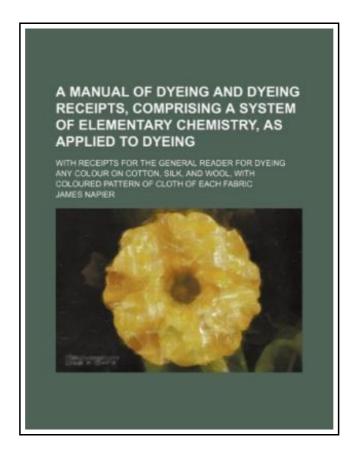
## A Manual of Dyeing and Dyeing Receipts, Comprising a System of Elementary Chemistry, as Applied to Dyeing; With Receipts for the General Reader for Dyeing Any Colour on Cotton, Silk, and Wool, with



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### Reviews

Merely no terms to spell out. We have read through and i also am confident that i will gonna read yet again again in the future. You will not sense monotony at anytime of your own time (that's what catalogs are for about should you question me).

(Pasquale Larkin I)

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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1875 Excerpt: .from personal experience, we can say that it is neither equally dark nor equally beautiful. It cannot be so dark, for gallic acid being much more insoluble than tannin falls to the bottom whenever it is formed, and consequently leaves the supernatant liquid much weaker in its dyeing properties. More recent discoveries have shewn that tannin is convertible into gallic acid by other and much more rapid means than being left to absorb oxygen: these are by the common processes of inducing fermentation. It is well known that fermentation is simply a derangement of the elements of certain complex compounds, and the re-arrangement of these elements in different positions and proportions, giving rise to new and altogether different compounds of a more simple nature. The primary compounds have been formed under the influence of the vital principle; but whenever this is withdrawn, the attraction of their elements seems too weak to enable them to resist any marked change of circumstances. Even a slight elevation of temperature is sufficient to overpower their affinities and induce change. Or if they are brought into contact with a body which is in the act of derangement, such as yeast, that body excites the same derangement in them, and the equilibrium being disturbed, the elements are left to arrange themselves according to their different attractions. If, for example, we dissolve a little sugar of grapes, which is composed of 12 carbon, 24 hydrogen, and 12 oxygen, in a little water, and...

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