

Thermal Expansion:-

When a body gets heat, its amplitude of vibration increases. So that the length, area and volume increases.

When Length increase  $\rightarrow$  Called Linear expansion.

When area increase  $\rightarrow$  called Superficial expansion

When Volume increase  $\rightarrow$  called Cubical expansion

[1] Co-efficient of Linear expansion (Linear expansivity) ( $\alpha$ ):-

Let us consider a rod of initial length

' $l_1$ ' at temperature ' $\theta_1$ '. When its temperature increase to ' $\theta_2$ ', then its length becomes ' $l_2$ '.

Now,

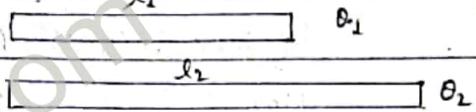


fig: Linear expansion of rod.

Change in Length of rod,  $\Delta l = l_2 - l_1$

change in temperature,  $\Delta \theta = \theta_2 - \theta_1$

Experimentally, it has been found that;

[2] Change in Length of rod is directly proportional to its original length.

$$\text{i.e., } \Delta l \propto l_1 \quad \text{--- (i)}$$

[3] Change in length of rod is directly proportional to change in temperature.

$$\text{i.e., } \Delta l \propto \Delta \theta \quad \text{--- (ii)}$$

Combining eqn (i) and eqn (ii), we get,

$$\Delta l = \alpha l_1 \Delta \theta \quad \text{--- (iii)}$$

Where,  $\alpha$  is proportionality constant called Co-efficient of linear expansion.

$$\text{From eqn (iii), } \frac{\Delta l}{\alpha} = \frac{l_1}{l_1 \Delta \theta}$$

$$\Rightarrow \alpha = \frac{\Delta l}{l_1 \Delta \theta} \quad [\because \text{If } l_1 = 1 \text{ m} \text{ & } \Delta \theta = 1^\circ\text{C or } 1\text{ K}]$$

Thus, co-efficient of linear expansion of the material of the rod is defined as the change in length per unit original length per unit change in temperature. And it is numerically equal to change in

Length

of a unit rod when its temperature change by  $1^\circ\text{C}$  or  $1\text{K}$ .

Also, from eqn (ii)

$$\begin{aligned} l_2 - l_1 &= \alpha l_1 (\theta_2 - \theta_1) \\ \Rightarrow l_2 &= l_1 \{ 1 + \alpha (\theta_2 - \theta_1) \} \quad \text{--- (iv)} \end{aligned}$$

[d] Co-efficient of Superficial expansion (Superficial Expansivity) ( $\beta$ ):-

Let us consider a solid square of initial

length ' $l_1$ ' & area ' $A_1$ ' at temperature ' $\theta_1$ '.

When its temperature increase to ' $\theta_2$ '. Then

its area becomes ' $A_2$ ' of length ' $l_2$ '.

Now,

change in area of Square,  $\Delta A = A_2 - A_1$

& change in temperature,  $\Delta \theta = \theta_2 - \theta_1$

Experimentally, It has been found that,

[I] Change in area of square is directly proportional to its original area.

$$\text{i.e., } \Delta A \propto A_1 \quad \text{--- (i)}$$

[II] Change in area of square is directly proportional to change in temperature.

$$\text{i.e., } \Delta A \propto \Delta \theta \quad \text{--- (ii)}$$

Combining eqn (i) & (ii), we get,

$$\Delta A \propto A_1 \Delta \theta$$

$$\Rightarrow \boxed{\Delta A = \beta A_1 \Delta \theta} \quad \text{--- (iii)}$$

Where,  $\beta$  is proportionality constant called Co-efficient of Superficial expansion:

$$\text{Also, from eqn (ii)} \quad \frac{\Delta A}{A_1 \Delta \theta}$$

$$\Rightarrow \boxed{\beta = \frac{\Delta A}{A_1 \Delta \theta}} \quad [\because \text{If } A_1 = 1 \text{ m}^2 \text{ & } \Delta \theta = 1^\circ\text{C or } 1\text{K}]$$

Thus, Co-efficient of Superficial expansion of solid square is defined as change in area per unit initial area per unit change in temperature. And numerically it is equal to change in Area of a unit square when its temperature change by  $1^\circ\text{C}$  or  $1\text{K}$ .

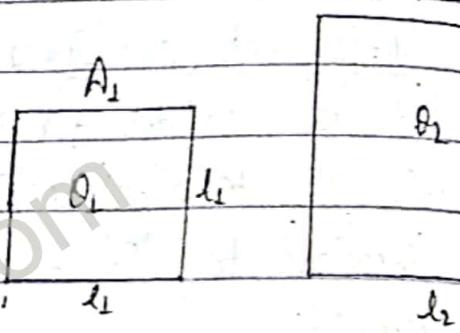


Fig: Superficial expansion of Solid

Also, from eq<sup>n</sup> (iii):

$$A_2 - A_1 = \beta A_1 (\theta_2 - \theta_1)$$

$$\Rightarrow A_2 = A_1 \{ 1 + \beta (\theta_2 - \theta_1) \} \quad \text{--- (iv)}$$

[3] Co-efficient of Cubical expansion ( $\gamma$ ) (Cubical expansivity):-

Let us consider a solid cube of initial Volume ' $V_1$ ' &

length ' $l_1$ ' at temperature ' $\theta_1$ '. When its temperature increases to ' $\theta_2$ ' then its volume becomes ' $V_2$ ' of length ' $l_2$ ' then, if tem-

Change in Volume of Cube,  $\Delta V = V_2 - V_1$

Change in temperature,  $\Delta \theta = \theta_2 - \theta_1$

Experimentally it has been found that,

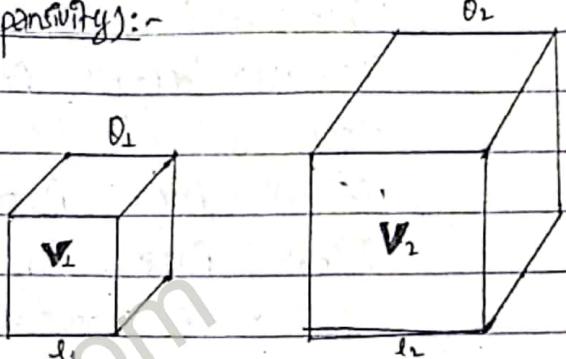


Fig: Cubical expansion of Solid Cube

(i) Change in Volume is directly proportional to

if original volume, i.e.,  $\Delta V \propto V_1 \quad \text{--- (i)}$

(ii) Change in Volume is directly proportional to

Change in temperature, i.e.,  $\Delta V \propto \Delta \theta \quad \text{--- (ii)}$

Combining eq<sup>n</sup> (i) & (ii),  $\Delta V \propto V_1 \Delta \theta$

$$\Rightarrow \boxed{\Delta V = \gamma V_1 \Delta \theta} \quad \text{--- (iii)}$$

Thus, co-efficient of Cubical Expansion

of Solid Cube is defined as change in

Volume per unit original Volume per m

Where,  $\gamma$  is proportionality Constant. Unit Change in temperature. And,

Called co-efficient of Cubical expansion.

numerically, it is equal to change in volume of unit when its

Also, from eq<sup>n</sup> (i):

temperature change by  $1^\circ\text{C}$  or  $1\text{K}$ .

$$\gamma = \frac{\Delta V}{V_1 \Delta \theta}$$

Also, from eq<sup>n</sup> (iii)

$$V_2 - V_1 = \gamma V_1 (\theta_2 - \theta_1)$$

If  $V_1 = 1 \text{ m}^3$  and  $\Delta \theta = 1^\circ\text{C}$  or  $1\text{K}$

then,

$$\boxed{\gamma = \Delta V}$$

$$\Rightarrow \boxed{V_2 = V_1 [1 + \gamma \{ \theta_2 - \theta_1 \}]} \quad \text{--- (iv)}$$

## # Relation between $\alpha$ , $\beta$ & $\gamma$ .

### [I] Relation between $\alpha$ & $\beta$ :-

Let us consider a solid square of initial length ' $l_1$ ' & area ' $A_1$ ' at temperature ' $\theta_1$ '. When its temperature increase to ' $\theta_2$ ', then, its area becomes ' $A_2$ ' of length ' $l_2$ '.

Now,

$$\text{Initial Area, } A_1 = l_1^2 \quad \dots \text{(i)}$$

$$\text{Final Area, } A_2 = l_2^2 \quad \dots \text{(ii)}$$

From Superficial expansivity;

$$A_2 = A_1 \{ 1 + \beta \Delta \theta \} \quad \dots \text{(iii)}$$

Also, from linear expansivity;

$$l_2 = l_1 \{ 1 + \alpha \Delta \theta \} \quad \dots \text{(iv)}$$

Using eqn (iv) in eqn (ii)

$$A_2 = [l_1 \{ 1 + \alpha \Delta \theta \}]^2$$

$$= l_1^2 \{ 1 + 2\alpha \Delta \theta + \alpha^2 \Delta \theta^2 \}$$

Since,  $\alpha$  is small quantity, so its higher term can be neglected,

$$A_2 = l_1^2 \{ 1 + 2\alpha \Delta \theta \}$$

$$A_2 = A_1 \{ 1 + 2\alpha \Delta \theta \} \quad \dots \text{(v)} \quad [\because \text{using eqn (i)}]$$

Comparing eqn (iii) and eqn (v), we get;

$$\Rightarrow \boxed{\beta = 2\alpha}$$

This is required relation between  $\alpha$  &  $\beta$ .

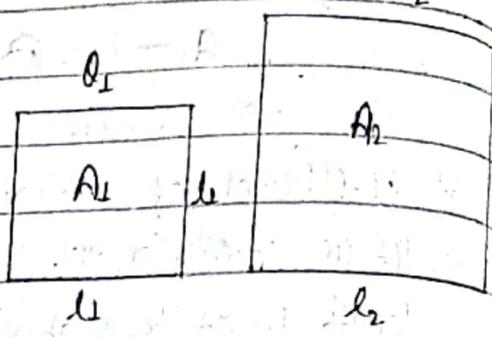


fig: Superficial expansion.

## [Q] Relation between $\alpha$ & $\gamma$ :

Let us consider a solid cube of initial Volume ' $V_1$ ' & length ' $l_1$ ' at temperature ' $\theta_1$ '. When its temperature increases to ' $\theta_2$ ', then its volume becomes ' $V_2$ ' of length ' $l_2$ '.

Now,

$$\text{Initial Volume, } V_1 = l_1^3 \quad \text{--- (i)}$$

$$\text{Final Volume, } V_2 = l_2^3 \quad \text{--- (ii)}$$

From Cubical expansivity:

$$V_2 = V_1 \{1 + \gamma \Delta \theta\} \quad \text{--- (iii)}$$

Also, from linear expansivity

$$l_2 = l_1 \{1 + \alpha \Delta \theta\} \quad \text{--- (iv)}$$

Using eqn (iv) in eqn (iii)

$$V_2 = [l_1 \{1 + \alpha \Delta \theta\}]^3 \\ = l_1^3 \{1 + 3\alpha \Delta \theta + 3\alpha^2 \Delta \theta^2 + \alpha^3 \Delta \theta^3\}$$

Since,  $\alpha$  is small quantity. So, its higher power term can be neglected.

$$V_2 = l_1^3 \{1 + 3\alpha \Delta \theta\}$$

$$V_2 = V_1 \{1 + 3\alpha \Delta \theta\} \quad \text{--- (v)} \quad [\because \text{using eqn (i)}$$

Comparing eqn (iii) & (v), we get,

$$\gamma = 3\alpha$$

This is required relation between  $\alpha$  and  $\gamma$ .

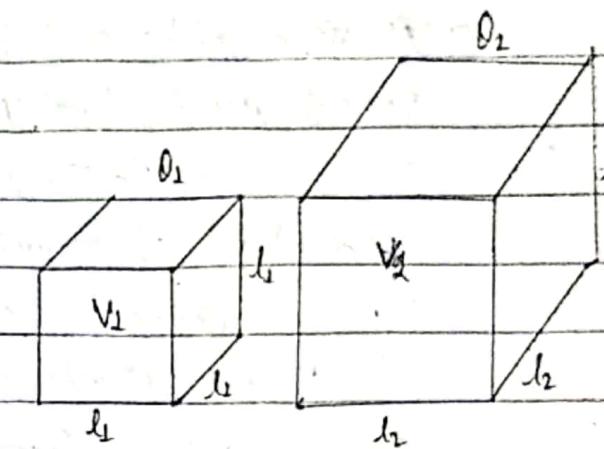


fig: Cubical expansion

We have,

$$\beta = 2\alpha$$

$$\therefore \alpha = \frac{\beta}{2} \quad \text{--- (1)}$$

Also, we have,

$$\gamma = 3\alpha$$

$$\therefore \alpha = \frac{\gamma}{3} \quad \text{--- (2)}$$

From eqn (1) & (2)

$$\Rightarrow \alpha = \frac{\beta}{2} = \frac{\gamma}{3}$$

This is required relation between  $\alpha$ ,  $\beta$  &  $\gamma$ .

## # Expansion of liquid:-

Let us Consider a liquid kept in a vessel up to Level 'A'. When the vessel is heated initially, the vessel expands and level of liquid falls to 'B'.

When, heating is continue the liquid, the liquid expands up to level 'C'. Here the expansion from initial level 'A' to final Level 'C' is called apparent expansion. Expansion from B to C is called real expansion.

And the expansion from A to B is called vessel expansion.

From figure, BC = AC + AB

$$\text{i.e., Real expansion} = \text{Apparent expansion} + \text{Vessel expansion}$$

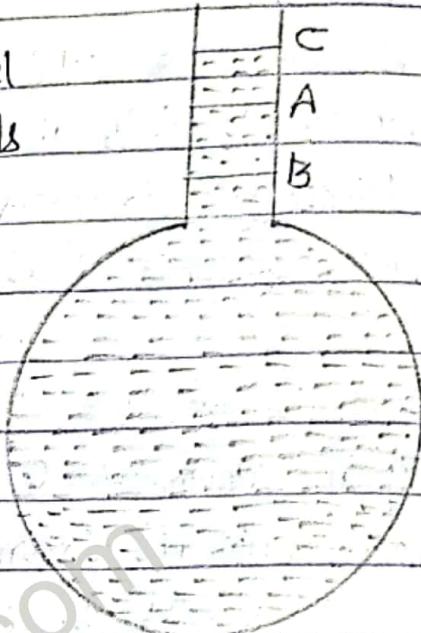


fig: Expansion of Liquid

## # Co-efficient of Real Expansion [Real expansivity] ( $\gamma_r$ ):-

It is defined as the, "Real change in volume of liquid per unit original volume per unit change in temperature." It is also called absolute expansivity. If  $\Delta V_r$  be the real change in volume of liquid with initial Volume 'V' for change in temperature ' $\Delta\theta$ ', then, Real expansivity is,

$$\gamma_r = \frac{\Delta V_r}{V \Delta\theta}$$

$$\Rightarrow \Delta V_r = \gamma_r V \Delta\theta \quad \dots \text{---(i)}$$

## # Co-efficient of Apparent Expansion [Apparent expansivity] ( $\gamma_a$ ):-

It is defined as the, "apparent change in volume of liquid per unit original volume per unit change in temperature".

If ' $\Delta V_a$ ' be the Apparent change in Volume of liquid with initial volume 'V' for change in temperature ' $\Delta\theta$ ', then, Apparent expansivity;

$$\gamma_a = \frac{\Delta V_a}{V \Delta\theta}$$

$$\Rightarrow \Delta V_a = \gamma_a V \Delta\theta \quad \text{--- (i)}$$

# Co-efficient of Cubical expansion of vessel [expansivity of vessel] ( $\gamma_v$ ): -

It is defined as the "change in Volume of vessel per unit original Volume per unit Change in temperature".

If ' $\Delta V_v$ ' be the change in Volume of vessel with initial Volume 'V' for change in temperature ' $\Delta\theta$ '. The, Expansivity of vessel;

$$\gamma_v = \frac{\Delta V_v}{V \Delta\theta}$$

$$\Rightarrow \Delta V_v = \gamma_v V \Delta\theta \quad \text{--- (ii)}$$

Since,

$$\Delta V_a = \Delta V_v + \Delta V_r$$

$$\text{or, } \gamma_a \underbrace{V \Delta\theta} = \gamma_v \underbrace{V \Delta\theta} + \gamma_r V \Delta\theta$$

$$\text{or, } \gamma_a = \gamma_v + \gamma_r$$

$$\Rightarrow \gamma_a = \gamma_v + 3\alpha_v$$

This is required relation between Real expansivity and Apparent Expansivity:

# Variation of density with temperature:-

Let us consider a body having mass 'M' and Volume 'V' at temperature ' $\theta_i$ '. Then, density is given by;

$$\rho_i = \frac{M}{V} \quad \text{--- (iii)}$$

Suppose the body is heated upto temperature ' $\theta_2$ ' and its volume becomes ' $V_2$ ' but its mass remains same so, density be,

$$\rho_2 = \frac{M}{V_2} \quad \text{--- (i)}$$

Dividing eqn (i) by (i)

$$\text{or, } \frac{\rho_2}{\rho_1} = \frac{V_1}{V_2} \quad \text{--- (ii)}$$

If 'Y' be the Cubical expansivity of the material/body then,

$$V_2 = V_1 \{1 + Y \Delta \theta\} \quad \text{--- (iii)}$$

$$\text{Where, } \Delta \theta = \theta_2 - \theta_1$$

Using eqn (iii) in eqn (ii)

$$\text{or, } \frac{\rho_2}{\rho_1} = \frac{V_1}{V_2 \{1 + Y \Delta \theta\}}$$

$$\Rightarrow \boxed{\rho_2 = \frac{\rho_1}{1 + Y \Delta \theta}} \rightarrow \text{for numerical}$$

$$\text{or, } \rho_2 = \rho_1 (1 + Y \Delta \theta)^{-1}$$

Using Binomial Expansion and Neglecting higher power term;

$$\boxed{\rho_2 = \rho_1 (1 - Y \Delta \theta)} \quad \text{--- (iv)}$$

Eqn (iv) gives Variation of density with temperature.

## # Dulong and Petit's method:-

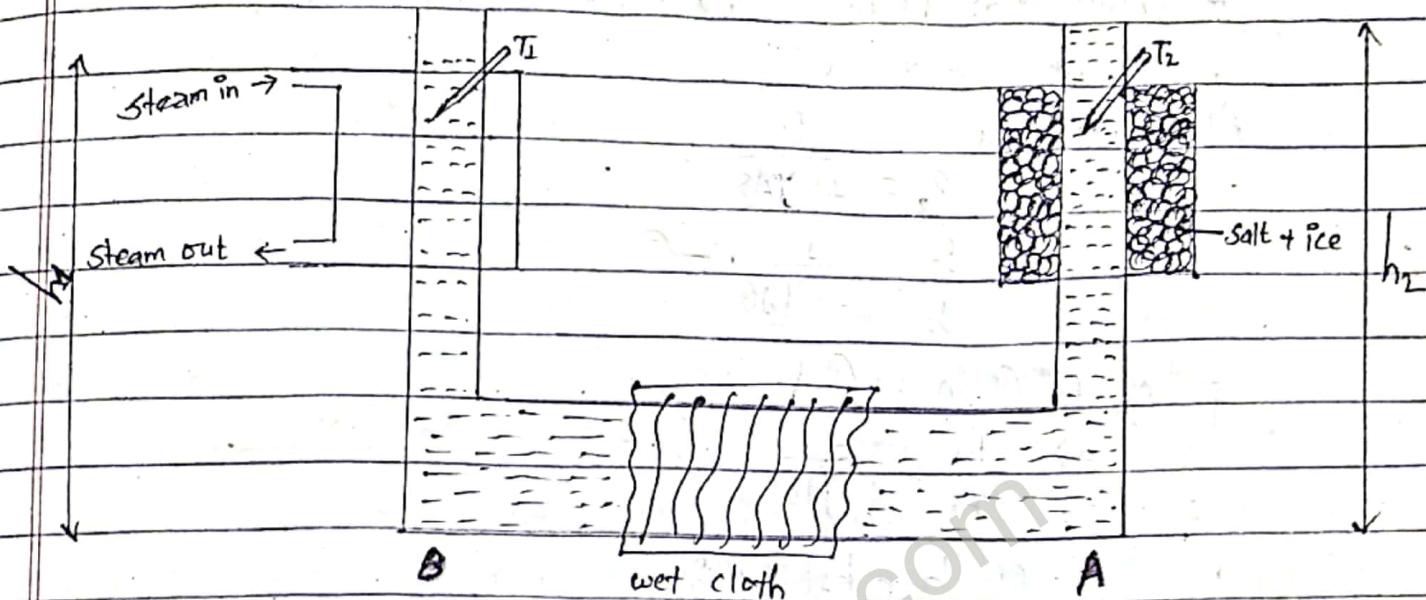


fig: Dulong & petit's method for measuring Real expansivity.

↳ The experiment of Dulong & Petit's method depends upon the principle of hydrostatic. It states that, "The height of liquid column that produce same pressure are inversely proportional to their density."

↳ Let us Consider a U-shaped glass tube filled with liquid. Its one side is kept inside the ice and salt. and other side is kept inside the Steam in and steam out. The horizontal portion of tube is wrapped with wet cloth to prevent exchange of heat with surrounding. two thermometer are provided to note the temperature of A & B side of the tube as shown in figure above.

Let ' $h_1$ ,  $\rho_1$ ,  $T_1$  and  $h_2$ ,  $\rho_2$ ,  $T_2$ ' be the height, density and temperature of liquid at B & A side of glass tube respectively.

Now, From principle of hydrostatic,

pressure of A = pressure of B

$$\Rightarrow h_2 \rho_2 g = h_1 \rho_1 g$$

$$\Rightarrow \frac{h_1}{h_2} = \frac{\rho_2}{\rho_1} \quad \text{--- (i)}$$

It shows that height is inversely proportional to density.

$$\text{i.e., } h \propto \frac{1}{\gamma}$$

Also, we have,

$$\frac{s_2}{s_1} = \frac{1}{1 + \gamma \Delta \theta}$$

$$\frac{s_2}{s_1} = \frac{1}{1 + \gamma \Delta \theta} \quad \text{--- (i)}$$

putting eqn (i) in eqn (i).

$$\Rightarrow \frac{h_2}{h_1} = \frac{1}{1 + \gamma \Delta \theta}$$

$$\text{or, } h_2 = h_1(1 + \gamma \Delta \theta)$$

$$\text{or, } \gamma \Delta \theta = \frac{h_2 - h_1}{h_1} = \frac{h_2 - h_1}{h_1}$$

$$\therefore \boxed{\gamma = \frac{h_2 - h_1}{h_1 \Delta \theta}} \quad \text{--- (ii)}$$

Using this eqn (ii), we can calculate the real expansivity of liquid.

## # Numericals :-

Q1 [A] A glass flask of volume  $400 \text{ cm}^3$  is just filled with mercury at  $0^\circ\text{C}$ . How much mercury will overflow when the temperature of the system rises to  $80^\circ\text{C}$ .

\* Soln:-

$$V_{2m} = ?, V_{1m} = V_{1f} = 400 \text{ cm}^3, \gamma_m = 1.8 \times 10^{-4} \text{ K}^{-1}, \gamma_f = 5.4 \times 10^{-6} \text{ K}^{-1}, V_{2f} = ?, \Delta \theta = 80^\circ\text{C}$$

Now,

$$V_{2m} = V_{1m} [1 + \gamma_m \Delta \theta]$$

$$= 400 [1 + (1.8 \times 10^{-4} \times 80)]$$

$$= 405.76 \text{ cm}^3$$

$$V_{2f} = V_{1f} [1 + \gamma_f \Delta \theta]$$

$$= 400 [1 + (5.4 \times 10^{-6} \times 80)]$$

$$= 400.1228 \text{ cm}^3$$

Now,

Volume of overflow mercury,

$$= V_{2m} - V_{2f}$$

$$= 405.76 \text{ cm}^3 - 400.1228 \text{ cm}^3$$

$$= 5.5872 \text{ cm}^3$$

#

Q. 1 [B] A glass vessel of Volume  $50 \text{ cm}^3$  is filled with mercury and is heated from  $20^\circ\text{C}$  to  $60^\circ\text{C}$ . What is the volume of mercury will overflow?

★ Soln:-

$$V_{2m} = V_{1m} [1 + \gamma_m \Delta \theta]$$

$$= 50 [1 + (1.8 \times 10^{-4} \times 10)]$$

$$= 50.0108 \text{ cm}^3$$

$$V_{2v} = V_{1v} [1 + \gamma_v \Delta \theta]$$

$$= 50 [1 + (5.4 \times 10^{-6} \times 40)]$$

$$= 50.3492 \text{ cm}^3$$

$$\text{Vol}^m \text{ of overflow mercury,}$$

$$= V_{2m} - V_{2v}$$

$$= 0.3492 \text{ cm}^3 \#$$

Q. 1 [C] A glass flask with volume  $200 \text{ cm}^3$  is filled to the brim with mercury at  $20^\circ\text{C}$ . How much mercury overflows when the temperature of the system is raised to  $100^\circ\text{C}$ ?

★ Soln:-

$$V_{2m} = V_{1m} [1 + \gamma_m \Delta \theta]$$

$$= 200 [1 + (1.8 \times 10^{-4} \times 80)]$$

$$= 202.88 \text{ cm}^3$$

$$V_{2f} = V_{1f} [1 + \gamma_f \Delta \theta]$$

$$= 200 [1 + (5.4 \times 10^{-6} \times 80)]$$

$$= 200.0864 \text{ cm}^3$$

$$\text{Vol}^m \text{ of mercury overflows,}$$

$$= V_{2m} - V_{2f}$$

$$= 2.7936 \text{ cm}^3$$

Q. 1 [D] A Copper vessel with a volume of exactly  $100 \text{ m}^3$  at a temperature of  $15^\circ\text{C}$  is filled with glycerin. If the temperature rises to  $25^\circ\text{C}$ , how much glycerin will spill out?

★ Soln:-

$$V_{1g} = V_{1g} [1 + \gamma_g \Delta \theta]$$

$$= 100 [1 + (1.0 \times 10^{-5} \times 10)]$$

$$= 100.49 \text{ m}^3$$

$$V_{2v} = V_{1v} [1 + \gamma_v \Delta \theta]$$

$$= 100 [1 + (5.4 \times 10^{-6} \times 10)]$$

$$= 100.051 \text{ m}^3$$

$$\text{Volume of spill out glycerin,}$$

$$= V_{2g} - V_{2v}$$

$$= 0.439 \text{ m}^3 \#$$

Q. 1 [E] A glass flask of volume  $500 \text{ cm}^3$  is just filled with mercury at  $0^\circ\text{C}$ . How much mercury overflows when the temperature of the system is raised to  $80^\circ\text{C}$ ?

★ ★

$$V_{2m} = V_{1m} [1 + \gamma_m \Delta \theta]$$

$$= 500 [1 + (1.8 \times 10^{-4} \times 80)]$$

$$= 507.2 \text{ m}^3$$

$$V_{2f} = V_{1f} [1 + \gamma_f \Delta \theta]$$

$$= 500 [1 + (5.4 \times 10^{-6} \times 80)]$$

$$= 500.116 \text{ m}^3$$

$$\text{Vol}^m \text{ of overflow mercury,}$$

$$= V_{2m} - V_{2f}$$

$$= 6.984 \text{ m}^3 \#$$

**Q. 2[A]** A steel wire having length 8m and diameter 4 mm is fixed between two rigid supports. Calculate increase in tension on a wire when temperature falls by  $10^{\circ}\text{C}$ . Where  $\gamma_w = 2 \times 10^{11} \text{ N/m}^2$ ,  $\alpha_s = 1.2 \times 10^{-5} \text{ K}^{-1}$ ,  $d = 4\text{mm} = 0.004 \text{ m}$ .

\* **SOL:** -  $\Delta\theta = 10^{\circ}\text{C}$

We have,

$$\gamma = \frac{F \times L}{A \times \frac{\Delta L}{L}} \Rightarrow F = \frac{\gamma_w A_w \Delta L_w}{L_w} = \frac{2 \times 10^{11} \times \pi d^2 \times \alpha_s l_w \Delta \theta}{L_w} = \frac{10^{11} \pi (0.004)^2}{2} \times 1.2 \times 10^{-5} \times 8$$

$$\therefore \text{Tension} = \frac{192 \pi}{2} = 96 \pi = 301.6 \text{ N} \#$$

**B. 2[B]** Two ends of a steel wire of length 8m and 2mm radius are fixed to two rigid supports. Calculate the increase in tension in the wire when temperature falls by  $10^{\circ}\text{C}$ .

\* **SOL:** -  $\gamma_w = 2 \times 10^{11}$ ;  $l_w = 8\text{m}$ ,  $r_w = 2\text{mm} = 0.002 \text{ m}$ ,  $\Delta\theta = 10^{\circ}\text{C}$ ,  $\alpha_s = 1.2 \times 10^{-5} \text{ K}^{-1}$

Now,

$$\gamma_w = \frac{F \times L_w}{A_w \frac{\Delta L_w}{L_w}} \Rightarrow F = \frac{\gamma_w A_w \Delta L_w}{L_w} = \frac{2 \times 10^{11} \times \pi r^2 \times \alpha_s l_w \Delta \theta}{L_w} = \frac{2 \pi \times 10^{11} \times (0.002)^2 \times 1.2 \times 10^{-5} \times 8}{8} = 96 \pi = 301.6 \text{ N} \#$$

Hence, Required tension is  $301.6 \text{ N} \#$

**Q. 3[A]** A iron rod of length 100m at  $10^{\circ}\text{C}$  is used to measure a distance of 2km on a day when the temperature is  $40^{\circ}\text{C}$ . Calculate the error in measuring the distance.

\* **SOL:** -

Length of rod at  $40^{\circ}\text{C}$ ,

$$= 100 \left[ 1 + (16 \times 10^{-6} \times 30) \right]$$

$$= 100.048 \text{ m}$$

$$\begin{aligned} \text{Now, } 100\text{m} &\xrightarrow{40^{\circ}\text{C}} 100.048 \text{ m} \\ 1\text{m} &\xrightarrow{40^{\circ}\text{C}} 1.00048 \text{ m} \\ 2\text{km} &\xrightarrow{40^{\circ}\text{C}} 1.00048 \times 2000 \text{ m} \\ &= 2000.96 \text{ m} \end{aligned}$$

Hence,

Error in measuring the distance,

$$= 2000.96 \text{ m} - 2000 \text{ m}$$

$$\Rightarrow 0.96 \text{ m} \#$$

Q. 4[A] The marking on an aluminium ruler and a brass ruler are perfectly aligned at  $0^\circ\text{C}$ . How far apart will the 20.0 cm marks be on the two rulers at  $100^\circ\text{C}$ , if precise alignment of the left hand ends of the rulers is maintained?

Soln:-

$$\begin{aligned} l_{2a} &= l_{1a}[1 + \alpha_a \Delta\theta] & l_{2b} &= l_{1b}[1 + \alpha_b \Delta\theta] & \text{difference in distance,} \\ &= 20[1 + (2.4 \times 10^{-5} \times 100)] & &= 20[1 + (2 \times 10^{-5} \times 100)] & = l_{2a} - l_{2b} \\ &= 20.048 \text{ cm} & &= 20.04 \text{ cm} & = 0.008 \text{ cm} \end{aligned}$$

Thus, 0.008 cm far will be 20.0 cm marked.

Q. 4[B] The length of an iron rod is measured by a brass scale. When both of them are at  $10^\circ\text{C}$ , the measured length is 50 cm. What is the length of the rod at  $40^\circ\text{C}$  when measured by the brass scale at  $40^\circ\text{C}$ ? ( $\alpha_b = 24 \times 10^{-6} \text{ C}^{-1}$ ,  $\alpha_i = 12 \times 10^{-6} \text{ C}^{-1}$ )

Soln:-  $\Delta\theta = 40^\circ\text{C} - 10^\circ\text{C} = 30^\circ\text{C}$

$$\begin{array}{lll} \text{Length of rod at } 40^\circ\text{C}, & \text{Length of scale at } 40^\circ\text{C}, & \text{difference in distance,} = 0.012 \text{ cm} \\ l_{2i} = l_{1i}[1 + \alpha_i \Delta\theta] & l_{2b} = l_{1b}[1 + \alpha_b \Delta\theta] & \text{Hence,} \\ = 50[1 + (12 \times 10^{-6} \times 30)] & = 50[1 + (24 \times 10^{-6} \times 30)] & \text{Length of rod at } 40^\circ\text{C measured by} \\ = 50.024 \text{ cm} & = 50.036 \text{ cm} & \text{Scale is } (50 - 0.012) \text{ cm} = 49.988 \text{ cm} \end{array}$$

Q. 4[C] A brass rod of length  $0.40 \text{ m}$  and steel rod of length  $0.60 \text{ m}$ , both are initially at  $0^\circ\text{C}$  are heated to  $75^\circ\text{C}$ . If the increase in lengths is the same for both the rods. Calculate the linear expansivity of brass. ( $\alpha_s = 12 \times 10^{-6} \text{ K}^{-1}$ )

Soln:-

$$(\Delta l)_{\text{brass}} = (\Delta l)_{\text{steel}}$$

$$\Rightarrow l_b \alpha_b \Delta\theta = l_s \alpha_s \Delta\theta = 0.6 \times 12 \times 10^{-6}$$

$$\alpha_b = \frac{7.2 \times 10^{-6}}{0.4} = 18 \times 10^{-6} \text{ K}^{-1}$$

thus, linear expansivity of brass (rod) =  $18 \times 10^{-6} \text{ K}^{-1}$ .

Q.4[D] An aluminium rod when measured with a steel scale, both being at  $25^\circ\text{C}$  appears to be 1m long. If the scale is correct at  $0^\circ\text{C}$ , what will be the length of rod at  $0^\circ\text{C}$ ? [ $\alpha_A = 26 \times 10^{-6} \text{ K}^{-1}$ ,  $\alpha_S = 12 \times 10^{-6} \text{ K}^{-1}$ ]

\* Soln:-

Length of steel scale at  $25^\circ\text{C}$ ,  $\Rightarrow l_{25A} = 1.0003 \text{ m}$ ,  $\Delta T = ?$

$$\begin{aligned} L_{25S} &= l_{25S}[1 + \alpha_S \Delta T] \\ &= 1 [1 + (12 \times 10^{-6} \times 25)] \\ &= 1.0003 \text{ m} \end{aligned}$$

$$\text{or, } l_{25A} = l_{25S}[1 + \alpha_A \Delta T]$$

$$\text{or, } 1.0003 = l_{25S}[1 + (26 \times 10^{-6} \times 25)] \\ \Rightarrow l_{25A} = 0.99 \text{ m}$$

Hence, length of rod at  $0^\circ\text{C}$  is accurately 0.99 m #

Q.2[C] A Copper wire of diameter 0.5 mm is stretched between two points at  $25^\circ\text{C}$ . Calculate the increase in tension in the wire if the temperature falls to  $0^\circ\text{C}$ . ( $\gamma_C = 1.2 \times 10^{11} \text{ N/m}^2$ ,  $\alpha_C = 18 \times 10^{-6} \text{ K}^{-1}$ )

\* Soln We have,

$$\gamma = \frac{F}{A} \times \frac{\Delta l}{l} \Rightarrow F = \frac{\gamma A l}{\Delta l} = \frac{1.2 \times 10^{11} \times (\pi r^2) \times \alpha C \Delta T}{l} = 1.2 \times 10^{11} \times (0.0005)^2 \times 25 \times 18 \times 10^{-6} = 10.6 \text{ N}$$

thus, increase in tension is 10.6 N #

Q.5[A] Using the following data, determine the temperature at which wood will just sink in benzene? [Density of benzene at  $0^\circ\text{C} = 9 \times 10^2 \text{ kg/m}^3$  &  $\beta_w^{0^\circ\text{C}} = 8.8 \times 10^2 \text{ kg/m}^3$ ]

\* Soln

$$\beta_{2W} = \beta_{2B}$$

$$\text{or, } \frac{\beta_{2W}}{1 + \gamma_W \Delta T} = \frac{\beta_{2B}}{1 + \gamma_B \Delta T}$$

$$\text{or, } \frac{8.8 \times 10^2}{1 + (1.5 \times 10^{-4} \times (T-0))} = \frac{9 \times 10^2}{1 + (1.2 \times 10^{-3} \times (T-0))}$$

$$\text{or, } 8.8 [1 + (1.2 T \times 10^{-3})] = 9 [1 + (1.5 T \times 10^{-4})]$$

$$\text{or, } 1 + (1.2 T \times 10^{-3}) = 1.02 [1 + (1.5 T \times 10^{-4})]$$

$$\text{or, } 1 - 1.02 + (1.2 T \times 10^{-3}) = [1.5 T \times 10^{-4}]$$

$$\text{or, } -0.02 + (1.2 \times 10^{-3})T = (1.5 \times 10^{-4})T$$

$$\text{or, } 0.00 + 1.8 T = 0.02$$

$$T =$$

$$\text{or, } 8.8 + (1.05 \times 10^{-3} T) = 9 + (1.35 \times 10^{-4} T)$$

$$\text{or, } 10.56 \times 10^{-3} T = 0.2 + (1.35 \times 10^{-4} T)$$

$$\text{or, } 0.00921 T = 0.2$$

$$\Rightarrow T = 21.71^\circ\text{C}$$

$$= 294.71 \text{ K} \#$$

Hence, temp of wood is  $21.71^\circ\text{C}$  #

Q.5[B] The density of Silver at  $0^{\circ}\text{C}$  is  $10310 \text{ kg/m}^3$  and the coefficient of linear expansion is  $0.000019 \text{ K}^{-1}$ . Calculate its density at  $100^{\circ}\text{C}$ .

\* Soln:-

$$\rho_0 = 10310 \text{ kg/m}^3, \alpha = 0.000019 \text{ K}^{-1}, \rho_{100} = ?, \Delta\theta = 100^{\circ}\text{C}$$

$$\text{Now, } \frac{\rho_0}{\rho_{100}} = \frac{10310}{1 + (\alpha \times 100)} = \frac{10310}{1 + (3 \times 0.000019 \times 100)} = \frac{10310}{1.0057} = 10251.56 \text{ kg/m}^3$$

Hence, density of Silver at  $100^{\circ}\text{C}$  is  $10251.56 \text{ kg/m}^3$  #

Q.6[A] A second pendulum made of brass keeps correct time at  $10^{\circ}\text{C}$ . How many seconds it will lose or gain per day when the temperature of its surrounding rises to  $35^{\circ}\text{C}$ ?

\* Soln:-

$$T_{10} = 2\pi \sqrt{\frac{l_{10}}{g}} \quad \text{(i)}$$

$$T_{35} = 2\pi \sqrt{\frac{l_{35}}{g}} \quad \text{(ii)}$$

Dividing eqn(ii) by eqn(i)

$$\frac{T_{35}}{T_{10}} = \sqrt{\frac{l_{35}}{l_{10}}} = \sqrt{\frac{l_{10}(1 + \alpha_b \Delta\theta)}{l_{10}}} = \sqrt{1 + (2 \times 10^{-5} \times 25)} = 1.000249969$$

Again, Required time =  $[1.000249969 - 1] \times 24 \times 60 \times 60 = 21.6 \text{ sec}$   $^{10^8}$  #

Q.6[B] A brass pendulum clock keeps correct time at  $15^{\circ}\text{C}$ . How many seconds per day it will lose or gain at  $0^{\circ}\text{C}$ ?

\* Soln:-

$$T_{15} = 2\pi \sqrt{\frac{l_{15}}{g}} \quad \text{(i)}$$

$$T_0 = 2\pi \sqrt{\frac{l_0}{g}} \quad \text{(ii)}$$

Dividing eqn(i) by eqn(ii)

$$\frac{T_{15}}{T_0} = \sqrt{\frac{l_{15}}{l_0}} = \sqrt{\frac{l_0(1 + \alpha \Delta\theta)}{l_0}} = \sqrt{1 + (2 \times 10^{-5} \times 15)} = 1.000149989$$

Now,

Required time =  $[1.000149989 - 1] \times 24 \times 60 \times 60 = 12.96 \text{ sec}$  gain #

Q.6(C) The pendulum of a clock is made of brass. If the clock keeps correct time at  $15^{\circ}\text{C}$ , how many seconds per day it will lose at  $20^{\circ}\text{C}$ ?

~~A.~~ Soln

$$T_{15} = 2\pi \sqrt{\frac{l_{15}}{g}} \quad \text{--- (i)}$$

$$T_{20} = 2\pi \sqrt{\frac{l_{20}}{g}} \quad \text{--- (ii)}$$

Dividing (ii) by (i);

$$\frac{T_{20}}{T_{15}} = \sqrt{\frac{l_{20}}{l_{15}}} = \sqrt{\frac{1.15}{1.05}} = \sqrt{1 + \alpha \Delta T} = \sqrt{1 + (2 \times 10^{-5} \times 5)} = 1.000049999$$

Again,

$$\text{Thus, Required loss time} = [1.000049999 - 1] \times 24 \times 60 \times 60 \text{ sec} = 4.32 \text{ sec} \quad \#$$

Q.6(D) A clock which has a brass pendulum beats seconds correctly when the temperature of the room is  $30^{\circ}\text{C}$ . How many seconds it will gain or lose per day when the temperature of the room falls to  $10^{\circ}\text{C}$ ? ( $\alpha_b = 0.000018 \text{ K}^{-1}$ )

~~A.~~ Soln:-

$$T_{30} = 2\pi \sqrt{\frac{l_{30}}{g}} \quad \text{--- (i)}$$

$$T_{10} = 2\pi \sqrt{\frac{l_{10}}{g}} \quad \text{--- (ii)}$$

Dividing eqn (i) by eqn (ii)

$$\begin{aligned} \text{Or, } \frac{T_{30}}{T_{10}} &= \sqrt{\frac{l_{30}}{l_{10}}} = \sqrt{\frac{1.05 \{1 + \alpha \Delta T\}}{1.00}} = \sqrt{1 + \alpha \Delta T} \\ &= \sqrt{1 + (0.000018 \times 20)} \\ &= 1.000179984 \end{aligned}$$

Again,

$$\begin{aligned} \text{Thus, Required time} &= [1.000179984 - 1] \times 24 \times 60 \times 60 \text{ sec} \\ &= 15.55 \text{ sec gain} \quad \# \end{aligned}$$