The	Micro-canonical	Algorithm

- · The preceding discussion leads to the algorithm
- We have a system (E.g. N harmonic oscillators) with total energy E (e.g. E=qtw), and want the temperature.
- We "just" need to count the number of ways $\Omega(E)$ for the system to Share (or partition) the energy. This determines the entropy S(E):

Then

This determines the relation between the temperature and energy E(T).

Example	of	Microcanonical	Algorithm
			-

- · You will do this in homework. So I will only sketch the steps here
- Take N harmonic oscillators with quinter of energy E=qtw. so the average number of vibrational quanta per oscillator is

$$\vec{n} = q = E$$
We will

N to wooder $\vec{n} = 1$

N = 400

X mo X mo X mo

Then
$$SC(q) = (N+q-1)!$$
 $q! (N-1)!$

(Compute S)

 $\Omega(q)$ is the number of states with q units of energy and N particles, which you will compute in

· Using the stirling approximation (homework!)

Then

- S(E) = kIn D(E) = NkB [(1+ n) In (1+ n) n In n]
- Then differentiating with respect to energy N = E

$$\frac{1}{T} = \frac{\partial S}{\partial E} = \frac{\partial S}{\partial n} \frac{1}{N t_{w_0}}$$

$$\frac{1}{K_B T} = \frac{1}{t_{w_0}} \frac{1}{n} \frac{1}{N t_{w_0}}$$
• So, for $n = 1$, $k_B T = \frac{1}{t_{w_0}} \frac{1}{N t_{w_0}} = \frac{1}{N t_{w_0}}$
• We can also express n in terms of the temperature
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Analysis of Thermal State

- We found the temperature of the system

 KBT = hwo/In2 or Bhw = In2
- · We can verify that this is correct numerically.
- Each site is an independent subsystem. The probability of a site having energy \mathcal{E}_n is $P(\mathcal{E}_n) \propto e^{-\mathcal{E}_n/k_BT}$

Energy is flowing in and out of every site. The probability that a site "steals" energy ϵ_n from the bath is given by the Boltzmann factor (see slides).

$$N(n) \propto e^{-n \ln 2}$$

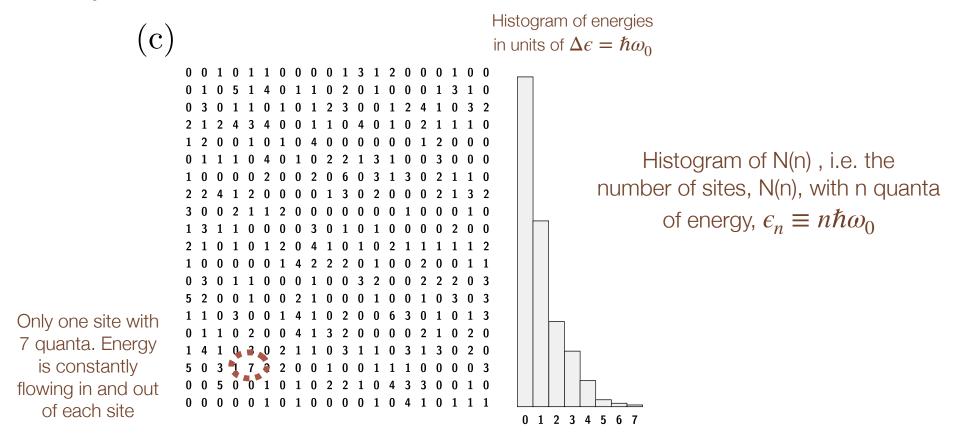
Each independent subsystem (one site in this case) has a probability distribution:

$$P_s = \frac{1}{Z}e^{-\epsilon_s/k_BT}$$

This is known as the **canonical ensemble**. The energy in each subsystem is variable, as each subsystem can "steal" energy from the others. Using the canonical ensemble we can calculate the mean properties of the subsystem and deduce the properties of the total system, i.e. the total system is just N copies of the subsystem.

The collection of subsystems (i.e. the 20x20 square of subsystems) is known as the **microcanonical ensemble**. The energy doesn't change in the total system. We can also calculate the properties of the total system by following the microcanonical approach, finding S(E) and using $(\partial S/\partial E) = 1/T$. The two approaches are identical. In the next section we will derive the Boltzmann factor from the microcanonical ensemble.

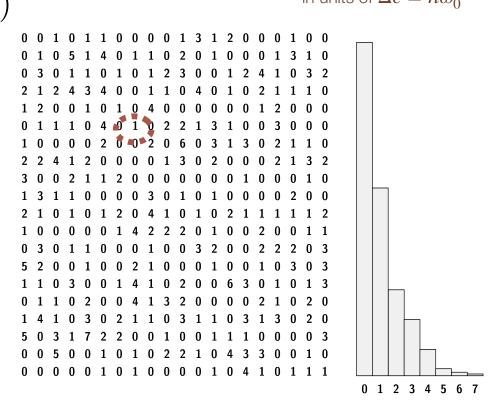
Analysis of the thermal state



A typical histogram of the number quanta is shown above: What is N(n)?

What is N(n)?





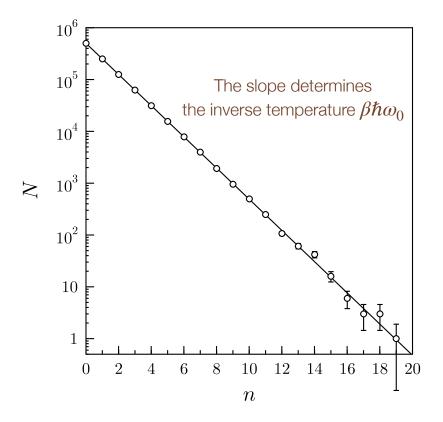
Pick a site: The remaining sites are the reservoir

Expect the probability for a site to have n quanta to be:

$$P(\epsilon_n) \propto e^{-\beta \epsilon_n} = e^{-n\beta\hbar\omega_0}$$

The histogram N(n) is the number of sites with n quanta, and should be P_n up to normalization

Numerical verification: number of sites, N(n), with n quanta on 1000x1000 grid



What you are seeing (on a log scale) is

$$N(n) = N_0 e^{-Cn}$$

The log of N(n) is the line you see

$$ln N(n) = ln N_0 - Cn$$

The slope should be $C=\beta\hbar\omega_0=\ln 2$ set by the temperature. It is!

We found temperature in this problem $k_BT=\hbar\omega_0/\ln 2$ by counting possibilities!