JEE 2023 Session-1 24th Jan to 1st Feb 2023

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Test Date	30/01/2023
Test Time	3:00 PM - 6:00 PM
Subject	B TECH

Section : **Physics Section A**

Q.1

Match List I with List II:

	List I		List II
A.	Attenuation	I.	Combination of a receiver and transmitter.
B.	Transducer	II.	process of retrieval of information from the carrier wave at receiver
C.	Demodulation	III.	converts one form of energy into another
D.	Repeater	IV.	Loss of strength of a signal while propogating through a medium.

Choose the **correct** answer from the options given below:

Options
1. A-I,   B-II,   C-III, D-IV
2. A-IV, B-III, C-II,   D-I
3. A-II, B-III, C-IV, D-I
4. A-IV, B-III, C-I,   D-II

Question Type : **MCQ**
Question ID : **3666942398**
Option 1 ID : **3666947570**
Option 2 ID : **3666947571**
Option 3 ID : **3666947572**
Option 4 ID : **3666947569**
Status : **Answered**
Chosen Option : **2**

Q.2 A machine gun of mass 10 kg fires 20 g bullets at the rate of 180 bullets per minute with a speed of 100 m s^{-1} each. The recoil velocity of the gun is

Options
1. 1.5 m/s
2. 0.02 m/s
3. 0.6 m/s
4. 2.5 m/s

Question Type : **MCQ**
Question ID : **3666942395**
Option 1 ID : **3666947559**
Option 2 ID : **3666947557**
Option 3 ID : **3666947558**
Option 4 ID : **3666947560**
Status : **Answered**
Chosen Option : **3**

Q.3 An object is allowed to fall from a height R above the earth, where R is the radius of earth. Its velocity when it strikes the earth's surface, ignoring air resistance, will be

Options 1. $2\sqrt{gR}$

2. $\sqrt{2gR}$

3. $\sqrt{\dfrac{gR}{2}}$

4. $\sqrt{gR}$

Question Type : **MCQ**
Question ID : **3666942394**
Option 1 ID : **3666947555**
Option 2 ID : **3666947553**
Option 3 ID : **3666947556**
Option 4 ID : **3666947554**
Status : **Answered**
Chosen Option : **2**

Q.4 A vehicle travels 4 km with speed of 3 km / h and another 4 km with speed of 5 km / h, then its average speed is

Options 1. 3.50 km / h

2. 4.00 km / h

3. 4.25 km / h

4. 3.75 km / h

Question Type : **MCQ**
Question ID : **3666942397**
Option 1 ID : **3666947565**
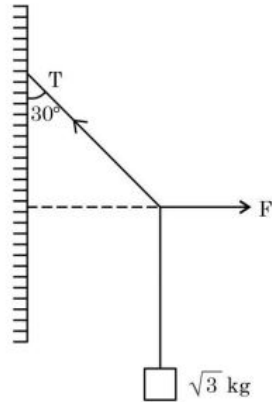Option 2 ID : **3666947567**
Option 3 ID : **3666947568**
Option 4 ID : **3666947566**
Status : **Answered**
Chosen Option : **4**

Q.5 A block of $\sqrt{3}$ kg is attached to a string whose other end is attached to the wall. An unknown force F is applied so that the string makes an angle of 30° with the wall. The tension T is: (Given g = 10 ms⁻²)

Options
1. 20 N

2. 25 N

3. 15 N

4. 10 N

Question Type : **MCQ**
Question ID : **3666942396**
Option 1 ID : **3666947563**
Option 2 ID : **3666947564**
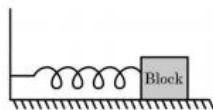Option 3 ID : **3666947562**
Option 4 ID : **3666947561**
Status : **Not Attempted and Marked For Review**
Chosen Option : **--**

Q.6 For a simple harmonic motion in a mass spring system shown, the surface is frictionless. When the mass of the block is 1 kg, the angular frequency is ω_1. When the mass block is 2 kg the angular frequency is ω_2. The ratio ω_2/ω_1 is

Options
1. $\sqrt{2}$

2. 2

3. $1/\sqrt{2}$

4. $1/2$

Question Type : **MCQ**
Question ID : **3666942410**
Option 1 ID : **3666947617**
Option 2 ID : **3666947619**
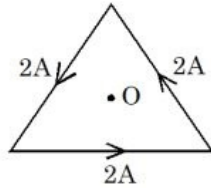Option 3 ID : **3666947618**
Option 4 ID : **3666947620**
Status : **Answered**
Chosen Option : **2**

Q.7 As shown in the figure, a current of 2A flowing in an equilateral triangle of side $4\sqrt{3}$ cm. The magnetic field at the centroid O of the triangle is

(Neglect the effect of earth's magnetic field)

Options

1. $4\sqrt{3} \times 10^{-5}$ T

2. $4\sqrt{3} \times 10^{-4}$ T

3. $3\sqrt{3} \times 10^{-5}$ T

4. $\sqrt{3} \times 10^{-4}$ T

Question Type : **MCQ**
Question ID : **3666942402**
Option 1 ID : **3666947586**
Option 2 ID : **3666947588**
Option 3 ID : **3666947585**
Option 4 ID : **3666947587**
Status : **Answered**
Chosen Option : **2**

Q.8 A thin prism P_1 with an angle 6° and made of glass of refractive index 1.54 is combined with another prism P_2 made from glass of refractive index 1.72 to produce dispersion without average deviation. The angle of prism P_2 is

Options

1. $7.8°$

2. $1.3°$

3. $6°$

4. $4.5°$

Question Type : **MCQ**
Question ID : **3666942406**
Option 1 ID : **3666947604**
Option 2 ID : **3666947601**
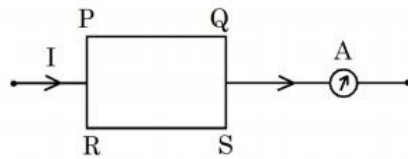Option 3 ID : **3666947602**
Option 4 ID : **3666947603**
Status : **Answered**
Chosen Option : **2**

Q.9 A current carrying rectangular loop $PQRS$ is made of uniform wire. The length $PR = QS = 5$ cm and $PQ = RS = 100$ cm. If ammeter current reading changes from I to $2I$, the ratio of magnetic forces per unit length on the wire PQ due to wire RS in the two cases respectively $\left(f_{PQ}^{I} : f_{PQ}^{2I}\right)$ is:

Options
1. $1:5$
2. $1:4$
3. $1:2$
4. $1:3$

Question Type : **MCQ**
Question ID : **3666942403**
Option 1 ID : **3666947592**
Option 2 ID : **3666947590**
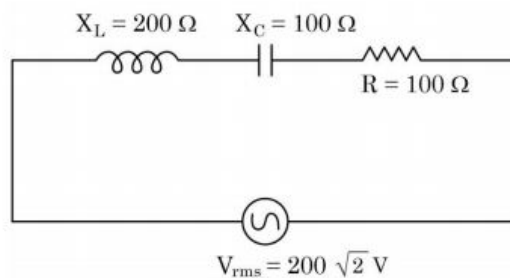Option 3 ID : **3666947589**
Option 4 ID : **3666947591**
Status : **Answered**
Chosen Option : **2**

Q.10 In the given circuit, rms value of current (I_{rms}) through the resistor R is:

Options
1. 20 A
2. $\dfrac{1}{2}$ A
3. 2 A
4. $2\sqrt{2}$ A

Question Type : **MCQ**
Question ID : **3666942404**
Option 1 ID : **3666947595**
Option 2 ID : **3666947594**
Option 3 ID : **3666947596**
Option 4 ID : **3666947593**
Status : **Answered**
Chosen Option : **3**

Q.11

Given below are two statements: one is labelled as **Assertion A** and the other is labelled as **Reason R**

Assertion A: The nuclear density of nuclides $^{10}_{5}B$, $^{6}_{3}Li$, $^{56}_{26}Fe$, $^{20}_{10}Ne$ and $^{209}_{83}Bi$ can be arranged as $\rho^N_{Bi} > \rho^N_{Fe} > \rho^N_{Ne} > \rho^N_{B} > \rho^N_{Li}$

Reason R: The radius R of nucleus is related to its mass number A as $R = R_0\, A^{1/3}$, where R_0 is a constant.

In the light of the above statements, choose the **correct** answer from the options given below

Options
1. Both **A** and **R** are true and **R** is the correct explanation of **A**

2. **A** is false but **R** is true

3.
Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**

4. **A** is true but **R** is false

Question Type : **MCQ**
Question ID : **3666942408**
Option 1 ID : **3666947609**
Option 2 ID : **3666947612**
Option 3 ID : **3666947610**
Option 4 ID : **3666947611**
Status : **Answered**
Chosen Option : **2**

Q.12

A point source of 100 W emits light with 5% efficiency. At a distance of 5 m from the source, the intensity produced by the electric field component is:

Options
1. $\dfrac{1}{40\pi}\dfrac{W}{m^2}$

2. $\dfrac{1}{10\pi}\dfrac{W}{m^2}$

3. $\dfrac{1}{2\pi}\dfrac{W}{m^2}$

4. $\dfrac{1}{20\pi}\dfrac{W}{m^2}$

Question Type : **MCQ**
Question ID : **3666942405**
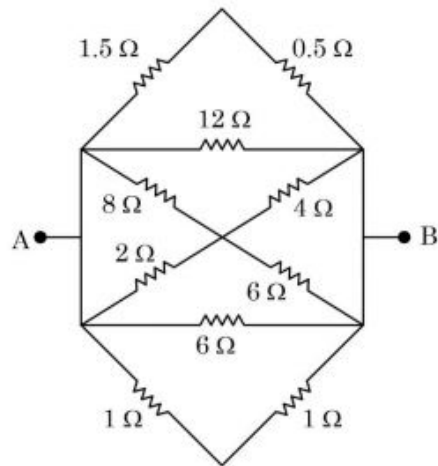Option 1 ID : **3666947599**
Option 2 ID : **3666947600**
Option 3 ID : **3666947598**
Option 4 ID : **3666947597**
Status : **Answered**
Chosen Option : **2**

Q.13

The equivalent resistance between A and B is _____ .

Options

1. $\dfrac{1}{3}\,\Omega$

2. $\dfrac{3}{2}\,\Omega$

3. $\dfrac{2}{3}\,\Omega$

4. $\dfrac{1}{2}\,\Omega$

Question Type : **MCQ**
Question ID : **3666942401**
Option 1 ID : **3666947581**
Option 2 ID : **3666947582**
Option 3 ID : **3666947583**
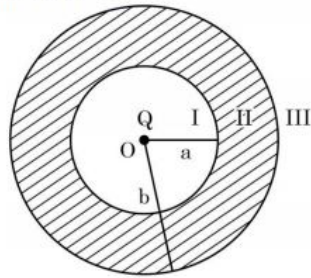Option 4 ID : **3666947584**
Status : **Answered**
Chosen Option : **3**

Q.14

As shown in the figure, a point charge Q is placed at the centre of conducting spherical shell of inner radius a and outer radius b. The electric field due to charge Q in three different regions I, II and III is given by:

(I : $r < a$, II : $a < r < b$, III: $r > b$)

Options

1. $E_I \neq 0, E_{II} = 0, E_{III} \neq 0$

2. $E_I = 0, E_{II} = 0, E_{III} \neq 0$

3. $E_I = 0, E_{II} = 0, E_{III} = 0$

4. $E_I \neq 0, E_{II} = 0, E_{III} = 0$

Question Type : **MCQ**
Question ID : **3666942399**
Option 1 ID : **3666947573**
Option 2 ID : **3666947574**
Option 3 ID : **3666947576**
Option 4 ID : **3666947575**
Status : **Answered**
Chosen Option : **2**

Q.15

other is labelled as **Reason R**

Assertion A: Efficiency of a reversible heat engine will be highest at -273°C temperature of cold reservoir.

Reason R: The efficiency of Carnot's engine depends not only on temperature of cold reservoir but it depends on the temperature of hot reservoir too and is given as $\eta = \left(1 - \dfrac{T_2}{T_1}\right)$.

In the light of the above statements, choose the **correct** answer from the options given below

Options

1. Both **A** and **R** are true and **R** is the correct explanation of **A**

2. **A** is true but **R** is false

3.

Both **A** and **R** are true but **R** is **NOT** the correct explanation of **A**

4. **A** is false but **R** is true

Question Type : **MCQ**
Question ID : **3666942392**
Option 1 ID : **3666947545**
Option 2 ID : **3666947547**
Option 3 ID : **3666947546**
Option 4 ID : **3666947548**
Status : **Answered**
Chosen Option : **1**

Q.16 Match List I with List II:

	List I		List II
A.	Torque	I.	kg m^{-1} s^{-2}
B.	Energy density	II.	kg ms^{-1}
C.	Pressure gradient	III.	kg m^{-2} s^{-2}
D.	Impulse	IV.	kg m^2 s^{-2}

Choose the *correct* answer from the options given below:

Options
1. A-IV, B-III, C-I, D-II

2. A-IV, B-I, C-III, D-II

3. A-IV, B-I, C-II, D-III

4. A-I, B-IV, C-III, D-II

Question Type : **MCQ**
Question ID : **3666942391**
Option 1 ID : **3666947541**
Option 2 ID : **3666947542**
Option 3 ID : **3666947543**
Option 4 ID : **3666947544**
Status : **Answered**
Chosen Option : **2**

Q.17 An electron accelerated through a potential difference V_1 has a de-Broglie wavelength of λ. When the potential is changed to V_2, its de-Broglie wavelength increases by 50%. The value of $\left(\dfrac{V_1}{V_2}\right)$ is equal to

Options
1. 3

2. $\dfrac{3}{2}$

3. $\dfrac{9}{4}$

4. 4

Question Type : **MCQ**
Question ID : **3666942407**
Option 1 ID : **3666947607**
Option 2 ID : **3666947606**
Option 3 ID : **3666947605**
Option 4 ID : **3666947608**
Status : **Answered**
Chosen Option : **3**

Q.18

A force is applied to a steel wire 'A', rigidly clamped at one end. As a result elongation in the wire is 0.2 mm. If same force is applied to another steel wire 'B' of double the length and a diameter 2.4 times that of the wire 'A', the elongation in the wire 'B' will be (wires having uniform circular cross sections)

Options 1. 6.9×10^{-2} mm

2. 6.06×10^{-2} mm

3. 3.0×10^{-2} mm

4. 2.77×10^{-2} mm

Question Type : **MCQ**
Question ID : **3666942393**
Option 1 ID : **3666947549**
Option 2 ID : **3666947552**
Option 3 ID : **3666947551**
Option 4 ID : **3666947550**
Status : **Answered**
Chosen Option : **1**

Q.19

A flask contains hydrogen and oxygen in the ratio of 2:1 by mass at temperature 27°C. The ratio of average kinetic energy per molecule of hydrogen and oxygen respectively is:

Options 1. $4 : 1$

2. $1 : 4$

3. $1 : 1$

4. $2 : 1$

Question Type : **MCQ**
Question ID : **3666942400**
Option 1 ID : **3666947579**
Option 2 ID : **3666947578**
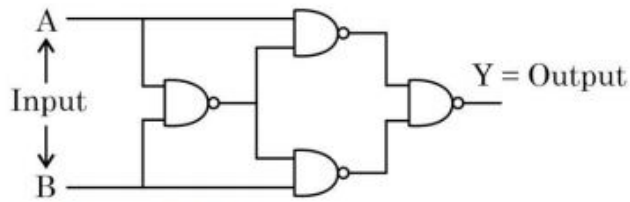Option 3 ID : **3666947580**
Option 4 ID : **3666947577**
Status : **Answered**
Chosen Option : **1**

Q.20

The output Y for the inputs A and B of circuit is given by

Truth table of the shown circuit is:

Options

1.

A	B	Y
0	0	1
0	1	1
1	0	1
1	1	0

2.

A	B	Y
0	0	0
0	1	1
1	0	1
1	1	0

3.

A	B	Y
0	0	1
0	1	0
1	0	0
1	1	1

4.

A	B	Y
0	0	0
0	1	1
1	0	1
1	1	1

Question Type : **MCQ**

Question ID : **3666942409**

Option 1 ID : **3666947616**

Option 2 ID : **3666947613**

Option 3 ID : **3666947615**

Option 4 ID : **3666947614**

Status : **Answered**

Chosen Option : **2**

Section : **Physics Section B**

Q.21

A stone tied to 180 cm long string at its end is making 28 revolutions in horizontal circle in every minute. The magnitude of acceleration of stone is $\frac{1936}{x}$ ms^{-2}. The value of x _____. (Take $\pi = \frac{22}{7}$)

Given **--**
Answer :

Question Type : **SA**

Question ID : **3666942414**

Status : **Not Attempted and Marked For Review**

Q.22 In an ac generator, a rectangular coil of 100 turns each having area $14 \times 10^{-2} \, m^2$ is rotated at 360 rev/min about an axis perpendicular to a uniform magnetic field of magnitude 3.0 T. The maximum value of the emf produced will be _____ V.

$\left(\text{Take } \pi = \dfrac{22}{7} \right)$

Given Answer : **44**

Question Type : **SA**
Question ID : **3666942417**
Status : **Answered**

Q.23 A radioactive nucleus decays by two different process. The half life of the first process is 5 minutes and that of the second process is 30 s. The effective half-life of the nucleus is calculated to be $\dfrac{\alpha}{11}$ s. The value of α is _____.

Given Answer : **--**

Question Type : **SA**
Question ID : **3666942419**
Status : **Not Attempted and Marked For Review**

Q.24 A faulty thermometer reads 5°C in melting ice and 95°C in stream. The correct temperature on absolute scale will be _____ K when the faulty thermometer reads 41°C.

Given Answer : **--**

Question Type : **SA**
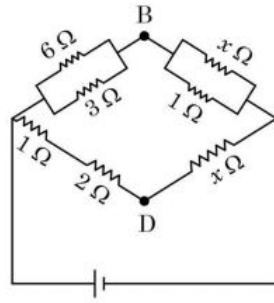Question ID : **3666942411**
Status : **Not Answered**

Q.25 In a Young's double slit experiment, the intensities at two points, for the path differences $\dfrac{\lambda}{4}$ and $\dfrac{\lambda}{3}$ (λ being the wavelength of light used) are I_1 and I_2 respectively. If I_0 denotes the intensity produced by each one of the individual slits, then $\dfrac{I_1 + I_2}{I_0} =$ _____.

Given Answer : **--**

Question Type : **SA**
Question ID : **3666942418**
Status : **Not Answered**

Q.26

If the potential difference between B and D is zero, the value of x is $\frac{1}{n}\Omega$. The value of n is _____.
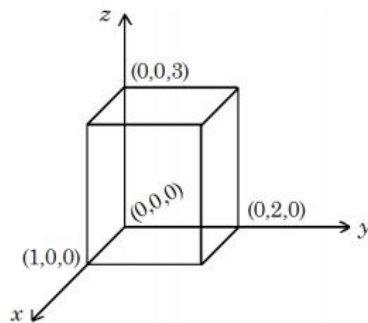

Given **2**
Answer :

Question Type : **SA**
Question ID : **3666942416**
Status : **Answered**

Q.27

As shown in figure, a cuboid lies in a region with electric field $E = 2x^2\hat{i} - 4y\hat{j} + 6\hat{k}$ N/C. The magnitude of charge within the cuboid is $n\,\epsilon_0\,C$.
The value of n is _____ (if dimension of cuboid is $1 \times 2 \times 3\,\text{m}^3$).

Given **--**
Answer :

Question Type : **SA**
Question ID : **3666942415**
Status : **Not Answered**

Q.28

A body of mass 2 kg is initially at rest. It starts moving unidirectionally under the influence of a source of constant power P. Its displacement in 4s is $\frac{1}{3}\alpha^2\sqrt{P}$ m. The value of α will be _____.
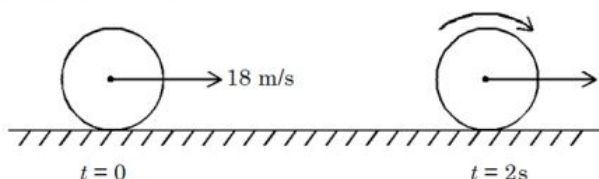
Given **--**
Answer :

Question Type : **SA**
Question ID : **3666942413**
Status : **Not Attempted and Marked For Review**

Q.29   A uniform disc of mass 0.5 kg and radius r is projected with velocity 18 m/s at $t = 0$s on a rough horizontal surface. It starts off with a purely sliding motion at $t = 0$s. After 2s it acquires a purely rolling motion (see figure). The total kinetic energy of the disc after 2s will be _____ J (given, coefficient of friction is 0.3 and $g = 10$ m/s^2).

Given --
Answer :

Question Type : **SA**
Question ID : **3666942412**
Status : **Not Answered**

Q.30   The velocity of a particle executing SHM varies with displacement (x) as $4v^2 = 50 - x^2$. The time period of oscillations is $\dfrac{x}{7}$ s. The value of x is _____.

$\left(\text{Take } \pi = \dfrac{22}{7} \right)$

Given --
Answer :

Question Type : **SA**
Question ID : **3666942420**
Status : **Not Answered**

Section : **Chemistry Section A**

Q.31   Given below are two statements: One is labelled as **Assertion A** and the other is labelled as **Reason R**.

Assertion A: Antihistamines do not affect the secretion of acid in stomach.

Reason R   : Antiallergic and antacid drugs work on different receptors.

In the light of the above statements, choose the **correct** answer from the options given below:

Options 1.
Both A and R are true but R is not the correct explanation of A

2. Both A and R are true and R is the correct explanation of A

3. A is false but R is true

4. A is true but R is false

Question Type : **MCQ**
Question ID : **3666942440**
Option 1 ID : **3666947708**
Option 2 ID : **3666947707**
Option 3 ID : **3666947710**
Option 4 ID : **3666947709**
Status : **Answered**
Chosen Option : **1**

Q.32
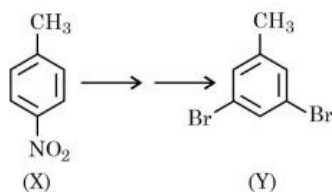
In the above conversion of compound (X) to product (Y), the sequence of reagents to be used will be:

Options

1. (i) $Br_2(aq)$ (ii) $LiAlH_4$ (iii) H_3O^+

2. (i) Fe, H^+ (ii) $Br_2(aq)$ (iii) HNO_2 (iv) $CuBr$

3. (i) Br_2, Fe (ii) Fe, H^+ (iii) $LiAlH_4$

4. (i) Fe, H^+ (ii) $Br_2(aq)$ (iii) HNO_2 (iv) H_3PO_2

Question Type : **MCQ**
Question ID : **3666942439**
Option 1 ID : **3666947704**
Option 2 ID : **3666947705**
Option 3 ID : **3666947703**
Option 4 ID : **3666947706**
Status : **Answered**
Chosen Option : **2**

Q.33

Given below are two statements:

Statement I: During Electrolytic refining, the pure metal is made to act as anode and its impure metallic form is used as cathode.

Statement II: During the Hall-Heroult electrolysis process, purified Al_2O_3 is mixed with Na_3AlF_6 to lower the melting point of the mixture.

In the light of the above statements, choose the **most appropriate** answer from the options given below:

Options

1. Statement I is correct but Statement II is incorrect

2. Statement I is incorrect but Statement II is correct

3. Both Statement I and Statement II are correct

4. Both Statement I and Statement II are incorrect

Question Type : **MCQ**
Question ID : **3666942425**
Option 1 ID : **3666947649**
Option 2 ID : **3666947650**
Option 3 ID : **3666947647**
Option 4 ID : **3666947648**
Status : **Answered**
Chosen Option : **2**

Q.34 The water quality of a pond was analysed and its BOD was found to be 4. The pond has

Options
1. Very clean water

2. Slightly polluted water

3. Highly polluted water

4. Water has high amount of fluoride compounds

Question Type : **MCQ**
Question ID : **3666942431**
Option 1 ID : **3666947671**
Option 2 ID : **3666947672**
Option 3 ID : **3666947673**
Option 4 ID : **3666947674**
Status : **Answered**
Chosen Option : **1**

Q.35 The wave function (Ψ) of 2s is given by

$$\Psi_{2s} = \frac{1}{2\sqrt{2\pi}}\left(\frac{1}{a_0}\right)^{1/2}\left(2 - \frac{r}{a_0}\right)e^{-r/2a_0}$$

At $r = r_0$, radial node is formed. Thus, r_0 in terms of a_0

Options
1. $r_0 = 2a_0$

2. $r_0 = \dfrac{a_0}{2}$

3. $r_0 = a_0$

4. $r_0 = 4a_0$

Question Type : **MCQ**
Question ID : **3666942422**
Option 1 ID : **3666947636**
Option 2 ID : **3666947637**
Option 3 ID : **3666947635**
Option 4 ID : **3666947638**
Status : **Answered**
Chosen Option : **2**

Q.36 Given below are two statements: One is labelled as Assertion A and the other is labelled as Reason R.

Assertion A: [structure of 2-hydroxy carbonyl compound with OH and O groups] can be easily reduced using Zn-Hg/HCl to [structure of reduced compound with OH].

Reason R: Zn-Hg/HCl is used to reduce carbonyl group to $-CH_2-$ group.

In the light of the above statements, choose the **correct** answer from the options given below:

Options 1. A is false but R is true

2. Both A and R are true and R is the correct explanation of A

3. A is true but R is false

4.
Both A and R are true but R is not the correct explanation of A

Question Type : **MCQ**
Question ID : **3666942438**
Option 1 ID : **3666947702**
Option 2 ID : **3666947699**
Option 3 ID : **3666947701**
Option 4 ID : **3666947700**
Status : **Answered**
Chosen Option : **2**

Q.37 Match List I with List II:

	List I (Mixture)		List II (Separation Technique)
A.	$CHCl_3 + C_6H_5NH_2$	I.	Steam distillation
B.	$C_6H_{14} + C_5H_{12}$	II.	Differential extraction
C.	$C_6H_5NH_2 + H_2O$	III.	Distillation
D.	Organic compound in H_2O	IV.	Fractional distillation

Options 1. A-III, B-IV, C-I,   D-II

2. A-II, B-I,   C-III, D-IV

3. A-IV, B-I,   C-III, D-II

4. A-III, B-I,   C-IV, D-II

Question Type : **MCQ**
Question ID : **3666942434**
Option 1 ID : **3666947684**
Option 2 ID : **3666947683**
Option 3 ID : **3666947685**
Option 4 ID : **3666947686**
Status : **Answered**
Chosen Option : **2**

Q.38 Which of the following reaction is correct?

Options
1. $2\,LiNO_3 \longrightarrow 2Li + 2NO_2 + O_2$

2. $4\,LiNO_3 \xrightarrow{\Delta} 2Li_2O + 2N_2O_4 + O_2$

3. $2\,LiNO_3 \xrightarrow{\Delta} 2NaNO_2 + O_2$

4. $4\,LiNO_3 \xrightarrow{\Delta} 2Li_2O + 4NO_2 + O_2$

Question Type : **MCQ**
Question ID : **3666942427**
Option 1 ID : **3666947658**
Option 2 ID : **3666947655**
Option 3 ID : **3666947657**
Option 4 ID : **3666947656**
Status : **Answered**
Chosen Option : **4**

Q.39 The correct order of pK_a values for the following compounds is:
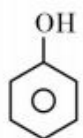
Options
1. $a > b > c > d$

2. $b > a > d > c$

3. $b > d > a > c$

4. $c > a > d > b$

Question Type : **MCQ**
Question ID : **3666942437**
Option 1 ID : **3666947695**
Option 2 ID : **3666947698**
Option 3 ID : **3666947696**
Option 4 ID : **3666947697**
Status : **Answered**
Chosen Option : **2**

Q.40 $KMnO_4$ oxidises I^- in acidic and neutral/faintly alkaline solution, respectively, to

Options 1. $IO_3^- \& IO_3^-$

2. $I_2 \& I_2$

3. $IO_3^- \& I_2$

4. $I_2 \& IO_3^-$

Question Type : **MCQ**
Question ID : **3666942429**
Option 1 ID : **3666947665**
Option 2 ID : **3666947663**
Option 3 ID : **3666947666**
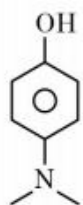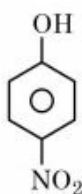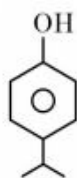Option 4 ID : **3666947664**
Status : **Answered**
Chosen Option : **2**

Q.41 Match List I with List II:

List I (Complexes)		List II (Hybridisation)	
A.	$[Ni(CO)_4]$	I.	sp^3
B.	$[Cu(NH_3)_4]^{2+}$	II.	dsp^2
C.	$[Fe(NH_3)_6]^{2+}$	III.	sp^3d^2
D.	$[Fe(H_2O)_6]^{2+}$	IV.	d^2sp^3

Options 1. A-II, B-I, C-III, D-IV

2. A-II, B-I, C-IV, D-III

3. A-I, B-II, C-III, D-IV

4. A-I, B-II, C-IV, D-III

Question Type : **MCQ**
Question ID : **3666942432**
Option 1 ID : **3666947677**
Option 2 ID : **3666947676**
Option 3 ID : **3666947675**
Option 4 ID : **3666947678**
Status : **Not Answered**
Chosen Option : **--**

Q.42 Formulae for Nessler's reagent is:

Options
1. HgI_2
2. $KHgI_3$
3. K_2HgI_4
4. KHg_2I_2

Question Type : **MCQ**
Question ID : **3666942433**
Option 1 ID : **3666947679**
Option 2 ID : **3666947682**
Option 3 ID : **3666947681**
Option 4 ID : **3666947680**
Status : **Not Answered**
Chosen Option : **--**

Q.43 The $Cl-Co-Cl$ bond angle values in a fac– $[Co(NH_3)_3Cl_3]$ complex is/are:

Options
1. $90°$
2. $180°$
3. $90° \& 180°$
4. $90° \& 120°$

Question Type : **MCQ**
Question ID : **3666942430**
Option 1 ID : **3666947667**
Option 2 ID : **3666947670**
Option 3 ID : **3666947668**
Option 4 ID : **3666947669**
Status : **Not Answered**
Chosen Option : **--**

Q.44 1 L, 0.02 M solution of $[Co(NH_3)_5SO_4]$ Br is mixed with 1 L, 0.02 M solution of $[Co(NH_3)_5Br]SO_4$. The resulting solution is divided into two equal parts (X) and treated with excess of $AgNO_3$ solution and $BaCl_2$ solution respectively as shown below:

1 L Solution (X) + $AgNO_3$ solution (excess) $\longrightarrow$ Y

1 L Solution (X) + $BaCl_2$ solution (excess) $\longrightarrow$ Z

The number of moles of Y and Z respectively are

Options
1. 0.02, 0.02

2. 0.02, 0.01

3. 0.01, 0.01

4. 0.01, 0.02

Question Type : **MCQ**
Question ID : **3666942421**
Option 1 ID : **3666947634**
Option 2 ID : **3666947632**
Option 3 ID : **3666947631**
Option 4 ID : **3666947633**
Status : **Not Answered**
Chosen Option : **--**

Q.45 Chlorides of which metal are soluble in organic solvents:

Options
1. Ca

2. Mg

3. K

4. Be

Question Type : **MCQ**
Question ID : **3666942426**
Option 1 ID : **3666947653**
Option 2 ID : **3666947654**
Option 3 ID : **3666947651**
Option 4 ID : **3666947652**
Status : **Answered**
Chosen Option : **3**

Q.46 Decreasing order towards SN 1 reaction for the following compounds is:

a   b   c   d

Options
1. $a > c > d > b$

2. $b > d > c > a$

3. $a > b > c > d$

4. $d > b > c > a$

Question Type : **MCQ**
Question ID : **3666942436**
Option 1 ID : **3666947692**
Option 2 ID : **3666947691**
Option 3 ID : **3666947693**
Option 4 ID : **3666947694**
Status : **Not Answered**
Chosen Option : **--**

Q.47 Bond dissociation energy of "E-H" bond of the "H_2E" hydrides of group 16 elements (given below), follows order.

A.   O

B.   S

C.   Se

D.   Te

Choose the correct from the options given below:

Options
1. $D > C > B > A$

2. $A > B > C > D$

3. $A > B > D > C$

4. $B > A > C > D$

Question Type : **MCQ**
Question ID : **3666942428**
Option 1 ID : **3666947659**
Option 2 ID : **3666947662**
Option 3 ID : **3666947661**
Option 4 ID : **3666947660**
Status : **Not Answered**
Chosen Option : **--**

Q.48 Maximum number of electrons that can be accommodated in shell with $n = 4$ are:

Options 1. 16

2. 50

3. 72

4. 32

Question Type : **MCQ**

Question ID : **3666942424**
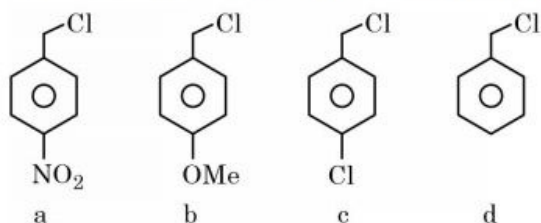
Option 1 ID : **3666947643**

Option 2 ID : **3666947645**

Option 3 ID : **3666947646**

Option 4 ID : **3666947644**

Status : **Not Attempted and Marked For Review**

Chosen Option : **--**

Q.49 The most stable carbocation for the following is:
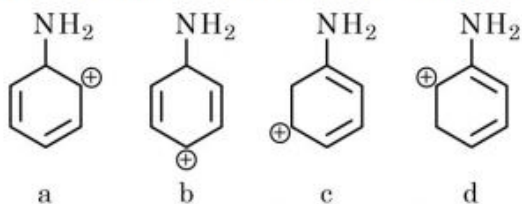
Options 1. b

2. a

3. c

4. d

Question Type : **MCQ**

Question ID : **3666942435**

Option 1 ID : **3666947688**

Option 2 ID : **3666947687**

Option 3 ID : **3666947689**

Option 4 ID : **3666947690**

Status : **Answered**

Chosen Option : **4**

Q.50 Boric acid is solid, whereas BF_3 is gas at room temperature because of

Options 1. Strong ionic bond in Boric acid

2. Strong hydrogen bond in Boric acid

3. Strong covalent bond in BF_3

4. Strong van der Waal's interaction in Boric acid

Question Type : **MCQ**
Question ID : **3666942423**
Option 1 ID : **3666947642**
Option 2 ID : **3666947640**
Option 3 ID : **3666947639**
Option 4 ID : **3666947641**
Status : **Answered**
Chosen Option : **2**

Section : **Chemistry Section B**

Q.51 1 mole of ideal gas is allowed to expand reversibly and adiabatically from a temperature of 27°C. The work done is 3 kJ mol^{-1}. The final temperature of the gas is _____ K (Nearest integer). Given $C_V = 20$ J mol^{-1} K^{-1}

Given --
Answer :

Question Type : **SA**
Question ID : **3666942442**
Status : **Not Attempted and Marked For Review**

Q.52 The electrode potential of the following half cell at 298 K

$X|X^{2+}(0.001\,M) \; || \; Y^{2+}(0.01\,M)|Y$ is _____ $\times 10^{-2}$ V (Nearest integer).

Given: $E^o_{X^{2+}|X} = -2.36$ V

$E^o_{Y^{2+}|Y} = +0.36$ V

$\dfrac{2.303RT}{F} = 0.06$ V

Given --
Answer :

Question Type : **SA**
Question ID : **3666942445**
Status : **Not Answered**

Q.53 The graph of $\log\dfrac{x}{m}$ vs log p for an adsorption process is a straight line inclined at an angle of 45° with intercept equal to 0.6020. The mass of gas adsorbed per unit mass of adsorbent at the pressure of 0.4 atm is _____ $\times 10^{-1}$ (Nearest integer).

Given: $\log 2 = 0.3010$

Given --
Answer :

Question Type : **SA**
Question ID : **3666942447**
Status : **Not Answered**

Q.54 Lead storage battery contains 38% by weight solution of H_2SO_4. The van't Hoff factor is 2.67 at this concentration. The temperature in Kelvin at which the solution in the battery will freeze is _____ (Nearest integer). Given $K_f = 1.8$ K kg mol^{-1}

Given --

Answer :

Question Type : **SA**

Question ID : **3666942443**

Status : **Not Answered**

Q.55 An organic compound undergoes first order decomposition. If the time taken for the 60% decomposition is 540 s, then the time required for 90% decomposition will be is _____ s. (Nearest integer).

Given: ln 10 = 2.3; log 2 = 0.3

Given --

Answer :

Question Type : **SA**

Question ID : **3666942446**

Status : **Not Attempted and Marked For Review**

Q.56 A short peptide on complete hydrolysis produces 3 moles of glycine (G), two moles of leucine (L) and two moles of valine (V) per mole of peptide. The number of peptide linkages in it are _____.

Given --

Answer :

Question Type : **SA**

Question ID : **3666942450**

Status : **Not Answered**

Q.57 Iron oxide FeO, crystallises in a cubic lattice with a unit cell edge length of 5.0 Å. If density of the FeO in the crystal is 4.0 g cm^{-3}, then the number of FeO units present per unit cell is _____. (Nearest integer)

Given: Molar mass of Fe and O is 56 and 16 g mol^{-1} respectively.
$N_A = 6.0 \times 10^{23}$ mol^{-1}

Given --

Answer :

Question Type : **SA**

Question ID : **3666942441**

Status : **Not Answered**

Q.58 The strength of 50 volume solution of hydrogen peroxide is _____ g/L (Nearest integer).
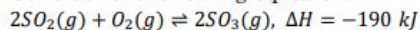
Given:

Molar mass of H_2O_2 is 34 g mol^{-1}

Molar volume of gas at STP = 22.7 L.

Given --

Answer :

Question Type : **SA**

Question ID : **3666942448**

Status : **Not Answered**

Q.59 Consider the following equation:

$$2SO_2(g) + O_2(g) \rightleftharpoons 2SO_3(g), \; \Delta H = -190 \; kJ$$

The number of factors which will increase the yield of SO_3 at equilibrium from the following is _____

A. Increasing temperature

B. Increasing pressure

C. Adding more SO_2

D. Adding more O_2

E. Addition of catalyst

Given --
Answer :

Question Type : **SA**

Question ID : **3666942444**

Status : **Not Answered**

Q.60 Number of compounds from the following which will not dissolve in cold $NaHCO_3$ and $NaOH$ solutions but will dissolve in hot $NaOH$ solution is

_____
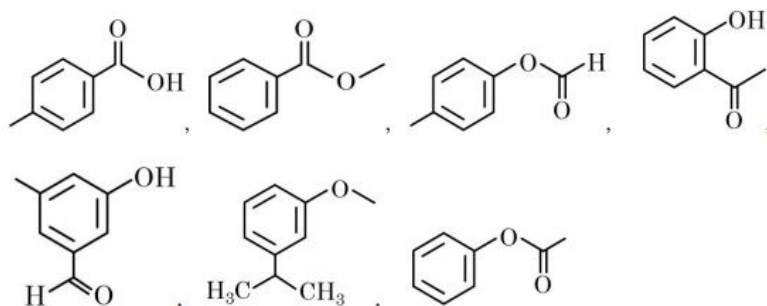
Given --
Answer :

Question Type : **SA**

Question ID : **3666942449**

Status : **Not Answered**

Section : **Mathematics Section A**

Q.61 The number of ways of selecting two numbers a and b, $a \in \{2, 4, 6, ..., 100\}$ and $b \in \{1, 3, 5, ..., 99\}$ such that 2 is the remainder when $a + b$ is divided by 23 is

Options 1. 268

2. 186

3. 108

4. 54

Question Type : **MCQ**

Question ID : **3666942455**

Option 1 ID : **3666947738**

Option 2 ID : **3666947737**

Option 3 ID : **3666947739**

Option 4 ID : **3666947740**

Status : **Not Answered**

Chosen Option : --

Q.62

The parabolas : $ax^2 + 2bx + cy = 0$ and $dx^2 + 2ex + fy = 0$ intersect on the line $y = 1$. If a, b, c, d, e, f are positive real numbers and a, b, c are in G.P., then

Options

1. d, e, f are in A.P.

2. $\dfrac{d}{a}, \dfrac{e}{b}, \dfrac{f}{c}$ are in G.P.

3. $\dfrac{d}{a}, \dfrac{e}{b}, \dfrac{f}{c}$ are in A.P.

4. d, e, f are in G.P.

Question Type : **MCQ**
Question ID : **3666942462**
Option 1 ID : **3666947767**
Option 2 ID : **3666947765**
Option 3 ID : **3666947766**
Option 4 ID : **3666947768**
Status : **Answered**
Chosen Option : **2**

Q.63

If P is a 3×3 real matrix such that $P^T = aP + (a-1)I$, where $a > 1$, then

Options

1. $|Adj\ P| = \dfrac{1}{2}$

2. $|Adj\ P| = 1$

3. $|Adj\ P| > 1$

4. P is a singular matrix

Question Type : **MCQ**
Question ID : **3666942454**
Option 1 ID : **3666947736**
Option 2 ID : **3666947735**
Option 3 ID : **3666947734**
Option 4 ID : **3666947733**
Status : **Answered**
Chosen Option : **2**

Q.64 The range of the function $f(x) = \sqrt{3-x} + \sqrt{2+x}$ is:

Options

1. $\left[\sqrt{2}, \sqrt{7}\right]$

2. $\left[\sqrt{5}, \sqrt{13}\right]$

3. $\left[2\sqrt{2}, \sqrt{11}\right]$

4. $\left[\sqrt{5}, \sqrt{10}\right]$

Question Type : **MCQ**
Question ID : **3666942451**
Option 1 ID : **3666947723**
Option 2 ID : **3666947724**
Option 3 ID : **3666947721**
Option 4 ID : **3666947722**
Status : **Answered**
Chosen Option : **2**

Q.65 Consider the following statements:

P : I have fever

Q : I will not take medicine

R : I will take rest.

The statement "If I have fever, then I will take medicine and I will take rest"
is equivalent to:

Options

1. $(P \vee Q) \wedge ((\sim P) \vee R)$

2. $(P \vee \sim Q) \wedge (P \vee \sim R)$

3. $((\sim P) \vee \sim Q) \wedge ((\sim P) \vee \sim R)$

4. $((\sim P) \vee \sim Q) \wedge ((\sim P) \vee R)$

Question Type : **MCQ**
Question ID : **3666942470**
Option 1 ID : **3666947797**
Option 2 ID : **3666947800**
Option 3 ID : **3666947799**
Option 4 ID : **3666947798**
Status : **Answered**
Chosen Option : **2**

Q.66

Let $a, b, c > 1$, a^3, b^3 and c^3 be in A.P., and $\log_a b$, $\log_c a$ and $\log_b c$ be in G.P. If the sum of first 20 terms of an A.P., whose first term is $\dfrac{a+4b+c}{3}$ and the common difference is $\dfrac{a-8b+c}{10}$ is -444, then abc is equal to:

Options

1. $\dfrac{125}{8}$

2. $\dfrac{343}{8}$

3. 343

4. 216

Question Type : **MCQ**
Question ID : **3666942457**
Option 1 ID : **3666947746**
Option 2 ID : **3666947748**
Option 3 ID : **3666947745**
Option 4 ID : **3666947747**
Status : **Answered**
Chosen Option : **2**

Q.67

Let $a_1 = 1, a_2, a_3, a_4, ...$ be consecutive natural numbers.

Then $\tan^{-1}\left(\dfrac{1}{1+a_1 a_2}\right) + \tan^{-1}\left(\dfrac{1}{1+a_2 a_3}\right) + ... + \tan^{-1}\left(\dfrac{1}{1+a_{2021} a_{2022}}\right)$ is equal to

Options

1. $\tan^{-1}(2022) - \dfrac{\pi}{4}$

2. $\cot^{-1}(2022) - \dfrac{\pi}{4}$

3. $\dfrac{\pi}{4} - \cot^{-1}(2022)$

4. $\dfrac{\pi}{4} - \tan^{-1}(2022)$

Question Type : **MCQ**
Question ID : **3666942468**
Option 1 ID : **3666947789**
Option 2 ID : **3666947791**
Option 3 ID : **3666947792**
Option 4 ID : **3666947790**
Status : **Answered**
Chosen Option : **2**

Q.68

The solution of the differential equation $\dfrac{dy}{dx} = -\left(\dfrac{x^2 + 3y^2}{3x^2 + y^2}\right)$, $y(1) = 0$ is

Options

1. $\log_e |x + y| - \dfrac{2xy}{(x+y)^2} = 0$

2. $\log_e |x + y| + \dfrac{xy}{(x+y)^2} = 0$

3. $\log_e |x + y| - \dfrac{xy}{(x+y)^2} = 0$

4. $\log_e |x + y| + \dfrac{2xy}{(x+y)^2} = 0$

Question Type : **MCQ**
Question ID : **3666942461**
Option 1 ID : **3666947762**
Option 2 ID : **3666947763**
Option 3 ID : **3666947761**
Option 4 ID : **3666947764**
Status : **Not Answered**
Chosen Option : **--**

Q.69

Let f, g and h be the real valued functions defined on $\mathbb{R}$ as

$$f(x) = \begin{cases} \dfrac{x}{|x|}, & x \neq 0 \\ 1, & x = 0 \end{cases}, \quad g(x) = \begin{cases} \dfrac{\sin(x+1)}{(x+1)}, & x \neq -1 \\ 1, & x = -1 \end{cases}$$

and $h(x) = 2[x] - f(x)$, where $[x]$ is the greatest integer $\leq x$.

Then the value of $\lim\limits_{x \to 1} g(h(x-1))$ is

Options

1. -1

2. $\sin(1)$

3. 1

4. 0

Question Type : **MCQ**
Question ID : **3666942458**
Option 1 ID : **3666947751**
Option 2 ID : **3666947750**
Option 3 ID : **3666947749**
Option 4 ID : **3666947752**
Status : **Not Answered**
Chosen Option : **--**

Q.70

For $\alpha, \beta \in \mathbb{R}$, suppose the system of linear equations

$x - y + z = 5$

$2x + 2y + \alpha z = 8$

$3x - y + 4z = \beta$

has infinitely many solutions. Then α and β are the roots of

Options

1. $x^2 - 18x + 56 = 0$

2. $x^2 - 10x + 16 = 0$

3. $x^2 + 18x + 56 = 0$

4. $x^2 + 14x + 24 = 0$

Question Type : **MCQ**
Question ID : **3666942453**
Option 1 ID : **3666947730**
Option 2 ID : **3666947731**
Option 3 ID : **3666947729**
Option 4 ID : **3666947732**
Status : **Not Answered**
Chosen Option : **--**

Q.71

If a plane passes through the points $(-1, k, 0), (2, k, -1), (1, 1, 2)$ and is parallel to the line $\dfrac{x-1}{1} = \dfrac{2y+1}{2} = \dfrac{z+1}{-1}$, then the value of $\dfrac{k^2+1}{(k-1)(k-2)}$ is

Options

1. $\dfrac{13}{6}$

2. $\dfrac{5}{17}$

3. $\dfrac{6}{13}$

4. $\dfrac{17}{5}$

Question Type : **MCQ**
Question ID : **3666942465**
Option 1 ID : **3666947777**
Option 2 ID : **3666947778**
Option 3 ID : **3666947779**
Option 4 ID : **3666947780**
Status : **Not Answered**
Chosen Option : **--**

Q.72

If the functions $f(x) = \dfrac{x^3}{3} + 2bx + \dfrac{ax^2}{2}$ and $g(x) = \dfrac{x^3}{3} + ax + bx^2$, $a \neq 2b$ have a common extreme point, then $a + 2b + 7$ is equal to:

Options

1. $\dfrac{3}{2}$

2. 3

3. 6

4. 4

Question Type : **MCQ**
Question ID : **3666942459**
Option 1 ID : **3666947753**
Option 2 ID : **3666947754**
Option 3 ID : **3666947756**
Option 4 ID : **3666947755**
Status : **Not Answered**
Chosen Option : **--**

Q.73

A vector $\vec{v}$ in the first octant is inclined to the x-axis at $60°$, to the y-axis at $45°$ and to the z-axis at an acute angle. If a plane passing through the points $\left(\sqrt{2}, -1, 1\right)$ and (a, b, c), is normal to $\vec{v}$, then

Options

1. $\sqrt{2}a + b + c = 1$

2. $a + \sqrt{2}b + c = 1$

3. $\sqrt{2}a - b + c = 1$

4. $a + b + \sqrt{2}c = 1$

Question Type : **MCQ**
Question ID : **3666942464**
Option 1 ID : **3666947773**
Option 2 ID : **3666947775**
Option 3 ID : **3666947774**
Option 4 ID : **3666947776**
Status : **Not Answered**
Chosen Option : **--**

Q.74

Let $\vec{a}$ and $\vec{b}$ be two vectors, Let $|\vec{a}|=1$, $|\vec{b}|=4$ and $\vec{a}\cdot\vec{b}=2$. If $\vec{c}=(2\vec{a}\times\vec{b})-3\vec{b}$, then the value of $\vec{b}\cdot\vec{c}$ is

Options

1. -60

2. -48

3. -24

4. -84

Question Type : **MCQ**

Question ID : **3666942466**

Option 1 ID : **3666947783**

Option 2 ID : **3666947782**

Option 3 ID : **3666947781**

Option 4 ID : **3666947784**

Status : **Not Answered**

Chosen Option : **--**

Q.75

Let S be the set of all values of a_1 for which the mean deviation about the mean of 100 consecutive positive integers $a_1, a_2, a_3,...., a_{100}$ is 25. Then S is

Options

1. ϕ

2. $\{99\}$

3. $\{9\}$

4. $\mathbb{N}$

Question Type : **MCQ**

Question ID : **3666942467**

Option 1 ID : **3666947788**

Option 2 ID : **3666947786**

Option 3 ID : **3666947785**

Option 4 ID : **3666947787**

Status : **Not Answered**

Chosen Option : **--**

Q.76

Let $\lambda \in \mathbb{R}$, $\vec{a} = \lambda\hat{i} + 2\hat{j} - 3\hat{k}$, $\vec{b} = \hat{i} - \lambda\hat{j} + 2\hat{k}$.

If $\left((\vec{a}+\vec{b}) \times (\vec{a} \times \vec{b})\right) \times (\vec{a}-\vec{b}) = 8\hat{i} - 40\hat{j} - 24\hat{k}$, then $\left|\lambda(\vec{a}+\vec{b}) \times (\vec{a}-\vec{b})\right|^2$ is equal to

Options

1. 140

2. 136

3. 144

4. 132

Question Type : **MCQ**

Question ID : **3666942469**

Option 1 ID : **3666947795**

Option 2 ID : **3666947794**

Option 3 ID : **3666947796**

Option 4 ID : **3666947793**

Status : **Not Answered**

Chosen Option : **--**

Q.77

Let $x = \left(8\sqrt{3} + 13\right)^{13}$ and $y = \left(7\sqrt{2} + 9\right)^{9}$. If $[t]$ denotes the greatest integer $\leq t$,

then

Options

1. $[x] + [y]$ is even

2. $[x]$ is even but $[y]$ is odd

3. $[x]$ and $[y]$ are both odd

4. $[x]$ is odd but $[y]$ is even

Question Type : **MCQ**

Question ID : **3666942456**

Option 1 ID : **3666947744**

Option 2 ID : **3666947742**

Option 3 ID : **3666947741**

Option 4 ID : **3666947743**

Status : **Not Answered**

Chosen Option : **--**

Q.78   Let q be the maximum integral value of p in $[0, 10]$ for which the roots of the equation $x^2 - px + \frac{5}{4}p = 0$ are rational. Then the area of the region

$$\{(x, y) : 0 \le y \le (x - q)^2, 0 \le x \le q\} \text{ is}$$

Options
1. 164

2. 25

3. 243

4. $\dfrac{125}{3}$

Question Type : **MCQ**
Question ID : **3666942452**
Option 1 ID : **3666947727**
Option 2 ID : **3666947726**
Option 3 ID : **3666947728**
Option 4 ID : **3666947725**
Status : **Not Answered**
Chosen Option : **--**

Q.79

$$\lim_{n \to \infty} \frac{3}{n} \left\{ 4 + \left(2 + \frac{1}{n}\right)^2 + \left(2 + \frac{2}{n}\right)^2 + \dots + \left(3 - \frac{1}{n}\right)^2 \right\} \text{ is equal to}$$

Options
1. 12

2. 0

3. 19

4. $\dfrac{19}{3}$

Question Type : **MCQ**
Question ID : **3666942460**
Option 1 ID : **3666947760**
Option 2 ID : **3666947759**
Option 3 ID : **3666947757**
Option 4 ID : **3666947758**
Status : **Not Answered**
Chosen Option : **--**

Q.80

Let A be a point on the x-axis. Common tangents are drawn from A to the curves $x^2 + y^2 = 8$ and $y^2 = 16x$. If one of these tangents touches the two curves at Q and R, then $(QR)^2$ is equal to

Options

1. 64

2. 76

3. 72

4. 81

Question Type : **MCQ**
Question ID : **3666942463**
Option 1 ID : **3666947769**
Option 2 ID : **3666947771**
Option 3 ID : **3666947770**
Option 4 ID : **3666947772**
Status : **Not Answered**
Chosen Option : **--**

Section : **Mathematics Section B**

Q.81

A bag contains six balls of different colours. Two balls are drawn in succession with replacement. The probability that both the balls are of the same colour is p. Next four balls are drawn in succession with replacement and the probability that exactly three balls are of the same colour is q. If $p : q = m : n$, where m and n are coprime, then $m + n$ is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **3666942480**
Status : **Not Answered**

Q.82

If the value of real number $a > 0$ for which $x^2 - 5ax + 1 = 0$ and $x^2 - ax - 5 = 0$ have a common real root is $\dfrac{3}{\sqrt{2\beta}}$ then β is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **3666942472**
Status : **Not Answered**

Q.83

Let $P(a_1, b_1)$ and $Q(a_2, b_2)$ be two distinct points on a circle with center $C(\sqrt{2}, \sqrt{3})$. Let O be the origin and OC be perpendicular to both CP and CQ. If the area of the triangle OCP is $\dfrac{\sqrt{35}}{2}$, then $a_1^2 + a_2^2 + b_1^2 + b_2^2$ is equal to _____.

Given --
Answer :

Question Type : **SA**
Question ID : **3666942478**
Status : **Not Answered**

Q.84

Let a line L pass through the point $P(2, 3, 1)$ and be parallel to the line $x + 3y - 2z - 2 = 0 = x - y + 2z$. If the distance of L from the point $(5, 3, 8)$ is α, then $3\alpha^2$ is equal to _____.

Given --
Answer :

Question Type : **SA**

Question ID : **3666942479**

Status : **Not Answered**

Q.85

The number of seven digits odd numbers, that can be formed using all the seven digits 1, 2, 2, 2, 3, 3, 5 is _____.

Given --
Answer :

Question Type : **SA**

Question ID : **3666942473**

Status : **Not Answered**

Q.86

Let $A = \{1, 2, 3, 5, 8, 9\}$. Then the number of possible functions $f : A \to A$ such that $f(m \cdot n) = f(m) \cdot f(n)$ for every $m, n \in A$ with $m \cdot n \in A$ is equal to

_____.

Given --
Answer :

Question Type : **SA**

Question ID : **3666942471**

Status : **Not Answered**

Q.87

Let A be the area of the region $\{(x, y): y \geq x^2, y \geq (1-x)^2, y \leq 2x(1-x)\}$. Then $540\,A$ is equal to _____.

Given --
Answer :

Question Type : **SA**

Question ID : **3666942476**

Status : **Not Answered**

Q.88

50th root of a number x is 12 and 50th root of another number y is 18. Then the remainder obtained on dividing $(x + y)$ by 25 is _____.

Given --
Answer :

Question Type : **SA**

Question ID : **3666942474**

Status : **Not Answered**

Q.89

The 8th common term of the series

$$S_1 = 3+7+11+15+19+.....,$$

$$S_2 = 1+6+11+16+21+.....$$

is _____.

Given --

Answer :

Question Type : **SA**

Question ID : **3666942475**

Status : **Not Answered**

Q.90

If $\int \sqrt{\sec 2x - 1}\, dx = \alpha \log_e \left| \cos 2x + \beta + \sqrt{\cos 2x \left(1 + \cos \dfrac{1}{\beta} x\right)} \right| +$ constant, then

$\beta - \alpha$ is equal to _____.

Given --

Answer :

Question Type : **SA**

Question ID : **3666942477**

Status : **Not Answered**