

Homework #5

PHYS 4D: Modern Physics

Donald Aingworth IV

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1 Questions

1.1 Question 1

Give the operators corresponding to the momentum and energy of a particle.

1.1.1 Answer

$$\hat{p} = -i\hbar \frac{\partial}{\partial x}; \hat{H} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial t^2} + V(x, t) \quad (1)$$

1.2 Question 2

Give the general form of an eigenvalue equation.

1.2.1 Answer

For operator \hat{A} , eigenfunction ψ , and eigenvalue A .

$$\hat{A}\psi = A\psi \quad (2)$$

1.3 Question 3

What significance do the eigenvalues have?

1.3.1 Answer

The eigenvalue is the formula for or value of the component being searched for within the bounds of the eigenfunction.

1.4 Question 4

What significance do the eigenfunctions have?

1.4.1 Answer

The eigenfunctions serve as the conditions governing the system.

1.5 Question 5

Write down the momentum eigenvalue equation.

1.5.1 Answer

This is a *solved* momentum eigenvalue equation.

$$-i\hbar \frac{\partial}{\partial x} \psi = \hbar k \psi \quad (3)$$

1.6 Question 6

Is the function $\cos(kx)$ an eigenfunction of the momentum operator?

1.6.1 Answer

No. The first derivative of $\cos(kx)$ is $-k \sin(kx)$, which does not contain $\cos(kx)$ to the point that the sets of terms in the two equations are practically disjoint.

1.7 Question 7

Is the function $\cos(kx)$ an eigenfunction of the operator corresponding to the kinetic energy?

1.7.1 Answer

Yes. The energy operator involves a second derivative. Taking the second derivative of $\cos(kx)$, we get $-k^2 \cos(kx)$. This does contain $\cos(kx)$, so it is an eigenfunction.

1.8 Question 8

An electron is described by the wave function $\psi(x) = Ae^{i\alpha x}$, where α denotes the Greek letter alpha. What is the momentum of the electron?

1.8.1 Answer

Use the equation with the momentum operator.

$$\hat{p}\psi = p\psi \quad (4)$$

Plug in the momentum operator $\hat{p} = -i\hbar \frac{\partial}{\partial x}$ and $\psi = Ae^{i\alpha x}$.

$$\hat{p}\psi = -i\hbar \frac{\partial}{\partial x} Ae^{i\alpha x} = \hbar\alpha Ae^{i\alpha x} \quad (5)$$

This does contain ψ , so we can divide that out of the equation.

$$p\psi = \hbar\alpha Ae^{i\alpha x} \quad (6)$$

$$\boxed{p = \hbar\alpha} \quad (7)$$

1.9 Question 9

How would you determine whether or not a particular wave function represented a state of the system having a definite value of the momentum?

1.9.1 Answer

Use the eigenfunction equation for the momentum ($\hat{p}\psi$). If the result is a multiple of ψ , then the wave function did represent a state of the system with a definitive momentum value.

1.10 Question 10

Calculate the dot product of the following two vectors.

$$A = \begin{pmatrix} 4 \\ 3 \\ 2 \\ 1 \end{pmatrix}; B = \begin{pmatrix} 1 \\ 1 \\ 2 \\ 2 \end{pmatrix} \quad (8)$$

1.10.1 Answer

$$A \cdot B = 4 * 1 + 3 * 1 + 2 * 2 + 1 * 2 \quad (9)$$

$$= 4 + 6 + 4 + 2 = \boxed{16} \quad (10)$$

1.11 Question 11

For the two vectors, A and B, given in the preceding question, evaluate item by item BA and from there $A \cdot BA$.

1.11.1 Answer

$$BA = \begin{pmatrix} 4 * 1 \\ 3 * 1 \\ 2 * 2 \\ 1 * 2 \end{pmatrix} = \boxed{\begin{pmatrix} 4 \\ 3 \\ 4 \\ 2 \end{pmatrix}} \quad A \cdot BA = \begin{pmatrix} 4 \\ 3 \\ 2 \\ 1 \end{pmatrix} \cdot \begin{pmatrix} 4 \\ 3 \\ 4 \\ 2 \end{pmatrix} = 16 + 9 + 8 + 2 = \boxed{35} \quad (11)$$

1.12 Question 12

Suppose a function $f(x)$ is defined in an interval between 0 and 10 and an equally spaced grid is created by the command,

`x = linspace(0.0, 10.0, 10)` and suppose the values of the function at the two Gauss quadrature points, ξ_{i1} and ξ_{i2} , within each interval are known. Write down a formula for determining the value of the following integral in terms of the values of the function at the Gauss points.

$$\int_0^{10} f(x) dx \quad (12)$$

1.13 Question 13

Suppose a differential equation is solved using spline collocation on a grid created by the command `x = linspace(0.0, 10.0, 10)`. How many rows and columns do the A, B, and C matrices have?

1.14 Question 14

Sketch the wave function for an electron incident upon a potential step when the energy E of the electron is greater than the step height V_0 .

1.15 Question 15

Sketch the wave function for an electron incident upon a potential step when the energy E of the electron is less than the step height V_0 .

1.16 Question 16

Is the wavelength of a particle that has passed over a potential barrier greater than or less than the wavelength of the incident particle?

1.16.1 Answer

While in the barrier, the wavelength would be larger. After passing through the barrier, the wavelength would be equal.

1.17 Question 17

Sketch the wave function of an electron which tunnels through a potential barrier located between 0 and L.

1.18 Question 18

Write down the Heisenberg uncertainty relations for the position and the momentum and for the time and the energy.

1.18.1 Answer

$$\Delta x \Delta p = \frac{\hbar}{2}; \Delta t \Delta E = \frac{\hbar}{2} \quad (13)$$

1.19 Question 19

Use the Heisenberg relation for the time and the energy to describe how the energy profile of an excited atomic state depends upon the lifetime of the state.

1.20 Question 20

How would the wave function $\psi(x)$ of a particle change as the width of the Fourier transform increases?

1.20.1 Answer

It would decrease.

1.21 Question 21

Suppose that a particle localized between a and b is described by the wave function $\psi(x)$. Write down an equation for the average value of the kinetic energy of the particle.

1.21.1 Answer

Assume that the operator for kinetic energy is \hat{K} .

$$\langle K \rangle = \int_a^b \psi^*(x) \hat{K} \psi(x) dx \quad (14)$$