Udaan 2025

Chemistry

Essential Chemistry Basics for Class 10th

DHA - 01

- **Q 1** Select the correct electronic configuration of Mg atom:
 - (A) 3,8,1
- (B) 1,8,3
- (C) 2,8,2
- (D) 2,2,8
- **Q 2** How many electrons can be accommodated in the third shell of an atom?
 - (A) 18
- **(**B) 8
- (C) 16
- **(**D) 28
- **Q 3** If a element have atomic number 7 then what will be the no of protons?

- (A) 7 (C) 9
- (B) 8
- (D) 6
- **Q 4** If the configuration of Na is 2, 8, 1 then the valency of Na will be-
 - (A) 1
- (B) 6
- (C) 0
- (D) 4
- **Q 5** What will be the valency of sulphur:
 - (A) 2
- (B) 3
- (C) 1
- (D) 4

Answer Key

Q1 C

Q2 A

Q3 A

Q4 A

Q5 **A**



Hints & Solutions

Q 1 Video Solution:



Q 2 Text Solution:

The maximum number of electrons present in a shell is given by : $2n^2$

Video Solution:



Q 3 Text Solution:

The number of protons in the nucleus of the atom is equal to the atomic number. Think about the correct answer.

Video Solution:



Q 4 Video Solution:



Q 5 Text Solution:

Valency is the capacity of combining with other elements. Think about it.

Video Solution:



