## Entropy Revisited:

· Previously we considered each microstate of the full system to be equally likely. Thus

probability to be in microstate in is constant

IP = 1 or C \( \Sigma 1 = 1 \)

The states

Or

$$CSZ(E) = I$$
 and  $P = I$ 

$$SZ(E)$$

$$S = k \ln \Omega = -k \ln P_m$$

Now we have a subsystem with probabilies Pm x e - E/KT. We would like to find the entropy for the subsystems with this probabily distribution (see slide)

The generalization is (as we show below)

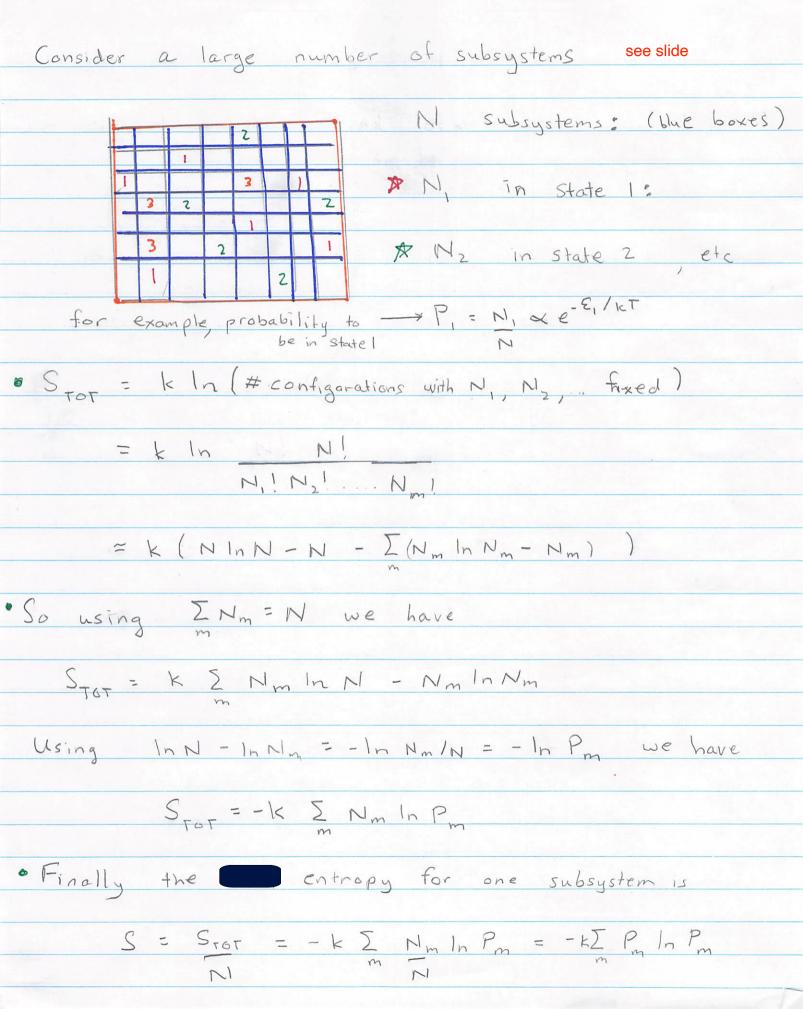
## Probability and entropy for one subsystem

Pick a subsystem (site):
The remaining sites are the reservoir

The probability for a site to have n quanta to be:

$$P(\epsilon_n) \propto e^{-\beta \epsilon_n} = e^{-n\beta\hbar\omega_0}$$

We will determine the entropy of the subsystem (one site) from this probability distribution. To find the total entropy, just multiply by the number of sites:  $S_{\rm tot} = NS$ 



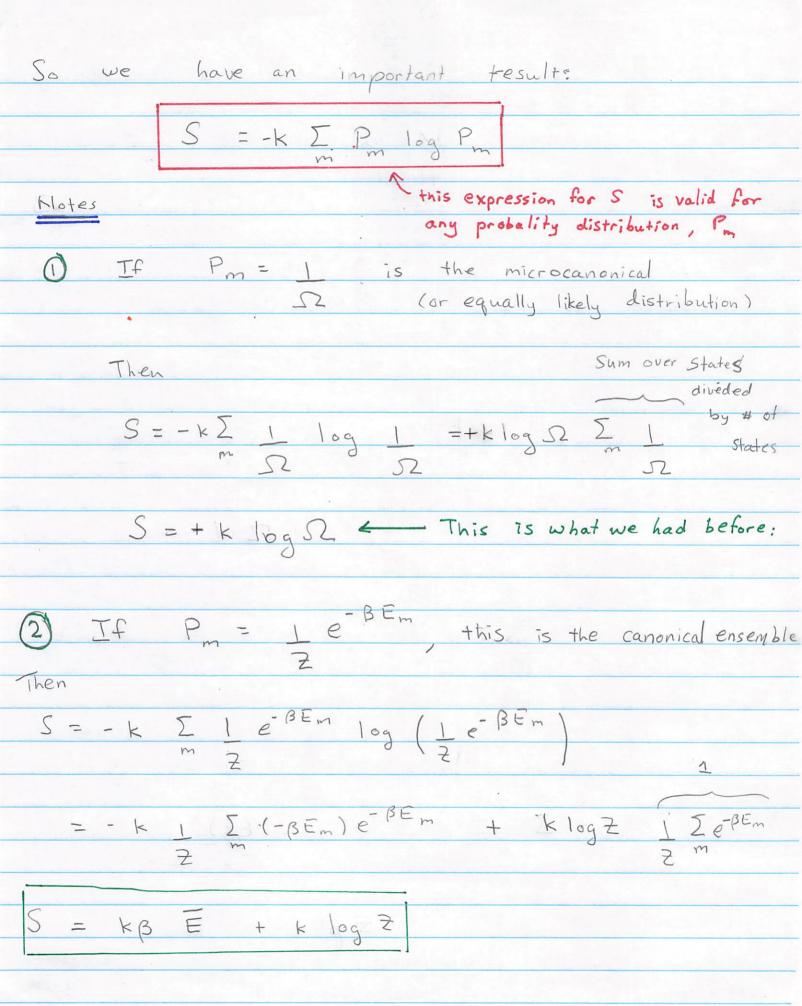
## Entropy of subsystem

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There are N sites in total

 $N_0$  are in state 0,  $N_1$  are in state 1,  $N_2$  are in state 2,  $N_3$  are in state 3...

 $P_r = N_r/N$  is the probability of a subsystem to be in the r-th state



$$S = E + k \log 2$$

This is how the entropy can be computed

From the partition function  $Z$ 
 $E = -10Z$ 
 $Z \partial B$ 

We also note that if I have two *independent* subsystems A and B, the entropy of the combined systems is the entropy A plus the entropy of B

$$S_{AB} = S_A + S_B$$

. You will prove this in homework directly from the formula  $S = -\sum_{m} P_{m} \ln P_{m}$ .

