



2001
HIGHER SCHOOL CERTIFICATE
TRIAL EXAMINATION

General Instructions

- Write your class and student number in the space provided.
- Attempt all questions 1 – 15
- Use a blue or black pen
- Select the alternative A, B, C, or D that best answers the question.
- Fill in the response oval completely.

Class _____

Student Number _____

Physics
Section I Part A
ANSWER SHEET

1. (A) (B) (C) (D)
2. (A) (B) (C) (D)
3. (A) (B) (C) (D)
4. (A) (B) (C) (D)
5. (A) (B) (C) (D)
6. (A) (B) (C) (D)
7. (A) (B) (C) (D)
8. (A) (B) (C) (D)
9. (A) (B) (C) (D)
10. (A) (B) (C) (D)
11. (A) (B) (C) (D)
12. (A) (B) (C) (D)
13. (A) (B) (C) (D)
14. (A) (B) (C) (D)
15. (A) (B) (C) (D)

PERIODIC TABLE OF THE ELEMENTS

Atomic Number	Symbol	Name	Atomic Weight
1	H	Hydrogen	1.008
2	He	Helium	4.003
3	Li	Lithium	6.941
4	Be	Beryllium	9.012
5	B	Boron	10.81
6	C	Carbon	12.01
7	N	Nitrogen	14.01
8	O	Oxygen	16.00
9	F	Fluorine	18.99
10	Ne	Neon	20.18
11	Na	Sodium	22.99
12	Mg	Magnesium	24.31
13	Al	Aluminum	26.98
14	Si	Silicon	28.09
15	P	Phosphorus	30.97
16	S	Sulfur	32.07
17	Cl	Chlorine	35.45
18	Ar	Argon	39.95
19	K	Potassium	39.10
20	Ca	Calcium	40.08
21	Sc	Scandium	44.96
22	Ti	Titanium	47.87
23	V	Vanadium	50.94
24	Cr	Chromium	52.00
25	Mn	Manganese	54.94
26	Fe	Iron	55.85
27	Co	Cobalt	58.93
28	Ni	Nickel	58.69
29	Cu	Copper	63.55
30	Zn	Zinc	65.39
31	Ga	Gallium	69.72
32	Ge	Germanium	72.61
33	As	Arsenic	74.92
34	Se	Selenium	78.96
35	Br	Bromine	79.90
36	Kr	Krypton	83.80
37	Rb	Rubidium	85.47
38	Sr	Strontium	87.62
39	Y	Yttrium	88.91
40	Zr	Zirconium	91.22
41	Nb	Niobium	92.91
42	Mo	Molybdenum	95.94
43	Tc	Technetium	98.91
44	Ru	Ruthenium	101.1
45	Rh	Rhodium	102.9
46	Pd	Palladium	106.4
47	Ag	Silver	107.9
48	Cd	Cadmium	112.4
49	In	Indium	114.8
50	Sn	Tin	118.7
51	Sb	Antimony	121.8
52	Te	Tellurium	127.6
53	I	Iodine	126.9
54	Xe	Xenon	131.3
55	Ba	Barium	137.3
56	La	Lanthanum	138.9
57	Ce	Cerium	140.1
58	Pr	Praseodymium	140.9
59	Nd	Neodymium	144.2
60	Pm	Promethium	144.9
61	Sm	Samarium	150.4
62	Eu	Europium	152.0
63	Gd	Gadolinium	157.3
64	Tb	Terbium	158.9
65	Dy	Dysprosium	162.5
66	Ho	Holmium	164.9
67	Er	Erbium	167.3
68	Tm	Thulium	168.9
69	Yb	Ytterbium	173.0
70	Lu	Lutetium	175.0
71	Hf	Hafnium	178.5
72	Ta	Tantalum	180.9
73	W	Tungsten	183.8
74	Re	Rhenium	186.2
75	Os	Osmium	190.2
76	Ir	Iridium	192.2
77	Pt	Platinum	195.1
78	Au	Gold	197.0
79	Hg	Mercury	200.6
80	Tl	Thallium	204.4
81	Pb	Lead	207.2
82	Bi	Bismuth	209.0
83	Po	Polonium	209.0
84	At	Astatine	210.0
85	Rn	Radon	222.0
86	Fr	Francium	223.0
87	Ra	Radium	226.0
88	Ac	Actinium	227.0
89	Th	Thorium	232.0
90	Pa	Protactinium	231.0
91	U	Uranium	238.0
92	Np	Neptunium	237.0
93	Pu	Plutonium	244.1
94	Am	Americium	243.1
95	Cm	Curium	247.1
96	Bk	Berkelium	247.1
97	Cf	Californium	251.1
98	Es	Einsteinium	252.1
99	Fm	Fermium	257.1
100	Md	Mendelevium	258.1
101	Nr	Nobelium	259.1
102	Lr	Lutetium	262.1

KEY

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100	Md	Mendelevium	258.1
101	Nr	Nobelium	259.1
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When the atomic weight is not known, the relative atomic mass of the most common radioactive isotope is shown in brackets.

The atomic weights of Pb and Tl are given for the isotopes ²⁰⁷Pb and ²⁰³Tl.

Question 16. (3 marks)

Marks	Marking criteria
2 marks	Uses $T = n B I A$ to show correctly $T = 24 \times 0.2 \times 25 \times 10^{-3} \times 8 \times 10^{-4}$ $= 9.6 \times 10^{-5} \text{ N.m (units)}$
1 mark	uses $T = n B I A \cos \theta$ and fully substitutes all values into equation but does not convert I (from mA to A) and/or A (from cm^2 to m^2) OR/ if fully substitutes into above equation but leaves answer with $\cos \theta$ OR/ correctly converts both I and A but leaves B out/or transposes incorrectly OR/ correctly converts both I and A but leaves n out. OR/ if fully substituted but area is taken as $(0.08)^2$ OR/ if fully substituted with correct conversion but area wrong value

0 marks	If both n and A incorrect OR If I incorrectly converted and A incorrect.
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Notes:

- (i) units for torque N.m
- (ii) some candidates calculated a value for B by using $\phi = BA$ and substituting $0.2 = B \times 0.08$

(b) 1 mark

Marks	Marking criteria
1 mark	using Ans(a) = degrees restoring torque $(= 2.0 \times 10^{-6})$ $= \frac{9.6 \times 10^{-5}}{2 \times 10^{-6}} = 48^\circ$

Question 17 (8 marks)

Marks	Marking criteria
2 marks	Explanation in terms of reducing eddy currents and therefore improves efficiency (or reduces energy losses).
1 mark	Mentions one of reducing eddy currents or improves efficiency only

Sample Answer:

The presence of a changing magnetic flux in the soft iron core causes eddy currents in the core (by Faraday's law of Electromagnetic Induction). The iron core is laminated to reduce the eddy currents, thereby improving the efficiency of the transformer by reducing energy losses due to the heating effects of eddy currents.

(b) 2 marks (maximum)

Marks	Marking Criteria
2 marks	<p>Candidate explains that the secondary coil needs to experience a change in flux to produce an induced emf.</p> <p>Explains that AC in the primary coil produces this changing magnetic flux whereas DC produces only a constant flux</p>
1 mark	Explains only one of the points above.

Sample Answer:

AC voltage sets up a changing magnetic flux in the core that is necessary to induce a voltage in the secondary coil. As DC is constant, the magnetic field would not be changing so $\Delta\Phi = 0$
 \therefore no emf induced in the secondary coil.

(c) 1 mark (maximum)

Marks	Marking criteria
1 mark.	<p>Identifies $V_p/V_s = N_p/N_s$ as the relevant relationship.</p> <p>Substitutes to show that.</p> $V_s = 240 \times 30/60 = 120V$

(d) 3 marks (maximum)

"Discuss..." identify issues and provide points for

Marks	Marking criteria
3 marks	<p>Candidate mentions (or implies) that the required voltage may be higher or lower than 240V.</p> <p>Describes at least <u>two</u> correct/accurate reasons as to why electrical appliances in the home connected to the mains supply use a transformer.</p> <p>Each reason is supported with a named appliance.</p> <p>Reasons include: voltage changes because appliance foreign made; maximised operating efficiency; appliance requires more current; impairs its function; make it safer; lower current due to delicate circuits; prevent overheating.</p>
2 marks	<p>Candidate mentions (or implies) that the required voltage may be higher or lower than 240V.</p> <p>Describes <u>one</u> reason as to why electrical appliances in the home connected to the mains supply use a transformer. The reason is supported with a named appliance.</p> <p>Reasons as per the above list.</p>
1 mark.	<p>Candidate mentions (or implies) that the required voltage may be higher or lower than 240V.</p> <p>but fails to give issues or if issue(s) given not supported with specific examples</p>

Sample Answer.

The required voltage for the appliance may be higher or lower than 240V. Portable electrical appliances contain a step-down transformer (e.g. computer circuitry) which converts the 240V domestic supply down to a lower, normal operating voltage for the correct and safe use of IC circuits. Televisions have step-up transformers to produce the high voltages needed to drive the electron gun in the picture tube.

Question 18 (2 marks)

Marks

The planet Mars has a mass of 6.42×10^{23} kg and a radius of 3.40×10^6 m. Calculate the escape velocity at the surface of Mars.

2

$$v = \sqrt{\frac{2Gm}{r}} \quad \therefore v = \sqrt{\frac{6.7 \times 10^{-11} \times 6.42 \times 10^{23}}{3.4 \times 10^6}}$$

① method

$$v = 6.03 \times 10^3 \text{ ms}^{-1} \quad \text{①}$$

Question 19 (4 marks)

A satellite of mass 100 kg performs a circular orbit, 1000 km above the surface of the Earth. The radius of the Earth is 6.40×10^6 m.

(a) Calculate the gravitational force acting on the satellite.

2

981 if
not adding
 1×10^6

$$F = \frac{G m_1 m_2}{r^2} = \frac{6.7 \times 10^{-11} \times 100 \times 6.0 \times 10^{24}}{(6.4 \times 10^6 + 1 \times 10^6)^2} \quad \text{① method}$$

$$\therefore F = 734.1 \text{ N} \quad \text{①}$$

(b) Calculate the time taken by the satellite to complete one revolution of the Earth.

2

①
method

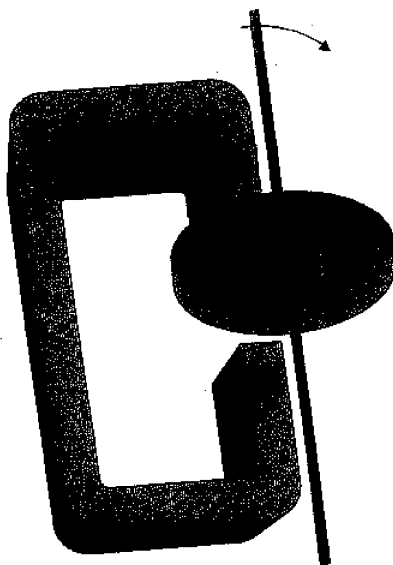
$$\frac{mv^2}{r} = 734.1 \quad \therefore \frac{m 4\pi^2 r}{T^2} = 734.1$$

$$\therefore T = \sqrt{\frac{100 \times 4\pi^2 \times (7.4 \times 10^6)^2}{734.1}} \quad \therefore T = 6.3 \times 10^3 \text{ s} \quad \text{①}$$

(1 hr 45 min)

Question 20 (3 marks)

Electromagnetic braking can be achieved by applying a strong magnetic field to a spinning metal disc attached to a shaft as shown below.



(a) Identify and explain how the magnetic field slows the spinning of the disc.

- motion of conductor in \vec{B} induces eddy currents. (1)
- force of \vec{B} on eddy currents opposes motion (Lenz's Law) (1)

(b) Would the brakes work if the disc was plastic instead of metal? Explain your answer. (1)

No. Plastic is insulating \therefore no eddy current \therefore no force or eddy current. (Lenz's Law doesn't apply.) (1)

Question 21 (2 marks)

Marks

2

Light of wavelength 6×10^{-9} m is incident on a sodium surface. The work function (i.e. the minimum energy required to emit an electron) of sodium is 2.9×10^{-19} J. Calculate the maximum kinetic energy of the electrons ejected from the sodium by this light.

$$\lambda = 6.0 \times 10^{-9} \text{ m} ; \phi = 2.9 \times 10^{-19} \text{ J}$$

$$E_k = hf - \phi = 6.626 \times 10^{-34} \times 3 \times 10^8 - 2.9 \times 10^{-19} \text{ J}$$

$$E_k = \frac{hc}{\lambda} - \phi ; E_k = 3.28 \times 10^{-17} \text{ J} \quad (1)$$

Question 22 (4 marks)

Give an example of a modern device that uses a cathode ray tube and outline its operation. (4)

(1) for most of the bits only 9

(1) for all of bits guide 1

(1) for good description of how it works \therefore no marks

(1) for how it conveys information i.e. how a signal is displayed. (1)

Question 23 (5 marks)

Marks

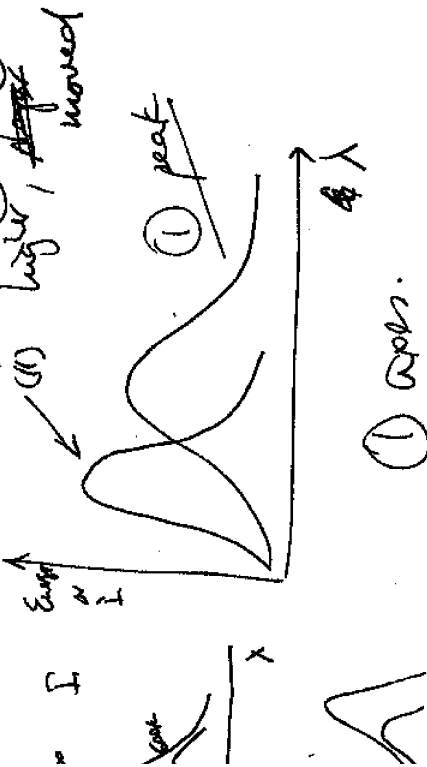
- (a) What do physicists mean by the term 'black body'?

A perfect emitter or absorber of radiant energy

1

- (b) (i) Sketch a graph to show how the intensity of light emitted by a black body depends upon the frequency (or wavelength) of the light.

2



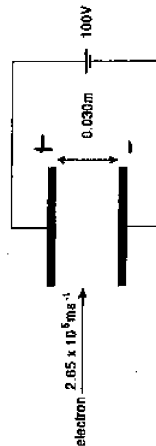
- (ii) Add to your graph a second sketch for the light intensity of the same body at a higher temperature. Make sure you distinguish clearly between the two sketches.

2

Question 24 (5 marks)

Marks

An electron travelling at a velocity of $2.65 \times 10^5 \text{ ms}^{-1}$ passes horizontally between two parallel, horizontal electric plates 0.030 m apart and connected to a potential difference of 100 V .



- (a) Calculate the electric field strength between the horizontal plates.

1

$$E = \frac{V}{d} = \frac{100}{0.03} = 3,333 \text{ Vm}^{-1} \quad (1)$$

- (b) What is the electrostatic force acting on the electron in the region between the plates?

2

$$F = Eq = 3,333 \times 1.6 \times 10^{-19} = 5.3 \times 10^{-16} \text{ N} \quad (1)$$

- (c) What magnetic field must be applied to the electron to allow it to pass between the plates undeflected?

2

$$(b) = Bqv \quad (1)$$

$$\therefore B = \frac{F}{qv} = \frac{5.3 \times 10^{-16}}{1.6 \times 10^{-19} \times 2.65 \times 10^5} = 1.26 \times 10^{-2} \text{ T} \quad (1)$$

Form VI Physics Trial Examination Crib –Questions 25-29

Some General Comments on the Open-Ended Questions

NB these questions were NOT marked on a 'mark per point' basis. Rather, they were marked in accordance with the Board of Studies' Performance Bands. They will *only* be accepted for remarking if they have been blatantly mismarked. If your interpretation of your answer differs from mine, my mark stands!

These questions were not well answered.

The most common failings were:

Not outlining significant concepts

e.g. discussing the photo-electric effect without saying what it is, or without defining what a photon is.

Ambiguity or Imprecision

e.g. 'the intensity is proportional to the photoemission'

the intensity of what? what aspect of the photoemission?

Non Sequiturs

e.g. 'Michelson-Morley experiment showed that the aether did not exist, therefore Einstein was proved correct'

the link between the two must be elucidated.

Not using diagrams

Writing two paragraphs of barely coherent text is **never** a substitute for a decent diagram. 'Describe' does not simply mean words!

Qualitative not Quantitative Answers

e.g. 'the energy of a photon depends on its frequency' rather than ' $E=hf$ '

25.

For full marks, the following were required:

1. MM attempted to determine the velocity of the Earth through the aether, by measuring the speed of light relative to the Earth.
2. Despite repeating the experiment six months later – when the velocity of the Earth relative to the aether might have been expected to have changed substantially – no change in the velocity of light relative to the Earth was observed.
3. This provided corroborating evidence for SR as it accorded with Einstein's suggestion that the speed of light is a constant for all observers.

Most common mistakes;

MM *proved the aether did not exist* – how can you prove something does not exist?

'the speed of light is constant' – must have 'for all observers' or similar

Some of the best answers started with the postulates of SR and showed how MM was consistent with them.

NB It is not historically true to say that MM *led to* SR. However, in the context of an otherwise correct answer, this was not penalised.

26.

Ans: 0.6c or 1.8×10^8 m/s

(1 mk for correct use of formula (i.e. /and / the right way round))

27.

a) 90 degrees

b) 5 marks for:
 curve starts at zero
 two complete periods shown
 correct shape (ie sine wave, not rectified)
 axes correct and labelled
 correct numerical values on both axes

c) either:

Energy considerations suggest that electrical energy is consumed only when a load is applied. Mechanical energy must therefore only be supplied when the bulb is connected i.e. work must be done to turn the generator.

or:

A current can only flow when a load is connected. The current produces a force within the coil that – from Lenz's Law – acts to oppose the change in motion, and therefore make the coil more difficult to turn.

One mark only if the answer does not explain why the coil is harder to turn.

28.

For full marks and answer should contain most or all of the following:

1. A lucid description of the experimental method, including a diagram.
2. An outline of what data should be taken and how.
3. An appreciation of the practicalities of the experiment.
4. An appreciation that, if the two directions are independent, then $a_H = 0$, $a_V = -g$.
5. A discussion of how the data can be quantitatively analysed to verify that the two directions are indeed independent.

Comments:

1. Too many written descriptions of the method were ambiguous. In most cases, diagrams would have improved the answer.
2. There was little regard to the practicalities of the experiment, e.g. 'shoot a person from a cannon ...'.
3. The phrase 'the data can be analysed to show that H and V are independent' is not a substitute for actually using Newton's Equations of Motion to show it yourself.

29.

For full marks, the following are required:

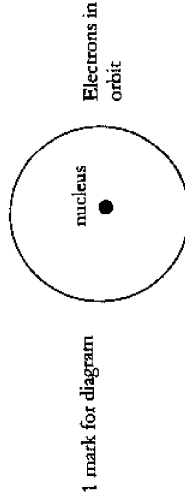
1. An outline of the photoelectric effect.
2. At least *two* pieces of experimental evidence that the wave model could not explain.
3. A description of a photon as a quanta of light energy, including the expression $E=hf$.
4. A discussion of how the photon model successfully explains the experimental observations given earlier.

Comments:

1. An incredible number of people did not bother to outline what the photoelectric effect is. Without a context, comments like *'as the frequency is increased, the stopping potential increases'* are meaningless.
2. Most people lost marks for failing to adequately explain *why* the photon model explained the observed effects. Simply stating *'the photon model accounts for this'* or something similar is not enough.

Quanta to Quarks crib SRW

- (a) Dense, tiny nucleus/electrons orbit nucleus/nucleus contains all of the positive charge and most of the mass (any two)



- b) Fired electrons at nickel and observed a diffraction/interference pattern (1 mark)
Electrons have wave properties (1 mark)
- c) i) Angular momentum of electrons is quantised and hence energy of electrons is quantised (1 mark) (Must mention that angular momentum is quantised. Just stating that the energy was quantised without any justification was not accepted)
Electrons lie in stationary states where they don't radiate (1 mark)
Energy in the form of e-m waves is emitted when electrons jump from a higher to lower orbit producing the Balmer spectrum (1 mark)
- ii) $1/\lambda = 1.097 \times 10^7 (1/2^2 - 1/3^2)$ (1 mark)
 $\lambda = 6.56 \times 10^{-7} \text{ m}$ (1 mark)
- If you had the wrong substitution you got 1 mark
- iii) $c = \lambda \nu = 4.57 \times 10^{14} \text{ Hz}$ (1 mark)
- d) i) Particles have wave properties given by $\lambda = h/p$ (1 mark)
Many candidates talked about DeBroglie/Schrodinger's model of the atom in terms of integral numbers of wavelength. This is not the DeBroglie hypothesis but a model of the atom derived from it.

The hypothesis was starting for many reasons

1. In classical physics particles and waves are completely separate and do not have a wave-particle duality. (1 mark)
or
 2. The proposal was made before there was experimental evidence (1 mark)
- ii) $\lambda = h/p = 7.27 \times 10^{-8} \text{ m}$ (1 mark)

- e) ${}_{36}^{92}\text{Kr}$ or Krypton -92 (1 mark)
- ii) Nuclear Fission (1 mark). I did not accept transmutation or chain reaction.
Transmutation is far too vague and chain reaction presupposes that the neutrons are going to hit other uranium atoms which is no where indicated in the equation.

iii) mass defect = $(3.344 + 5.0089) \times 10^{-27} \text{ kg} - (6.6463 + 1.6749) \times 10^{-27} \text{ kg}$ (1 mark)
 mass defect = $0.0317 \times 10^{-27} \text{ kg}$ (1 mark)

iv) $E = \text{mass defect} \times c^2 = 2.853 \times 10^{-12} \text{ J}$

f) In Beta decay it was found that the following conservation laws did not appear at first to hold true. $n \rightarrow p + e^- + ?$

1. Momentum was not conserved (1 mark)
2. Kinetic energy was not conserved (1 mark)
3. The Kinetic energy of the electron was distributed across a range of values whereas mechanics predicts it should have just one energy. (1 mark)
4. Angular momentum as given by the spin of the particles $+\frac{1}{2}$ was conserved. (1 mark).

Maximum of three marks.

All of the above led Pauli to propose the existence of a third neutral particle.

(Many candidates talked about mass defect. This is not sensible as in all nuclear reactions there is a mass defect. The mass of the neutrino is so small anyway that its mass could not have even been detected at the time. What is important however is the apparent energy loss)

g) In a controlled fission reaction the numbers of neutrons which then go on to cause fission in other Uranium atoms is limited by control rods made from Cadmium or Boron which absorb neutrons (1 mark)/

(many candidates confused moderators with control rods. Moderators will actually speed up the reaction as they slow down the neutrons so that they can more efficiently cause fission in Uranium)

In an uncontrolled fission reaction the neutrons emitted are highly likely to cause subsequent fission reactions and since 2 or 3 are emitted at a time this results in a rapid build up of neutrons and fission reactions releasing an enormous amount of energy. (1 mark)