$

$$E = \frac{1}{2}mv^2 = \frac{1}{2}(9.11 \times 10^{-31} \text{ kg})(8.555 \times 10^4 \text{ m/s})^2 = 3.334 \times 10^{-21} \text{ J}$$

$$= 0.0208 \text{ eV}$$

(b) electron with $v = 8 \times 10^5 \text{ cm/s}$

$$p = mv = (9.11 \times 10^{-31} \text{ kg})(8 \times 10^5 \text{ m/s}) = \boxed{7.288 \times 10^{-27} \frac{\text{kg} \cdot \text{m}}{\text{s}}}$$

$$E = \frac{p^2}{2m} = \frac{(7.288 \times 10^{-27} \frac{\text{kg} \cdot \text{m}}{\text{s}})^2}{2(9.11 \times 10^{-31} \text{ kg})} = 2.915 \times 10^{-23} \text{ J}$$
$$\boxed{= 1.822 \times 10^{-4} \text{ eV}}$$

$$\lambda = \frac{h}{p} = \frac{6.625 \times 10^{-34} \text{ J} \cdot \text{s}}{7.288 \times 10^{-27} \frac{\text{kg} \cdot \text{m}}{\text{s}}}$$
$$= 9.09 \times 10^{-8} \text{ m} = \boxed{90.9 \text{ nm}} = 909 \text{ \AA}$$

2.11 want $\lambda = 1 \text{ \AA}$ from x-ray

(a) what potential voltage must an electron accelerate through to produce $\lambda = 1 \text{ \AA}$?

we need $E_{\text{photon}} = E_{\text{electron}}$

$$\Rightarrow \frac{hc}{\lambda} = qV \quad \text{both sides in eV}$$

$$\text{so } V = \frac{hc}{\lambda q} = \frac{(4.135 \times 10^{-15} \text{ eV} \cdot \text{s})(3 \times 10^8 \text{ m/s})}{(1 \times 10^{-10} \text{ m})(1.602 \times 10^{-19} \text{ C})}$$

$$= 7.743 \times 10^{22} \frac{\text{eV}}{\text{C}} \quad \text{but volt} = \frac{\text{J}}{\text{C}}$$

$$\text{so } \rightarrow 7.743 \times 10^{22} \frac{\text{eV}}{\text{C}} \cdot \frac{1.6 \times 10^{-19} \text{ J}}{\text{eV}}$$

$$\boxed{= 12,389 \text{ V}}$$

(b) Just before it hits the target it has maximum energy and it is all kinetic

$$E = qV = (1.6 \times 10^{-19} \text{ C})(12,389 \text{ J/C}) = 1.982 \times 10^{-15} \text{ J}$$

$$E = \frac{p^2}{2m} \Rightarrow p = \sqrt{2mE}$$

$$p = \sqrt{2(9.11 \times 10^{-31} \text{ kg})(1.982 \times 10^{-15} \text{ J})} = 6.01 \times 10^{-23} \frac{\text{kg} \cdot \text{m}}{\text{s}}$$

$$\text{so } \lambda = \frac{h}{p} = \frac{6.625 \times 10^{-34} \text{ J}\cdot\text{s}}{6.01 \times 10^{-23} \text{ kg m/s}} = \frac{1.10 \times 10^{-11} \text{ m}}{= 0.11 \text{ \AA}}$$

2.15

$$(a) \Delta E \Delta t \geq \hbar$$

$$\Delta E \leq 0.8 \text{ eV maximum}$$

$$\text{so } \Delta t \text{ minimum} = \frac{\hbar}{\Delta E} = \frac{(\frac{1}{2\pi})(4.135 \times 10^{-15} \text{ eV}\cdot\text{s})}{0.8 \text{ eV}}$$

$$\Delta t = 8.23 \times 10^{-16} \text{ s}$$

$$(b) \Delta p \cdot \Delta x \geq \hbar$$

$$\Delta x \leq 1.5 \text{ \AA}$$

$$\text{so } \Delta p \text{ minimum} = \frac{\hbar}{\Delta x} = \frac{(\frac{1}{2\pi})(6.625 \times 10^{-34} \text{ J}\cdot\text{s})}{(1.5 \times 10^{-10} \text{ m})}$$

$$\Delta p = 7.03 \times 10^{-25} \frac{\text{kg}\cdot\text{m}}{\text{s}}$$

2.17

$$\Psi(x, t) = A \cos\left(\frac{\pi x}{2}\right) e^{-j\omega t} \quad \text{For } -1 \leq x \leq 3$$

$$\text{Determine } A \text{ so } \int_{-1}^3 |\Psi(x, t)|^2 dx = 1$$

$$\begin{aligned} |\Psi(x, t)|^2 &= \Psi(x, t) \cdot \Psi^*(x, t) = A \cos\left(\frac{\pi x}{2}\right) e^{-j\omega t} \cdot A^* \cos\left(\frac{\pi x}{2}\right) e^{j\omega t} \\ &= A \cdot A^* \cos^2\left(\frac{\pi x}{2}\right) = |A|^2 \cos^2\left(\frac{\pi x}{2}\right) \end{aligned}$$

$$\text{so } \int_{-1}^3 |A|^2 \cos^2\left(\frac{\pi x}{2}\right) dx = 1$$

using $\cos^2(x) = \frac{1}{2} + \frac{1}{2} \cos(2x)$

$$|A|^2 \int_{-1}^3 \left[\frac{1}{2} + \frac{1}{2} \cos \pi x \right] dx = 1$$

$$\text{so } |A|^2 = \frac{1}{\frac{1}{2} \left[\int_{-1}^3 dx + \int_{-1}^3 \cos \pi x dx \right]}$$

performing the integral

$$\frac{1}{2} \left[x \Big|_{-1}^3 + \frac{\sin \pi x}{\pi} \Big|_{-1}^3 \right] \rightarrow 0$$

$$= \frac{1}{2} [3 - (-1)] = \frac{4}{2} = 2$$

$$\text{so } |A|^2 = \frac{1}{2} \quad \text{so } |A| = \frac{1}{\sqrt{2}} = \boxed{\frac{\sqrt{2}}{2}}$$

2.20 $\psi(x) = \sqrt{\frac{2}{a}} \cos\left(\frac{\pi x}{a}\right)$ for $-\frac{a}{2} < x < \frac{a}{2}$
 $\psi(x) = 0$ elsewhere

- Find probability of finding the electron between

(a) $0 < x < \frac{a}{4}$

the probability of finding it between $0 < x < \frac{a}{4}$

is $\int_0^{a/4} |\psi(x)|^2 dx$

$$|\psi(x)|^2 = \psi(x) \cdot \psi^*(x) = \frac{2}{a} \cos^2\left(\frac{\pi x}{a}\right)$$

so we have $\frac{2}{a} \int_0^{a/4} \cos^2\left(\frac{\pi x}{a}\right) dx$

using $\cos^2(x) = \frac{1}{2} + \frac{1}{2} \cos(2x)$

$$\frac{2}{a} \int_0^{a/4} \left[\frac{1}{2} + \frac{1}{2} \cos\left(\frac{2\pi x}{a}\right) \right] dx$$

$$\frac{1}{a} \left[x + \frac{a}{2\pi} \sin\left(\frac{2\pi x}{a}\right) \right]_0^{a/4}$$

$$= \frac{1}{a} \left[\frac{a}{4} + \frac{a}{2\pi} \sin\left(\frac{\pi}{2}\right) \right]$$

$$= \frac{1}{4} + \frac{1}{2\pi} = \boxed{0.409}$$

(b) evaluating between $\frac{a}{4} < x < \frac{a}{2}$

$$\frac{1}{a} \left[x + \frac{a}{2\pi} \sin\left(\frac{2\pi x}{a}\right) \right]_{a/4}^{a/2}$$

$$= \frac{1}{a} \left[\frac{a}{4} + \frac{a}{2\pi} \sin(\pi) - \frac{a}{2\pi} \sin\left(\frac{\pi}{2}\right) \right]$$

$$= \frac{1}{a} \left[\frac{a}{4} - \frac{a}{2\pi} \right] = \frac{1}{4} - \frac{1}{2\pi} = \boxed{0.091}$$

(c) evaluating between $-\frac{a}{2} < x < \frac{a}{2}$

* it should be = 1, since the particle must be somewhere described by the wavefunction

verify

$$\frac{1}{a} \left[x + \frac{a}{2\pi} \sin\left(\frac{2\pi x}{a}\right) \right]_{-a/2}^{a/2}$$

$$= \frac{1}{a} \left[a + \frac{a}{2\pi} [\sin(\pi) - \sin(-\pi)] \right] = \boxed{1}$$

2.22 electron in Free space: $\psi(x,t) = A e^{j(kx - \omega t)}$

* This problem is ill-defined!

For a free electron: $k = \sqrt{\frac{2mE}{\hbar^2}}$ From Schrodinger Egn.

$$\text{so } \Rightarrow E = \frac{\hbar^2 k^2}{2m} = \hbar \omega$$

$$\Rightarrow \omega = \frac{\hbar k^2}{2m}$$

$$\text{so if } k = 8 \times 10^8 \text{ m}^{-1} \rightarrow \underline{\omega = 3.7 \times 10^{13} \text{ rad/s}}$$

- ω cannot be $8 \times 10^{12} \text{ rad/s}$ as given in the problem statement!

so I will use $k = 8 \times 10^8 \text{ m}^{-1}$ and $\omega = 3.7 \times 10^{13} \text{ rad/s}$

$$(a) \quad v_{\text{phase}} = \frac{\omega}{k} = \frac{3.7 \times 10^{13} \text{ rad/s}}{8 \times 10^8 \text{ m}^{-1}} = 46,250 \text{ m/s}$$

$$\lambda = \frac{2\pi}{|k|} = \frac{2\pi}{8 \times 10^8 \text{ m}^{-1}} = 7.85 \times 10^{-9} \text{ m} = \boxed{7.85 \text{ nm}}$$

For momentum, we need the particle velocity

↳ For a free electron this is the group velocity

$$v_g = \frac{d\omega}{dk} \quad \text{or you can say } p = mv = \hbar k$$

$$\text{so } v = \frac{\hbar k}{m} = \frac{(1.054 \times 10^{-34} \text{ F.s})(8 \times 10^8 \text{ m}^{-1})}{(9.11 \times 10^{-31} \text{ kg})} = 92,558 \text{ m/s} \\ = 2 v_p$$

some rounding error here

$$\text{to see why } v_g = v \quad \text{Find } \frac{d\omega}{dk} = \frac{d}{dk} \left(\frac{\hbar k^2}{2m} \right) = \frac{\hbar k}{m} = v \quad \checkmark$$

$$\text{so } p = mv = (9.11 \times 10^{-31} \text{ kg})(92,558 \text{ m/s}) = \boxed{8.43 \times 10^{-26} \frac{\text{kg m}}{\text{s}}}$$

$$\text{now } E = \frac{1}{2}mv^2 = (0.5)(9.11 \times 10^{-31} \text{ kg})(92,558 \text{ m/s})^2 = 3.9 \times 10^{-21} \text{ J} \\ = 0.0244 \text{ eV}$$

$$\text{or } E = \hbar \omega = (1.054 \times 10^{-34} \text{ J}\cdot\text{s})(3.7 \times 10^{13} \text{ rad/s}) \\ = 3.9 \times 10^{-21} \text{ J} = 0.0244 \text{ eV} \quad \checkmark$$

(b) Again, the problem is ill-defined

$$\text{if } k = 1.5 \times 10^9 \text{ m}^{-1} \Rightarrow \omega \text{ should be } 1.3 \times 10^{14} \text{ rad/s}$$

$$\text{so } v_{\text{phase}} = \frac{\omega}{k} = \frac{1.3 \times 10^{14} \text{ rad/s}}{-1.5 \times 10^9 \text{ m}^{-1}} = \boxed{-86,667 \text{ m/s}}$$

$$\text{so } v_{\text{group}} = 2 \cdot v_p = 173,333 \text{ m/s} = v$$

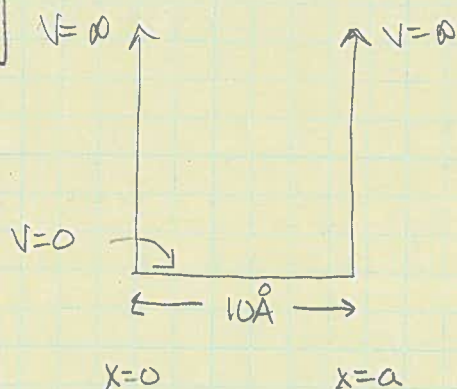
$$\lambda = \frac{2\pi}{|k|} = \frac{2\pi}{1.5 \times 10^9 \text{ m}^{-1}} = \boxed{4.19 \text{ nm}}$$

$$p = m \cdot v = (9.11 \times 10^{-31} \text{ kg})(173,333 \text{ m/s}) = \boxed{-1.58 \times 10^{-25} \frac{\text{kg m}}{\text{s}}}$$

$$E = \frac{1}{2}mv^2 = (0.5)(9.11 \times 10^{-31} \text{ kg})(173,333 \text{ m/s})^2 = 1.369 \times 10^{-20} \text{ J} \\ = \boxed{0.0855 \text{ eV}}$$

* note that - sign means particle moves in -x direction

2.26



(a) The 1st three energy levels are calculated using (2.38)

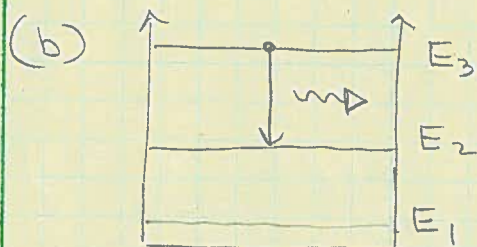
$$E_n = \frac{\hbar^2 n^2 \pi^2}{2ma^2} \quad \text{For } n=1, 2, 3$$

$$E_1 = \frac{(1.054 \times 10^{-34} \text{ F.s})^2 (1)^2 \pi^2}{2(9.11 \times 10^{-31} \text{ kg})(10 \times 10^{-10} \text{ m})^2} = 6.02 \times 10^{-20} \text{ F}$$

$$= 0.376 \text{ eV}$$

$$E_2 = E_1 \cdot 4 = 1.504 \text{ eV}$$

$$E_3 = E_1 \cdot 9 = 3.384 \text{ eV}$$



$$\Delta E = E_3 - E_2 = 3.384 \text{ eV} - 1.504 \text{ eV}$$

$$\Delta E = 1.88 \text{ eV}$$

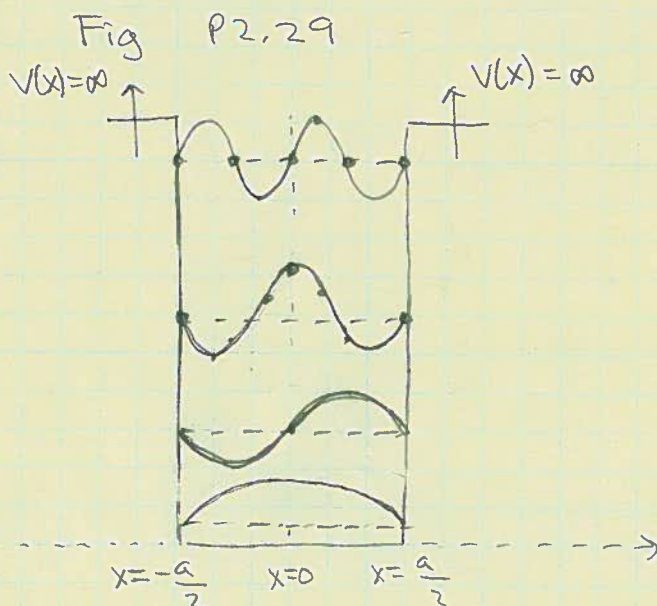
must equal photon energy to conserve energy during the transition

$$\text{so } \Delta E = \frac{hc}{\lambda} \Rightarrow \lambda = \frac{hc}{\Delta E}$$

$$\lambda = \frac{(4.135 \times 10^{-15} \text{ eV.s})(3 \times 10^8 \text{ m/s})}{1.88 \text{ eV}}$$

$$= 6.60 \times 10^{-7} \text{ m} = 660 \text{ nm}$$

2.29



- Derive and sketch wave functions for four lowest energy levels ($n=1, 2, 3, 4$)

- We must consider the new boundary conditions

$$\psi(x = -\frac{a}{2}) = \psi(x = \frac{a}{2}) = 0$$

- The general solution to the time-independent Schrodinger equation for an infinite potential well is

$$\psi(x) = A \cos kx + B \sin kx$$

$$\text{where } k = \sqrt{\frac{2mE}{\hbar^2}}$$

- Applying the boundary condition at $x = \frac{a}{2}$ gives

$$\psi(\frac{a}{2}) = A \cos\left(\frac{ka}{2}\right) + B \sin\left(\frac{ka}{2}\right) = 0$$

$$= A \cos\left(\frac{ka}{2}\right) + B \sin\left(\frac{ka}{2}\right) = 0$$

→ this can be zero if : even multiple of $\frac{\pi}{2}$ (i.e., integer multiple of π)

$$A = 0 \text{ and } \frac{ka}{2} = n\pi \Rightarrow k = \frac{2n\pi}{a} \quad n=1, 2, 3, \dots$$

$$\text{or } B = 0 \text{ and } \frac{ka}{2} = (2n-1)\frac{\pi}{2} \Rightarrow k = \frac{(2n-1)\pi}{a} \quad n=1, 2, 3, \dots$$

↑
odd multiple of $\frac{\pi}{2}$

- so we have even parity and odd parity solutions

$$\psi(x) = A \cos(kx) \quad (\text{even})$$

$$\psi(x) = B \sin(kx) \quad (\text{odd})$$

- and we can say the $\cos(kx)$ solutions are for

$$\frac{ka}{2} = n \frac{\pi}{2} \quad \text{with } n = 1, 3, 5, \dots$$

and the $\sin(kx)$ solutions are for

$$\frac{ka}{2} = n \frac{\pi}{2} \quad \text{with } n = 2, 4, 6, \dots$$

so

$$\psi_1(x) = A_1 \cos\left(\frac{\pi}{a}x\right)$$

$$\psi_2(x) = B_2 \sin\left(\frac{2\pi}{a}x\right)$$

$$\psi_3(x) = A_3 \cos\left(\frac{3\pi}{a}x\right)$$

$$\psi_4(x) = B_4 \sin\left(\frac{4\pi}{a}x\right)$$

$$E_1 = \frac{\hbar^2 \pi^2}{2ma^2}$$

$$E_2 = \frac{4\hbar^2 \pi^2}{2ma^2}$$

$$E_3 = \frac{9\hbar^2 \pi^2}{2ma^2}$$

$$E_4 = \frac{16\hbar^2 \pi^2}{2ma^2}$$

- see beginning for sketches.

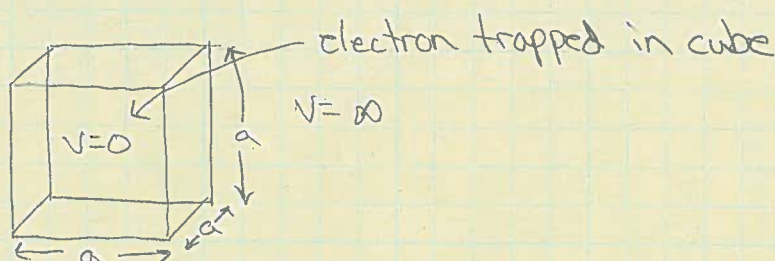
2.30

3D infinite potential well

$$V(x) = 0 \quad \text{for} \quad \begin{aligned} 0 < x < a \\ 0 < y < a \\ 0 < z < a \end{aligned}$$

$$V(x) = \infty \quad \text{elsewhere}$$

* Note: this potential profile is similar to that of the experimentally achievable "quantum dot" or "quantum box"



show $E_{n_x n_y n_z} = \frac{\hbar^2 \pi^2}{2ma^2} (n_x^2 + n_y^2 + n_z^2)$

$$\begin{aligned} n_x &= 1, 2, 3, \dots \\ n_y &= 1, 2, 3, \dots \\ n_z &= 1, 2, 3, \dots \end{aligned}$$

- In 3D, Schrodinger's equation is

$$-\frac{\hbar^2}{2m} \nabla^2 \psi(\vec{r}, t) + V(\vec{r}) \psi(\vec{r}, t) = i\hbar \frac{\partial \psi(\vec{r}, t)}{\partial t}$$

where $\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$ is the Laplacian operator

$$\text{and } \vec{r} = x \hat{e}_x + y \hat{e}_y + z \hat{e}_z$$

↑
unit vector in
z direction

so $\psi(\vec{r}, t) = \psi(x, y, z, t)$

so we have :

$$\begin{aligned} -\frac{\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right] \psi(x, y, z, t) + V(x, y, z) \psi(x, y, z, t) \\ = i\hbar \frac{\partial \psi(x, y, z, t)}{\partial t} \end{aligned}$$

- we can guess a solution to SE of

$$\psi(x, y, z, t) = X(x) Y(y) Z(z) \Phi(t)$$

\nwarrow only depends upon x \uparrow only depends upon y \nearrow only depends upon z \leftarrow only depends upon t

- plugging the solution into SE in 3D we get

$$-\frac{\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right] X(x) Y(y) Z(z) \Phi(t) + V(x, y, z) X(x) Y(y) Z(z) \Phi(t) = j\hbar \frac{\partial}{\partial t} X(x) Y(y) Z(z) \Phi(t)$$

- inside the box $V(x, y, z) = 0$

$$-\frac{\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right] X(x) Y(y) Z(z) \Phi(t) = j\hbar \frac{\partial}{\partial t} X(x) Y(y) Z(z) \Phi(t)$$

$$-\frac{\hbar^2}{2m} \left[Y(z) Z(z) \Phi(t) \frac{\partial^2 X(x)}{\partial x^2} + X(x) Z(z) \Phi(t) \frac{\partial^2 Y(y)}{\partial y^2} + X(x) Y(y) \Phi(t) \frac{\partial^2 Z(z)}{\partial z^2} \right] = j\hbar X(x) Y(y) Z(z) \frac{\partial \Phi(t)}{\partial t}$$

now divide both sides by $X(x) Y(y) Z(z) \Phi(t)$

$$-\frac{\hbar^2}{2m} \left[\frac{1}{X(x)} \frac{\partial^2 X(x)}{\partial x^2} + \frac{1}{Y(y)} \frac{\partial^2 Y(y)}{\partial y^2} + \frac{1}{Z(z)} \frac{\partial^2 Z(z)}{\partial z^2} \right] = j\hbar \frac{1}{\Phi(t)} \frac{\partial \Phi(t)}{\partial t}$$

- each side must equal a constant, call it "E"
- for the quantization and energy levels we care about the time-independent part

$$-\frac{\hbar^2}{2m} \left[\frac{1}{X(x)} \frac{\partial^2 X(x)}{\partial x^2} + \frac{1}{Y(y)} \frac{\partial^2 Y(y)}{\partial y^2} + \frac{1}{Z(z)} \frac{\partial^2 Z(z)}{\partial z^2} \right] = E$$

if we let $k = \sqrt{\frac{2mE}{\hbar^2}}$ as is done in the 1D case described in the book

$$\frac{1}{X(x)} \frac{\partial^2 X(x)}{\partial x^2} + \frac{1}{Y(y)} \frac{\partial^2 Y(y)}{\partial y^2} + \frac{1}{Z(z)} \frac{\partial^2 Z(z)}{\partial z^2} = -k^2$$

we have

- now $k^2 = k_x^2 + k_y^2 + k_z^2$

- we get three equations

$$(1) \quad \frac{\partial^2 X(x)}{\partial x^2} + k_x^2 X(x) = 0$$

$$(2) \quad \frac{\partial^2 Y(y)}{\partial y^2} + k_y^2 Y(y) = 0$$

$$(3) \quad \frac{\partial^2 Z(z)}{\partial z^2} + k_z^2 Z(z) = 0$$

general solution to one of these is

$$X(x) = A_1 \sin(k_x x) + A_2 \cos(k_x x)$$

since $X(0) = 0$ from boundary conditions $\Rightarrow A_2 = 0$

$$\text{so } X(x) = A_1 \sin(k_x x)$$

also since $X(a)$ must be zero we get the condition

$$X(a) = A_1 \sin(k_x a) = 0 \Rightarrow k_x = \frac{n_x \pi}{a} \quad n_x = 1, 2, 3, \dots$$

- similarly for $Y(y)$ and $Z(z)$

$$Y(y) = A_3 \sin(k_y y) \Rightarrow k_y = \frac{n_y \pi}{a} \quad n_y = 1, 2, 3, \dots$$

$$Z(z) = A_5 \sin(k_z z) \Rightarrow k_z = \frac{n_z \pi}{a} \quad n_z = 1, 2, 3, \dots$$

$$\text{so } \Psi(x, y, z) = A \sin(k_x x) \sin(k_y y) \sin(k_z z)$$

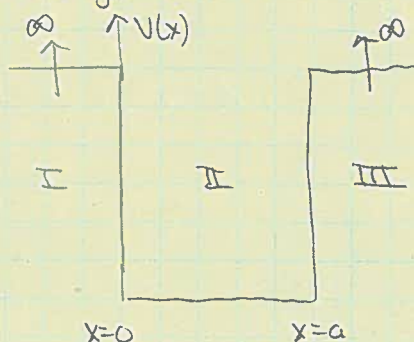
$$\text{and } E = \frac{\hbar^2 k^2}{2m} \quad ; \quad k^2 = k_x^2 + k_y^2 + k_z^2$$

$$\Rightarrow E = \frac{\hbar^2 (k_x^2 + k_y^2 + k_z^2)}{2m} = \frac{\hbar^2 \pi^2}{2m a^2} (n_x^2 + n_y^2 + n_z^2)$$

$$E = \frac{\hbar^2 \pi^2}{2m a^2} (n_x^2 + n_y^2 + n_z^2) \quad \begin{matrix} n_x = 1, 2, 3, \dots \\ n_y = 1, 2, 3, \dots \\ n_z = 1, 2, 3, \dots \end{matrix}$$

2.32

Consider a proton in a 1D infinite potential well in Fig. 2.6



- (a) The fact that we are considering a proton instead of an electron only changes the mass, so the derivation is the same as the one given in section 2.3.2 in the book

$$E_n = \frac{\hbar^2 n^2 \pi^2}{2ma^2} \quad n=1, 2, 3, \dots$$

* m = mass of proton

- (b) - lowest possible energy is when $n=1$
 - next highest is for $n=2$

$$\text{so we have } \Delta E = \frac{4\hbar^2 \pi^2}{2ma^2} - \frac{\hbar^2 \pi^2}{2ma^2} = \frac{3\hbar^2 \pi^2}{2ma^2}$$

For (i) $a = 4 \text{ \AA}$ we have

$$\Delta E = \frac{3(1.054 \times 10^{-34} \text{ J}\cdot\text{s})^2 \pi^2}{2(1.67 \times 10^{-27} \text{ kg})(4 \times 10^{-10} \text{ m})^2} = 6.155 \times 10^{-22} \text{ J}$$

$$\boxed{= 0.00385 \text{ eV}}$$

For (ii) $a = 0.5 \text{ cm}$

$$\Delta E = \frac{3(1.054 \times 10^{-34} \text{ J}\cdot\text{s})^2 \pi^2}{2(1.67 \times 10^{-27} \text{ kg})(0.5 \times 10^{-2} \text{ m})^2} = 3.939 \times 10^{-36} \text{ J}$$

$$\boxed{= 2.46 \times 10^{-17} \text{ eV}}$$

* so as the well gets wider, the energy difference goes down, that is there is almost a continuum of energies for a significantly wide well!