Exponential Distribution

a) 
$$\int dx A e^{-X/L} = 1$$

Changing Variables  $u = x/e$ 

Al  $\int dx e^{-X/L} = 1$ 

Al  $\int du e^{-u} = 1$ 

I proved below

$$A = VL$$

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$$A = \int dx \times e^{-X/L}$$

$$\langle x^2 \rangle = \int_0^\infty dx \ x^2 \ e^{-x/l}$$

$$(\chi^2) = l^2 \int \frac{dx}{2} \left(\frac{x}{2}\right)^2 e^{-x/2}$$

$$(x^2) = \int_0^2 \int_0^2 du \, u^2 e^{-u} = \int_0^2 2!$$

$$(8x^2) = (x^2) - (x)^2 = 2l^2 - l^2 = (8x^2)$$

$$\langle e^{f \times} \rangle = \int_{0}^{\infty} dx \ e^{f \times} e^{-x}$$

$$= \int_{0}^{\infty} dx e^{-(1-f)x}$$

Now according to the generating for method

$$\langle e^{f \times} \rangle = 1 + \langle x \rangle f + \langle x^2 \rangle f^2 + \langle x^3 \rangle f^3 + ...$$

The explicit computation gives

$$\langle e^{f_{x}} \rangle = \frac{1}{1-f} = \frac{1}{1-f} + \frac{1}{f^{2}} + \frac{1}{f^{3}} + \dots$$

So, for instance, comparing the coefficient of f3 we conclude

$$(x^3) f^3 = f^3 \text{ or } (x^3) = 3!$$

· More generally

$$\frac{n!}{\langle x_{\nu} \rangle t_{\nu}} = t_{\nu} \quad \text{or} \quad \langle x_{\nu} \rangle = u'$$

Above we used the following integrals

$$T = \int_{0}^{\infty} e^{-tx} dx = -e^{-tx} \Big|_{0}^{\infty} = T$$

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$$T = \int_{0}^{\infty} e^{-tx} dx = e^{-tx} dx$$

$$= -e^{-tx} \int_{0}^{\infty} e^{-tx} dx = 0$$

$$T = \int_{0}^{\infty} e^{-tx} d$$

$$n! = \int dx e^{-x} x^n$$

$$= \int \frac{dx}{dx} e^{-x} x^{n+1} = \Gamma(n+1)$$

$$\Gamma(1/2) = \int_{\overline{X}}^{\infty} dx e^{-x} x^{1/2}$$

writing 
$$y = \sqrt{x}$$
,  $dy = 1$   $dx$ , or  $2\sqrt{x}$ 

$$2 dy = dx$$

$$y \times x$$

$$\Gamma(1/2) = 2 \int_{0}^{1} dy e^{-y^{2}} dy = \int_{-\infty}^{\infty} dy e^{-y^{2}} = \int_{-\infty}^{\infty} dy e^{-y^{2}} dy = \int_{-\infty}^{\infty} dy e^{-y} dy = \int_{-\infty}^{\infty} dy e^{$$

This is a gaussian integral 
$$\int_{-\infty}^{\infty} dx e^{-\frac{1}{2}x^2} = \sqrt{2\pi\sigma^2}$$

with 
$$\sigma^2 = V_2$$
 so  $\Gamma(V_2) = \sqrt{\Pi}$ 

$$\Gamma(x+1) = \int_0^x du e^{-u} u^x$$

$$= e^{-u} \times | + \int_{0}^{\infty} e^{-u} \times u^{\times -1}$$

$$= 0 + \times \int_{0}^{\infty} e^{-u} u^{x-1}$$

[d) So if 
$$\Gamma(7/2) = \frac{5}{2}\Gamma(\frac{5}{2}) = \frac{5}{2} \cdot \frac{3}{2}\Gamma(\frac{3}{2})$$
  
=  $\frac{5}{2} \cdot \frac{3}{2} \cdot \frac{1}{2}\Gamma(\frac{1}{2})$ 

$$= \frac{15}{8} \sqrt{\pi} \approx 3.3$$

$$2! < 15 \sqrt{15} < 3!$$
 or  $2 < 3.3 < 6$ 

(e) 
$$A_2 = 2 \pi^{2/2} \Gamma = 2 \pi \Gamma$$

$$A_3 = 2 \pi^{3/2} r^2 = 2\pi^{3/2} r^2$$

$$\frac{1}{2} \Gamma(1/2)$$

## Combinatorics and Stirling The number of selections is $\frac{N_A \cdot C_r}{\left(\frac{1}{3}N_A\right)! \left(\frac{2}{3}N_A\right)!} \quad \text{with } r = \frac{1}{3}N_A$ · Taking the log log NAC = log NA! - log ((= NA)!) - log ((= NA)!) = NA log NA - NA - (1 NA log (1 NA) - 1 NA) - (3 NA log (3 NA) - 3 NA) $= -\frac{1}{3} N_A \log(3) + \frac{2}{3} N_A \log(\frac{3}{2})$ = NA log (27) = 0.64 NA

NAC = e 0.64 NA = (e log lo) 0.64 Na/log lo = 10 0.64 NA/log lo = 10 1.66 × 10 23 Random Walk

(x) = 
$$pa - (1-p)a$$
  
(x) =  $a(2p-1)$ 

$$\langle x^2 \rangle = pa^2 + (1-p)a^2 = a^2$$

So

$$(x^{2})^{2} = a^{2}(1 - (2p-1)^{2})$$

$$= a^{2}(1 - 4p^{2} + 4p - 1)$$

$$\langle \chi \rangle = n \langle \chi \rangle = n (2p-1) \alpha$$

$$\times$$
 > 20 $\times$ 

Or n (2p-1)a > 2/4p(1-p) /n a · So √n > 4/p(1-p) 2(p-1/2) 17 P=1/2, so the mumerator is approx: · So if P = 1 + 0,0001 , we have 1 × 108

Speed of Nitrogen Gas. No diatomic nitrogen

$$PV = N \cdot k_BT$$

So

 $k_BT = PV = (5 \times 10^5 \,\text{N/m}^2) (6 \times (0.1 \,\text{m})^3)$ 
 $2 \times 6 \times 10^{23}$ 
 $= 2.5 \times 10^{21} \,\text{J} \quad k_B = V_{44} \,\text{eV}$ 
 $= 0.016 \,\text{eV} \quad T = 180^{\circ} \,\text{K}$ 

So the molecule has five degrees of freedom

 $E = 5 \, k_BT \simeq 0.040 \,\text{eV}$ 

The translational motion is 3-degrees of freedom
$$\langle 1 m v^2 \rangle = 3 \times 1 k_B T$$

$$S_{0}$$

$$\sqrt{\langle V^{2} \rangle} = \sqrt{3 k_{0} N_{A} T}$$

$$V_{rms} = \sqrt{3 k_{0} N_{A} T}$$

$$m N_{A}$$

Now 
$$k_B N_A = 8.32 \text{ J}$$
  $T = 180^{\circ} \text{K}$   
So  $m N_A = \text{molar mass} \approx 28 \text{ g} = 2 \times 14 \text{ g}$   
 $V_{rms} = \frac{3.8.32 \text{ J}}{\text{K}} \frac{180^{\circ} \text{ K}}{\text{V}} \frac{1}{2}$