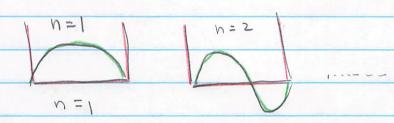
Why h in d3-d3p/h3?

· Now we will turn to why we have \$3 in

$$\int \frac{d^3r d^3p}{h^3}$$

· First consider a quantum mechanical "particle in a box" in Id. The wave functions are shown below



$$2V_n = \sqrt{\frac{2}{L}} \sin(k_n x)$$

$$2\nu_n = \sqrt{2} \sin(k_n x) \qquad E_n = \frac{1}{2} k_n^2 \qquad k_n = \overline{11} n$$

• The energy eigenstates are super-position of waves with momentum

$$Sink_{n}x = e^{ipx/t} - e^{-ipx/t}$$

· Then

Spacing between Energies is small.

Now

$$E_n = \frac{1}{4} \left(\frac{\pi n}{L} \right)^2 \propto \frac{1}{L^2} \left(\frac{\pi n}{L} \right)^2 \sim \frac{1}{L^2} \left(\frac{\pi n}{L} \right)^2 \sim$$

So since L is big can be replaced by an integral

The typical quantum number is when $E_n \sim k_B T$ $\frac{t^2 n}{2m L^2} \sim kT$ or $n \sim L^2 \frac{2mT}{t^2} \sim L^2 \gg 1$ sum over $\frac{t^2}{t^2} \sim L^2 \approx 1$ · So since the typical n in the sum over states is large we can replace the sum with an integral Z = (dne-E(n)/r now K= FTn so dn = Ldk and $\frac{2}{2} = \int \frac{Ldk}{T} e^{-E(k)/kT}$ to (-00, 00) not (0, ∞) · finally we can change the integration limits and divide by two, calling p= thk, to find $\frac{2}{2\pi} = \int \frac{d^{2}p}{2\pi t} e^{-\frac{E(p)}{kT}}$ $\frac{2}{2\pi} = \int \frac{d^{2}p}{h} e^{-\frac{E(p)}{kT}}$

are summing over state: We The sum over states becomes an integral whenever L/2+ >>1 , i.e. dimensions $E_n = \frac{1}{2} \left(k_x^2 + k_y^2 + k_z^2 \right)$ Kx = Thx And S wo over states becomes an integral over phase space with $1/h^3$ Kz=TTnz/L $\sum_{K_{x}} \sum_{k_{y}} \sum_{K_{z}} \frac{1^{3} dk_{x} dk_{y} dk_{z}}{T^{3}}$ phase his
space

to (-00, -00) for kx, ky and kz

and get 1/23

Why N!

Consider the same two level system (see slide)

Composed of two distinguishable subsystems. For example, the two subsystems might be a different physical locations, When we considered Natoms forming a crystal this was the case; i.e. one atom per site:

The states are

Then the partition function

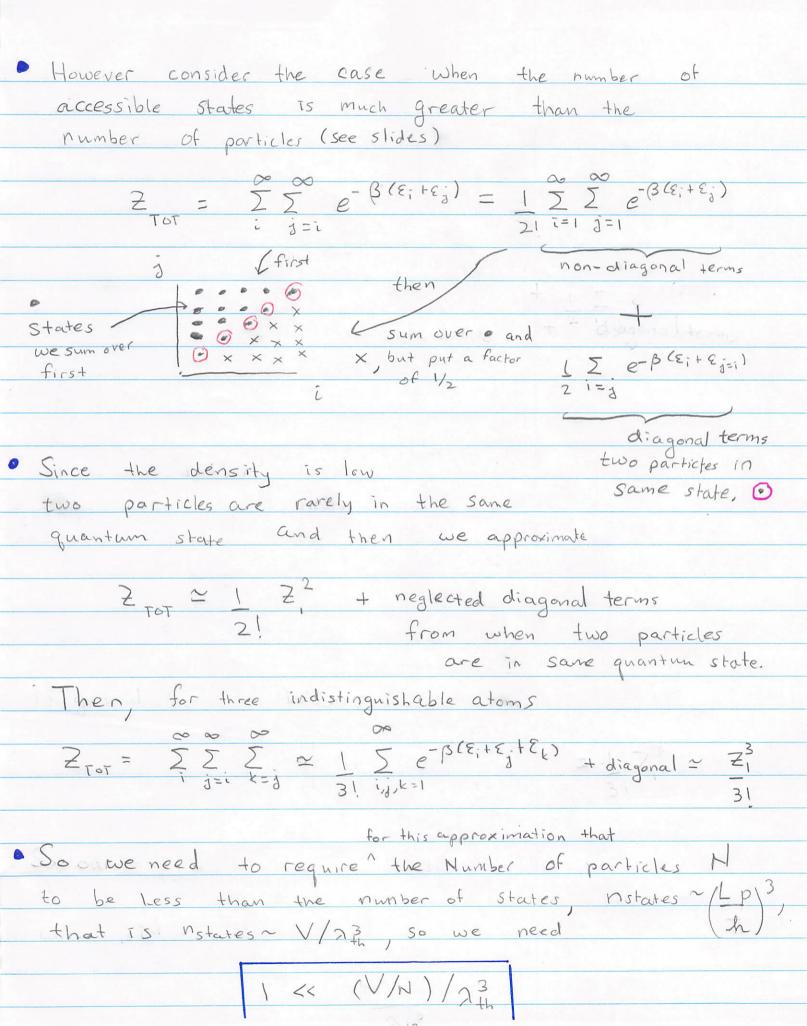
$$Z_{\overline{ior}} = e^{c} + e^{-\beta\Delta} + e^{-\beta\Delta} + e^{-2\beta\Delta}$$

$$= \sum_{i,j} e^{-\beta(\xi_{i}^{A} + \xi_{j}^{B})} = \sum_{i} e^{-\beta\xi_{i}^{A}} \sum_{i} e^{-\beta\xi_{j}^{B}} = (Z_{i}^{A})^{2}$$

With N particles, distinguishable, we get Zi

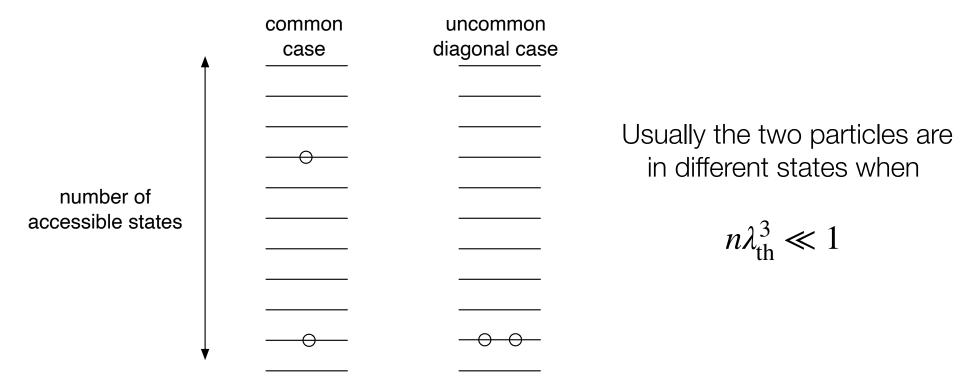
· However in the gas the particles are indistinguishable

$$Z = e^{0} + e^{-\beta \Delta} + e^{-2\beta \Delta} = \sum_{i=0}^{\infty} \sum_{j=i}^{\infty} e^{-\beta (E_{i} + E_{j})} + Z_{i}^{2}$$



Ideal Gas Limit:

The number of particles is much smaller than the number of quantum states



| So | in words we are requiring that the volume particle is large compared to the (debroglie 2) |
|------|---|
| Then | we have approximately |
| | $\sum_{\text{States}} e^{-\beta(\xi_1 + \dots + \xi_N)/kT} \simeq \sum_{\text{NI}} Z_1^{\text{NI}}$ |
| | of identical |
| | indistinguishable particles |
| | |