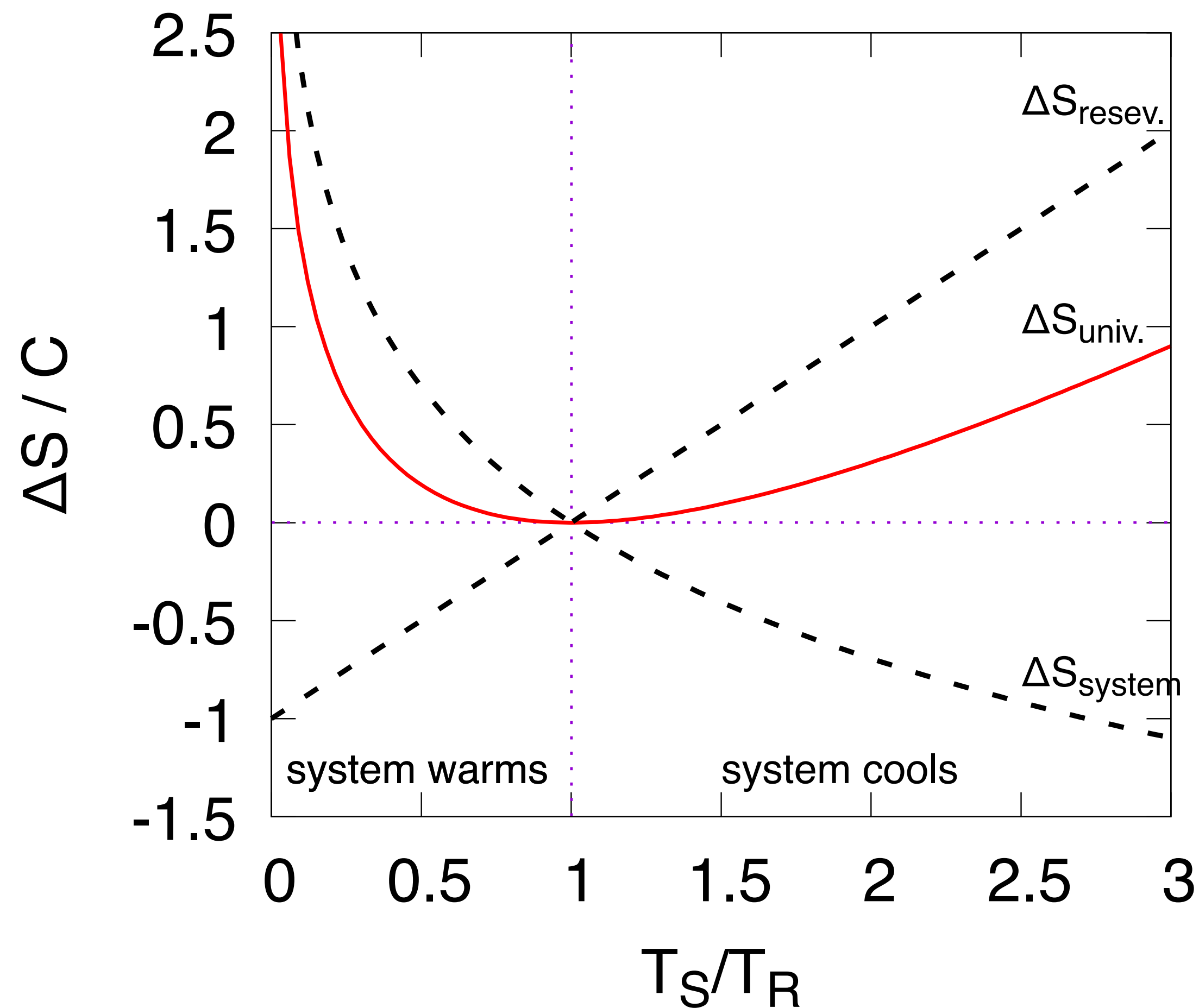


Change in Entropy Ball in Lake: Blundell Example 14.1



System = Ball

The ball has (initial) temperature T_S

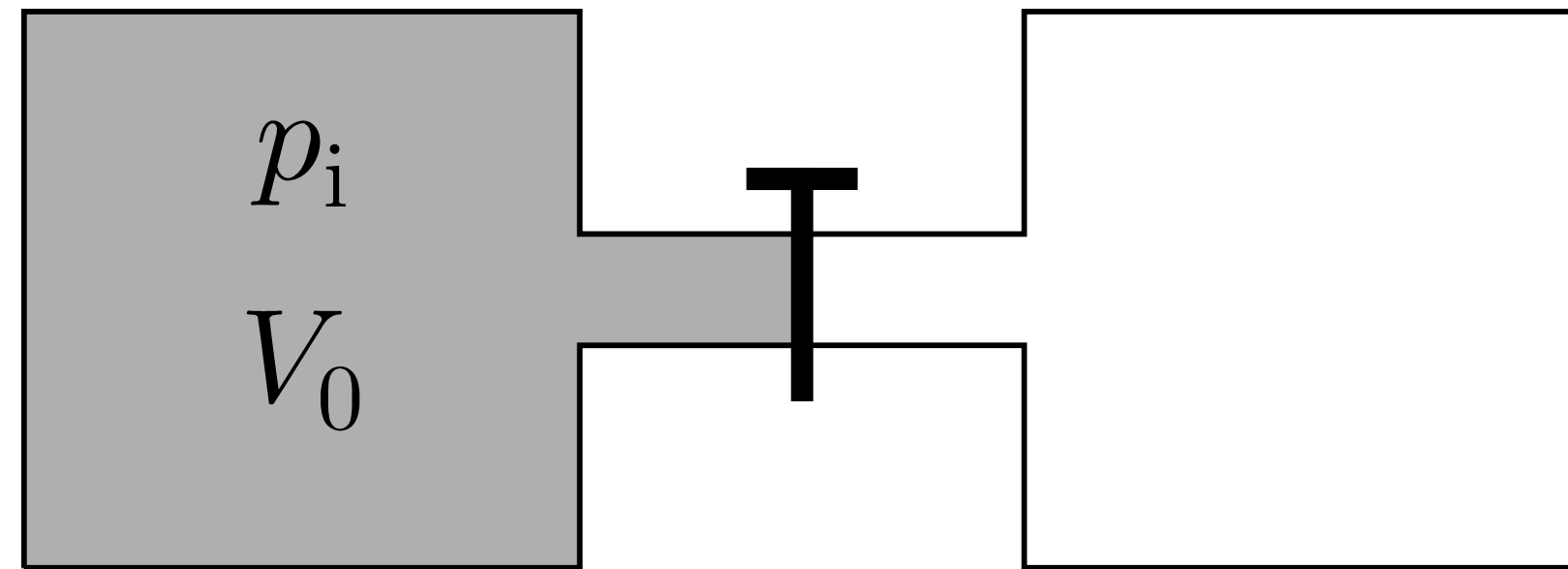
Reservoir = Lake

The reservoir has constant temperature T_R

Universe is the ball and lake

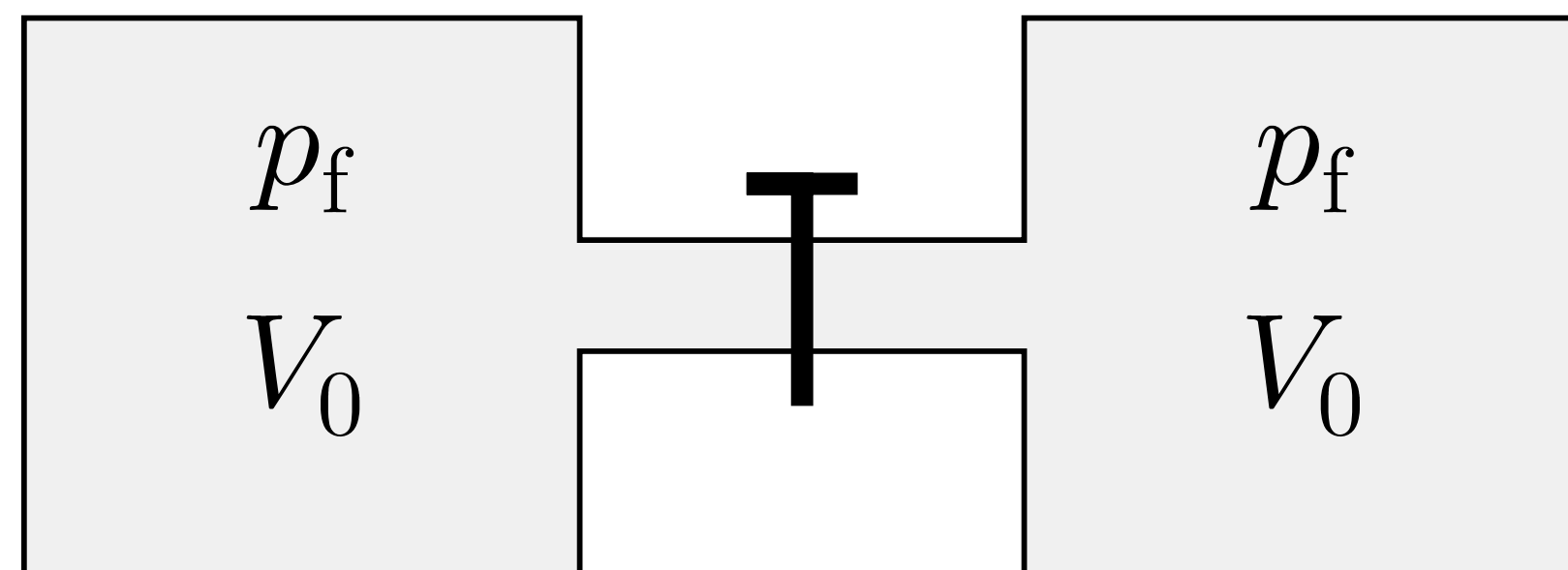
Joule Expansion

(a)



The expansion is a highly non-equilibrium process.

(b)



During the expansion no heat enters the system. Thus the energy initial equals the final energy