

All thermal problems need access to the following data:

Specific Heats, Thermal conductivities and linear expansion coefficients of selected metals

Substance	Specific Heat c , ($\text{J kg}^{-1} \text{K}^{-1}$)	Thermal conductivity k , ($\text{W m}^{-1} \text{K}^{-1}$)	Linear expansion coefficient α , ($^{\circ}\text{C}^{-1}$)	Density (gcm^{-3}) at 0°C
Aluminium	910	205.0	24×10^{-6}	2.70
Brass	377	109.0	19×10^{-6}	8.40
Copper	390	385.0	17×10^{-6}	8.96
Lead	130	34.7	29×10^{-6}	11.3
Steel	456	50.2	11×10^{-6}	7.85

Water

Quantity	Value
Specific Heat (liquid)	$4186 \text{ Jkg}^{-1}\text{K}^{-1}$
Latent heat of Fusion	$3.33 \times 10^5 \text{ Jkg}^{-1}$
Latent heat of vapourization	$2.26 \times 10^6 \text{ Jkg}^{-1}$
Density (at 4.00°C)	1000 kgm^{-3}
Melting point (at 1 atm)	0.000°C
Boiling point (at 1 atm)	100.0°C
Volume expansion coefficient (β) (at 20°C : you may assume it is constant between 15°C and 100°C)	$207 \times 10^{-6} (^{\circ}\text{C})^{-1}$