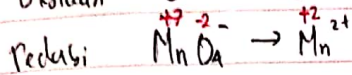
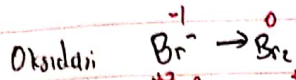
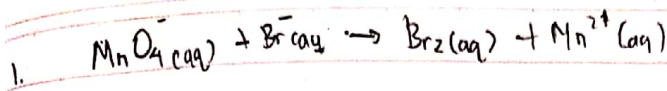
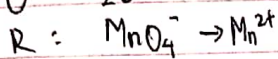
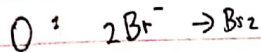


Nama : Ricky
NIM : 21060117120003

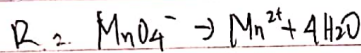
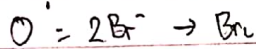
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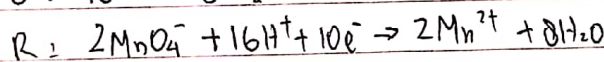
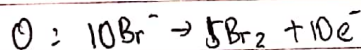
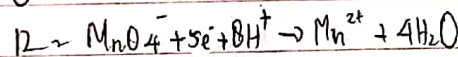
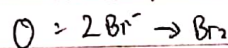
Seimbangkan



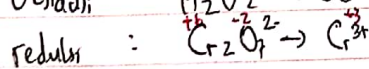
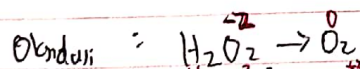
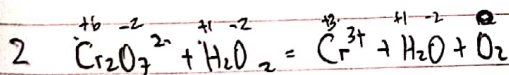
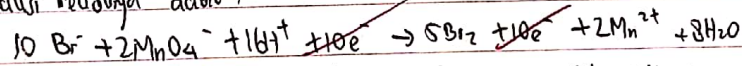
Menambahkan H_2O



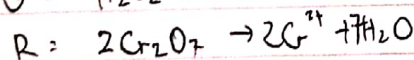
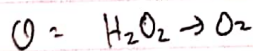
menambahkan H^+ + electron yg hilang. dan kembalikan



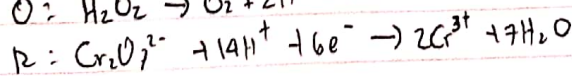
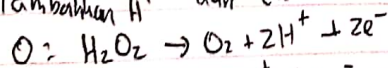
maka reaksi redoksnya adalah



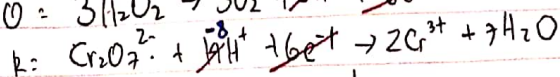
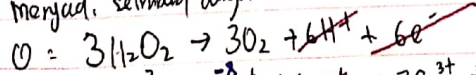
Setarakan siklus dan tambahkan H_2O



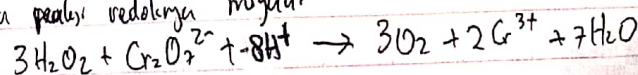
Tambahkan H^+ dan electron



menjadi seimbang dengan

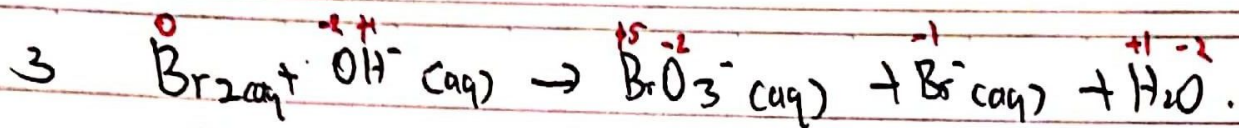


maka reaksi redoksnya menjadi

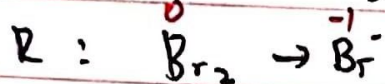
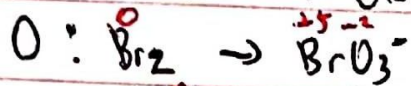


Nama: Ricky

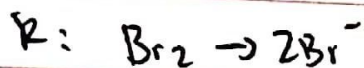
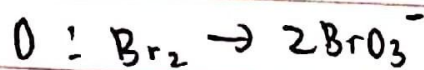
NIM: 21060117170003



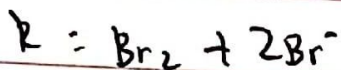
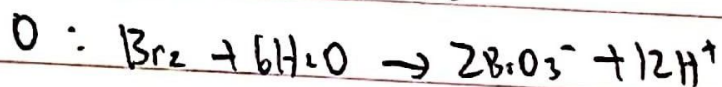
Tentukan Reduksi dan Oksidasi



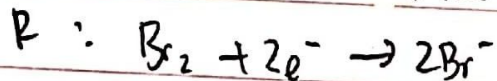
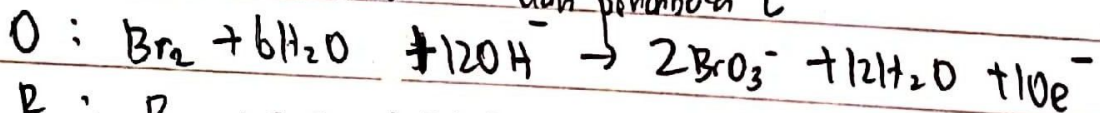
Setimbangkan



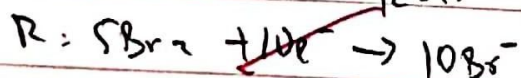
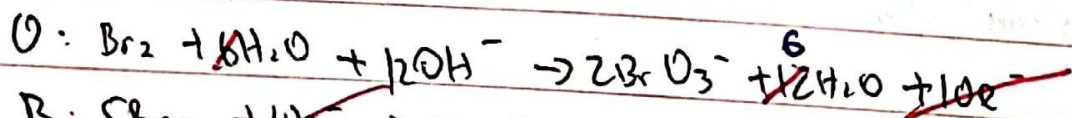
Penambahan H^+ dan H_2O



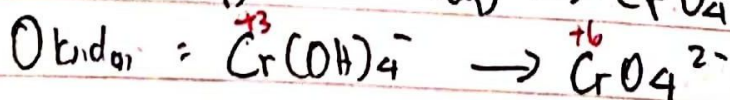
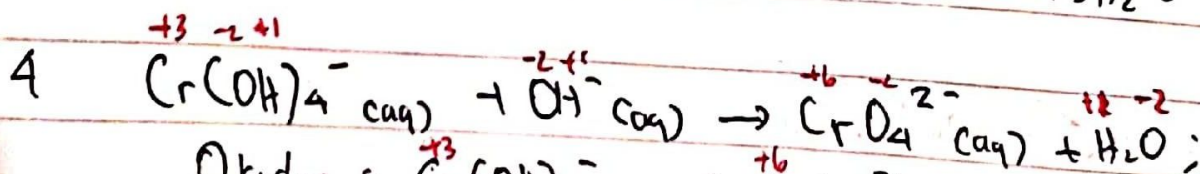
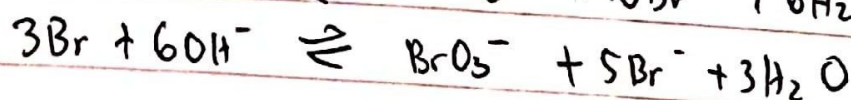
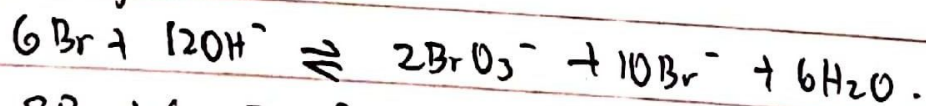
penambahan OH^- disisi lain dan penambahan e^-



Setarakan reaksi



reaksi redoksnya adalah



Merupakan reaksi non redoks karena tidak ada yang dioksidasi