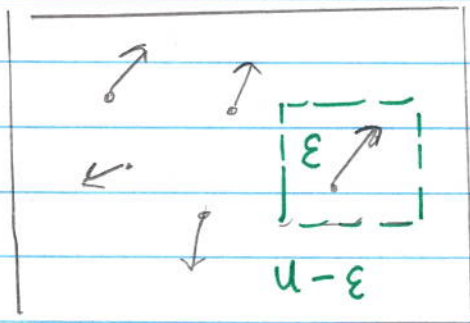


The Boltzmann Factor

- This is based on Sect. 4.6. We will return to Chapter 4 later. Our goal is to give only an initial understanding.
- Consider a system with total energy U



- System has energy U
 - Subsystem has energy ϵ
 - Rest has energy $U - \epsilon$
- Pick a small subsystem which is independent of the larger system except through the exchange of energy. In the case of an ideal gas, each molecule is independent of all others, and can be considered an independent subsystem
 - What is the probability that the subsystem will have energy ϵ ?

For instance, it is possible, though extremely unlikely, that the one molecule (subsystem) will have all of the energy of the gas.

- The probability to find the subsystem with energy ϵ is

$$P(\epsilon) \propto e^{-\epsilon/k_B T} \leftarrow \text{Boltzmann factor}$$

- We get tired of writing $1/k_B T$ so define $\beta \equiv 1/k_B T$, i.e.

$$P(\epsilon) = C e^{-\beta \epsilon}$$

$$\beta \equiv \frac{1}{k_B T}$$

- If you have some set of microscopic states $i = 1 \dots N$, then since $\sum C e^{-\beta \epsilon_i} = 1$ (since the sum of probabilities is one) we have

$$P(\text{state } r) = \frac{e^{-\beta \epsilon_r}}{\sum_i e^{-\beta \epsilon_i}}$$

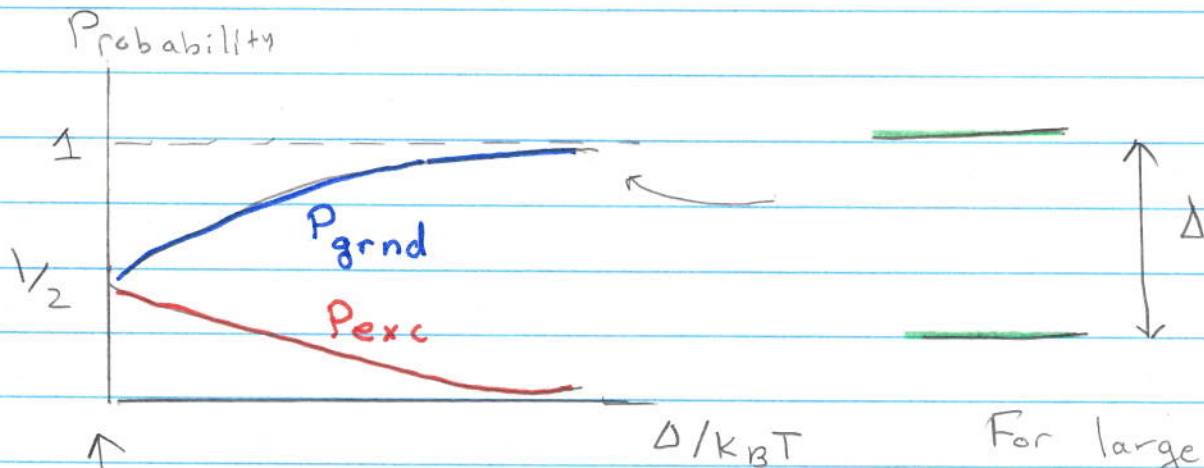
↑
probability to find the subsystem in state r with energy ϵ_r

Ex 1

Consider an "atom" consisting of a ground state with energy 0, and an excited state of energy Δ . At temperature T determine the probability to be in the ground and excited state

$$P_{\text{ground}} = \frac{e^{-\beta 0}}{e^{-\beta 0} + e^{-\beta \Delta}} = \frac{1}{1 + e^{-\beta \Delta}}$$

$$P_{\text{excit}} = \frac{e^{-\beta \Delta}}{e^{-\beta 0} + e^{-\beta \Delta}} = \frac{e^{-\beta \Delta}}{(1 + e^{-\beta \Delta})}$$



small Δ compared
to $k_B T$, the

system can easily
jump from ground to
excited and back

and thus $P \approx 1/2$

For large Δ compared
to $k_B T$ the system
does not ^{have} enough
thermal energy to
jump to the
higher state

and thus all the atoms are in
the ground state, $P_{\text{grnd}} \approx 1$