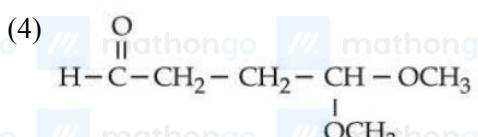
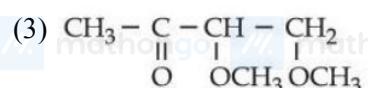
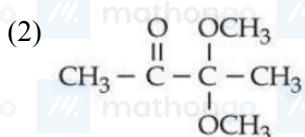
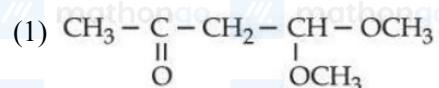


**Q1. 21 January Shift 1**

An organic compound " P " of molecular formula  $C_6H_{12}O_3$  gives positive Iodoform test but negative Tollen's test.

When " P " is treated with dilute acid, it produces " Q ". " Q " gives positive Tollen's test and also iodoform test. The structure of " P " is :

**Q2. 21 January Shift 2**

Match List - I with List - II.

List - I

Reagents

A.  $H_2$ , Pd –  $BaSO_4$

B.  $SnCl_2$ ,  $HCl$

C.  $CrO_2Cl_2$ ,  $CS_2$

D.  $CO$ ,  $HCl$ , Anhyd.  $AlCl_3$

List - II

Reaction Name (Involving aldehydes)

I. Etard Reaction

II. Rosenmund Reduction

III. Gatterman - Koch Reaction

IV. Stephen Reaction

Choose the correct answer from the options given below :

(1) A-IV, B-III, C-I, D-II

(2) A-II, B-IV, C-I, D-III

(3) A-IV, B-I, C-II, D-III

(4) A-II, B-III, C-IV, D-I

**Q3. 22 January Shift 1**

Match the LIST-I with LIST-II

	<b>List-I (Reagents)</b>		<b>List-II (Name of reaction involving carbonyl compounds)</b>
A.	$NH_2-NH_2$ , $KOH$	I.	Tollen's Test
B.	$Ag(NH_3)_2OH$	II.	Clemmensen Reduction
C.	Aq. $CuSO_4$ , Sodium potassium tartarate, $KOH$	III.	Wolff-Kishner Reduction
D.	$Zn-Hg$ , $HCl$	IV.	Fehling's Test

Choose the correct answer from the options given below:

(1) A-II, B-I, C-IV, D-III

(2) A-III, B-IV, C-I, D-II

(3) A-III, B-I, C-IV, D-II

(4) A-IV, B-III, C-II, D-I

**Q4. 22 January Shift 2**

The compound A, C<sub>8</sub>H<sub>8</sub>O<sub>2</sub> reacts with acetophenone to form a single product via cross-Aldol condensation. The compound A on reaction with conc. NaOH forms a substituted benzyl alcohol as one of the two products. The compound A is :

- (1) 2-hydroxy acetophenone      (2) 4-methyl benzoic acid  
 (3) 4-methoxy benzaldehyde      (4) 4-hydroxy benzaldehyde

**Q5. 23 January Shift 1**

'x' is the product which is obtained from propanenitrile and stannous chloride in the presence of hydrochloric acid followed by hydrolysis. 'y' is the product which is obtained from the but-2-ene by the ozonolysis followed by hydrolysis. From the following, which product is not obtained when one mole of 'x' and one mole of 'y' react with each other in the presence of alkali followed by heating?

- (1) 2-Methylpent-2-enal      (2) Pent-2-enal  
 (3) 2-Methylbut-2-enal      (4) 3-Methylbut-2-enal

**Q6. 23 January Shift 2**

Which of the following statements are TRUE about Haloform reaction?:

- A. Sodium hypochlorite reacts with KI to give KOI.  
 B. KOI is a reducing agent.  
 C.  $\alpha, \beta$ -unsaturated methylketone  $\left( \text{CH}_3 - \text{CH} = \text{CH} - \overset{\text{O}}{\underset{\text{CH}_3}{\text{C}}} \right)$  will give iodoform reaction.  
 D. Isopropyl alcohol will not give iodoform test.  
 E. Methanoic acid will give positive iodoform test.

Choose the correct answer from the options given below:

- (1) A, C & E Only      (2) B, D & E Only      (3) A & C Only      (4) A, B & C Only

**Q7. 23 January Shift 2**

Iodoform test can differentiate between

- A. Methanol and Ethanol  
 B. CH<sub>3</sub>COOH and CH<sub>3</sub>CH<sub>2</sub>COOH  
 C. Cyclohexene and cyclohexanone  
 D. Diethyl ether and Pentan-3-one  
 E. Anisole and acetone

Choose the correct answer from the options given below:

- (1) B, C & E Only      (2) A & D Only      (3) A & E Only      (4) A, B & E Only

**Q8. 24 January Shift 2**

Given below are two statements:

**Statement I:** Cross aldol condensation between two different aldehydes will always produce four different products.

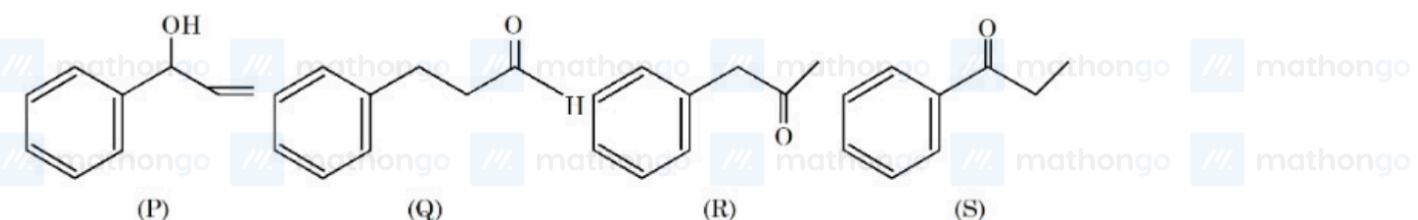
**Statement II:** When semicarbazide reacts with a mixture of benzaldehyde and acetophenone under optimum pH, it forms a condensation product with acetophenone only.

In the light of the above statements, choose the correct answer from the options given below

- (1) Statement I is false but Statement II is true      (2) Both Statement I and Statement II are false  
 (3) Both Statement I and Statement II are true      (4) Statement I is true but Statement II is false

**Q9. 28 January Shift 1**

Given below are the four isomeric compounds (P, Q, R, S)



Identify correct statements from below.

- A. Q, R and S will give precipitate with 2, 4 – DNP.  
 B. P and Q will give positive Bayer's test.  
 C. Q and R will give sooty flame.  
 D. R and S will give yellow precipitate with  $I_2/NaOH$ .  
 E. Q alone will deposit silver with Tollen's reagent Choose the correct option.  
 (1) A, B, D and E only      (2) A and E only  
 (3) C and E only      (4) A, C and E only

**ANSWER KEYS**

1. (1)    2. (2)    3. (3)    4. (3)    5. (2)    6. (3)    7. (3)    8. (2)  
 9. (4)