## First Principle 2017-Fall Homework 3 Solution

Kai-Hsin Wu (吳愷訢)\*

Department of Physics and Center of Theoretical Sciences, National Taiwan University, Taipei 10607, Taiwan

- 1. The following shows the result of Al and Na:
  - Al
    - (1)  $a_0$  using volume optimization:

$$a_0 = 4.05000$$

(2) Variation with different  $a_0$ :

a<sub>0</sub> E 3.90 -14.541085 3.95 -14.665726 4.00 -14.735369 4.05 -14.757538 4.10 -14.738699 4.15 -14.684395 4.20 -14.599834

(3) the following figure shows the energy (E) v.s. V:

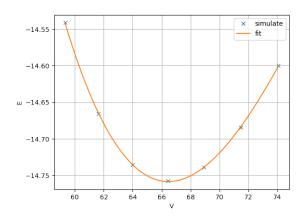


FIG. 1. Al-fcc E-V

<sup>\*</sup> r05222003@ntu.edu.tw

By using third order polyfit, and with the following formula, we can get the bulk-modulus B and the minimum  $a_0$ :

$$B = V \frac{\partial^2}{\partial V^2} E$$
$$V = a_0^3$$

$$a_0 = 4.050723 \text{ Å}$$
   
  $B = 74.608739 \text{ }GPar$ 

(4) We start with HF energy density and seek for the minimum of  $r_s$ :

$$e^{HF} = \frac{2.21}{r_s^2} - \frac{0.916}{r_s}$$

Where the unit of energy is  ${\it Ry}$  and  ${\it r_s}$  is in unit of bohr radius. The minimum is at :

$$r_s = 2.553467 \text{ Å}$$

using following relation by which we consider 1-free electron per unit-cell, we can estimate the lattice constant  $a_0$ :

$$\frac{4\pi}{3}r_s^3 = n^{-1}$$
$$\frac{N_{free}}{a_0^3} = n$$

$$a_0 = \left(\frac{4\pi}{3}\right)^{1/3} r_s$$

$$\approx 4.11616 \text{ Å}$$