

# iit Jee 2010 Paper 1 Offline 84Questions

## Question 001

### Numerical

#### QUESTION

Based on VSEPR theory, the number of 90 degree F-Br-F angles in  $\text{BrF}_5$  is

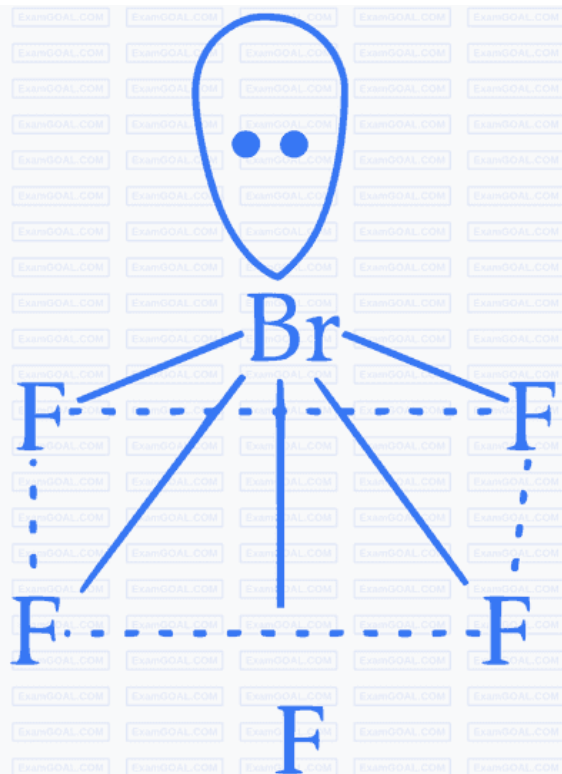
#### SOURCE

Chemistry • chemical-bonding-and-molecular-structure

#### EXPLANATION

According to VSEPR,  $\text{BrF}_5$  has square pyramidal structure with axial plane containing a lone pair and fluorine. The other four fluorine are arranged in square planar configuration around central metal atom. Thus,  $\text{BrF}_5$  assumes square pyramidal shape where the valence electron pairs surrounding an atom tend to repel each other and will, therefore, adopt an arrangement that minimises this repulsion, thus, determining the molecule's geometry. All four planar bonds ( $\text{F} - \text{Br} - \text{F}$ ) will reduce from  $90^\circ$  to  $84.8^\circ$  after lone pair - bond pair repulsion.

So, there are no 90 -degree  $\text{F} - \text{Br} - \text{F}$  angles in  $\text{BrF}_5$ .



Question 002 **MCQ**

**QUESTION**

The species which by definition has **ZERO** standard molar enthalpy of formation at 298 K is

- A**  $\text{Br}_2 g$
- B**  $\text{Cl}_2 g$
- C**  $\text{H}_2\text{O} g$
- D**  $\text{CH}_4 g$

### CORRECT OPTION

**B**  $\text{Cl}_2 g$

### SOURCE

Chemistry • thermodynamics

### EXPLANATION

The standard molar enthalpy of formation, denoted as

$$\Delta H_f^\circ$$

, is defined as the change in enthalpy when one mole of a compound is formed from its constituent elements in their standard states under standard conditions *298 K and 1 atm pressure*. By definition, the standard molar enthalpy of formation of a pure element in its most stable form at 298 K is zero.

Looking at the options:

- **$\text{Br}_2 g$**  - Bromine's standard state at room temperature is liquid ( $\text{Br}_2 l$ ), not gas.
- **$\text{Cl}_2 g$**  - Chlorine's standard state at room temperature is a gas, which means the standard molar enthalpy of formation for  $\text{Cl}_2 g$  is indeed zero.
- **$\text{H}_2\text{O} g$**  - Water in gaseous form is not a basic element but a compound of hydrogen and oxygen, so its enthalpy of formation is not zero.
- **$\text{CH}_4 g$**  - Methane, represented here, is also a compound composed of carbon and hydrogen, thus its enthalpy of formation is also not zero.

Therefore, the correct option is **Option B**,  $\text{Cl}_2 g$ , as chlorine gas is a diatomic molecule and an element in its standard state at 298 K, and as such, its standard molar enthalpy of formation is zero.

### QUESTION

Among the following, the intensive property is *properties are*

A

molar conductivity

B

electromotive force

C

resistance

D

heat capacity

### CORRECT OPTION

A

molar conductivity

### SOURCE

Chemistry • thermodynamics

### EXPLANATION

An intensive property is a property that is independent of the amount of mass or size of the system. In contrast, an extensive property is dependent on the size or amount of mass in the system. Let's analyze each option based on this definition:

- **Molar Conductivity:** Molar conductivity, denoted as

$$\Lambda_m$$

, is a measure of the ionic conductivity of an electrolyte solution divided by the concentration of the electrolyte. Since it is normalized by the amount of

substance, it does not depend on the total amount of the substance present in the solution. Thus, molar conductivity is an intensive property.

- **Electromotive Force  $EMF$ :** EMF is the voltage generated by a battery or by the magnetic force according to Faraday's Law of electromagnetic induction. It refers to a potential difference and does not depend on the quantity of material or size of the battery. Therefore, EMF is considered an intensive property.
- **Resistance:** Resistance, denoted by

$$R$$

, measures how much a material opposes the flow of electric current. It depends on the material's length, cross-sectional area, and resistivity. For instance, a longer wire has greater resistance. Therefore, resistance is an extensive property.

- **Heat Capacity:** Heat capacity is the amount of heat required to change the temperature of a substance by a certain temperature interval. It is dependent on the amount of substance present, and therefore, it is an extensive property. The specific heat capacity, however, is intensive since it is the heat capacity per unit mass.

From the options provided:

- Option A *MolarConductivity* is intensive.
- Option B *ElectromotiveForce* is intensive.
- Option C *Resistance* is extensive.
- Option D *HeatCapacity* is extensive.

Therefore, the intensive properties among the given options are Option A *MolarConductivity* and Option B *ElectromotiveForce*.

#### Question 004 MCQ

##### QUESTION

Aqueous solution of  $\text{HNO}_3$ ,  $\text{KOH}$  and  $\text{CH}_3\text{COOH}$  and  $\text{CH}_3\text{COONa}$  of identical concentrations are provided. The pairs of solutions which form a buffer upon

mixing is *are*

A

$\text{HNO}_3$  and  $\text{CH}_3\text{COOH}$

B

$\text{KOH}$  and  $\text{CH}_3\text{COONa}$

C

$\text{HNO}_3$  and  $\text{CH}_3\text{COONa}$

D

$\text{CH}_3\text{COOH}$  and  $\text{CH}_3\text{COONa}$

**CORRECT OPTION**

C

$\text{HNO}_3$  and  $\text{CH}_3\text{COONa}$

**SOURCE**

Chemistry • ionic-equilibrium

**Question 005**

Numerical

**QUESTION**

Amongst the following the total number of compounds whose aqueous solution turns red litmus paper blue is

$\text{KCN}$ ,  $\text{K}_2\text{SO}_4$ ,  $(\text{NH}_4)_2\text{C}_2\text{O}_4$ ,  $\text{NaCl}$ ,  $\text{Zn}(\text{NO}_3)_2$ ,  $\text{FeCl}_3$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{NH}_4\text{NO}_3$  and  $\text{LiCN}$

**SOURCE**

Chemistry • ionic-equilibrium

**Question 006**

MCQ

**QUESTION**

The reagent *s* used for softening the temporary hardness of water is *are*

**A**  $\text{Ca}_3(\text{PO}_4)_2$

**B**  $\text{CaOH}_2$

**C**  $\text{Na}_2\text{CO}_3$

**D**  $\text{NaOCl}$

**CORRECT OPTION**

**B**  $\text{CaOH}_2$

**SOURCE**

Chemistry • hydrogen

**Question 007**

Numerical

**QUESTION**

The total number of cyclic isomers possible for a hydrocarbon with the molecular formula  $\text{C}_4\text{H}_6$  is

**SOURCE**

## Question 008 MCQ

## QUESTION

The synthesis of 3-octyne is achieved by adding a bromoalkane into a mixture of sodium amide and alkyne

A

 $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$  and  $\text{CH}_2\text{CH}_2\text{C}$  $\equiv$  $\text{CH}$ 

B

 $\text{BrCH}_2\text{CH}_2\text{CH}_3$  and  $\text{CH}_3\text{CH}_2\text{CH}_2\text{C}$  $\equiv$  $\text{CH}$ 

C

 $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$  and  $\text{CH}_3\text{C}$  $\equiv$  $\text{CH}$ 

D

 $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$  and  $\text{CH}_3\text{CH}_2\text{C}$  $\equiv$  $\text{CH}$ 

## CORRECT OPTION

D

 $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$  and  $\text{CH}_3\text{CH}_2\text{C}$  $\equiv$



**SOURCE**

Chemistry • hydrocarbons

**EXPLANATION**

To get appropriate bromoalkane an alkyne that were involved in the synthesis, we have to cleave the molecule and we know that sodium amide is involved, so the synthesis of 3-octyne using the alkyne and bromoalkane was  $S_N2$  reaction. And as the sodium amide is involved, the alkyne involved must be terminal then only it could abstract the acidic hydrogen 3-octyne is a molecule that will have eight carbon atoms in their parent chain since the prefix-oct represents 8 and as in the name it is given that 3 octyne, ' 3 ' represents the position of triple bond. Thus suffix-one represents triple bond.

So, in the molecule triple bond will be at the third carbon. Structure of 3-octyne :

Now, cleave the molecule between  $C_2 - C_3$  or  $C_4 - C_5$ .

Now, see the products formed if cleavage takes place between  $C_2-C_3$ , the reactants will be

If cleavage takes place between  $C_4-C_5$ , the reactants will be :

**Then correct option is D.**

**Question 009**

MCQ

**QUESTION**

The bond energy (in  $\text{kcal mol}^{-1}$ ) of a C-C single bond is approximately

A 1

B 10

C 100

D 1000

**CORRECT OPTION**

C 100

**SOURCE**

Chemistry • thermodynamics

**EXPLANATION**

The bond energy of a C-C *carbon – carbon* single bond is a measure of the strength of that bond, or how much energy is required to break one mole of such bonds in a gaseous substance. This value is a fundamental concept in chemistry, particularly in the study of organic molecules and reactions.

For a C-C single bond, the bond energy is typically around 83 kcal/mol *kcalpermole*. This places the value closest to option C amongst the given choices.

Therefore, the correct answer is:

**Option C:** 100

### QUESTION

The concentration of potassium ions inside a biological cell is at least twenty times higher than the outside. The resulting potential difference across the cell is important in several processes such as transmission of nerve impulses and maintaining the ion balance. A simple model for such a concentration cell involving a metal M is :



For the above electrolytic cell the magnitude of the cell potential  $|E_{\text{cell}}| = 70 \text{ mV}$ .

For the above cell :

**A**  $E_{\text{cell}} < 0 ;$   $\Delta G > 0$

**B**  $E_{\text{cell}} > 0 ;$   $\Delta G < 0$

**C**  $E_{\text{cell}} < 0 ;$   $\Delta G^{\circ} > 0$

**D**  $E_{\text{cell}} > 0 ;$   $\Delta G^{\circ} > 0$

### CORRECT OPTION

**B**  $E_{\text{cell}} > 0 ;$   $\Delta G < 0$

## SOURCE

Chemistry • electrochemistry

## EXPLANATION

To determine the right option for the given concentration cell, we must first understand the concepts of cell potential, Gibbs free energy, and the relationship between these quantities.

The cell potential,

$$E_{cell}$$

, for a concentration cell like the one described is directly related to the concentrations of the ions on both sides of the cell. It's calculated using the Nernst equation:

$$E_{cell} = E^{\circ} - \frac{RT}{nF} \ln \frac{[M^{+}](\text{low concentration})}{[M^{+}](\text{high concentration})}$$

Since the given cell involves the same metal on both sides in its standard state, the standard cell potential,

$$E^{\circ}$$

, is zero. Therefore, the equation simplifies to:

$$E_{cell} = -\frac{RT}{nF} \ln \frac{0.05}{1}$$

Given that

$$R$$

*the ideal gas constant* is approximately 8.314 J/mol·K,

$$F$$

*the Faraday constant* is about 96485 C/mol,

$$T$$

is the temperature in Kelvin *assuming standard temperature of 298 K*, and

$$n$$

the number of moles of electron transferred per mole of reaction is 1, the equation further simplifies to:

$$E_{cell} = -\frac{8.314 \times 298}{96485} \ln \left( \frac{0.05}{1} \right)$$

Calculate the ln term:

$$\ln \left( \frac{0.05}{1} \right) = \ln(0.05) \approx -2.9957$$

Then the equation is:

$$E_{cell} = -\frac{2476.422}{96485} \times -2.9957 \approx 0.077 \text{ volts} = 77 \text{ mV}$$

Here, the calculated

$$E_{cell} > 0$$

, which matches the given cell potential of 70 mV. It differs slightly due to rounding and exact values used in constants. This cell potential being positive indicates spontaneous reaction tendency to go from high to low concentration.

We now apply this to the relationship between the cell potential and Gibbs free energy, which is given by:

$$\Delta G = -nFE_{cell}$$

Since

$$E_{cell} > 0$$

,

$$\Delta G < 0$$

which indicates that the process is thermodynamically favorable *spontaneous*.

Looking at the provided options:

- Option A:  $E_{cell} < 0$  ;

$$\Delta G > 0$$

*Incorrect; as  $E_{\text{cell}}$  is positive and  $\Delta G$  is negative*

- Option B:  $E_{\text{cell}} > 0$  ;

$$\Delta G < 0$$

*Correct; matches our calculation*

- Option C:  $E_{\text{cell}} < 0$  ;

$$\Delta G^{\circ} > 0$$

*Incorrect*

- Option D:  $E_{\text{cell}} > 0$  ;

$$\Delta G^{\circ} > 0$$

*Incorrect,  $\Delta G^{\circ}$  not directly relevant here, but centralized on the star*

Thus, the correct answer is **Option B**.

### Question 011

MCQ

#### QUESTION

The concentration of potassium ions inside a biological cell is at least twenty times higher than the outside. The resulting potential difference across the cell is important in several processes such as transmission of nerve impulses and maintaining the ion balance. A simple model for such a concentration cell involving a metal M is :



For the above electrolytic cell the magnitude of the cell potential  $|E_{\text{cell}}| = 70 \text{ mV}$ .

If the 0.05 molar solution of  $M^{+}$  is replaced by a 0.0025 molar  $M^{+}$  solution, then the magnitude of the cell potential would be :

**A** 35 mV

**B** 70 mV

**C** 140 mV

**D** 700 mV

#### CORRECT OPTION

**C** 140 mV

#### SOURCE

Chemistry • electrochemistry

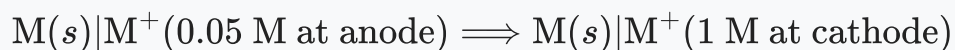
#### EXPLANATION

To understand how the concentration of ions affects the cell potential, we will first use the Nernst equation. The Nernst equation gives us a way to calculate the potential of a cell under non-standard conditions and is represented as follows:

$$E = E^0 - \frac{RT}{nF} \ln \frac{a_{\text{Red}}}{a_{\text{Ox}}}$$

In our case, we're analyzing a concentration cell where the metal M is the same in both the anode and the cathode but with different concentrations of  $M^+$ . Here, the standard potential  $E^0$  is zero because the same substance is used as both the anode and cathode.

The cell reaction becomes:



Since  $E^0 = 0$ , the Nernst equation simplifies to:

$$E = -\frac{RT}{nF} \ln \frac{[M^+ (\text{cathode})]}{[M^+ (\text{anode})]}$$

Which turns into:

$$E = -\frac{RT}{nF} \ln \frac{1}{0.05},$$

and we can further simplify using  $\ln\left(\frac{1}{0.05}\right) = -\ln(0.05)$ . Assume the reaction involves the transfer of 1 mole of electrons  $n = 1$ , the value of F

*Faraday constant* is approximately 96485 C/mol, and the temperature T is 298K *room temperature*. The gas constant R is 8.314 J/mol · K. Plugging in these values:

$$E \approx -\frac{(8.314 \text{ J/mol}\cdot\text{K})(298 \text{ K})}{(1)(96485 \text{ C/mol})} \ln(0.05)$$

This equation can be used to find the potential in volts when the concentration at anode was 0.05 M and was resulting in 70 mV.

Now, switching the anode concentration to 0.0025 M, we reapply the Nernst equation:

$$E = -\frac{RT}{nF} \ln \frac{1}{0.0025}$$

We can see the ratio of concentrations has changed, moving from 1/0.05 to 1/0.0025. Thus, the concentration difference across the membrane has increased, which should increase the voltage according to the Nernst equation. Specifically, the ratio changed by a factor of  $0.05/0.0025 = 20$  times.

Given that  $\ln\left(\frac{1}{x}\right)$  is proportional to the voltage, when the ratio of concentration changes by twenty-fold, and because  $\ln(0.0025) = -\ln(400)$  which is twice the  $\ln(20)$ , the potential will double. Since the initial 70 mV doubles, the new cell potential will be:

$$E \approx 2 \times 70 \text{ mV} = 140 \text{ mV}.$$

Therefore, the correct answer is:

**Option C: 140 mV**



### QUESTION

The concentration of R in the reaction R



P was measured as a function of time and the following data is obtained

$R$ molar	1.0	0.75	0.40	0.10
$t \text{ min.}$	0.0	0.05	0.12	0.18

The order of reaction is

### SOURCE

Chemistry • chemical-kinetics-and-nuclear-chemistry

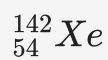
## Question 013 Numerical

### QUESTION

The number of neutrons emitted when



undergoes controlled nuclear fission to



and



is

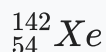
### SOURCE

**EXPLANATION**

To determine the number of neutrons emitted during the nuclear fission of



resulting in the products



and



, we need to ensure the conservation of mass number and atomic number.

The mass number  $A$  and atomic number  $Z$  have to be conserved. This means that the sum of the mass numbers and atomic numbers of the products *including any neutron emitted* must equal those of the uranium nucleus undergoing fission.

Let's start by writing down the conservation of mass number and atomic number:

1. Conservation of Mass Number:

$$235 = 142 + 90 + n \times 1$$

Here,

$$n$$

represents the number of neutrons released. We can now calculate

$$n$$

:

$$n = 235 - (142 + 90) = 235 - 232 = 3$$

1. Conservation of Atomic Number:

$$92 = 54 + 38 + 0 \times n$$

Neutrons do not contribute to the atomic number as they have no charge.

This calculation shows that three neutrons are needed to satisfy the conservation of mass number. The atomic number conservation also coincides, as neutrons do not alter it. Therefore, the number of neutrons emitted during this fission process is

3

### Question 014 Numerical

#### QUESTION

A student performs a titration with different burettes and finds titre values of 25.2 mL, 25.25 mL, and 25.0 mL. The number of significant figures in the average titre value is

#### SOURCE

Chemistry • some-basic-concepts-of-chemistry

#### EXPLANATION

To find the average titre value, first add up the three measurements provided and then divide by the number of measurements.

The sum of the measurements is:

$$25.2 \text{ mL} + 25.25 \text{ mL} + 25.0 \text{ mL} = 75.45 \text{ mL}$$

Since there are three measurements, divide this sum by 3 to calculate the average:

$$\text{Average titre value} = \frac{75.45 \text{ mL}}{3} = 25.15 \text{ mL}$$

When reporting the average, we must consider the significant figures of the original measurements. The number of significant figures is determined by the least precise measurement, which in this case is 25.0 mL with three significant

figures. Therefore, we should report the average value to three significant figures as well.

The average value of 25.15 mL has four significant figures, so we need to round it to three significant figures. However, this is slightly tricky since 25.15 already appears to be rounded to four significant figures. We should consult the original measurements to decide on the best course of action.

Looking at the individual measurements 25.2, 25.25, and 25.0, we should consider the lowest decimal place which they all have in common, which is the first decimal place. The third measurement has no second decimal place, indicating its level of precision. Thus, the number of significant figures for the average titre value should be in line with this level of precision. Since the average calculated is 25.15, when we adjust to the first decimal place for consistent significant figures, the average is 25.1 mL with three significant figures.

Corrected Average titre value = 25.1 mL

Therefore, the number of significant figures in the average titre value is three: 25.1 mL.

### Question 015 MCQ

#### QUESTION

The correct statement about the following disaccharide is :

Ring  $\alpha$  is pyranose with

A

$\alpha$

-glycosidic link.

Ring  $\alpha$  is furanose with

**B** $\alpha$ 

-glycosidic link.

**C**

Ring *b* is furanose with

 $\alpha$ 

-glycosidic link.

**D**

Ring *b* is pyranose with

 $\alpha$ 

-glycosidic link.

**CORRECT OPTION****A**

Ring *a* is pyranose with

 $\alpha$ 

-glycosidic link.

**SOURCE**

Chemistry • biomolecules

**EXPLANATION**

Pyranose structure is a six-membered heterocyclic ring containing five carbon atoms and one oxygen atom. A glycosidic bond is formed between the hemiacetal group of a saccharide and the hydroxyl group of some organic compound such as an alcohol. C-1 of ring *a* has an

 $\alpha$ 

-glycosidic linkage.

Ring *b* has 5-members and hence is a furanose.

**QUESTION**

In the reaction

the products are :

A

B

C

D

**CORRECT OPTION**

D

**SOURCE**

Chemistry • alcohols-phenols-and-ethers

**EXPLANATION**

Alkyl aryl ethers are cleaved at the alkyl-oxygen bond due to stronger aryl-oxygen bond. The reaction yields phenol and alkyl halide. Thus,

The reaction is

## QUESTION

Plots showing the variation of the rate constant  $k$  with temperature  $T$  are given below. The point that follows Arrhenius equation is

A

B

C

D

## CORRECT OPTION

A

## SOURCE

Chemistry • chemical-kinetics-and-nuclear-chemistry

## EXPLANATION

According to the Arrhenius equation

$$k = Ae^{-E_a/RT}$$

where,  $k$  = rate constant,  $E_a$  = activation energy and  $T$  = temperature

From the expression, we have that as temperature increases, the rate constant  $k$  increases exponentially.

### Question 018 MCQ

#### QUESTION

The correct structure of ethylenediaminetetraacetic acid *EDTA* is

A

B

C

D

#### CORRECT OPTION

C

#### SOURCE

Chemistry • coordination-compounds

#### EXPLANATION

EDTA *ethylenediaminetetracetic acid* is a hexadentate ligand. It is also called the amino poly-carboxylic acid.



Its structure consists of the ethane group which is attached to tertiary amine groups at the terminal positions forming an ethylene diamine. Tertiary amine group is situated by the acetic acid groups ( $-\text{CH}_2\text{COOH}$ ).

Question 019 MCQ

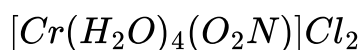
QUESTION

The ionisation isomer of

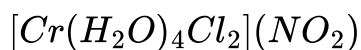


is

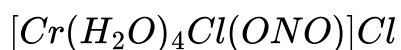
A



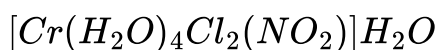
B



C

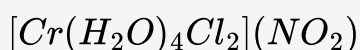


D



CORRECT OPTION

B



SOURCE

**EXPLANATION**

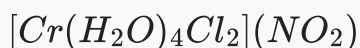
The ionisation isomers give different ions in solution. In the complex given in option *B*, Cl

is replaced by NO

in ionisation sphere.



and



have different ions inside and outside the coordinate sphere and they are isomers. Therefore, they are ionisation isomers.

**Question 020** MCQ**QUESTION**

In the Newman projection for 2,2-dimethylbutane, X and Y can, respectively, be

**A** H and H

H and C

**A**

H

5

C

2

**C**

H

5

and H

CH

3

**D**

and CH

3

#### CORRECT OPTION

H and C

2

**B**

H

5

#### SOURCE

Chemistry • basics-of-organic-chemistry

#### EXPLANATION

In the structure of 2,2-dimethylbutane:

On C

2

C

3

bond axis X = CH

3

, Y = CH

3

.

On C

1

C

2

bond axis X = H, Y = C

2

H

5

.

### Question 021 MCQ

#### QUESTION

In the reaction

The intermediate *s* is *are*

A

B

C

D

#### CORRECT OPTION

A

#### SOURCE

Chemistry • alcohols-phenols-and-ethers

#### EXPLANATION

In presence of base, phenol forms phenoxide ion which is

*ortho*

-

*para*

directing because in the resonating structures, the electron density increases at ortho and para positions. The phenoxide ion further undergoes electrophilic substitution with Br

2

at

*ortho*

and

*para*

positions.

### Question 022

MCQ

#### QUESTION

Partial roasting of chalcopyrite produces

A

Cu

2

S and FeO

B

Cu

2

O and FeO

C

CuS and Fe

2

O

3

D

Cu

2

O and Fe

2

O

3

### CORRECT OPTION

Cu

A

2

S and FeO

### SOURCE

Chemistry • isolation-of-elements

### EXPLANATION

Partial roasting of chalcopyrite gives

$2\text{CuFeS}$

2

+ O

2

→

Cu

2

S + 2FeS + SO

2

$2\text{CuFeS}$

2

+ 4O

2

→

Cu

2



2

Since iron is more electropositive than copper, so its sulphide is oxidised in preference. Cu

2

S remains mostly unaffected and the small amount of Cu

2

O formed on oxidation, reacts with FeS to give back Cu

2

S.

2Cu

2

S + 3O

2

→

2Cu

2

O + 2SO

2

Cu

2

O + FeS

→

Cu

2

S + FeO



**Question 023**

MCQ

**QUESTION**

Iron is removed from chalcoppyrite as

**A** FeO

**B** FeS

**C**  $\begin{matrix} \text{Fe} & & 2 \\ \text{O} & & 3 \end{matrix}$

**D** FeSiO  $\begin{matrix} 3 \end{matrix}$

**CORRECT OPTION**

**D** FeSiO  $\begin{matrix} 3 \end{matrix}$

**SOURCE**

Chemistry • isolation-of-elements

**EXPLANATION**

Iron obtained in form of its oxide is removed by slag formation with SiO



### Question 024 MCQ

#### QUESTION

In self-reduction, the reducing species is

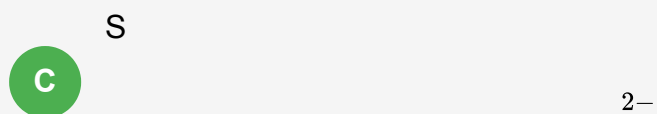
**A** S

**B**  $\text{O}^{2-}$

**C**  $\text{S}^{2-}$

**D**  $\text{SO}_2$

### CORRECT OPTION

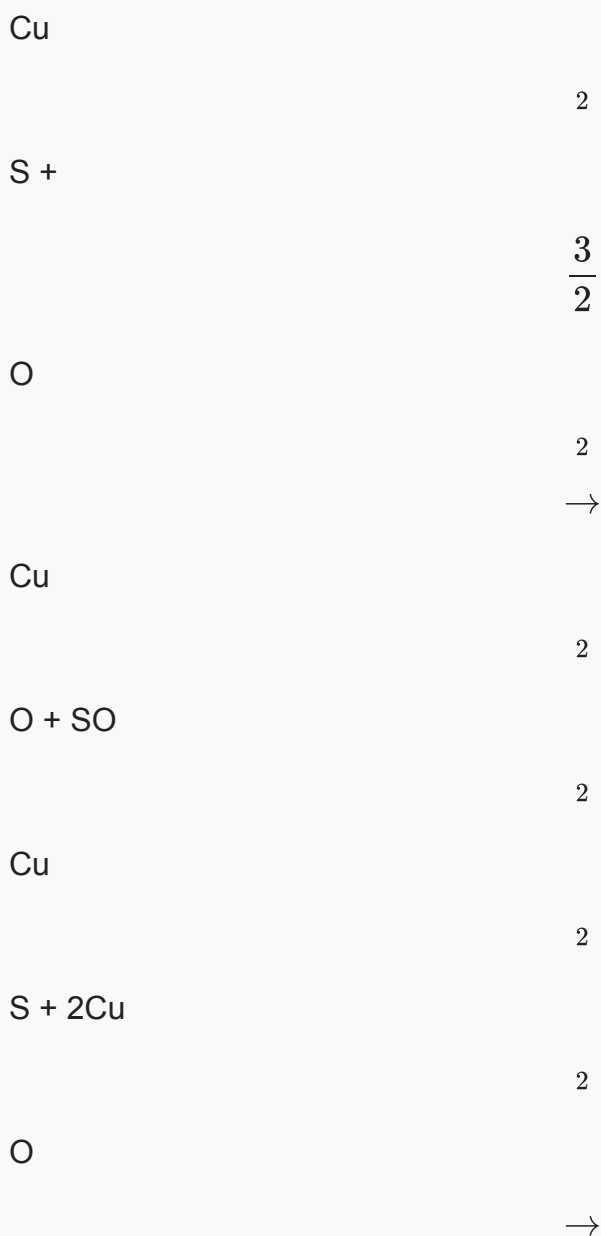


### SOURCE

Chemistry • isolation-of-elements

### EXPLANATION

In the self-reduction step,



$6\text{Cu(l)}$   
Blister copper

+ SO

2

In the reaction, the oxidation number of sulphur  $\text{S}^{2-}$  changes to  $\text{S}^{4+}$  so, S

2-

is the reducing agent.

### Question 025 Numerical

#### QUESTION

The total number of basic groups in the following form of lysine is

#### SOURCE

Chemistry • biomolecules

#### EXPLANATION

There are two basic groups in lysine

### Question 026 Numerical

#### QUESTION

In the scheme given below, the total number of intra molecular aldol condensation products formed from Y is \_\_\_\_\_.

#### SOURCE

Chemistry • aldehydes-ketones-and-carboxylic-acids

#### EXPLANATION

There is only one product formed in intramolecular aldol condensation, which is explained as follows:

### Question 027 Numerical

#### QUESTION

Amongst the following, the total number of compounds soluble in aqueous NaOH is \_\_\_\_\_.

#### SOURCE

Chemistry • aldehydes-ketones-and-carboxylic-acids

#### EXPLANATION

Aromatic alcohols and carboxylic acids forms salt with NaOH, will dissolve in aqueous NaOH :

Benzylic alcohol is less acidic than water and hence does dissolve in aqueous NaOH.

### Question 028

Numerical

#### QUESTION

The value of

$n$

in the molecular formula



is \_\_\_\_\_.

#### SOURCE

Chemistry • p-block-elements

#### EXPLANATION

In the given molecular formula



, according to charge balance in a molecule, we get

$$2n + 2(+3) + 6(+4) - 18(2) = 0$$

$$\Rightarrow n = 3$$

So, the formula is



**Question 029****Numerical****QUESTION**

The number of values of

$$\theta$$

in the interval,

$$\left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$$

such that

$$\theta \neq \frac{n\pi}{5}$$

for

$$n = 0, \pm 1, \pm 2$$

and

$$\tan \theta = \cot 5\theta$$

as well as

$$\sin 2\theta = \cos 4\theta$$

is

**SOURCE**

Mathematics • trigonometric-functions-and-equations

**EXPLANATION**

Given,

$$\tan \theta = \cot 5\theta$$

$$\Rightarrow \tan \theta = \tan \left( \frac{\pi}{2} - 5\theta \right)$$

$$\Rightarrow \frac{\pi}{2} - 5\theta = n\pi + \theta$$

$$\Rightarrow 6\theta = \frac{\pi}{2} - \frac{n\pi}{6}$$

$$\Rightarrow \theta = \frac{\pi}{12} - \frac{n\pi}{6}$$

Also

$$\cos 4\theta = \sin 2\theta = \cos \left( \frac{\pi}{2} - 2\theta \right)$$

$$\Rightarrow 4\theta = 2n\pi \pm \left( \frac{\pi}{2} - 2\theta \right)$$

Taking positive

$$6\theta = 2n\pi + \frac{\pi}{2} \Rightarrow \theta = \frac{n\pi}{3} + \frac{\pi}{12}$$

Taking negative

$$2\theta = 2n\pi - \frac{\pi}{2} \Rightarrow \theta = n\pi - \frac{\pi}{4}$$

Above values of

$$\theta$$

suggests that there are only 3 common solutions.

### Question 030

Numerical

#### QUESTION

The number of all possible values of

$$\theta$$



where

$$0 < \theta < \pi,$$

for which the system of equations

$$(y + z) \cos 3\theta = (xyz) \sin 3\theta$$

$$x \sin 3\theta = \frac{2 \cos 3\theta}{y} + \frac{2 \sin 3\theta}{z}$$

$$(xyz) \sin 3\theta = (y + 2z) \cos 3\theta + y \sin 3\theta$$

\$

have a solution

$$(x_0, y_0, z_0)$$

with

$$y_0 z_0 \neq 0,$$

is

#### SOURCE

Mathematics • trigonometric-functions-and-equations

#### EXPLANATION

View the equation in  $xyz$ ,  $y$  and  $t$ .

We have,

$$(xyz) \sin 3\theta - y \cos 3\theta - z \cos 3\theta = 0$$

$$(xyz) \sin 3\theta - 2y \sin 3\theta - 2z \cos 3\theta = 0$$

$$(xyz) \sin 3\theta - y(\cos 3\theta + \sin 3\theta) - 2z \cos 3\theta = 0$$

$$xyz \neq 0$$

Hence, the equation has non-trivial solution which gives

$$\begin{vmatrix} \sin 3\theta & -\cos 3\theta & -\cos 3\theta \\ \sin 3\theta & -2\sin 3\theta & -2\cos 3\theta \\ \sin 3\theta & -(\cos 3\theta + \sin 3\theta) & -2\cos 3\theta \end{vmatrix} = 0$$

$$\Rightarrow \sin 3\theta \cos 3\theta (\sin 3\theta - \cos 3\theta) = 0$$

$$\Rightarrow \sin 3\theta = 0$$

then

$$xyz = 0$$

*not possible*

$$\cos 3\theta = 0$$

not possible

$$\sin 3\theta = \cos 3\theta \Rightarrow \tan 3\theta = 1$$

$$3\theta = n\pi + \frac{\pi}{4}, n \in \mathbb{Z}$$

$$\theta = \frac{n\pi}{3} + \frac{\pi}{12}$$

;

$$\theta = \frac{\pi}{12}, \frac{5\pi}{12}, \frac{9\pi}{12}$$

Thus there are 3 solutions.

### Question 031 Numerical

#### QUESTION

The maximum value of the expression

$$\frac{1}{\sin^2 \theta + 3 \sin \theta \cos \theta + 5 \cos^2 \theta}$$

is

### SOURCE

Mathematics • trigonometric-functions-and-equations

### EXPLANATION

Let

$$f(\theta) = \frac{1}{\sin^2 \theta + 3 \sin \theta \cos \theta + 5 \cos^2 \theta}$$

Again let

$$\begin{aligned} g(\theta) &= \sin^2 \theta + 3 \sin \theta \cos \theta + 5 \cos^2 \theta \\ &= \frac{1 - \cos 2\theta}{2} + 5 \left( \frac{1 + \cos 2\theta}{2} \right) + \frac{3}{2} \sin 2\theta \\ &= 3 + 2 \cos 2\theta + \frac{3}{2} \sin 2\theta \end{aligned}$$

$\therefore$

$$g(\theta)_{\min} = 3 - \sqrt{4 + \frac{9}{4}} = 3 - \frac{5}{2} = \frac{1}{2}$$

$\therefore$

$$f(\theta) = \frac{1}{g(\theta)_{\min}} = 2$$

### Question 032 MCQ

### QUESTION

Let

$$z_1$$

and

$$z_2$$

be two distinct complex number and let  $z = 1 - t$

$$z_1$$

+ t

$$z_2$$

for some real number t with  $0 < t < 1$ . If  $\text{Arg } w$  denote the principal argument of a non-zero complex number w, then

A

$$|z - z_1| + |z - z_2| = |z_1 - z_2|$$

B

= Arg

$$(z - z_1)$$

$$(z - z_2)$$

C

= 0

$$\begin{vmatrix} z - z_1 & \bar{z} - \bar{z}_1 \\ z_2 - z_1 & \bar{z}_2 - \bar{z}_1 \end{vmatrix}$$

D

= Arg

$$(z - z_1)$$

$$(z_2 - z_1)$$

### CORRECT OPTION

Arg

$$(z - z_1)$$

**D**

= Arg

$$(z_2 - z_1)$$

### SOURCE

Mathematics • complex-numbers

### EXPLANATION

Given,

$$z = \frac{(1-t)z_1 + tz_2}{(1-t) + t}$$

Clearly,  $z$  divides  $z_1$  and  $z_2$  in the ratio of  $t : 1-t$ ,  $0 < t < 1$

$$\Rightarrow$$

$$AP + BP = AB$$

i.e.,

$$|z - z_1| + |z - z_2| = |z_1 - z_2|$$

$$\Rightarrow$$

Option  $a$  is true.

and

$$\arg(z - z_1) = \arg(z_2 - z) = \arg(z_2 - z_1)$$

$$\Rightarrow$$

$b$  is false and  $d$  is true.

Also,

$$\arg(z - z_1) = \arg(z_2 - z_1)$$

$$\Rightarrow \arg\left(\frac{z - z_1}{z_2 - z_1}\right) = 0$$

$\therefore$

$$\frac{z - z_1}{z_2 - z_1}$$

is purely real.

$$\Rightarrow \frac{z - z_1}{z_2 - z_1} = \frac{\bar{z} - \bar{z}_1}{\bar{z}_2 - \bar{z}_1}$$

or,

$$\left| \frac{z - z_1}{z_2 - z_1} \cdot \frac{\bar{z}_2 - \bar{z}_1}{\bar{z} - \bar{z}_1} \right| = 0$$

$\therefore$

Option  $c$  is correct.

Hence,  $a, c, d$  is the correct option.

### Question 033 MCQ

#### QUESTION

Let

$$p$$

and

$$q$$

be real numbers such that

$$p \neq 0, p^3 \neq q$$

and

$$p^3 \neq -q.$$

If

$$p^3 \neq -q.$$

and

$$\beta$$

are nonzero complex numbers satisfying

$$\alpha + \beta = -p$$

and

$$\alpha^3 + \beta^3 = q,$$

then a quadratic equation having

$$\frac{\alpha}{\beta}$$

and

$$\frac{\beta}{\alpha}$$

as its roots is

A

$$(p^3 + q)x^2 - (p^3 + 2q)x + (p^3 + q) = 0$$

B

$$(p^3 + q)x^2 - (p^3 - 2q)x + (p^3 + q) = 0$$

C

$$(p^3 - q)x^2 - (5p^3 - 2q)x + (p^3 - q) = 0$$

D

$$(p^3 - q)x^2 - (5p^3 + 2q)x + (p^3 - q) = 0$$

### CORRECT OPTION

B

$$(p^3 + q)x^2 - (p^3 - 2q)x + (p^3 + q) = 0$$

### SOURCE

Mathematics • quadratic-equation-and-inequalities

### EXPLANATION

Sum of roots =

$$\frac{\alpha^2 + \beta^2}{\alpha\beta}$$

and product = 1

Given,

$$\alpha$$

+

$$\beta$$

=

$$-$$

p and

$$\alpha$$

<sup>3</sup> +

$$\beta$$

<sup>3</sup> = q



$$\Rightarrow (\alpha + \beta)(\alpha^2 - \alpha\beta + \beta^2) = q$$

$\therefore$

$$\alpha^2 + \beta^2 - \alpha\beta = \frac{-q}{p}$$

..... *i*

and

$$(\alpha + \beta)^2 = p^2$$

$$\Rightarrow \alpha^2 + \beta^2 + 2\alpha\beta = p^2$$

..... *ii*

From Eq. *i* and *ii*, we get

$$\alpha^2 + \beta^2 = \frac{p^3 - 2q}{3p}$$

and

$$\alpha\beta = \frac{p^3 + q}{3p}$$

$\therefore$

Required equation is

$$x^2 - \frac{(p^2 - 2q)x}{(p^3 + q)} + 1 = 0$$

$$\Rightarrow (p^3 + q)x^2 - (p^3 - 2q)x + (p^3 + q) = 0$$

### Question 034

Numerical

#### QUESTION

Let

$$S_k$$

$= 1, 2, \dots, 100$ , denote the sum of the infinite geometric series whose first term is

$$\frac{k-1}{k!}$$

and the common ratio is

$$\frac{1}{k}$$

. Then the value of

$$\frac{100^2}{100!} + \sum_{k=1}^{100} |(k^2 - 3k + 1) S_k|$$

is

#### SOURCE

Mathematics • sequences-and-series

#### EXPLANATION

Using  $S_{\infty} = \frac{a}{1-r}$ , we get

$$S_k = \begin{cases} 0, & k = 1 \\ \frac{1}{(k-1)!}, & k \geq 2 \end{cases}$$

$$\text{Now } \sum_{k=1}^{100} |(k^2 - 3k + 1) S_k| = \sum_{k=2}^{100} |(k^2 - 3k + 1)| \frac{1}{(k-1)!}$$

$$= |-1| + \sum_{k=3}^{100} \frac{(k^2 - 1) + 1 - 3(k-1) - 2}{(k-1)!}$$

$$\text{as } k^2 - 3k + 1 > 0 \forall k \geq 3$$

$$\begin{aligned}
&= 1 + \sum_{k=3}^{100} \left( \frac{1}{(k-3)!} - \frac{1}{(k-1)!} \right) \\
&= 1 + \left( 1 - \frac{1}{2!} \right) + \left( \frac{1}{1!} - \frac{1}{3!} \right) + \left( \frac{1}{2!} - \frac{1}{4!} \right) + \dots + \\
&\quad \left( \frac{1}{96!} - \frac{1}{98!} \right) + \left( \frac{1}{97!} - \frac{1}{99!} \right) \\
&= 3 - \frac{1}{98!} - \frac{1}{99!} = 3 - \frac{9900}{100!} - \frac{100}{100!} \\
&= 3 - \frac{10000}{100!} = 3 - \frac{(100)^2}{100!} \\
\therefore \frac{100^2}{100!} + \sum_{k=1}^{100} |(k^2 - 3k + 1)S_k| &= 3
\end{aligned}$$

### Question 035

MCQ

#### QUESTION

Let

$A$

and

$B$

be two distinct points on the parabola

$$y^2 = 4x$$

. If the axis of the parabola touches a circle of radius

$r$

having

$AB$

as its diameter, then the slope of the line joining

$A$

and

$B$

can be

A

$$-\frac{1}{r}$$

B

$$\frac{1}{r}$$

C

$$\frac{2}{r}$$

D

$$-\frac{2}{r}$$

#### CORRECT OPTION

C

$$\frac{2}{r}$$

#### SOURCE

Mathematics • parabola

#### EXPLANATION

Let A

$\equiv$

(t

$\frac{2}{1}$

, 2t<sub>1</sub>) and B

$\equiv$

(t

$\frac{2}{2}$

, 2t<sub>2</sub>)

The centre of the circle =

$$\left( \frac{t_1^2 + t_2^2}{2}, t_1 + t_2 \right)$$

As the circle touches the x-axis thus

$$t_1 + t_2 = \pm r$$

Slope of

$$AB = \frac{2}{t_1 + t_2} = \pm \frac{2}{r}$$

### Question 036 MCQ

#### QUESTION

The circle

$$x^2 + y^2 - 8x = 0$$

and hyperbola

$$\frac{x^2}{9} - \frac{y^2}{4} = 1$$

intersect at the points

$A$

and

$B$

Equation of a common tangent with positive slope to the circle as well as to the hyperbola is

A

$$2x - \sqrt{5y} - 20 = 0$$

B

$$2x - \sqrt{5y} + 4 = 0$$

C

$$3x - 4y + 8 = 0$$

D

$$4x - 3y + 4 = 0$$

#### CORRECT OPTION

B

$$2x - \sqrt{5y} + 4 = 0$$

#### SOURCE

Mathematics • hyperbola

## EXPLANATION

A tangent to

$$\frac{x^2}{9} - \frac{y^2}{4} = 1$$

is

$$y = mx + \sqrt{9m^2 - 4}, m > 0$$

.... 1

A tangent to

$$(x - 4)^2 + y^2 = 16$$

is

$$xy = m(x - 4) + 4\sqrt{1 + m^2}$$

..... 2

Comparing 1 and 2,

$$\sqrt{9m^2 - 4} = -4m + 4\sqrt{1 + m^2} \Rightarrow \sqrt{9 - \frac{4}{m^2}} = -4 + 4\sqrt{1 + \frac{1}{m^2}}$$

Let

$$\frac{1}{m^2} = t$$

, we have

$$\sqrt{9 - 4t} = -4 + 4\sqrt{1 + t}$$

Squaring, we have

$$\Rightarrow 9 - 4t = 16 + 16(1 + t) - 32\sqrt{1 + t} \Rightarrow 32\sqrt{1 + t} = 23 + 20t$$

Again squaring

$$1024(1 + t) = 529 + 920t + 400t^2$$

$$\Rightarrow 400t^2 - 104t - 495 = 0 \Rightarrow t = \frac{5}{4}$$

Thus

$$m^2 = \frac{4}{5}, m = \frac{2}{\sqrt{5}}$$

The tangent is

$$y = \frac{2}{\sqrt{5}}x + \frac{4}{\sqrt{5}}$$

i.e.

$$2x - \sqrt{5}y + 4 = 0$$

### Question 037 MCQ

#### QUESTION

The circle

$$x^2 + y^2 - 8x = 0$$

and hyperbola

$$\frac{x^2}{9} - \frac{y^2}{4} = 1$$

intersect at the points

$A$

and

$B$

.

Equation of the circle with

$AB$



as its diameter is

A

$$x^2 + y^2 - 12x + 24 = 0$$

B

$$x^2 + y^2 + 12x + 24 = 0$$

C

$$x^2 + y^2 + 24x - 12 = 0$$

D

$$x^2 + y^2 - 24x - 12 = 0$$

#### CORRECT OPTION

A

$$x^2 + y^2 - 12x + 24 = 0$$

#### SOURCE

Mathematics • hyperbola

#### EXPLANATION

A point on hyperbola is  $3\sec\theta, 2\tan\theta$ .

It lies on the circle, so

$$9\sec^2\theta + 4\tan^2\theta - 24\sec\theta = 0$$

.

$$\Rightarrow 13\sec^2\theta - 24\sec\theta - 4 = 0 \Rightarrow \sec\theta = 2, -\frac{2}{13}$$

Therefore,

$$\sec \theta = 2 \Rightarrow \tan \theta = \sqrt{3}$$

The point of intersection are

$$A(6, 2\sqrt{3})$$

and

$$B(6, -2\sqrt{3})$$

Hence, The circle with AB as diameter is

$$(x - 6)^2 + y^2 = (2\sqrt{3})^2 \Rightarrow x^2 + y^2 - 12x + 24 = 0$$

### Question 038 Numerical

#### QUESTION

The line

$$2x + y = 1$$

is tangent to the hyperbola

$$\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$$

.

If this line passes through the point of intersection of the nearest directrix and the

$x$

-axis, then the eccentricity of the hyperbola is

#### SOURCE

Mathematics • hyperbola

## EXPLANATION

On substituting

$$\left(\frac{a}{e}, 0\right)$$

in

$$y = -2x + 1$$

,

we get

$$0 = -\frac{2a}{e} + 1$$

$$\Rightarrow \frac{a}{e} = \frac{1}{2}$$

Also,

$$y = -2x + 1$$

is tangent to hyperbola

$\therefore$

$$1 = 4a^2 - b^2$$

$$\Rightarrow \frac{1}{a^2} = 4 - (e^2 - 1)$$

$$\Rightarrow \frac{4}{e^2} = 5 - e^2$$

$$\Rightarrow e^4 - 5e^2 + 4 = 0$$

$$\Rightarrow (e^2 - 4)(e^2 - 1) = 0$$

$\Rightarrow$

$$e = 2, e = 1$$

$e = 1$  gives the conic as parabola. But conic is given as hyperbola, hence  $e = 2$ .

### Question 039

MCQ

#### QUESTION

If the angles

$$A, B$$

and

$$C$$

of a triangle are in an arithmetic progression and if

$$a, b$$

and

$$c$$

denote the lengths of the sides opposite to

$$A, B$$

and

$$C$$

respectively, then the value of the expression

$$\frac{a}{c} \sin 2C + \frac{c}{a} \sin 2A$$

is

A

$$\frac{1}{2}$$

B

$$\frac{\sqrt{3}}{2}$$

C

$$1$$

D

$$\sqrt{3}$$

#### CORRECT OPTION

D

$$\sqrt{3}$$

#### SOURCE

Mathematics • properties-of-triangle

#### EXPLANATION

Since, A, B, C are in AP

$\Rightarrow$

$$2B = A + C \text{ i.e.,}$$

$\angle$

$$B = 60$$

$\circ$

$\therefore$

$$\frac{a}{c}$$

$$2\sin C \cos C +$$

$$\frac{c}{a}$$

$$2\sin A \cos A$$

$$= 2k a \cos C + c \cos A$$

$$\text{using, } \frac{a}{\sin A} = \frac{b}{\sin B} = \frac{c}{\sin C} = \frac{1}{k}$$

$$= 2k b$$

$$= 2 \sin B$$

$$\text{using, } b = a \cos C + c \cos A$$

$$=$$

$$\sqrt{3}$$

## Question 040

Numerical

### QUESTION

Let

$$f$$

be a real-valued differentiable function on

$$R$$

the set of all real numbers such that

$$f(1) = 1$$

. If the

$$y$$

-intercept of the tangent at any point

$$P(x, y)$$

on the curve

$$y = f(x)$$

is equal to the cube of the abscissa of

$$P$$

, then find the value of

$$f(-3)$$

### SOURCE

Mathematics • application-of-derivatives

### EXPLANATION

The equation of the tangent at  $x, y$  to the given curve  $y = f(x)$  is

$$Y - y = \frac{dy}{dx}(X - x)$$

Y-intercept

$$= y - x \frac{dy}{dx}$$

According to the question

$$\begin{aligned} x^3 &= y - x \frac{dy}{dx} \\ \Rightarrow \frac{dy}{dx} - \frac{y}{x} &= -x^2 \end{aligned}$$

which is linear in  $x$ .

$$\begin{aligned} IF &= e^{\int \frac{-1}{x} dx} = \frac{1}{x} \\ \therefore \end{aligned}$$

Required solution is

$$\begin{aligned} y \cdot \frac{1}{x} &= \int -x^2 \cdot \frac{1}{x} dx \\ \Rightarrow \frac{y}{x} &= \frac{-x^2}{2} + c \end{aligned}$$

$$\Rightarrow y = \frac{-x^3}{2} + cx$$

At  $x = 1, y = 1,$

$$1 = \frac{-1}{2} + c$$

$$\Rightarrow c = \frac{3}{2}$$

Now,

$$\begin{aligned} f(-3) &= \frac{27}{2} + \frac{3}{2}(-3) \\ &= \frac{27 - 9}{2} = 9 \end{aligned}$$

### Question 041 Numerical

#### QUESTION

If the distance between the plane

$$Ax - 2y + z = d$$

and the plane containing the lines

$$\frac{x - 1}{2} = \frac{y - 2}{3} = \frac{z - 3}{4}$$

and

$$\frac{x - 2}{3} = \frac{y - 3}{4} = \frac{z - 4}{5}$$

is

$$\sqrt{6},$$

then

$$|d|$$



is \_\_\_\_\_.

## SOURCE

Mathematics • 3d-geometry

## EXPLANATION

We have a plane

$$Ax - 2y + z = d$$

and another plane that contains the two lines

$$\text{Line 1: } \frac{x-1}{2} = \frac{y-2}{3} = \frac{z-3}{4},$$

$$\text{Line 2: } \frac{x-2}{3} = \frac{y-3}{4} = \frac{z-4}{5}.$$

We know the distance between these two planes is  $\sqrt{6}$ , and we want to find  $|d|$ .

## 1. Find the equation of the plane containing the two given lines

### Step 1a: Parametric forms of the lines

**Line 1:** Let the parameter be  $t$ . Then

$$\frac{x-1}{2} = \frac{y-2}{3} = \frac{z-3}{4} = t \implies \begin{cases} x = 1 + 2t, \\ y = 2 + 3t, \\ z = 3 + 4t. \end{cases}$$

A direction vector for Line 1 is  $\mathbf{v}_1 = (2, 3, 4)$ .

A point on Line 1 is  $\mathbf{P}_1 = (1, 2, 3)$ .

**Line 2:** Let the parameter be  $s$ . Then

$$\frac{x-2}{3} = \frac{y-3}{4} = \frac{z-4}{5} = s \implies \begin{cases} x = 2 + 3s, \\ y = 3 + 4s, \\ z = 4 + 5s. \end{cases}$$

A direction vector for Line 2 is  $\mathbf{v}_2 = (3, 4, 5)$ .

A point on Line 2 is  $\mathbf{P}_2 = (2, 3, 4)$ .

### Step 1b: Normal to the plane containing these lines

A plane that contains both lines must contain their direction vectors  $\mathbf{v}_1$  and  $\mathbf{v}_2$ . Therefore, a normal to this plane is given by the cross product  $\mathbf{v}_1 \times \mathbf{v}_2$ .

$$\mathbf{v}_1 = (2, 3, 4), \quad \mathbf{v}_2 = (3, 4, 5).$$

Compute the cross product:

$$\mathbf{v}_1 \times \mathbf{v}_2 = \det \begin{pmatrix} \mathbf{i} & \mathbf{j} & \mathbf{k} \\ 2 & 3 & 4 \\ 3 & 4 & 5 \end{pmatrix} = (3 \cdot 5 - 4 \cdot 4, 4 \cdot 3 - 2 \cdot 5, 2 \cdot 4 - 3 \cdot 3) = (15$$

Hence a normal vector to the plane is  $\mathbf{n} = (-1, 2, -1)$ . Equivalently, we can multiply by  $-1$  *which does not change the plane* to get  $\mathbf{n} = (1, -2, 1)$ .

Thus the plane containing the two lines has the form

$$1 \cdot x - 2 \cdot y + 1 \cdot z = K,$$

i.e.

$$x - 2y + z = K.$$

### Step 1c: Find the constant $K$

To find  $K$ , just plug in any point on either line. For instance, the point  $\mathbf{P}_1 = (1, 2, 3)$  on Line 1:

$$1(1) - 2(2) + 1(3) = 1 - 4 + 3 = 0.$$

So  $K = 0$ .

Check also with  $\mathbf{P}_2 = (2, 3, 4)$  from Line 2:

$$2 - 2 \cdot 3 + 4 = 2 - 6 + 4 = 0.$$

That also gives 0. So indeed the plane containing both lines is

$$\boxed{x - 2y + z = 0}.$$

## 2. Determine $A$ so that the planes can be parallel

We are given the plane

$$Ax - 2y + z = d$$

and have found that the plane containing the lines is

$$x - 2y + z = 0.$$

For these two planes to have a finite, nonzero distance between them, they must be parallel. Two planes are parallel precisely when their normal vectors are scalar multiples of each other.

The normal to  $Ax - 2y + z = d$  is  $(A, -2, 1)$ .

The normal to  $x - 2y + z = 0$  is  $(1, -2, 1)$ .

Set

$$(A, -2, 1) = \lambda (1, -2, 1).$$

Matching components:

$$A = \lambda \cdot 1 = \lambda.$$

$$-2 = \lambda \cdot (-2) \implies \lambda = 1.$$

$$1 = \lambda \cdot (1) \implies \lambda = 1.$$

Hence  $\lambda = 1$  and  $A = 1$ .

Therefore, the given plane must be

$$\boxed{x - 2y + z = d}.$$

## 3. Use the formula for the distance between two parallel planes

Now we have two parallel planes:

$$x - 2y + z = 0,$$

$$x - 2y + z = d.$$

The normal vector to both is  $\mathbf{n} = (1, -2, 1)$ . Its magnitude is

$$\|\mathbf{n}\| = \sqrt{1^2 + (-2)^2 + 1^2} = \sqrt{1 + 4 + 1} = \sqrt{6}.$$

The distance  $D$  between two parallel planes

$$\alpha_1 : \mathbf{n} \cdot \mathbf{x} = k_1, \quad \alpha_2 : \mathbf{n} \cdot \mathbf{x} = k_2$$

is given by

$$D = \frac{|k_1 - k_2|}{\|\mathbf{n}\|}.$$

In our case:

For the plane  $x - 2y + z = 0$ , we have  $k_1 = 0$ .

For the plane  $x - 2y + z = d$ , we have  $k_2 = d$ .

The distance is given to be  $\sqrt{6}$ .

Thus

$$\sqrt{6} = \frac{|0 - d|}{\sqrt{6}} = \frac{|d|}{\sqrt{6}} \implies |d| = 6.$$

## 4. Conclusion

$$\boxed{|d| = 6}.$$

### Question 042 Numerical

#### QUESTION

If

$$\vec{a}$$

and

$$\vec{b}$$

are vectors in space given by

$$\vec{a} = \frac{\hat{i} - 2\hat{j}}{\sqrt{5}}$$

and

$$\vec{b} = \frac{2\hat{i} + \hat{j} + 3\hat{k}}{\sqrt{14}},$$

then find the value of

$$(2\vec{a} + \vec{b}) \cdot [(\vec{a} \times \vec{b}) \times (\vec{a} - 2\vec{b})].$$

#### SOURCE

Mathematics • vector-algebra

#### EXPLANATION

$$(2\vec{a} + \vec{b})[(\vec{a} \times \vec{b}) \times (\vec{a} - 2\vec{b})]$$

$$= 4(\vec{a} \cdot \vec{a}) + \vec{b} \cdot \vec{b} + 4\vec{a} \cdot \vec{b} \text{ where}$$

$$\vec{a} \cdot \vec{b} = \frac{2 - 2}{\sqrt{70}} = 0 \quad |\vec{a}| = 1 \text{ and } |\vec{b}| = 1$$

$$= 5$$

**QUESTION**

Equation of the plane containing the straight line

$$\frac{x}{2} = \frac{y}{3} = \frac{z}{4}$$

and perpendicular to the plane containing the straight lines

$$\frac{x}{3} = \frac{y}{4} = \frac{z}{2}$$

and

$$\frac{x}{4} = \frac{y}{2} = \frac{z}{3}$$

is

**A**

$$x + 2y - 2z = 0$$

**B**

$$3x + 2y - 2z = 0$$

**C**

$$x - 2y + z = 0$$

**D**

$$5x + 2y - 4z = 0$$

**CORRECT OPTION****C**

$$x - 2y + z = 0$$

**SOURCE**

**EXPLANATION**

Plane 1 :

$$ax + by + cz = 0$$

contains line

$$\frac{x}{2} = \frac{y}{3} = \frac{z}{4}$$

. Therefore,

$$2a + 3b + 4c = 0$$

..... 1

Plane 2 :

$$a'x + b'y + c'z = 0$$

is perpendicular to plane containing line

$$\frac{x}{3} = \frac{y}{4} = \frac{z}{2}$$

and

$$\frac{x}{4} = \frac{y}{2} = \frac{z}{3}$$

.

Hence,

$$3a' + 4b' + 2c' = 0$$

and

$$4a' + 2b' + 3c' = 0$$

.

$$\Rightarrow \frac{a'}{12 - 4} = \frac{b'}{8 - 9} = \frac{c'}{6 - 16}$$

$$\Rightarrow 8a - b - 10c = 0$$

.... 2

From Eq. 1 and 2, we get

$$\frac{a}{-30 + 4} = \frac{b}{32 + 20} = \frac{c}{-2 - 24}$$

$$\Rightarrow$$

Equation of plane is

$$x - 2y + z = 0$$

#### Question 044 MCQ

##### QUESTION

Let

$$P, Q, R$$

and

$$S$$

be the points on the plane with position vectors

$$-2\hat{i} - \hat{j}, 4\hat{i}, 3\hat{i} + 3\hat{j}$$

and

$$-3\hat{i} + 2\hat{j}$$

respectively. The quadrilateral

$$PQRS$$

must be a



**A** parallelogram, which is neither a rhombus nor a rectangle

**B** square

**C** rectangle, but not a square

**D** rhombus, but not a square

#### CORRECT OPTION

**A** parallelogram, which is neither a rhombus nor a rectangle

#### SOURCE

Mathematics • vector-algebra

#### EXPLANATION

We have

$$PS = \sqrt{1^2 + 3^2} = \sqrt{10}$$

$$SR = \sqrt{6^2 + 1^2} = \sqrt{37}$$

,

$$RQ = \sqrt{1^2 + 3^2} = \sqrt{10}$$

$$QP = \sqrt{6^2 + 1^2} = \sqrt{37}$$

Then PQRS is a parallelogram. It can be a rectangle, so let's check the product of slopes of PS and SR

$$(m_{PS}) \cdot (m_{SR}) = \frac{3}{-1} \times \frac{1}{6} = -\frac{1}{2} \neq -1$$

Thus PS and SR are not perpendicular. So, it's not a rectangle.

Also,

$$m_{PR} \cdot m_{QS} = \frac{4}{5} \times \frac{2}{-7} = -\frac{8}{35} \neq -1$$

Thus, PR and QS are also not perpendicular. So it's not a rhombus either.

**Question 045** **MCQ**

**QUESTION**

Let

$$\omega$$

be a complex cube root of unity with

$$\omega \neq 1.$$

A fair die is thrown three times. If

$$r_1,$$

$$r_2$$

and

$$r_3$$

are the numbers obtained on the die, then the probability that

$$\omega^{r_1} + \omega^{r_2} + \omega^{r_3} = 0$$

is

**A**

$$\frac{1}{18}$$

**B**

$$\frac{1}{9}$$

**C**

$$\frac{2}{9}$$

**D**

$$\frac{1}{36}$$

**CORRECT OPTION****C**

$$\frac{2}{9}$$

**SOURCE**

Mathematics • probability

**EXPLANATION**

Sample space A dice is thrown thrice,

$$n(s) = 6 \times 6 \times 6$$

.

Favorable events

$$\omega^{r_1} + \omega^{r_2} + \omega^{r_3} = 0$$

i.e.

$$(r_1, r_2, r_3)$$

are ordered 3-triples which can take values,

$$\begin{aligned} & (1, 2, 3), (1, 5, 3), (4, 2, 3), (4, 5, 3) \} \\ & (1, 2, 6), (1, 5, 6), (4, 2, 6), (4, 5, 6) \} \end{aligned}$$

i.e. 8 ordered pairs and each can be arranged in  $3!$  ways = 6

$\therefore$

$$n(E) = 8 \times 6$$

$$\Rightarrow P(E) = \frac{8 \times 6}{6 \times 6 \times 6} = \frac{2}{9}$$

### Question 046 Numerical

#### QUESTION

For any real number

$$x,$$

let

$$[x]$$

denote the largest integer less than or equal to

$$x.$$

Let

$$f$$

be a real valued function defined on the interval

$$[-10, 10]$$

by

$$f(x) = \begin{cases} x - [x] & \text{if } [x] \text{ is odd,} \\ 1 + [x] - x & \text{if } [x] \text{ is even} \end{cases}$$

\$

Then the value of

$$\frac{\pi^2}{10} \int_{-10}^{10} f(x) \cos \pi x \, dx$$

is

### SOURCE

Mathematics • definite-integration

### EXPLANATION

Case 1 :

Let

$$0 \leq x < 1$$

then

$$[x] = 0$$

, which is even

$$\therefore$$

$$\begin{aligned} f(x) &= 1 + [x] - x \\ &= 1 + 0 - x \\ &= 1 - x \end{aligned}$$

Case 2 :

Let

$$1 \leq x < 2$$

then

$$[x] = 1$$

, which is odd

$$\therefore$$

$$f(x) = x - [x]$$

$$= x - 1$$

Case 3 :

Let

$$2 \leq x < 3$$

then

$$[x] = 2$$

, which is even

$$\therefore$$

$$f(x) = 1 + [x] - x$$

$$= 1 + 2 - x$$

$$= 3 - x$$

Case 4 :

Let

$$3 \leq x < 4$$

then

$$[x] = 3$$

, which is odd

$$\therefore$$

$$f(x) = x - [x]$$

$$= x - 3$$

$$\therefore$$

$$f(x) = \begin{cases} 1 - x & ; \quad 0 \leq x < 1 \\ x - 1 & ; \quad 1 \leq x < 2 \\ 3 - x & ; \quad 2 \leq x < 3 \\ x - 3 & ; \quad 3 \leq x < 4 \end{cases}$$

$$\therefore$$

$$f(x)$$

is periodic and period of

$$f(x) = 2$$

And period of

$$\cos \pi x = \frac{2\pi}{\pi} = 2$$

$\therefore$

Period of

$$f(x) \cos \pi x = 2$$

Now,

$$\begin{aligned} I &= \frac{\pi^2}{10} \int_{-10}^{10} f(x) \cos \pi x \, dx \\ &= \frac{\pi^2}{10} \int_{-10}^{-10+10 \times 2} f(x) \cos \pi x \, dx \\ &= \frac{\pi^2}{10} \int_0^{10 \times 2} f(x) \cos \pi x \, dx \\ &= \frac{\pi^2}{10} \times 10 \int_0^2 f(x) \cos \pi x \, dx \\ &= \pi^2 \int_0^2 f(x) \cos \pi x \, dx \\ &\therefore \end{aligned}$$

$$\begin{aligned} I &= \pi^2 \left[ \int_0^1 f(x) \cos \pi x \, dx + \int_1^2 f(x) \cos \pi x \, dx \right] \\ &= \pi^2 \left[ \int_0^1 (1-x) \cos \pi x \, dx + \int_1^2 (x-1) \cos \pi x \, dx \right] \\ &= \pi^2 \left[ \int_0^1 \cos \pi x \, dx - \int_0^1 x \cos \pi x \, dx + \int_1^2 x \cos \pi x \, dx - \int_1^2 \cos \pi x \, dx \right] \\ &= \pi^2 \left[ \frac{1}{\pi} [\sin \pi x]_0^1 - \int_0^1 x \cos \pi x \, dx + \int_1^2 x \cos \pi x \, dx - \frac{1}{\pi} [\sin \pi x]_1^2 \right] \end{aligned}$$

$$\begin{aligned}
&= \pi^2 \left[ 0 - \int_0^1 x \cos \pi x \, dx + \int_1^2 x \cos \pi x \, dx - 0 \right] \\
&= \pi^2 \left[ - \left[ x \frac{\sin \pi x}{\pi} + \frac{1}{\pi^2} \cos \pi x \right]_0^1 + \left[ x \frac{\sin \pi x}{\pi} + \frac{1}{\pi^2} \cos \pi x \right]_1^2 \right] \\
&\left[ \text{As } \int x \cos \pi x \, dx = x \cdot \int \cos \pi x - \int \left( 1 \cdot \frac{\sin \pi x}{\pi} \right) dx = x \cdot \frac{\sin \pi x}{\pi} + \frac{1}{\pi^2} \cos \pi x \right] \\
&= \pi^2 \left[ - \left[ \left( 1 \cdot \frac{\sin \pi}{\pi} + \frac{1}{\pi^2} \cdot \cos \pi \right) - \left( 0 + \frac{1}{\pi^2} \cdot \cos 0 \right) \right] + \left[ \left( 2 \cdot \frac{\sin 2\pi}{\pi} + \frac{1}{\pi^2} \cos 2\pi \right) - \left( 1 \cdot \frac{\sin \pi}{\pi} + \frac{1}{\pi^2} \cos \pi \right) \right] \right] \\
&= \pi^2 \left[ - \left\{ \left( -\frac{1}{\pi^2} \right) - \left( \frac{1}{\pi^2} \right) \right\} + \left\{ \left( +\frac{1}{\pi^2} \right) - \left( -\frac{1}{\pi^2} \right) \right\} \right] \\
&= \pi^2 \left[ - \left( -\frac{2}{\pi^2} \right) + \frac{2}{\pi^2} \right] \\
&= \pi^2 \left[ \frac{2}{\pi^2} + \frac{2}{\pi^2} \right] \\
&= \pi^2 \times \frac{4}{\pi^2} \\
&= 4
\end{aligned}$$

#### Question 047 MCQ

##### QUESTION

Let

$$f$$

be a real-valued function defined on the interval

$$(0, \infty)$$

by



$$f(x) = \ln x + \int_0^x \sqrt{1 + \sin t} \, dt.$$

then which of the following  
statements *s* is *are* true?

$$f''(x)$$

**A** exists for all

$$x \in (0, \infty)$$

$$f'(x)$$

exists for all

$$x \in (0, \infty)$$

and

**B**

$$f'$$

is continuous on

$$(0, \infty)$$

, but not differentiable on

$$(0, \infty)$$

there exists

$$\alpha > 1$$

such that

**C**

$$|f'(x)| < |f(x)|$$

for all

$$x \in (\alpha, \infty)$$

there exists

$$\beta > 0$$

such that

D

$$|f(x)| + |f'(x)| \leq \beta$$

for all

$$x \in (0, \infty)$$

#### CORRECT OPTION

there exists

$$\alpha > 1$$

such that

C

$$|f'(x)| < |f(x)|$$

for all

$$x \in (\alpha, \infty)$$

#### SOURCE

Mathematics • application-of-integration

#### EXPLANATION

$$f(x) = \ln x + \int_0^x \sqrt{1 + \sin t} \, dt$$

$$f'(x) = \frac{1}{x} + \sqrt{1 + \sin x}$$

$f$  is not differentiable at  $\sin x =$

—

1

i.e.

$$x = 2n\pi - \frac{\pi}{2}, n \in \mathbb{N}$$

in the interval  $0, \infty$

$$f''(x) = -\frac{1}{x^2} + \frac{\cos x}{2\sqrt{1 + \sin x}}$$

$f'$  does not exist for all  $x$

$\in$

$0, \infty$

$f'$  exist for  $x > 0$

we have

$$\frac{1}{x} + \sqrt{1 + \sin x} < \ln x + \int_0^x \sqrt{1 + \sin x} dx$$

because L.H.S. is bounded and R.H.S. is not bounded so

$\exists$

some

$\alpha$

beyond which R.H.S. is greater than L.H.S.

i.e.

$$|f'(x)| < |f(x)|$$

for all  $x$

$\in$

\$\$\$a\$\$\$, \$\$\$\infty\$\$\$

$$|f| + |f'| \leq \beta$$

is wrong as  $f$  is unbounded while  $f'$  is bounded.

### Question 048 MCQ

#### QUESTION

The value of

$$\int_0^1 \frac{x^4(1-x)^4}{1+x^2} dx$$

is *are*

A

$$\frac{22}{7} - \pi$$

B

$$\frac{2}{105}$$

C

$$0$$

D

$$\frac{71}{15} - \frac{3\pi}{2}$$

#### CORRECT OPTION

**SOURCE**

Mathematics • definite-integration

**EXPLANATION**

$$\begin{aligned}
\int_0^1 \frac{x^4(1-x)^4}{1+x^2} dx &= \int_0^1 \frac{x^4\{(1+x^2) - 2x\}^2}{1+x^2} dx \\
&= \int_0^1 x^4 \frac{(1+x^2)^2 - 4x(1+x^2) + 4x^2}{1+x^2} dx \\
&= \int_0^1 x^4 \left[ 1+x^2 - 4x + \frac{4\{1+x^2-1\}}{1+x^2} \right] dx \\
&= \int_0^1 x^4 \left[ 1+x^2 - 4x + 4 - \frac{4}{1+x^2} \right] dx \\
&= \int_0^1 \left[ x^6 - 4x^5 + 5x^4 - 4 \frac{x^4 - 1 + 1}{1+x^2} \right] dx \\
&= \int_0^1 (x^6 - 4x^5 + 5x^4 - 4x^2 + 4) dx - 4 \int_0^1 \frac{dx}{1+x^2} \\
&= \left[ \frac{x^7}{7} - \frac{2x^6}{3} + x^5 - \frac{4x^3}{3} + 4x \right]_0^1 - 4[\tan^{-1}x]_0^1 \\
&= \left[ \frac{1}{7} - \frac{2}{3} + 1 - \frac{4}{3} + 4 \right] - \pi = \frac{22}{7} - \pi
\end{aligned}$$

## QUESTION

The value of

$$\lim_{x \rightarrow 0} \frac{1}{x^3} \int_0^x \frac{t \ln(1+t)}{t^4 + 4} dt$$

is

A

0

B

 $\frac{1}{12}$ 

C

 $\frac{1}{24}$ 

D

 $\frac{1}{64}$ 

## CORRECT OPTION

B

 $\frac{1}{12}$ 

## SOURCE

Mathematics • definite-integration

### EXPLANATION

$$\lim_{x \rightarrow 0} \frac{1}{x^3} \int_0^x \frac{t \log(1+t)}{4+t^4} dt$$

Using L' Hospital's rule,

$$= \lim_{x \rightarrow 0} \frac{\frac{x \log(1+x)}{4+x^4}}{3x^2}$$

$$= \lim_{x \rightarrow 0} \frac{\log(1+x)}{3x} \cdot \frac{1}{4+x^4}$$

$$= \frac{1}{3} \cdot \frac{1}{4} = \frac{1}{12}$$

$$\text{using, } \lim_{x \rightarrow 0} \frac{\log(1+x)}{x} = 1$$

### Question 050 MCQ

#### QUESTION

Let

$$ABC$$

be a triangle such that

$$\angle ACB = \frac{\pi}{6}$$

and let

$$a, b$$

and

$$c$$

denote the lengths of the sides opposite to

$A$

,

$B$

and

$C$

respectively. The value  $s$  of

$x$

for which

$$a = x^2 + x + 1, \quad b = x^2 - 1$$

and

$$c = 2x + 1$$

is *are*

A

$$- (2 + \sqrt{3})$$

B

$$1 + \sqrt{3}$$

C

$$2 + \sqrt{3}$$

D

$$4\sqrt{3}$$

**CORRECT OPTION**



**SOURCE**

Mathematics • properties-of-triangle

**EXPLANATION**

Using,

$$\cos C = \frac{a^2 + b^2 - c^2}{2ab}$$

$$\Rightarrow \frac{\sqrt{3}}{2} = \frac{(x^2 + x + 1)^2 + (x^2 - 1)^2 - (2x + 1)^2}{2(x^2 + x + 1)(x^2 - 1)}$$

$$\Rightarrow (x + 2)(x + 1)(x - 1)x + (x^2 - 1)^2 = \sqrt{3}(x^2 + x + 1)(x^2 - 1)$$

$$\Rightarrow x^2 + 2x + (x^2 - 1) = \sqrt{3}(x^2 + x + 1)$$

$$\Rightarrow (2 - \sqrt{3})x^2 + (2 - \sqrt{3})x - (\sqrt{3} + 1) = 0$$

$$\Rightarrow x = -(2 + \sqrt{3})$$

and

$$x = 1 + \sqrt{3}$$

But,

$$x = -(2 + \sqrt{3}) \Rightarrow c$$

is negative.

 $\therefore$ 

$$x = 1 + \sqrt{3}$$

is the only solution.

Hence,  $b$  is the correct option.

### Question 051 MCQ

#### QUESTION

The number of  $3 \times 3$  matrices  $A$  whose entries are either 0 or 1 and for which the system

$$A \begin{bmatrix} x \\ y \\ z \end{bmatrix} = \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix} \text{ has exactly two distinct solutions, is}$$

**A** 0

**B**  $2^9 - 1$

**C** 168

**D** 2

#### CORRECT OPTION

**A** 0

#### SOURCE

Mathematics • matrices-and-determinants

#### EXPLANATION

Given,

A  $\begin{bmatrix} x \\ y \\ z \end{bmatrix} = \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix}$  has two distinct solution we know that three Planes cannot intersect at two distinct point.

Hence, number of  $3 \times 3$  matrix  $A$  is zero.

**Question 052** **MCQ**

**QUESTION**

Let  $f, g$  and  $h$  be real valued functions defined on the interval  $[0, 1]$  by

$$f(x) = e^{x^2} + e^{-x^2},$$

$$g(x) = xe^{x^2} + e^{-x^2}$$

$$\text{and } h(x) = x^2e^{x^2} + e^{-x^2}.$$

If  $a, b$  and  $c$  denote, respectively, the absolute maximum of  $f, g$  and  $h$  on  $[0, 1]$ , then :

**A**  $a = b$  and  $c \neq b$

**B**  $a = c$  and  $a \neq b$

**C**  $a \neq b$  and  $c \neq b$

**D**

$$a = b = c$$

**CORRECT OPTION****D**

$$a = b = c$$

**SOURCE**

Mathematics • functions

**EXPLANATION**

$$f(x) = e^{x^2} + e^{-x^2}$$

$$f'(x) = e^{x^2} \frac{d}{dx}(x^2) + e^{-x^2} \frac{d}{dx}(-x^2)$$

$$= e^{x^2}(2x) + e^{-x^2}(-2x)$$

$$= 2x(e^{x^2} - e^{-x^2}) \geq 0 \forall x \in [0, 1]$$

$$g(x) = xe^{x^2} + e^{-x^2}$$

$$h(x) = x^2e^{x^2} + e^{-x^2}$$

Clearly for  $0 \leq x \leq 1$   $f(x) \geq g(x) \geq h(x)$

$\therefore f(1) = g(1) = h(1) = e + \frac{1}{e}$  and  $f(1)$  is greatest

$$\therefore a = b = c = e + \frac{1}{e}$$

### QUESTION

Let  $z_1$  and  $z_2$  be two distinct complex numbers let  $z = (1 - t)z_1 + tz_2$  for some real number  $t$  with  $0 < t < 1$ .

If  $\text{Arg}(w)$  denotes the principal argument of a nonzero complex number  $w$ , then :

**A**  $|z - z_1| + |z - z_2| = |z_1 - z_2|$

**B**  $\text{Arg}(z - z_1) = \text{Arg}(z - z_2)$

**C**  $\begin{vmatrix} z - z_1 & \bar{z} - \bar{z}_1 \\ z_2 - z_1 & \bar{z}_2 - \bar{z}_1 \end{vmatrix} = 0$

**D**  $\text{Arg}(z - z_1) = \text{Arg}(z_2 - z_1)$

### CORRECT OPTION

**A**  $|z - z_1| + |z - z_2| = |z_1 - z_2|$

### SOURCE

Mathematics • complex-numbers

### EXPLANATION

$$\text{Given, } z = (1 - t)z_1 + tz_2$$

$$\Rightarrow \frac{z - z_1}{z_2 - z_1} = t$$

$$\Rightarrow \arg\left(\frac{z - z_1}{z_2 - z_1}\right) = 0 \dots\dots\dots (i)$$

$$\Rightarrow \arg(z - z_1) = \arg(z_2 - z_1)$$

$$\frac{z - z_1}{z_2 - z_1} = \frac{\bar{z} - \bar{z}_1}{\bar{z}_2 - \bar{z}_1}$$

$$\begin{vmatrix} z - z_1 & \bar{z} - \bar{z}_1 \\ z_2 - z_1 & \bar{z}_2 - \bar{z}_1 \end{vmatrix} = 0$$

$$AP + PB = AB$$

$$\Rightarrow |z - z_1| + |z - z_2| = |z_1 - z_2|$$

#### Question 054 MCQ

##### QUESTION

The number of  $A$  in  $T_p$  such that  $A$  is either symmetric or skew-symmetric or both, and  $\det(A)$  divisible by  $p$  is :

**A**  $(p - 1)^2$

**B**  $2(p - 1)$

**C**  $(p - 1)^2 + 1$

**D**  $2p - 1$

##### CORRECT OPTION

**D**  $2p - 1$

## SOURCE

Mathematics • matrices-and-determinants

## EXPLANATION

We must have  $a^2 - b^2 = 1 < p$

$$(a + b)(a - b) = 1 < p$$

Either  $a - b = 0$  or  $a + b$  is a multiple of  $p$  when  $a = b$  number of matrices is  $p$  and when  $a + b = \text{multiple of } p$ .

$$\Rightarrow a, b \text{ has } p - 1$$

$$\text{Total number of matrices} = p + p - 1$$

$$= 2p - 1$$

## Question 055 MCQ

### QUESTION

The number of  $A$  in  $T_p$  such that the trace of  $A$  is not divisible by  $p$  but  $\det(A)$  is divisible by  $p$  is

[Note : The trace of a matrix is the sum of its diagonal entries.]

**A**  $(p - 1)(p^2 - p + 1)$

**B**  $p^3 - (p - 1)^2$

**C**  $(p - 1)^2$

**D**  $(p-1)(p^2-2)$

**CORRECT OPTION**

**C**  $(p-1)^2$

**SOURCE**

Mathematics • matrices-and-determinants

**EXPLANATION**

We have an odd prime  $p$ . Consider the set

$$T_p = \left\{ A = \begin{pmatrix} a & b \\ c & a \end{pmatrix} : a, b, c \in \{0, 1, 2, \dots, p-1\} \right\}.$$

We want to count the number of such matrices  $A$  that satisfy:

$$\text{trace}(A) \not\equiv 0 \pmod{p}.$$

Since  $\text{trace}(A) = a + a = 2a$ , this means

$$2a \not\equiv 0 \pmod{p}.$$

Because  $p$  is an odd prime, 2 is invertible mod  $p$ .

Hence  $2a \equiv 0 \pmod{p}$  if and only if  $a \equiv 0 \pmod{p}$ .

Thus the condition

$$\text{trace}(A) \not\equiv 0 \pmod{p}$$

is equivalent to

$$a \not\equiv 0 \pmod{p}.$$

In other words,  $a$  must be a nonzero element modulo  $p$ ; so

$$a \in \{1, 2, \dots, p-1\}.$$



$$\det(A) \equiv 0 \pmod{p}.$$

For the matrix

$$\begin{pmatrix} a & b \\ c & a \end{pmatrix}, \text{ we have}$$

$$\det(A) = a^2 - bc.$$

The condition  $\det(A) \equiv 0 \pmod{p}$  means

$$a^2 - bc \equiv 0 \pmod{p} \iff bc \equiv a^2 \pmod{p}.$$

Putting these two pieces together:

$$a \text{ is nonzero } \$a \in \{1, \dots, p-1\} \$.$$

$$bc \equiv a^2 \pmod{p}.$$

## Counting the solutions

We must count the number of triples  $(a, b, c)$  with  $a \neq 0$  and  $bc \equiv a^2$ .

### Range for $a$

Since  $a \in \{1, 2, \dots, p-1\}$ , there are  $p-1$  possible values of  $a$ .

### Condition $bc \equiv a^2$ for given $a$

Since  $a \neq 0$ ,  $a^2$  is also nonzero mod  $p$ .

If  $b = 0$ , then  $bc \equiv 0$ , which would only be equal to  $a^2$  if  $a^2 \equiv 0$ . But  $a^2 \neq 0$  since  $a \neq 0$ . So  $b = 0$  gives no solutions.

Similarly, if  $c = 0$ , then  $bc = 0$ , which again cannot equal the nonzero  $a^2$ . So  $c = 0$  gives no solutions either.

Therefore  $b$  and  $c$  must both be nonzero modulo  $p$ .

Once  $b$  is chosen to be any nonzero element in  $\{1, \dots, p-1\}$ , there is a unique  $c \equiv a^2 b^{-1} \pmod{p}$ . Hence:

For each nonzero  $b$  *that is*,  $b \in \{1, \dots, p-1\}$ , there is exactly **one**  $c \in \{1, \dots, p-1\}$  satisfying  $bc \equiv a^2$ .

Consequently, for each fixed nonzero  $a$ , the number of  $(b, c)$ -pairs is  $p-1$ .

### Putting it all together

We have  $p-1$  choices for  $a \neq 0$ .

For each such  $a$ , there are  $p-1$  pairs  $(b, c)$  satisfying  $bc \equiv a^2$ .

Therefore, in total, the number of matrices is

$$(p-1) \times (p-1) = (p-1)^2.$$

Hence the correct answer matches option **C**, which is

$(p-1)^2.$

## Question 056 MCQ

### QUESTION

The number of  $A$  in  $T_p$  such that  $\det(A)$  is not divisible by  $p$  is :

**A**  $2p^2$

**B**  $p^3 - 5p$

**C**  $p^3 - 3p$

**D**  $p^3 - p^2$

### CORRECT OPTION

**D**  $p^3 - p^2$

### SOURCE

Mathematics • matrices-and-determinants

### EXPLANATION

Let  $p$  be an odd prime number, and consider the set

$$T_p = \left\{ A = \begin{pmatrix} a & b \\ c & a \end{pmatrix} : a, b, c \in \{0, 1, 2, \dots, p-1\} \right\}.$$

We want to count the number of matrices  $A \in T_p$  whose determinant is **not** divisible by  $p$ .

Recall that for

$$A = \begin{pmatrix} a & b \\ c & a \end{pmatrix},$$

the determinant is

$$\det(A) = a^2 - bc.$$

Hence  $\det(A)$  is **not** divisible by  $p$  precisely when

$$a^2 - bc \not\equiv 0 \pmod{p}, \quad \text{i.e.,} \quad a^2 \not\equiv bc \pmod{p}.$$

---

### Total number of matrices

Since  $a, b, c$  each range over  $\{0, 1, \dots, p-1\}$ , there are

$$p^3$$

total possible matrices in  $T_p$ .

---

**Step 1: Count how many  $(a, b, c)$  do satisfy  $a^2 \equiv bc \pmod{p}$**

It will be easier to first count the number of triples  $(a, b, c)$  for which

$$a^2 \equiv bc \pmod{p},$$

and then subtract that from  $p^3$ .

---

**Case A:**  $a = 0$

Then  $a^2 = 0$ . We need

$$bc \equiv 0 \pmod{p}.$$

Over the field  $\mathbf{F}_p$ , the product  $bc \equiv 0$  if and only if at least one of  $b$  or  $c$  is 0 *since  $p$  is prime*.

If  $b = 0$ , then  $c$  can be anything in  $\{0, \dots, p-1\}$ .

This gives  $p$  possibilities.

If  $c = 0$ , then  $b$  can be anything in  $\{0, \dots, p-1\}$ .

This also gives  $p$  possibilities.

However, the pair  $(b = 0, c = 0)$  is counted in both cases, so we must subtract it out once.

Hence, for  $a = 0$ , the number of  $(b, c)$  such that  $bc \equiv 0$  is

$$p + p - 1 = 2p - 1.$$

---

**Case B:**  $a \neq 0$

Then  $a$  can be any nonzero element in  $\{1, 2, \dots, p-1\}$ ; there are  $p-1$  such choices. In that scenario,  $a^2 \not\equiv 0 \pmod{p}$ . We want

$$bc \equiv a^2 \pmod{p}.$$

Over  $\mathbf{F}_p$ , a nonzero right-hand side  $a^2$  can be written as  $bc$  in the following way:

If  $b = 0$ , then  $bc = 0$ , which cannot equal  $a^2 \neq 0$ .

No solutions in that subcase.

If  $b \neq 0$ , then for each nonzero  $b$  there is **exactly one**  $c$  satisfying  $bc \equiv a^2 \pmod{p}$ , namely  $c \equiv a^2 b^{-1} \pmod{p}$ .

Since  $b$  can be any of the  $p - 1$  nonzero elements, we get  $p - 1$  solutions  $(b, c)$  for each nonzero  $a$ .

Thus, for each  $a \neq 0$ , there are  $p - 1$  pairs  $(b, c)$ . Hence, for  $p - 1$  values of  $a$ , the total number of solutions in this case is

$$(p - 1) \times (p - 1) = (p - 1)^2.$$

---

**Total number of solutions to  $a^2 = bc$**

Summing these two cases:

Case A ( $a = 0$ ):  $2p - 1$  solutions.

Case B ( $a \neq 0$ ):  $(p - 1)^2$  solutions.

Hence the total count of  $(a, b, c)$  satisfying  $a^2 \equiv bc$  is

$$(2p - 1) + (p - 1)^2 = (2p - 1) + (p^2 - 2p + 1) = p^2.$$

*A classic result : there are exactly  $p^2$  solutions to  $bc = a^2$  over  $\mathbf{F}_p^3$ .*

---

**Step 2: Count how many  $(a, b, c)$  satisfy  $a^2 \neq bc$**

Since there are  $p^3$  total triples  $(a, b, c)$ , the number that satisfy

$$a^2 \neq bc \pmod{p}$$

is simply

$$p^3 - p^2.$$

This directly corresponds to matrices  $A$  whose determinant  $a^2 - bc$  is  $\not\equiv 0 \pmod{p}$ .

Hence, the number of matrices in  $T_p$  with  $\det(A)$  **not** divisible by  $p$  is

$$p^3 - p^2.$$

The expression  $p^3 - p^2$  matches **Option D**.

### Question 057

Numerical

#### QUESTION

A 0.1 kg mass is suspended from a wire of negligible mass. The length of the wire is 1 m and its cross-sectional area is  $4.9$

×

$10^{-7} \text{ m}^2$ . If the mass is pulled a little in the vertically downward direction and released, it performs simple harmonic motion of angular frequency  $140 \text{ rad s}^{-1}$ . If the Young's modulus of the material of the wire is  $n$

×

$10^9 \text{ Nm}^{-2}$ , the value of  $n$  is

#### SOURCE

Physics • simple-harmonic-motion

#### EXPLANATION

When mass  $m$  is pulled by a force  $F$ , the wire elongation  $x$ , length  $l$ , cross-sectional area  $A$ , and Young's modulus of wire material  $Y$  are related by

$$Y = \frac{F/A}{x/l}$$

i.e.,

$$F = (Y A/l)x$$

The restoring force by the wire is equal but opposite to F i.e.,  $F_r =$

—

F. Apply Newton's second law to get

$$m d^2 x / dt^2 = -(YA/l)x = -\omega^2 x$$

This equation represents SHM with an angular frequency

$$\omega = \sqrt{YA/(lm)}$$

. Substitute the values to get

$$Y = \omega^2 lm / A = 4 \times 10^9$$

N/m<sup>2</sup>.

### Question 058 MCQ

#### QUESTION

A block of mass  $m$  is on an inclined plane of angle  $\theta$ . The coefficient of friction between the block and the plane is  $\mu$  and  $\tan \theta > \mu$ . The block is held stationary by applying a force  $P$  parallel to the plane. The direction of force pointing up the plane is taken to be positive. As  $P$  is varied from  $P_1 = mg \sin \theta - \mu \cos \theta$  to  $P_2 = mg \sin \theta + \mu \cos \theta$ , the frictional force  $f$  versus  $P$  graph will look like

A

B



D

#### CORRECT OPTION

A

#### SOURCE

Physics • laws-of-motion

#### EXPLANATION

The forces acting on the block are its weight  $mg$ , normal reaction  $N$ , applied force  $P$  and frictional force  $f$ .

Resolve  $mg$  along and normal to the inclined plane and apply Newton's second law to get

$$0 = P + f - mg \sin \theta$$

,

which gives

$$f = -P + mg \sin \theta$$

. ..... 1

This is a straight line with slope

—

1. Substitute the values of  $P_1$  and  $P_2$  in equation 1 to get the frictional force at those points i.e.,

$$f_1 = \mu mg \cos \theta$$

and

$$f_2 = -\mu mg \cos \theta$$



**Question 059****MCQ****QUESTION**

A point mass of 1 kg collides elastically with a stationary point mass of 5 kg. After their collision, the 1 kg mass reverses its direction and moves with a speed of  $2 \text{ ms}^{-1}$ . Which of the following statement *s* is *are* correct for the system of these two masses?

**A**Total momentum of the system is  $3 \text{ kg ms}^{-1}$ **B**Momentum of 5 kg mass after collision is  $4 \text{ kg ms}^{-1}$ **C**

Kinetic energy of the centre of mass is 0.75 J

**D**

Total kinetic energy of the system is 4 J

**CORRECT OPTION****A**Total momentum of the system is  $3 \text{ kg ms}^{-1}$ **SOURCE**

Physics • impulse-and-momentum

**EXPLANATION**

Here,  $m_1 = 1 \text{ kg}$ ,  $m_2 = 5 \text{ kg}$

$$u_1 = u, u_2 = 0$$

$$v_1 =$$

—

$$2 \text{ m s}$$

—

$$^1, v_2 = v$$

By the law of conservation of linear momentum, we get

$$m_1 u_1 + m_2 u_2 = m_1 u_1 + m_2 u_2$$

$$1 \times u + 5 \times 0 = 1 \times (-2) + 5 \times v$$

$$u = 5v - 2$$

..... *i*

By the definition of coefficient of restitution,

$$e = \frac{v_2 - v_1}{u_1 - u_2}$$

For a perfectly elastic collision,  $e = 1$

$\therefore$

$$1 = \frac{v + 2}{u}$$

or

$$u = v + 2$$

..... *ii*

Solving equations *i* and *ii*, we get

$$u = 3 \text{ m s}$$

—

$$^1, v = 1 \text{ m s}$$

1

Before collision,

Total momentum of the system = 1

×

3 + 5

×

0 = 3 kg m s

—

1

After collision,

Total momentum of the system = 1

×

$3 - 2 + 5$

×

1 = 3 kg m s

—

1

Hence, option *a* is correct.

Momentum of 5 kg mass after collision = 5

×

1 = 5 kg m s

—

1.

Hence, option *b* is incorrect.

Velocity of centre mass is

$$v_{cm} = \frac{1 \times 3 + 5 \times 0}{1 + 5} = \frac{1}{2}$$

m s

—

1

Kinetic energy of the centre of mass

$$= \frac{1}{2} m_{system} v_{CM}^2 = \frac{1}{2} \times 6 \times \left(\frac{1}{2}\right)^2 = 0.75$$

J

Hence, option *c* is correct.

Before collision,

Total kinetic energy of the system

$$= \frac{1}{2} \times 1 \times 3^2 + \frac{1}{2} \times 5 \times 0^2 = 4.5$$

J

After collision,

Total kinetic energy of the system

$$= \frac{1}{2} \times 1 \times (-2)^2 + \frac{1}{2} \times 5 \times (1)^2 = 4.5$$

J

Hence, option *d* is incorrect.

## Question 060

Numerical

QUESTION

A binary star consists of two stars A (mass  $2.2M_s$ ) and B (mass  $11M_s$ ), where  $M_s$  is the mass of the sun. They are separated by distance  $d$  and are rotating about their centre of mass, which is stationary. The ratio of the total angular momentum of the binary star to the angular momentum of star B about the centre of mass is

### SOURCE

Physics • gravitation

### EXPLANATION

Let stars A and B are rotating about their centre of mass with angular velocity

$$\omega$$

.

Let distance of stars A and B from the centre of mass be  $r_A$  and  $r_B$  respectively as shown in the figure.

Total angular momentum of the binary stars about the centre of mass is

$$L = M_A r_A^2 \omega + M_B r_B^2 \omega$$

Angular momentum of the star B about centre of mass is

$$L_B = M_B r_B^2 \omega$$

$$\therefore$$

$$\frac{L}{L_B} = \frac{(M_A r_A^2 + M_B r_B^2) \omega}{M_B r_B^2 \omega} = \left( \frac{M_A}{M_B} \right) \left( \frac{r_A}{r_B} \right)^2 + 1$$

Since

$$M_A r_A = M_B r_B$$

or,

$$\frac{r_A}{r_B} = \frac{M_B}{M_A}$$

$$\therefore$$

$$\frac{L}{L_B} = \frac{M_B}{M_A} + 1 = \frac{11M_S}{2.2M_S} + 1 = \frac{11 + 2.2}{2.2} = 6$$

## Question 061

### Numerical

#### QUESTION

Gravitational acceleration on the surface of a planet is

$$\frac{\sqrt{6}}{11}g$$

, where

$$g$$

is the gravitational acceleration on the surface of the earth. The average mass density of the planet is

$$\frac{2}{3}$$

times that of the earth. If the escape speed on the surface of the earth is taken to be  $11 \text{ kms}^{-1}$ , the escape speed on the surface of the planet in  $\text{kms}^{-1}$  will be

#### SOURCE

Physics • gravitation

#### EXPLANATION

On the planet,

$$g_p = \frac{GM_p}{R_p^2} = \frac{G}{R_p^2} \left( \frac{4}{3}\pi R_p^3 \rho_p \right) = \frac{4}{3}G\pi R_p \rho_p$$

On the earth,

$$g_e = \frac{GM_e}{R_e^2} = \frac{G}{R_e^2} \left( \frac{4}{3}\pi R_e^3 \rho_e \right) = \frac{4}{3}G\pi R_e \rho_e$$

$\therefore$

$$\frac{g_p}{g_e} = \frac{R_p \rho_p}{R_e \rho_e}$$

or

$$\frac{R_p}{R_e} = \frac{g_p \rho_p}{g_e \rho_e}$$

..... *i*

On the planet,

$$v_p = \sqrt{2g_p R_p}$$

On the earth,

$$v_e = \sqrt{2g_e R_e}$$

$\therefore$

$$\frac{v_p}{v_e} = \sqrt{\frac{g_p R_p}{g_e R_e}} = \frac{g_p}{g_e} \sqrt{\frac{\rho_e}{\rho_p}}$$

*Using(i)*

Here,

$$\rho_p = \frac{2}{3} \rho_e$$

,

$$g_p = \frac{\sqrt{6}}{11} g_e$$

$\therefore$

$$\frac{v_p}{v_e} = \frac{\sqrt{6}}{11} \sqrt{\frac{3}{2}}$$

or,

$$v_p = 11 \times \frac{\sqrt{6}}{11} \times \sqrt{\frac{3}{2}}$$

(

∴

$$v_e = 11 \text{ km s}^{-1}$$

—

<sup>1</sup> *Given*)

$$= 3 \text{ km s}^{-1}$$

—

1

### Question 062 MCQ

#### QUESTION

A real gas behaves like an ideal gas if its

- ☐ A pressure and temperature are both high
- ☐ B pressure and temperature are both low
- ☐ C pressure is high and temperature is low
- ☐ D pressure is low and temperature is high

#### CORRECT OPTION

- ☒ D pressure is low and temperature is high



### SOURCE

Physics • heat-and-thermodynamics

### EXPLANATION

In an ideal gas, the average force of attraction between the molecules and volume of the molecules *incomparisontovolumeofthegas* are negligibly small. These conditions are satisfied for a real gas when pressure is low and temperature is high.

## Question 063

Numerical

### QUESTION

Two spherical bodies A *radius*6cm and B *radius*18cm are at temperature  $T_1$  and  $T_2$ , respectively. The maximum intensity in the emission spectrum of A is at 500 nm and in that of B is at 1500 nm. Considering them to be black bodies, what will be the ratio of the rate of total energy radiated by A to that of B?

### SOURCE

Physics • heat-and-thermodynamics

### EXPLANATION

According to Wien's displacement law,

$$\lambda_m T = \text{constant}$$

$$\lambda_m T = \text{constant}$$

$$\therefore$$

$$(\lambda_m)_A T_A = (\lambda_m)_B T_B$$

or,

$$\frac{T_A}{T_B} = \frac{(\lambda_m)_B}{(\lambda_m)_A} = \frac{1500 \text{ nm}}{500 \text{ nm}}$$

or

$$\frac{T_A}{T_B} = 3$$

.....  $i$

According to Stefan Boltzmann law, rate of energy radiated by a black body

$$E = \sigma AT^4 = \sigma 4\pi R^2 T^4$$

$$\text{Here, } A = 4\pi R^2$$

$\therefore$

$$\frac{E_A}{E_B} = \left(\frac{R_A}{R_B}\right)^2 \left(\frac{T_A}{T_B}\right)^4 = \left(\frac{6 \text{ cm}}{18 \text{ cm}}\right)^2 (3)^4$$

Using (i)

$$= 9$$

### Question 064 Numerical

#### QUESTION

A piece of ice (heat capacity =  $2100 \text{ J kg}^{-1} \text{ }^\circ\text{C}^{-1}$  and latent heat =  $3.36$

$\times$

$10^5 \text{ J kg}^{-1}$ ) of mass  $m$  grams is at  $-5 \text{ }^\circ\text{C}$  at atmospheric pressure. It is given  $420 \text{ J}$  of heat so that the ice starts melting. Finally when the ice-water mixture is in equilibrium, it is found that  $1 \text{ gm}$  of ice has melted. Assuming there is no other heat exchange in the process, the value of  $m$  is

#### SOURCE

Physics • heat-and-thermodynamics

### EXPLANATION

In the final state, ice-water mixture is in equilibrium. Thus, the temperature of  $m$  grams of ice is raised from

5

C to 0

C. The heat absorbed in this process is

$$Q_1 = mS$$

T. .... 1

The state of  $m_1 = 1$  g of ice is changed from solid to liquid. The heat absorbed in the melting process is

$$Q_2 = m_1 L. \dots\dots\dots 2$$

The heat supplied is  $Q = 420$  J. By energy conservation,  $Q = Q_1 + Q_2$ . Substitute  $Q_1$  and  $Q_2$  from equations 1 and 2 to get

$$m = \frac{Q - m_1 L}{S \Delta T} = \frac{420 - (10^{-3})(3.36 \times 10^5)}{(2100)(5)} \\ = 8 \times 10^{-3}$$

kg = 8 g.

### QUESTION

A few electric field lines for a system of two charges

$$Q_1$$

and

$$Q_2$$

fixed at two different points on the

$$x$$

-axis are shown in the figure. These lines suggest that

A

$$|Q_1| > |Q_2|$$

B

$$|Q_1| < |Q_2|$$

C

at a finite distance to the left of

$$Q_1$$

the electric field is zero

D

at a finite distance to the right of

$$Q_2$$

the electric field is zero

### CORRECT OPTION

$$|Q_1| > |Q_2|$$

## SOURCE

Physics • electrostatics

## EXPLANATION

Number of electric field lines originating from  $Q_1$  is more than terminating at  $Q_2$ .

$\therefore$

$$|Q_1| > |Q_2|$$

Here,  $Q_1$  is positive while  $Q_2$  is negative.

The electric field at a distance  $x$  towards the right of  $Q_2$  is given by

$$|\vec{E}| = \frac{1}{4\pi\epsilon_0} \frac{Q_1}{(d+x)^2} - \frac{1}{4\pi\epsilon_0} \frac{Q_2}{x^2}$$

,

where  $d$  is the separation between  $Q_1$  and  $Q_2$ . Since

$$Q_1 > Q_2$$

,

$$|\vec{E}|$$

becomes zero for some finite  $x$ .

## Question 066 MCQ

### QUESTION

A student uses a simple pendulum of exactly 1m length to determine  $g$ , the acceleration due to gravity. He uses a stop watch with the least count of 1 sec for

this and records 40 seconds for 20 oscillations. For this observation, which of the following statement *s* is *are* true?

**A** Error  $\Delta T$  in measuring  $T$ , the time period, is 0.05 seconds

**B** Error  $\Delta T$  in measuring  $T$ , the time period, is 1 second

**C** Percentage error in the determination of  $g$  is 5%

**D** Percentage error in the determination of  $g$  is 2.5%

#### CORRECT OPTION

**C** Percentage error in the determination of  $g$  is 5%

#### SOURCE

Physics • units-and-measurements

#### EXPLANATION

Relative error in measurement of time,

$$\frac{\Delta t}{t} = \frac{1 \text{ s}}{40 \text{ s}} = \frac{1}{40}$$

Time period,

$$T = \frac{40 \text{ s}}{20} = 2 \text{ s}$$

Error in measurement of time period,

$$\Delta T = T \times \frac{\Delta t}{t} = 2 \text{ s} \times \frac{1}{40} = 0.05 \text{ s}$$

The time period of simple pendulum is

$$T = 2\pi\sqrt{\frac{l}{g}}$$

or,

$$T^2 = \frac{4\pi^2 l}{g}$$

or,

$$g = \frac{4\pi^2 l}{T^2}$$

$\therefore$

$$\frac{\Delta g}{g} = \frac{2\Delta T}{T} = 2 \times \frac{1}{40} = \frac{1}{20}$$

$$\therefore \frac{\Delta T}{T} = \frac{\Delta t}{t}$$

Percentage error in determination of g is

$$\frac{\Delta g}{g} \times 100 = \frac{1}{20} \times 100 = 5\%$$

### Question 067 MCQ

#### QUESTION

Incandescent bulbs are designed by keeping in mind that the resistance of their filament increases with the increase in temperature. If at room temperature, 100, 60 and 40 W bulbs have filament resistances  $R_{100}$ ,  $R_{60}$  and  $R_{40}$  respectively, the relation between these resistances is

A

$$\frac{1}{R_{100}} = \frac{1}{R_{40}} + \frac{1}{R_{60}}$$

**B**

$$R_{100} = R_{40} + R_{60}$$

**C**

$$R_{100} > R_{60} > R_{40}$$

**D**

$$\frac{1}{R_{100}} > \frac{1}{R_{60}} > \frac{1}{R_{40}}$$

**CORRECT OPTION****D**

$$\frac{1}{R_{100}} > \frac{1}{R_{60}} > \frac{1}{R_{40}}$$

**SOURCE**

Physics • current-electricity

**EXPLANATION**

The power of the bulb is

$$P = \frac{V^2}{R}$$

Therefore,

$$100 = \frac{V^2}{R_{100}} \Rightarrow \frac{1}{R_{100}} = \frac{100}{V^2}$$

where  $R_{100}$  is the resistance *at any temperature* corresponds to 100 W.

Similarly,

$$60 = \frac{V^2}{R_{60}} \Rightarrow \frac{1}{R_{60}} = \frac{60}{V^2}$$



and

$$40 = \frac{V^2}{R_{40}} \Rightarrow \frac{1}{R_{40}} = \frac{40}{V^2}$$

From these equations, we get

$$P_{100} > P_{60} > P_{40} \Rightarrow \frac{1}{R_{100}} > \frac{1}{R_{60}} > \frac{1}{R_{40}}$$

### Question 068 MCQ

#### QUESTION

To verify Ohm's law, a student is provided with a test resistor  $R_T$ , a high resistance  $R_1$ , a small resistance  $R_2$ , two identical galvanometers  $G_1$  and  $G_2$ , and a variable voltage source  $V$ . The correct circuit to carry out the experiment is

A

B

C

D

#### CORRECT OPTION

C

#### SOURCE

**EXPLANATION**

To verify Ohm's law, we need to measure the voltage across the test resistance  $R_T$  and current passing through it. The voltage can be measured by connecting a high resistance  $R_1$  in series with galvanometer. This combination becomes a voltmeter and should be connected in parallel to  $R_T$ . The current can be measured by connecting a low resistance  $R_2$  *shunt* in parallel with galvanometer. This combination becomes an ammeter and should be connected in series to measure the current through  $R_T$ .

**Question 069****MCQ****QUESTION**

An AC voltage source of variable angular frequency

$$\omega$$

and fixed amplitude  $V_0$  is connected in series with a capacitance  $C$  and an electric bulb of resistance  $R$  *inductance zero*. When

$$\omega$$

is increased

**A**

the bulb glows dimmer.

**B**

the bulb glows brighter.

**C**

total impedance of the circuit is unchanged.

**D** total impedance of the circuit increases.

**CORRECT OPTION**

**B** the bulb glows brighter.

**SOURCE**

Physics • alternating-current

**EXPLANATION**

Impedance of the circuit,

$$Z = \sqrt{(X_C)^2 + (R)^2} = \sqrt{\left(\frac{1}{\omega C}\right)^2 + R^2}$$

As

$$\omega$$

increases,  $Z$  decreases.

Current in the circuit,

$$I = \frac{V_0}{Z}$$

When

$$\omega$$

is increased, the impedance of the circuit decreases and the current through the bulb increases. Therefore the bulb glows brighter.

### QUESTION

A thin flexible wire of length  $L$  is connected to two adjacent fixed points and carries a current  $I$  in the clockwise direction, as shown in the figure. When the system is put in a uniform magnetic field of strength  $B$  going into the plane of the paper, the wire takes the shape of a circle. The tension in the wire is

A  $IBL$

B

$$\frac{IBL}{\pi}$$

C

$$\frac{IBL}{2\pi}$$

D

$$\frac{IBL}{4\pi}$$

### CORRECT OPTION

C

$$\frac{IBL}{2\pi}$$

### SOURCE

Physics • magnetism

### EXPLANATION

Consider an small element AB of length  $dl$  of the circle of radius  $R$  subtending an angle

$$\theta$$

at the centre O.

If  $T$  is the tension in the wire, then force towards the centre will be equal to

$$2T \sin \left( \frac{\theta}{2} \right)$$

which is balanced by outward magnetic force on the current carrying element

$$(= I dl B )$$

$$2T \sin \left( \frac{\theta}{2} \right) = I dl B$$

For small angle

$$\theta$$

,

$$\sin \frac{\theta}{2} \approx \frac{\theta}{2}$$

or,

$$T = \frac{IBdl}{\theta} = IBR$$

$$\therefore \theta = \frac{dl}{R}$$

$$= \frac{IBL}{2\pi}$$

$$\therefore R = \frac{L}{2\pi}$$

Question 071

MCQ

### QUESTION

A thin uniform annular disc *see figure* of mass  $M$  has outer radius  $4R$  and inner radius  $3R$ . The work required to take a unit mass from point  $P$  on its axis to infinity is

A

$$\frac{2GM}{7R}(4\sqrt{2} - 5)$$

B

$$-\frac{2GM}{7R}(4\sqrt{2} - 5)$$

C

$$\frac{GM}{4R}$$

D

$$\frac{2GM}{5R}(\sqrt{2} - 1)$$

### CORRECT OPTION

A

$$\frac{2GM}{7R}(4\sqrt{2} - 5)$$

### SOURCE

Physics • gravitation

### EXPLANATION

We need to find gravitational potential energy of a unit mass placed at the point P.

The surface mass density of the annular disc is

$$\sigma = M/(16\pi R^2 - 9\pi R^2) = M/(7\pi R^2)$$

.

Consider a small ring of radius  $r$  and thickness  $dr$ . The mass of the ring is  $dm = 2$

$\pi$

$r$

$\sigma$

dr. As distance of any point of the ring from P is same, the potential at P due to the ring is

$$V_P = -\frac{G(2\pi r\sigma dr)}{\sqrt{16R^2 + r^2}}$$

Integrate from  $r = 3R$  to  $r = 4R$  to get the potential energy of the unit mass placed at P

$$V_P = \int_{3R}^{4R} -\frac{GdM}{\sqrt{(4R)^2 + (r)^2}} = -\frac{GM2\pi}{7\pi R^2} \int_{3R}^{4R} \frac{rdr}{\sqrt{16R^2 + r^2}}$$

Solving, we get

$$V_P = -\frac{GM2\pi}{7\pi R^2} \left[ \sqrt{16R^2 + r^2} \right]_{3R}^{4R} = -\frac{2GM}{7R} (4\sqrt{2} - 5)$$

Work done in moving a unit mass from P to

$\infty$

= V

$\infty$

—

$V_P$

$$= 0 - \left( \frac{-2GM}{7R} (4\sqrt{2} - 5) \right) = \frac{2GM}{7R} (4\sqrt{2} - 5)$$

### Question 072 MCQ

#### QUESTION

Consider a thin square sheet of side  $L$  and thickness, made of a material of resistivity

$$\rho$$

. The resistance between two opposite faces, shown by the shaded areas in the figure is

- ☐ A directly proportional to  $L$ .
- ☐ B directly proportional to  $t$ .
- ☐ C independent to  $L$ .
- ☐ D independent of  $t$ .

#### CORRECT OPTION

- ☒ C independent to  $L$ .

#### SOURCE

Physics • current-electricity



### EXPLANATION

The sheet is of square shape with thickness  $t$ , width

$w$

$= L$ , and length  $l = L$ . The resistance between the two opposite faces is given by

$$R = \frac{\rho l}{A} = \frac{\rho l}{wt} = \frac{\rho L}{Lt} = \frac{\rho}{t}$$

### Question 073 MCQ

#### QUESTION

One mole of an ideal gas in initial state A undergoes a cyclic process ABCA, as shown in the figure. Its pressure at A is  $P_0$ . Choose the correct option  $s$  from the following:

- A** Internal energies at A and B are the same.
- B** Work done by the gas in process AB is  $P_0 V_0 \ln 4$ .
- C** Pressure at C is  $P_0/4$ .
- D** Temperature at C is  $T_0/4$ .

#### CORRECT OPTION

A

Internal energies at A and B are the same.

**SOURCE**

Physics • heat-and-thermodynamics

**EXPLANATION**

The internal energy of one mole of an ideal gas at temperature  $T$  is given by  $U = 3RT/2$ . The process AB is isothermal i.e.,  $T_A = T_B = T_0$ , which makes  $U_A = U_B$ .

The work done in the isothermal process AB is

$$W_{AB} = nRT \ln(V_B/V_A)$$

$$= nRT_0 \ln(4V_0/V_0) = p_0V_0 \ln 4.$$

Information regarding  $p$  and  $T$  at C can not be obtained from the given graph. Unless it is mentioned that line BC passes through origin or not.

Hence, the correct options are  $a$  and  $b$ .

Note:

If we assume that the line BC pass through the origin. In the process BC, the slope  $V/T$  is constant. Thus, BC is an isobaric process

*as  $V/T = nR/p$ , by the ideal gas equation.* Thus,

$$p_C = p_B = RT_B / V_B$$

$$= RT_0/(4V_0) = (RT_0/V_0)/4 = p_0/4.$$

Apply the ideal gas equation for the states A and C to get

$$T_C = \frac{p_C V_C}{p_0 V_0} T_0 = \frac{(p_0/4) V_0}{p_0 V_0} T_0 = T_0/4$$

.

## QUESTION

A ray OP of monochromatic light is incident on the face AB of prism ABCD near vertex B at an incident angle of  $60^\circ$

○

*see figure*. If the refractive index of the material of the prism is

$$\sqrt{3}$$

, which of the following is *are* correct?

A

The ray gets totally internally reflected at face CD.

B

The ray comes out through face AD.

C

The angle between the incident ray and the emergent ray is  $90^\circ$

○

D

The angle between the incident ray and the emergent ray is  $120^\circ$

○

## CORRECT OPTION

A

The ray gets totally internally reflected at face CD.

## SOURCE

**EXPLANATION**

Consider the refraction at the face AB. Snell's law,

$$\sin i / \sin r = \sin 60^\circ / \sin r = \sqrt{3}$$

,

gives

$$r = 30^\circ$$

.

The geometry in BCQP gives

$$\angle BPQ = 30^\circ + 90^\circ = 120^\circ$$

$$\angle CQP = 360^\circ - (135^\circ + 60^\circ + 120^\circ) = 45^\circ$$

.

Thus, the angle of incidence at Q is

$$i_1 = 45^\circ$$

. The critical angle for prism to air refraction is given by

$$\sin i_c = 1/\sqrt{3}$$

. Since

$$\sin i_1 = 1/\sqrt{2} > 1/\sqrt{3}$$

, we get

$$i_1 > i_c$$

i.e., the angle of incidence is greater than the critical angle. Thus, the ray undergoes total internal reflection at Q. The laws of reflection gives

$$r_1 = i_1 = 45^\circ$$

. In triangle QRD,

$$\angle QRD = 60^\circ$$

and hence the angle of incidence at R is

$$i_2 = 30^\circ$$

.

Applying Snell's law at face AD, we get

$$\sqrt{3} \times \sin 30^\circ = 1 \times \sin e$$

or,

$$\sqrt{3} \times \frac{1}{2} = \sin e$$

$$\sin e = \frac{\sqrt{3}}{2}$$

or,

$$e = \sin^{-1} \left( \frac{\sqrt{3}}{2} \right) = 60^\circ$$

From figure,

The angle between the incident ray and the emergent ray is 90

◦

.

Hence, option *c* is correct and option *d* is incorrect.

Note : Angle between incident and emergent rays is the same as the angle between the two faces = 90

◦

.

## QUESTION

In the graph below, the resistance  $R$  of a superconductor is shown as a function of its temperature  $T$  for two different magnetic fields  $B_1$  *solidline* and  $B_2$  *dashedline*. If  $B_2$  is larger than  $B_1$  which of the following graphs shows the correct variation of  $R$  with  $T$  in these fields?

A

B

C

D

## CORRECT OPTION

A

## SOURCE

Physics • magnetism

## EXPLANATION

From the given figure,  $T_c B$  is a monotonically decreasing function of  $B$ . Thus,  $B_2 > B_1$  implies  $T_c(B_2) < T_c(B_1)$ . Hence resistance with  $B_2$  will become zero at lower temperature in comparison to  $B_1$ .

## QUESTION

A superconductor has  $T_c 0 = 100$  K. When a magnetic field of 7.5 T is applied, its  $T_c$  decreases to 75 K. For this material, one can definitely say that when

A

 $B = 5$  T,  $T_c B = 80$  K

B

 $B = 5$  T,  $75$  K  $< T_c B < 100$  K

C

 $B = 10$  T,  $75$  K  $< T_c < 100$  K

D

 $B = 10$ ,  $T_c = 70$  K

## CORRECT OPTION

B

 $B = 5$  T,  $75$  K  $< T_c B < 100$  K

## SOURCE

Physics • magnetism

## EXPLANATION

It is given that  $T_c 0 = 100$  K and  $T_c 7.5 = 75$  K. Since  $T_c B$  is a monotonically decreasing function of  $B$ ,  $T_c 5 < T_c 0$  and  $T_c 5 > T_c 7.5$ . Thus,  $75$  K  $< T_c 5 < 100$  K.

## QUESTION

If the total energy of the particle is  $E$ , it will perform periodic motion only if

**A**  $E < 0$

**B**  $E > 0$

**C**  $V_0 > E > 0$

**D**  $E > V_0$

## CORRECT OPTION

**C**  $V_0 > E > 0$

## SOURCE

Physics • simple-harmonic-motion

## EXPLANATION

The kinetic energy of the particle cannot be negative. The total energy  $E$  is the sum of the kinetic energy  $Kx$  and potential energy  $Vx$  i.e.,

$$E = Kx + Vx \dots\dots\dots 1$$

From the given figure,  $Vx$

$$\geq$$



0 for all  $x$ . If  $E$

$$\leq$$

0 for some  $x$  then kinetic energy  $Kx = E$

$$-$$

$Vx$

$$\leq$$

0 for those  $x$  and hence the motion of the particle is not allowed for those  $x$ . On the other hand, if  $E$

$$\geq$$

$V_0$  then  $Kx = E$

$$-$$

$Vx$

$$\geq$$

0 for all  $x$  and particle is allowed to move for all  $x$ , including infinity  
*in this case the particle will escape to infinity*. Thus, for the particle to have a periodic motion,  $V_0 > E > 0$ . In this case, particle is allowed at points where

$Kx = E$

$$-$$

$Vx = E$

$$-$$

$$\alpha$$

$x^4$

$$\geq$$

0 i.e.,

$|x|$

$$\leq$$

$$E \propto \alpha^{1/4}.$$

**Question 078** MCQ**QUESTION**

For periodic motion of small amplitude  $A$ , the time period  $T$  of this particle is proportional to

**A**

$$A\sqrt{m/\alpha}$$

**B**

$$\frac{1}{A}\sqrt{m/\alpha}$$

**C**

$$A\sqrt{\alpha/m}$$

**D**

$$\frac{1}{A}\sqrt{\alpha/m}$$

**CORRECT OPTION****B**

$$\frac{1}{A}\sqrt{m/\alpha}$$

**SOURCE**

## EXPLANATION

As

$$V = ax^4$$

$$[\alpha] = \frac{[V]}{[x^4]} = \frac{[ML^2T^{-2}]}{[L^4]} = [ML^{-2}T^{-2}]$$

By method of dimensions,

$$\left[ \frac{1}{A} \sqrt{\frac{m}{\alpha}} \right] = \frac{[M]^{1/2}}{[L][ML^{-2}T^{-2}]^{1/2}} = [T]$$

Only option *b* has the dimensions of time.

## Question 079 MCQ

## QUESTION

The acceleration of this particle for

$$|x| > X_0$$

is

A

proportional to  $V_0$ .

B

proportional to  $V_0/mX_0$ .

proportional to

C

$$\sqrt{V_0/mX_0}$$

D

zero.

### CORRECT OPTION

D

zero.

### SOURCE

Physics • simple-harmonic-motion

### EXPLANATION

For

$$|x| > X_0$$

,

$$V = V_0$$

= constant

Force

$$= -\frac{dV}{dx} = 0$$

Hence, acceleration of the particle is zero for

$$|x| > X_0$$

.

## QUESTION

A stationary source is emitting sound at a fixed frequency  $f_0$ , which is reflected by two cars approaching the source. The difference between the frequencies of sound reflected from the cars is 1.2% of  $f_0$ . What is the difference in the speeds of the cars *in km per hour* to the nearest integer? The cars are moving at constant speeds much smaller than the speed of sound which is 330 ms

—

1.

## SOURCE

Physics • waves

## EXPLANATION

Let car B be the observer *moving towards S*.

The frequency observed is

$$f_1 = f_0 \left( \frac{c + v}{c} \right)$$

When sound gets reflected, the frequency observed by source S is

$$f_2 = f_1 \left( \frac{c}{c - v} \right)$$

where  $v$  is the speed of car and  $c$  is the speed of sound. Therefore,

$$f_2 = f_0 \left( \frac{c + v}{c - v} \right)$$

Now,

$$df_x = f_0 \left[ \frac{(c-v)dv - (c+v)(-dv)}{(c-v)^2} \right]$$

$$= \frac{2f_0 c dv}{(c-v)^2}$$

That is,

$$\frac{2f_0 c dv}{(c-v)^2} = \left( \frac{1.2}{100} \right) f_0$$

$$\Rightarrow dv = \frac{1.2}{100} \times \frac{(c-v)^2}{2c}$$

Since,  $v \ll c$ , we get  $c$

—

$v$

$\simeq$

$c$ .

Therefore,

$$dv = \frac{1.2}{100} \times \frac{c}{2} = 1.98$$

m/s

$$= 1.98 \times \frac{18}{5}$$

km/h = 7 km/h.

## Question 081

Numerical

### QUESTION

The focal length of a thin biconvex lens is 20 cm. When an object is moved from a distance of 25 cm in front of it to 50 cm, the magnification of its image changes from  $m_{25}$  to  $m_{50}$ . The ratio

$$\frac{m_{25}}{m_{50}}$$

is \_\_\_\_\_.

#### SOURCE

Physics • geometrical-optics

#### EXPLANATION

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$

or,

$$\frac{u}{v} - 1 = \frac{u}{f}$$

or,

$$\frac{u}{v} = \left( \frac{u + f}{f} \right)$$

$\therefore$

$$m = \frac{v}{u} = \left( \frac{f}{u + f} \right)$$

$$\frac{m_{25}}{m_{50}} = \frac{\left( \frac{20}{-25+20} \right)}{\left( \frac{20}{-50+20} \right)} = 6$$

### QUESTION

An

$\alpha$

-particle and a proton are accelerated from the rest by a potential difference of 100 V. After this, their de Broglie wavelengths are

$\lambda$

$\alpha$

and

$\lambda$

$p$ , respectively. The ratio

$$\frac{\lambda_p}{\lambda_\alpha}$$

, to the nearest integer, is \_\_\_\_\_.

### SOURCE

Physics • dual-nature-of-radiation

### EXPLANATION

The de Broglie wavelength of a particle with momentum  $p$  is given by

$\lambda$

$= h/p$ .

The momentum and kinetic energy of a particle of mass  $m$  are related by

$$p = \sqrt{2mK}$$

.



The kinetic energy of a charge  $q$ , accelerated through potential  $V$ , is given by  $K = qV$ . Thus,

$$\lambda = h/\sqrt{2mK} = h/\sqrt{2mqV}$$

,

which gives

$$\begin{aligned}\frac{\lambda_p}{\lambda_\alpha} &= \sqrt{\frac{2m_\alpha q_\alpha V}{2m_p q_p V}} = \sqrt{\frac{2 \cdot 4u \cdot 2e \cdot 100}{2 \cdot 1u \cdot 1e \cdot 100}} \\ &= \sqrt{8} = 2.8 \approx 3\end{aligned}$$

### Question 083

Numerical

#### QUESTION

When two identical batteries of internal resistance  $1\ \Omega$

$$\Omega$$

each are connected in series across a resistor  $R$ , the rate of heat produced in  $R$  is  $J_1$ . When the same batteries are connected in parallel across  $R$ , the rate is  $J_2$ . If  $J_1 = 2.25 J_2$ , then the value of  $R$  in

$$\Omega$$

is \_\_\_\_\_.

#### SOURCE

Physics • current-electricity

#### EXPLANATION

In series : When the batteries are connected in series, we have

$$J_1 = \left( \frac{2E}{R+2} \right)^2 R$$

In parallel : When the batteries are connected in parallel, we have

$$J_2 = \left( \frac{E}{R + (1/2)} \right)^2 R$$

It is given that,

$$\frac{J_1}{J_2} = 2.25$$

$$\Rightarrow \frac{4}{(R+2)^2} \times \frac{(2R+1)^2}{4} = 2.25 \Rightarrow \frac{2R+1}{R+2} = 1.5$$

$$\Rightarrow 2R + 1 = 1.5R + 3 \Rightarrow 0.5R = 2$$

Therefore,

$$R = 4\Omega$$

#### Question 084 Numerical

##### QUESTION

When two progressive waves

$$y_1 = 4 \sin(2x - 6t)$$

and

$$y_2 = 3 \sin \left( 2x - 6t - \frac{\pi}{2} \right)$$

are superimposed, the amplitude of the resultant wave is \_\_\_\_\_.

## SOURCE

Physics • waves

## EXPLANATION

Here,

$$y_1 = 4 \sin(2x - 6t)$$

$$y_2 = 3 \sin \left( 2x - 6t - \frac{\pi}{2} \right)$$

The phase difference between two waves is

$$\phi = \frac{\pi}{2}$$

The amplitude of the resultant wave is

$$\begin{aligned} A &= \sqrt{A_1^2 + A_2^2 + 2A_1A_2 \cos \phi} \\ &= \sqrt{4^2 + 3^2 + 2 \times 4 \times 3 \times \cos \frac{\pi}{2}} = 5 \end{aligned}$$