	Suk	BE - SE oject Co	SUMMER- 2018 Date: 18-05-2018 Total Marks: 70				
1	Subject Name: Chemistry Time: 2:30 PM to 05:00 PM Instructions:						
1	. •	Questio	on No. 1 is compulsory. Attempt a	ny fou	r o	ut of remaining Six	questions.
2	2. Make suitable assumptions wherever no		•		C	•	
3	i.	Figures	s to the right indicate full marks.				
_							Marks
Q.1. (a)							7
1	•	Ultravio	olet rays are used in water treatment	for			
		a)	Illumination	b)		Disinfection	
		c)	Coagulation	d)		Sedimentation	
2).	High gr	ade steel is manufactured using				
		a)	Open hearth furnace	b)		Bessemer converter	
		c)	Electric furnace	d)		Blast furnace	
3	.	White P	Portland cement does not contain				
		a)	${ m SiO}_2$	b)		Al_2O_3	
		c)	Fe_2O_3	d)		MgO	
4		A good	fuel should possess				
		a)	Very high ignition temperature		b)	Very low igniti	on temperature
		c)	Moderate ignition temperature		d)	•	-
5	5.	Which o	of the following polymers are hard				
		a)	Linear	b)		Branched	
		c)	Cross-linked	d)		Thermoplastic	
6).	Porosity	y in the refractory brick generally de	ecrease	S		
		a)	Strength	b)		Resistance to abreac	ction
		c)	Resistance to spalling	d)		All of the above	
7	' .	Enzyme	e Rennin is used in	,			
		a)	Brewing industry	b)		Dairy industry	
		c)	Bio-fuel	d)		Paper industry	
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Q.1. ((b)		/
		Low ionization energy and high electron affinity will favor the formation ofbond.	
	2.	Dissolved CO ₂ may be removed from water by addition	
	3.	The presence of the nitrogen in the organic compound can be detected by to	est.
	4.	Polymerization between styrene and butadiene would lead to the formation of a	
	5.	The acid is responsible for the ripening of the guava fruit.	
	6.	is used on tips of drilling and cutting tools.	
	7.	Chromatography is the technique of chemical compound.	
Q.2.			
	a)	Write about the 12 principles of the green chemistry.	3
	b)	Write a shot note on the co-ordinate bond.	4
	c)	What is the principle of EDTA titration? How the permanent hardness of water is determined using EDTA method?	ined 7
Q.3.			
	a)	What are the three methods of preparation of alloys?	3
	b)	Describe the process for the production of steel from cast iron.	4
	c)	Define corrosion. Explain the protection of iron metal though cathodic and anodic protect	ion.
			7
Q.4.			
	a)	How tricalcium silicates react with water? Write the chemical equations.	3
	b)	What is the vulcanization of rubber? How does it improve the properties of the natural rubb	ber?
	c)	Explain the manufacturing process of lime. Write its properties and significance of lime.	7
Q.5.			
	a)	What are the abrasives? Give important uses of abrasives.	3
	b)	What are the thermal insulators? Name three organic and inorganic thermal insulators.	4
	c)	How is ethanol biosynthesized from molasses, starch and cellulosic materials?	7

Q.6.						
	a)	Calculate the mass of air needed for complete combustion of $5.0 \ kg$ of coal containing 80%				
		carbon, 15% hydrogen and the rest oxygen.	3			
	b)	Write the short note on annealing and hardening.	4			
	c)	What is the composition of Portland cement? Discuss the manufacturing process of Portland cement.	nent.			
Q.7.			,			
	a)	What is meant by the carbonization? How is carried out in a bee-hive oven?	3			
	b)	State two disadvantages of using hard water for domestic and industrial purpose.	4			
	c)	Discuss the various methods of controlling corrosion. Explain any one.	7			