

9th International Conference Interdisciplinarity in Engineering, INTER-ENG 2015, 8-9 October 2015, Tirgu-Mures, Romania

## The Heat Treatment Influence of 1.3343 High Speed Steel on Content of Residual Austenite

Maria Stoicanescu<sup>a,\*</sup>, Emilian Ene<sup>a</sup>, Adriana Zara<sup>a</sup>, Ioan Giacomelli<sup>a</sup>, Aurel Crisan<sup>a</sup>

<sup>a</sup>*Department of Materials Science, Transylvania University, Brasov, Romania*

---

### Abstract

The heat treatment applied to high speed steel is in large measure dictated by the necessity of decreasing the residual austenite content. After hardening, a large amount of residual austenite it is found in the structure; this is unwanted because it is less hard than practical necessities and also unstable in time. As a result, the heat treatment has the purpose to conferee to tools mechanical properties that are required in exploitation, and in the same time, stable from structural point of view and of mechanical characteristics. In this purpose there are known several processes, as for example multiple tempering and treatment below zero degrees. In this paper, it is proposed the analysis of residual austenite from 1.3343 high steel speeds subject to multiple heat treatment operations, by determining after each, the quantity of residual austenite; this was made with X ray diffraction using a software package program. Research has revealed the decrease of residual austenite content with the number of tempering as well as with the treatment below zero degrees.

© 2016 The Authors. Published by Elsevier Ltd. This is an open access article under the CC BY-NC-ND license (<http://creativecommons.org/licenses/by-nc-nd/4.0/>).

Peer-review under responsibility of the “Petru Maior” University of Tirgu Mures, Faculty of Engineering

**Keywords:** heat treatment; multiple temperings; residual austenite.

---

### 1. Introduction

The stability in time of the properties and of the dimensions on high speed steel tools represents a condition at heat treatment outcome [1-11]. In the same time due to chemical composition, the martensitic critical points at these high speed steels are at low temperatures, Ms, being below zero degrees [12-18]. Thus after hardening, in the

---

\* Corresponding author.

E-mail address: [stoican.m@unitbv.ro](mailto:stoican.m@unitbv.ro)

structure, residual austenite it is found in appreciable proportion, at several dozens of percentage. The low hardness and the reduced stability can affect the tools behavior in exploitation. The heat treatment subsequent hardening has as principal purpose, the conferment of the mechanical and dimensional characteristics to these products as they are required [19-23]. Heat treatments for these are multiple tempering and treatment below zero degrees.

The influence of thermal treatment of rapid steels regarding the content of residual austenite and its determination in various stages of heat treatment represents an important topic in the studies and research carried out [24-28].

## 2. Experimental research

In terms of determination of residual austenite in various stages of heat treatment it has been chosen 1.3343 (Rp5) high speed steel with the following chemical composition shown in Table 1.

Table 1. The chemical composition

Steel type	Chemical composition in percent								
	C	Si	Mn	P	S	Cr	Mo	V	W
1.3343 (Rp5)	0,92	0,28	0,39	0,021	0,0011	3,43	4,71	1,71	5,95

In order to achieve results, high speed steel samples were subjected to hardening using equipment for vacuum heat treatment; the cooling was performed using an installation with purified nitrogen, strongly recycled. Working parameters at hardening are indicated in Table 2.

Table 2 Hardening parameters

Steel type	Preheating			Final heating			Cooling	Hardness [HRC]
	Press. [torr]	Temp. [°C]	Time [min]	Press. [torr]	Temp. [°C]	Time [min]		
1.3343	10 <sup>-1</sup>	870	30	0,2...0,3	1190	7	purified nitrogen	60,5

Following the process the hardening samples were subjected to heat treatment operations destined to reduce the amount of residual austenite. Thus, after consecutive hardenings, there were applied tempering operations, according to Table 3.

Table 3 Tempering parameters

Steel type	Tempering working parameters			
	Press. [torr]	Temp. [°C]	Time [min]	Cooling
1.3343	400	560	60	purified nitrogen

There has been effectuated multiple tempering, from 1 to 6, for several samples in parallel. A set of samples were also after hardening, subject to heat treatment below zero degrees, respectively at -40° C. In the following figures, the microstructures of samples treated as above, are presented.

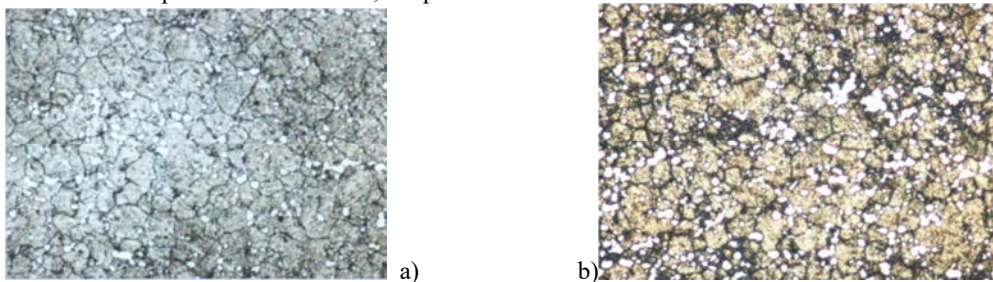


Fig. 1. a) High speed steel 1.3343 vacuum hardening from 1190° C. NITAL Attack; 1000:1; b) High speed steel 1.3343 vacuum hardening from 1190° C and 1 tempering. NITAL Attack: 1000:1.

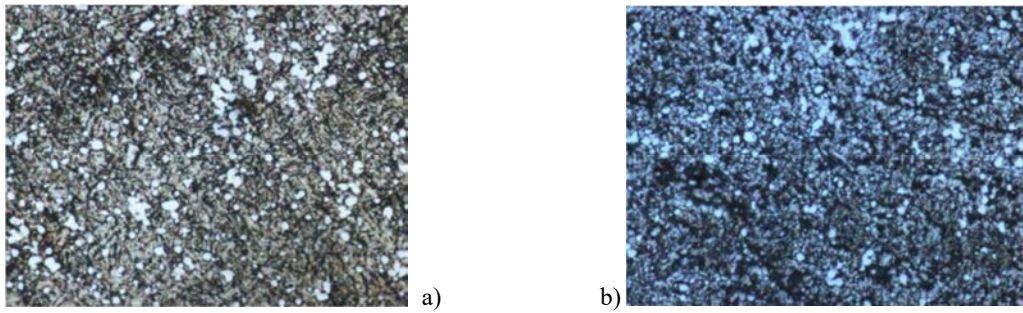


Fig. 2. a) High speed steel 1.3343 vacuum hardening from 1190° C and 3 consecutive temperings. NITAL Attack: 1000 :1; b) High speed steel 1.3343 vacuum hardening at 1190° C and 6 consecutive temperings. NITAL Attack: 1000:1.

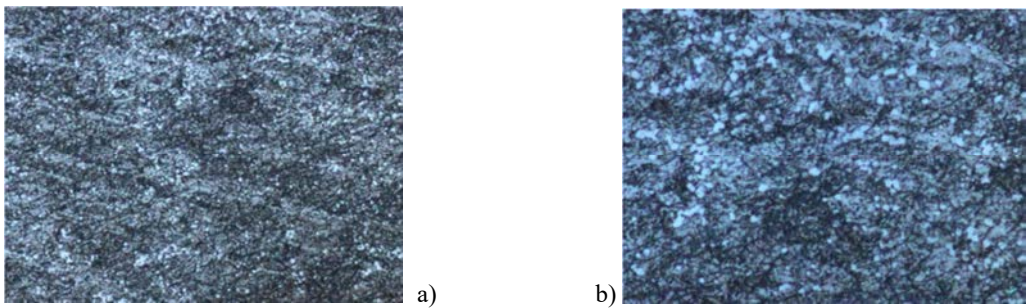


Fig. 3. High speed steel 1.3343 vacuum hardening from 1190° C and cooling at -40 C, 1h. Nital Attack: (a) 500:1; (b) 1000:1.

### 3. Determining the content of residual austenite

The high speed steel samples 1.3343 treated in many variations, as described above, were carried out by X-ray diffraction analysis using a PANalytical diffractometer XPert MPD. The diffraction spectra are given below.

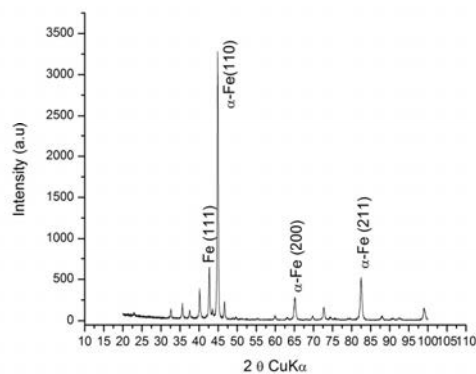


Fig. 4. X-ray diffraction spectra of 1.3343 high speed steel sample, after vacuum hardening

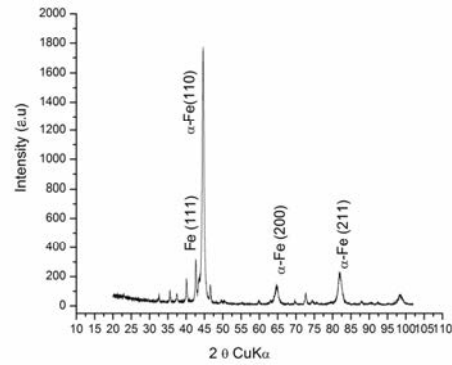


Fig. 5. X-ray diffraction spectra of 1.3343 high speed steel samples, vacuum hardening and 1 tempering at  $560^\circ\text{C}$ .

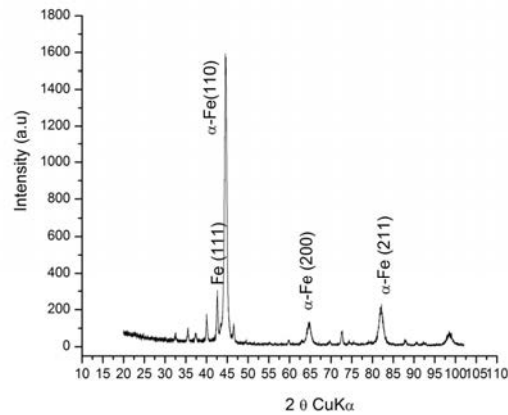


Fig. 6. X-ray diffraction spectra of 1.3343 high speed steel sample, vacuum hardening and 2 consecutive temperings at  $560^\circ\text{C}$ .

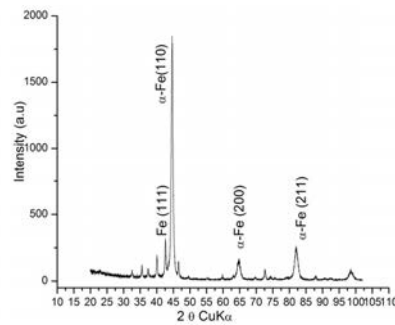


Fig. 7. X-ray diffraction spectra of 1.3343 high speed steel sample, vacuum hardening and 3 consecutive temperings at  $560^\circ\text{C}$ .

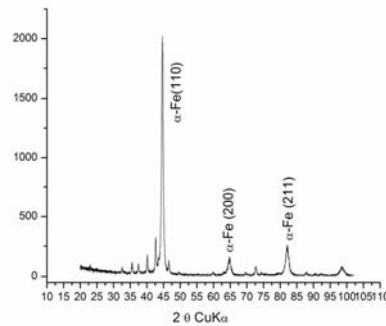


Fig. 8. X-ray diffraction spectra of 1.3343 high speed steel sample, vacuum hardening and 6 consecutive temperings at 560° C

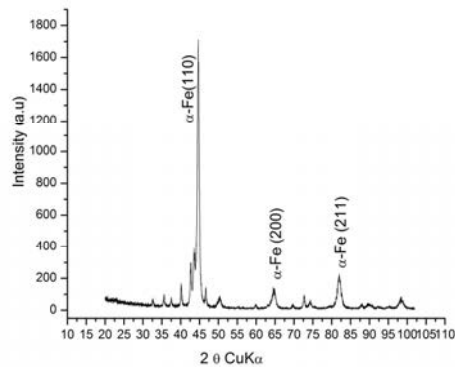


Fig. 9. X-ray diffraction spectra of 1.3343 high speed steel sample, vacuum hardening and cooling at -40°C for 1h

Considering that, to austenite with crystalline lattice cfc is corresponding the peak of the plane (111) from the diffractograms above, from qualitatively point of view, there is found a decrease of residual austenite content with the increasing of the number of temperings, respectively with treatment below zero degrees. For quantitative determinations were carried out spectra analysis using the software program package XpertHighScorePlus. The identification of crystalline phases was made on standard base XRD, JCPDS of International Centre of Diffraction, SUA. The investigation results regarding the hardness and the content of residual austenite are shown in table 4, where: H-hardness, and T-tempering.

Table 4. The hardness and the content of residual austenite in different stages

Properties	The initial state	After hardening (H)	H+1T	H+2 T	H+3 T	H+6 T	H+ treat. below 0°C
Hardness [HRC]	42,5	60,5	63,40	63,87	64,13	64,20	64,83
Res.austenite [%]	-	43,8	7,8	7,4	7,1	7,0	7,1

#### 4. Conclusions

The results obtained based on the experimental attempts allow the following conclusions: the high content of residual austenite found after hardening, confirms data from the specialized literature; the tempering applied shortly after hardening, has a sensitive effect upon the residual austenite transformation; the major quantity of transformed structure occurs after the first tempering; along in the process, on subsequent temperings, gradually the percentage of transformed residual austenite is decreasing; the residual austenite transformation is accompanied by the hardness increasing, more visible after the first tempering and relatively insignificant following the process; the heat treatment below zero degrees causes in a single stage the drastically decreasing of residual austenite content, with visible effects on hardness. A subsequent tempering, with the stress relieving role, would lead to a slight decrease in

hardness. In the same time, it is appreciated that through profound coolings the steel resilience decreases. As a result, the treatment below zero degrees is applied to the products at which the shocks in exploitation are trifling.

## References

- [1] Reed-Hill R, Abbaschian R. Physical metallurgy principles. 3 rd Edition. Boston: PWS-Kent Publishing.(1991).
- [2] Christian, J.W. The theory of transformations in metals and alloys. Pergamon, p. 546. (2002)
- [3] Balteş L, Țierean M, Olah A. Ways to decrease the residual austenite content through tempering in magnetic field, for speed steels cutting tools. 7 th International research/ expert conference – TMT (2003), Barcelona, Spain, pag. 205-208, ISBN 9958-617-18-8.
- [4] Kruglov EP, Tabolenko PP. Vacuum heat treatment of high speed and resistant martensitic steels. Metal Science and Heat Treatment. Vol.44, Nov.1-2, (2002).
- [5] Dumitrescu C., Saban R. Physical metallurgy and heat treatment. Ed.Fair Partners, (2001).
- [6] Lusk MT, Wang W, Sun X, Lee YK. On the role of kinematics in constructing predictive models of austenite decomposition, austenite formation and decomposition. (Eds. E.B. Damm and M.J. Merwin), Minerals, Metals and Materials Society, Warrendale, PA (2003).
- [7] Popescu VJ, Chiriac C. Rapid Steel. Ed. AGIR Bucharest, (2002).
- [8] Gill SS, Singh J, et al. Metallurgical principles of cryogenically treated tool steels—a review on the current state of science. International Journal of Advanced Manufacturing Technology 2010; 54(1-4): 59-82.
- [9] Torodoc N, Giacomelli I. Contributions of heat and thermochemical treatments to the improvement of the performances of high speed steel tools. Advanced Technologies and Materials, p. 234 - 236, Vol. I, ISSN 1843-5807.
- [10] Tisza M. Physical metallurgy for engineers . ASM International 2002 Page 348-350.
- [11] Druga L, Ghelec E, Manea V. How to increase the efficiency of thermochemical treatments. International Conference on Materials Science & Engineering BRAMAT 2003, p. 30 – 33 Brasov. (2003).
- [12] Krauss G. Principles of heat treatment of steel. ASM, Materials Park, OH, (1980), p. 52.
- [13] Bhadeshia H. K. D. H. TRIP-assisted steels. ISIJ International, 42 (2002), p.1059-1060.
- [14] Reisner G, Werner EA, Fischer FD. Micromechanical modelling of martensitic transformation in random microstructures. Int. Journal of Solids & Structures 1998; 35(19):2457-2473.
- [15] Mitelea I, Lugscheider E, Tillmann W. Materials science in mechanical engineering. Publisher Welding, Timișoara (1999).
- [16] Olson GB, Owen WS. Martensite. ASM International (1992).
- [17] Dobrzanski LA, Trzaska J. Application of neural networks to forecasting the CCT diagrams. Journal of Materials Processing Technology 2004; 157-158:107-113.
- [18] Bhadeshia HKDH, Edmonds DV. Tempered Martensite Embrittlement: Role of Retained Austenite and Cementite. Metal Science 1979; 13:325-334.
- [19] Tisza M. Physical metallurgy for engineers. ASM International (2002) p. 232.
- [20] Dossett JL, Boyer HE. Practical heat treating. ASM International (2006) p. 112.
- [21] Zhong N, Wang XD, Huang BX, Rong YH, Wang L. Microstructures and mechanical property of quenched and partitioned Fe-C-Mn-Si steel. The 3rd International Conference on Advanced Structural Steels (ICASS 2006), H. C. Lee (Gyeongju, Korea, Korean Institute of Metals and Materials and POSCO, (2006), p. 885-891.
- [22] Lambers HG, Tschumak S, Maier HJ, Canadine D. Role of austenitization and pre-deformation on the kinetics of the isothermal bainitic transformation. Metal Mater Trans A 40 (6) (2009). p. 1355.
- [23] Zhou H, Li Y, Qi J. An Investigation of the Auto Tempering of Low Carbon Martensite. Chinese Journal of Mechanical Engineering, 20 (1984), p.1-12.
- [24] Samuels LE. Light Microscopy of Carbon Steels. ASM International, Materials Park, OH, (1999), p. 273.
- [25] Ershov VM, Nekrasova LS. Transformation of cementite into austenite. Metal Sci Heat Treat. **24** (1) p. 9–11.
- [26] Krauss G. Martensitic transformation, structure and properties in hardenable steels, in Hardenability Concepts with Applications to Steel, (Jan 1982).
- [27] Campbell FC. Elements of metallurgy and engineering alloys. ASM International (2008) p. 195-196.
- [28] Speich GR, Leslie WC. Met. Trans., Vol. 3, (1972), p. 1,043.