Chapter 2

SELF-ASSESSMENT

Select the letter of the best answer below.

- **1.** K/U Which reaction would be performed to convert an alkyne to alcohol?
 - a. substitution
 - **b.** elimination
 - c. esterification
 - d. addition
 - **e.** hydrolysis
- **2.** K/U Which reaction is used in industry to convert alkanes to alkenes?
 - a. esterification
 - **b.** polymerization
 - c. cracking
 - **d.** breaking
 - e. elimination
- **3.** K/U Which pair of reactions are the reverse of one another?
 - a. addition and substitution
 - **b.** hydrolysis and condensation
 - c. elimination and substitution
 - **d.** esterification and condensation
 - e. oxidation and addition
- **4. K/U** Identify the name of the major product for an addition reaction with the following reactants.

- a. 2-methylbut-2-ene
- **b.** 2-chloro-3-methylbutane
- c. 3-chloro-2-methylbutane
- **d.** 2-chloro-2-methylbutane
- e. 2-chloro-3-methylbutene
- **5.** K/U Which statement about oxidation reactions is
 - **a.** Oxidation reactions can change an aldehyde to a carboxylic acid.
 - **b.** Oxidation reactions can change a ketone to an alcohol.
 - **c.** Oxidation reactions can change an alkene to an alkane.
 - d. Oxidation reactions can change an alkyne to an
 - e. Oxidation reactions can change an aldehyde to an alcohol.

- **6.** K/U Which reaction would be used to convert an alkyl halide to an alcohol?
 - a. reduction
 - **b.** addition
 - c. condensation
 - **d.** elimination
 - e. substitution
- **7. K**/**U** Which is *not* a solution to petroleum-based plastics' inability to degrade easily?
 - **a.** Reduce the use of plastics.
 - **b.** Burn plastics when they are discarded.
 - **c.** Re-use plastics as often as possible.
 - **d.** Recycle plastics as often as possible.
 - e. Replace petroleum-based plastics with degradable plastics.
- **8.** K/U Which are the monomers of proteins?
 - a. glucose
 - **b.** alkanes
 - c. nucleotides
 - d. amino acids
 - e. sucrose
- **9.** K/U Which two functional groups react together in a condensation reaction to form an amide and water?
 - a. an amine and an alcohol
 - **b.** a carboxylic acid and an amine
 - c. a carboxylic acid and an amide
 - d. a carboxylic acid and an alcohol
 - e. an ester and an alcohol
- **10.** K/U Crosslinking is seen in which of the following polymers?
 - a. polystyrene
 - b. PVC
 - c. polyacrylamide
 - d. PET
 - e. Kevlar®

Use sentences and diagrams as appropriate to answer the questions below.

- **11.** A What monomer would you expect to be an active ingredient in muscle-building supplements? Explain your answer.
- **12.** T/I Determine the products of the following condensation and esterification reactions.
 - **a.** 2,2-dimethylbutanoic acid + 2-methylpropan-2-ol $\xrightarrow{\text{H}_2SO_4}$
 - **b.** 3-chlorobutanoic acid + propan-1-amine $\xrightarrow{\text{H}_2\text{SO}_4}$

- **13.** A Propose a synthesis, including any necessary conditions, for cycloheptanone from an alkene.
- **14.** T/I Determine the monomers for the following polymers and identify the polymers as addition or condensation polymers.

- **15.** K/U What are the similarities and differences between cellulose and starch?
- **16.** Write a laboratory procedure (main reactions only) to perform an addition polymerization if your starting material is a carboxylic acid.

- **17.** Determine the products of the following addition and elimination reactions. Identify the major product in each case.
 - **a.** hept-2-ene + HCl \rightarrow
 - **b.** 2-methylheptan-4-ol $\xrightarrow{\text{H}_2\text{SO}_4}$
- **18.** Write a short essay explaining how plastics contribute to your life. Include your opinion on whether the benefits outweigh the consequences of their use.
- **19.** T/I Determine the products of the following reaction, then balance the equation.

$$(CH_3)_2CHCH = CHCH_3 + O_2 \rightarrow$$

- **20.** Is a protein an example of a polyester or a polyamide? Explain your answer.
- **21. K/U** Explain what happens during reduction of an organic molecule.
- **22.** T/I Determine the products for the following reactions.
 - **a.** $CH_3CH_2C\equiv CH+Cl_2 \rightarrow ?+Cl_2 \rightarrow ?$ **b.** cyclopentanol $\xrightarrow{H_2SO_4}$

 - c. Cl— $CH_2CH_2CH_3 + OH^- \rightarrow$
- **23.** A Propose a synthesis, including any necessary conditions, for a four-carbon alkyl halide from butanone.
- **24.** T/I Determine the reactants in the following reactions.
 - **a.** $? + ? \rightarrow \text{propanoic acid} + \text{hexanol}$
 - **b.** $\xrightarrow{\text{H}_2\text{SO}_4}$ cyclobutene + water
- **25.** Propose a one-step synthesis of the following compound from a non-cyclic reactant.

Self-Check

If you missed question	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
Review section(s)	2.1	2.2	2.1	2.1	2.1	2.1	2.2	2.2	2.1	2.2	2.2	2.1	2.1	2.2	2.2	2.2	2.1	2.2	2.1	2.2	2.1	2.1	2.1	2.1	2.1