# Numerical solution of the Schrödinger equation for the hydrogen atom

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#### Introduction

The nonrelativistic Schrödinger equation for the hydrogen atom is solved numerically using the finite volume method. Central differencing is applied to the diffusion terms. Time discretization is fully implicit. The resulting linear system is solved using the Gauss-Seidel method.

## Model equations

The three-dimensional Schrödinger equation is:

$$i\hbar\frac{\partial\psi}{\partial t} = -\frac{h^2}{2m}\nabla^2\psi - \frac{k}{r}\psi\tag{1}$$

Where i is the imaginary number, h a constant, m the particle mass, k a constant depending on the charge of the electron and radial coordinate r,  $\psi$  the dependent variable and t time. Expanding the Laplacian in (1) in spherical coordinates gives:

$$ih\frac{\partial\psi}{\partial t} = -\frac{h^2}{2m} \left[ \frac{1}{r^2} \left( \frac{\partial}{\partial r} \frac{r^2 \partial\psi}{\partial r} \right) + \frac{1}{r^2 \sin(\phi)} \frac{\partial}{\partial \phi} \left( \sin(\phi) \frac{\partial\psi}{\partial \phi} \right) + \frac{1}{r^2 \sin^2(\phi)} \frac{\partial^2\psi}{\partial \theta^2} \right] - \frac{k}{r} \psi$$
(2)

Where  $\phi$  is the angle between the direction vector  $\mathbf{r}$  and the z axis and  $\theta$  the angle between the projection of  $\mathbf{r}$  onto the xy plane and the x axis. The boundary conditions of the system are  $\psi = 0$  at r = 0 and at r = R with R the length of the domain in the r direction.

#### Discretization

The domain V of the system is partitioned into  $n_r$  nodes in the radial direction,  $n_\theta$  nodes in the  $\theta$  direction and  $n_\phi$  nodes in the  $\phi$  direction. Two additional linear grids represent the 'polar' regions. The nodes of these linear grids lie on the z axis, i.e.: at  $\phi = 0, \pi$ .

### Verification

In order to verify that the computations proceed correctly it is verified that the linear system obtained from discretization of equation (1) is satisfied. Analysis of results show that the linear system is satisfied with an error in the order of 1e-12.