

Unravelling the Impact of SARS-CoV-2 on Hemostatic and Complement Systems: A Systems Biology Perspective

Didar Murad¹, Rehan Zafar Paracha^{1*}, Iftikhar Ahmad^{1,2}, Maryum Nisar¹

1 School of Interdisciplinary Engineering and Sciences(SINES), Department of Sciences, National University of Sciences and Technology(NUST), Islamabad, Pakistan

2 School of Electrical Engineering and Computer Science (SEECS), Institute of National University of Sciences and Technology (NUST), Islamabad, Pakistan

* corresponding authors:

rehan@sines.nust.edu.pk

Dummy mathematical model

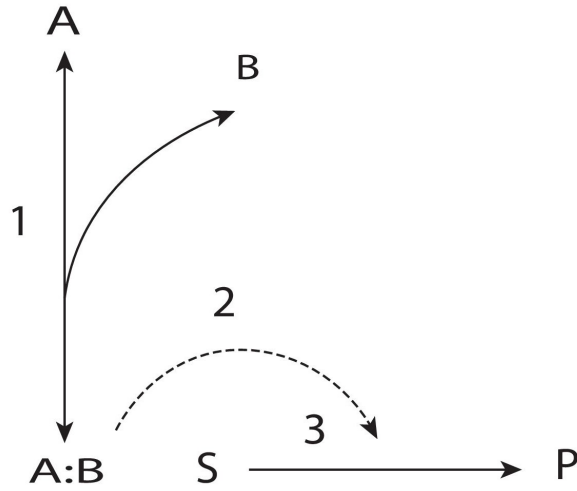


Fig 1. Reaction network diagram of dummy mathematical model.

$$\frac{d(A)}{dt} = -[k^+] * A * B + [k^-] * AB$$

$$\frac{d(B)}{dt} = -[k^+] * A * B + [k^-] * AB$$

$$\frac{d(AB)}{dt} = [k^+] * A * B - [k^-] * AB$$

$$\frac{d(S)}{dt} = -\frac{[k_{catS}^{AB}] * S * AB}{([k_{mS}^{AB}] + S)}$$

$$\frac{d(P)}{dt} = \frac{[k_{catS}^{AB}] * S * AB}{([k_{mS}^{AB}] + S)}$$

Assumed species A, B represent genes/proteins, and AB their complex. S and P indicate substrate and product. By kinetic law of mass action suppose association rate (k^+) and disassociation rate (k^-).

Assumed, AB cleaved substrate S, the cleavage rate is represented by $\frac{[k_{catS}^{AB}] * S * AB}{([k_{mS}^{AB}] + S)}$. Where k_{catS}^{AB} implies catalytic rate and k_{mS}^{AB} implies Michaelis constants such that $k_{mS}^{AB} = \frac{k^- + k_{catS}^{AB}}{k^+}$, disassociation constant $k_d = \frac{k^-}{k^+}$ [1].

References

1. Liu B, Zhang J, Tan PY, Hsu D, Blom AM, Leong B, Sethi S, Ho B, Ding JL, Thiagarajan PS. A computational and experimental study of the regulatory mechanisms of the complement system. PLoS Computational Biology. 2011 Jan 20;7(1):e1001059. <https://doi.org/10.1371/journal.pcbi.1001059> PMID: [21283780](https://pubmed.ncbi.nlm.nih.gov/21283780/)