

## Partition Functions: A recap

- We have described several thermodynamic functions which are useful

$$U, \quad F = U - TS, \quad H = U + PV, \quad G = U + PV - TS$$

Now we will discuss how to compute these systematically using partition functions.

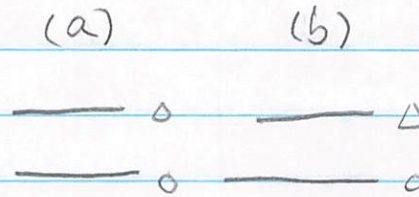
### Partition Function when Energy is a sum of energies:

atom (a) and  
atom (b)

- First consider a system consisting of two atoms, each of which can be in one of two states of energy

$$\epsilon_i^{(a)} = 0, \Delta \quad \text{or} \quad \epsilon_j^{(b)} = 0, \Delta$$

$$E_{\text{Tot}} = \epsilon_i^{(a)} + \epsilon_j^{(b)}$$



- There are four states

$$(0, 0), (0, \Delta), (\Delta, 0), (\Delta, \Delta)$$

- Now

$$Z_{\text{Tot}} = \sum_{i,j=0}^{\Delta} e^{-E_{\text{Tot}}/kT}$$

$$= \sum_{i,j} e^{-(\epsilon_i^{(a)} + \epsilon_j^{(b)})/kT}$$

$$= \sum_{i,j} e^{-\epsilon_i^{(a)}/kT} e^{-\epsilon_j^{(b)}/kT}$$

$$= \sum_i e^{-\epsilon_i^{(a)}/kT} \sum_j e^{-\epsilon_j^{(b)}/kT} = Z^{(a)} Z^{(b)}$$

(Technically we are assuming that (a) and (b) are distinguishable)  
Thus (0, Δ) and (Δ, 0) are two states, not the same state.

- So if the energy breaks up into a sum of energies, the partition function factorizes into a product

$$Z_{\text{TOT}} = Z^a Z^b$$

- Now the free energy is

$$F_{\text{TOT}} = -k_B T \ln Z_{\text{TOT}} \quad (\text{see homework})$$

$$F_{\text{TOT}} = -k_B T \ln Z^{(a)} - k_B T \ln Z^{(b)}$$

$$F_{\text{TOT}} = F^{(a)} + F^{(b)}$$

- If System (a) and (b) have the same energy levels (as in this example) then  $F^{(a)} = F^{(b)}$  and

$$F_{\text{TOT}} = 2 F^{(a)}$$

- Clearly this generalizes to  $N^{\text{distinguishable}}$  systems. If the energy is a sum of energies (with the same energy level scheme, for simplicity)

$$E = \epsilon_i^{(1)} + \epsilon_j^{(2)} + \epsilon_k^{(3)} + \dots \epsilon^{(N)}$$

$$Z = \sum_{ijk\dots} e^{-E/kT} = Z_1^N$$

Where

$$Z_1 = \sum_i e^{-\epsilon_i/kT}$$



- So in this case

$$F_{\text{TOT}} = -k_B T \ln Z_{\text{tot}} = -k_B T \ln Z_1^N = -N k_B T \ln Z_1$$

$$\boxed{F_{\text{TOT}} = N F_1}$$

(for total system of  $N^{\text{distinguishable}}$  subsystems with the same level scheme)

- We had this in homework. For one harmonic oscillator we had

$$Z_1 = \frac{1}{1 - e^{-\hbar\omega\beta}}$$

The oscillators are separated in space and are distinguishable

- For  $N$  harmonic oscillators

$$F_{\text{TOT}} = -N k_B T \ln Z_1 = + N \overbrace{k_B T \ln (1 - e^{-\hbar\omega\beta})}^{\equiv F_1}$$

- Then

$$S_{\text{TOT}} = - \frac{\partial F_{\text{TOT}}}{\partial T} = N \left( - \frac{\partial F_1}{\partial T} \right) = N S_1 = N \left( \frac{\beta \hbar \omega}{e^{\beta \hbar \omega} - 1} - \ln (1 - e^{-\beta \hbar \omega}) \right)$$

homework

- Note we have  $F_{\text{TOT}} = U_{\text{TOT}} - T S_{\text{TOT}}$  so

$$U_{\text{TOT}} = F_{\text{TOT}} + T S_{\text{TOT}} = N \frac{\hbar \omega}{e^{\beta \hbar \omega} - 1} = N U_1$$

So  $S_{\text{TOT}} = N S_1$  and  $U_{\text{TOT}} = N U_1$  etc  
since  $Z_{\text{TOT}} = Z_1^N$ .