Entropy Revisited:

• Previously we considered each microstate of the full system to be equally likely. Thus

probability to be in microstate in is constant

IP = 1 or C \(\Sigma 1 = 1 \)

This litterally counts

the states

Or

 $C\Omega(E) = 1$ and P = 1 $\Omega(E)$

$$S = k \ln \Omega = -k \ln P_m$$

Now we have a subsystem with probabilies

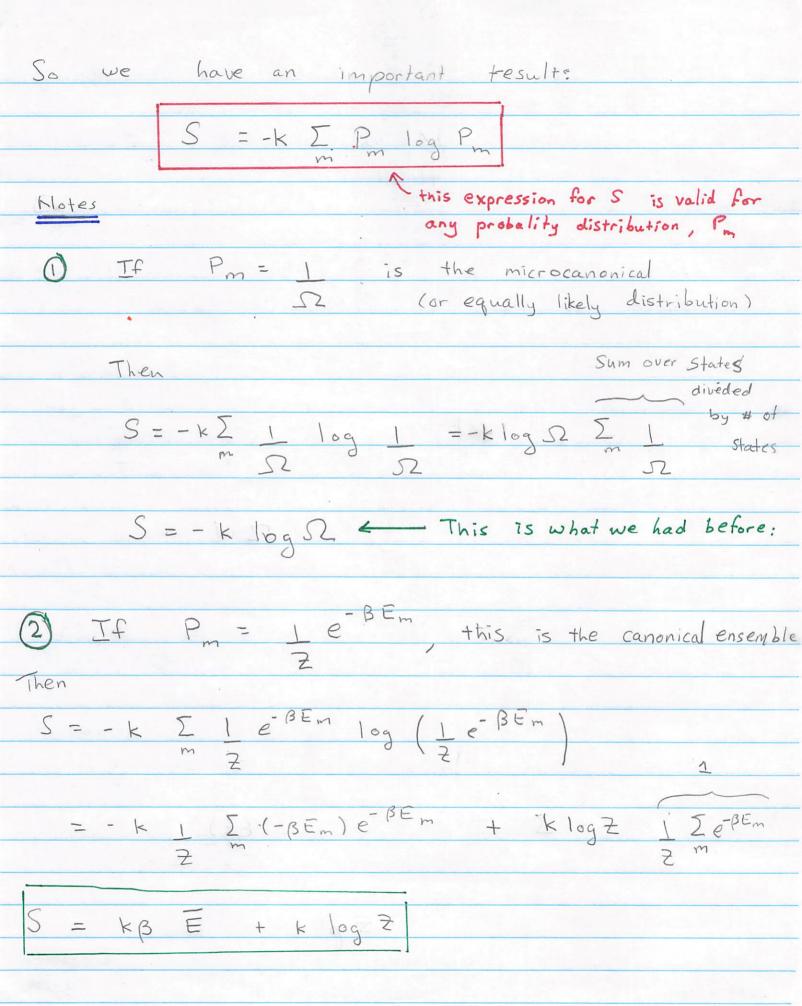
Pm x e- EXKT. We would to find the mean entropy for the subsystems, with this probability distribution

The generalization is (as we show below)

$$S = -\langle k \ln P_m \rangle = -k \sum_{m} P_m \ln P_m$$

Let's work it out!

Consider a large number of subsystems 1 N subsystems: (blue boxes) 3 2 1 × N, in State 1: 3 2 1 × N2 in State 2, etc for example, probability to -> P = N, x e = E,/kT Stor = Kln (# configurations with N, Nz, fixed) = k In N,! N, N, = K (NInN-N - I(Nm In Nm - Nm)) · So using INm = N we have Stor = K & Nm In N - Nm In Nm Using InN-InNm = - In Nm/N = - In Pm we have Stor = - K Z Nm In Pm · Finally the mean entropy for one subsystem is S = Stot = -k I Nm In Pm = -k I Pm In Pm



$$S = \frac{1}{T} + k \log 2$$

this is how the entropy can be computed from the partition function ?

