

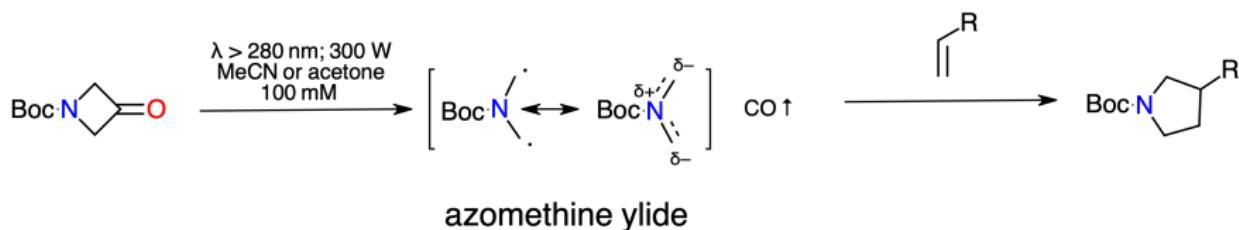
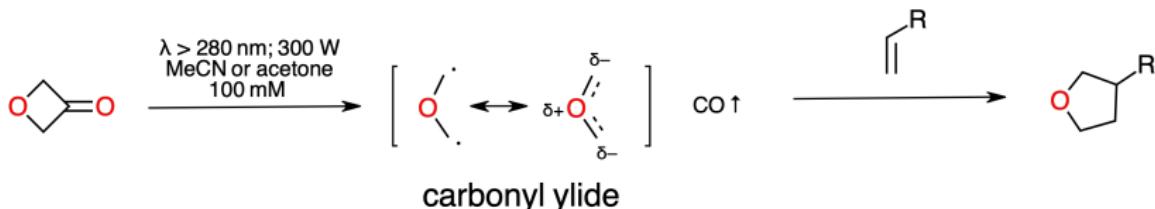
Photodecarbonylation of strained heterocycles: Synthetic applications, experimental insights, and computational models

Zarko Boskovic

Department of Medicinal Chemistry
University of Kansas

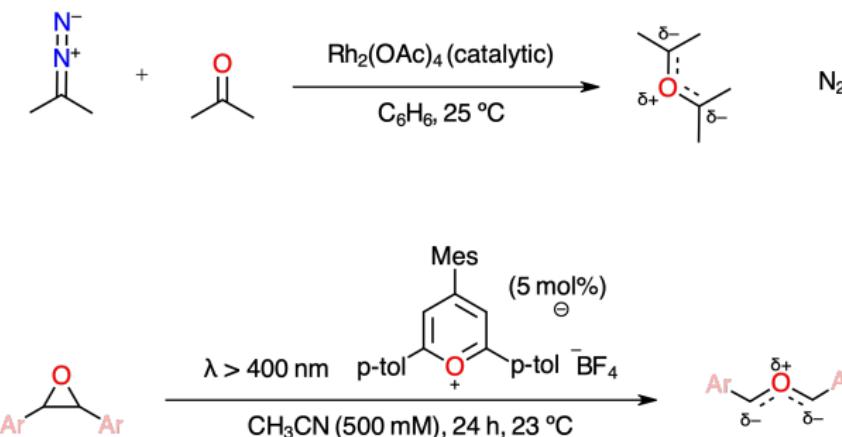
March 27, 2023
ACS Spring National Meeting
Indianapolis

Access to the simplest ylides¹



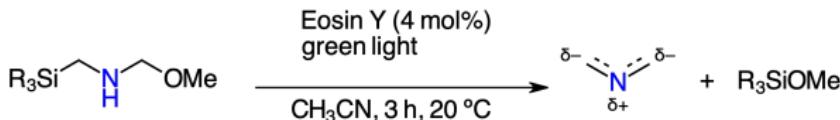
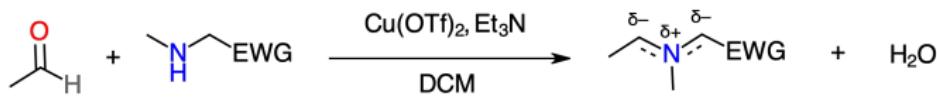
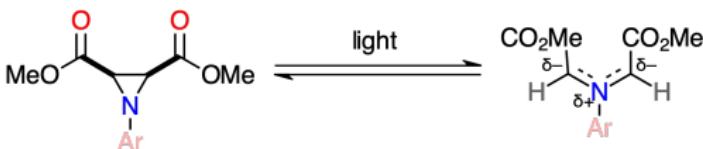
¹ Manvendra Singh et al. "Photochemical Decarbonylation of Oxetanone and Azetidinone: Spectroscopy, Computational Models, and Synthetic Applications". In: Angewandte Chemie International Edition 62.3 (2023), e202215856.

Methods of carbonyl ylide synthesis²



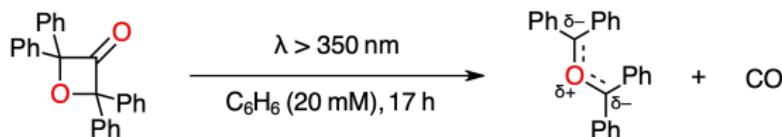
²Albert Padwa. "Generation and utilization of carbonyl ylides via the tandem cyclization-cycloaddition method". en. In: *Accounts of Chemical Research* 24.1 (Jan. 1991), pp. 22–28, Edwin Alfonzo, Felix Steven Alfonso, and Aaron B. Beeler. "Redesign of a Pyrylium Photoredox Catalyst and Its Application to the Generation of Carbonyl Ylides". In: *Organic Letters* 19.11 (June 2017), pp. 2989–2992.

Azomethine ylide synthesis³



³Rolf. Huisgen, Wolfgang. Scheer, and Helmut. Huber. "Stereospecific Conversion of *cis-trans* Isomeric Aziridines to Open-Chain Azomethine Ylides". In: Journal of the American Chemical Society 89.7 (Mar. 1967), pp. 1753–1755, Robert M. Williams et al. "Asymmetric [1,3]-dipolar cycloaddition reactions: synthesis of highly substituted proline derivatives". In: The Journal of Organic Chemistry 57.24 (Nov. 1992), pp. 6527–6532, Pat N. Confalone and Edward M. Huie. "The stabilized iminium ylide-olefin [3+2] cycloaddition reaction. Total synthesis of Sceletium alkaloid A4". In: Journal of the American Chemical Society 106.23 (Nov. 1984), pp. 7175–7178.

Griffin-Joullié observation⁴

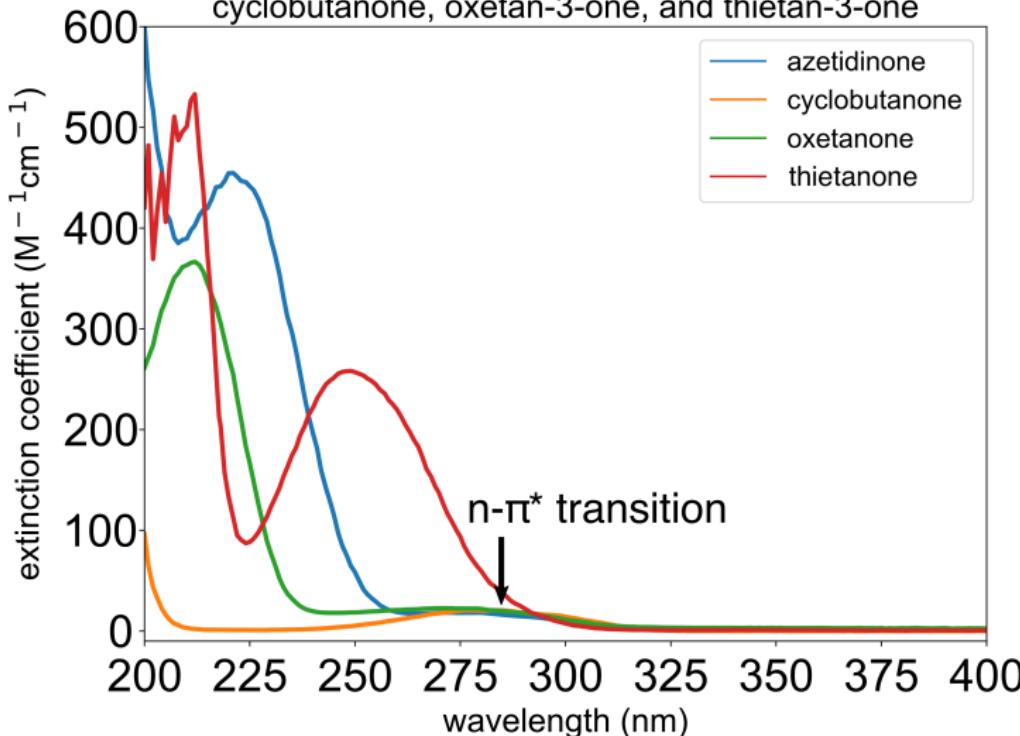


No synthetic application.

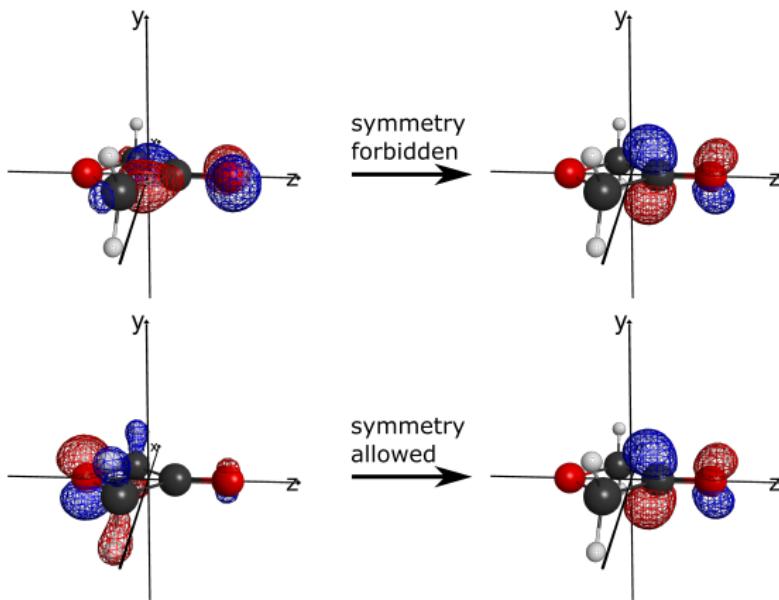
⁴J. P. Wasacz et al. "Photochemistry of 2,2,4,4-tetraphenyloxetan-3-one. Intermediates in the photofragmentation of aryl substituted oxiranes". en. In: The Journal of Organic Chemistry 41.3 (Feb. 1976), pp. 572–574.

$\pi^* \leftarrow n$ absorption

Molar extinction coefficient of azetidin-3-one,
cyclobutanone, oxetan-3-one, and thietan-3-one

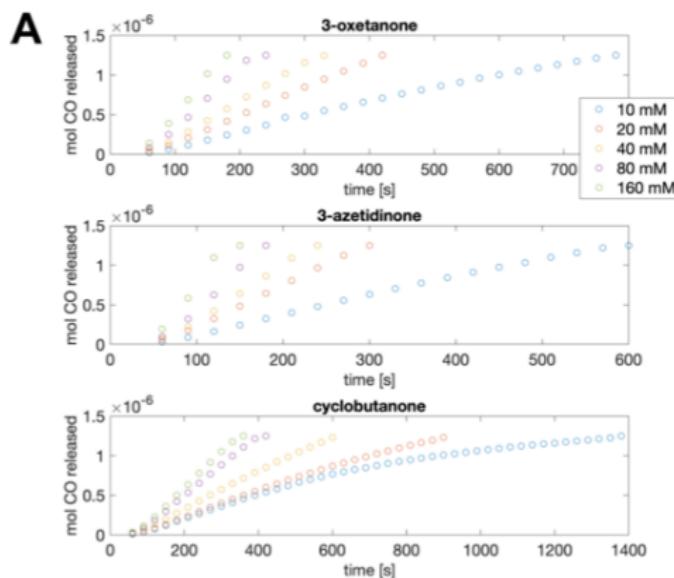


Symmetry controlled excitation

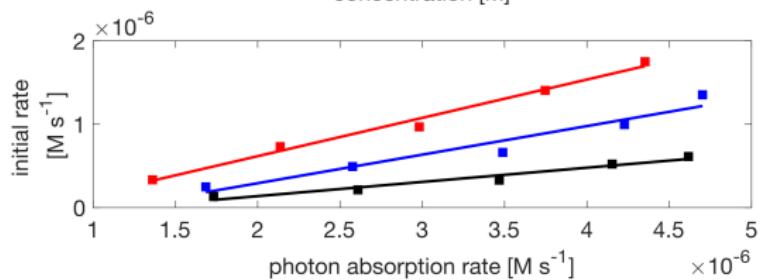
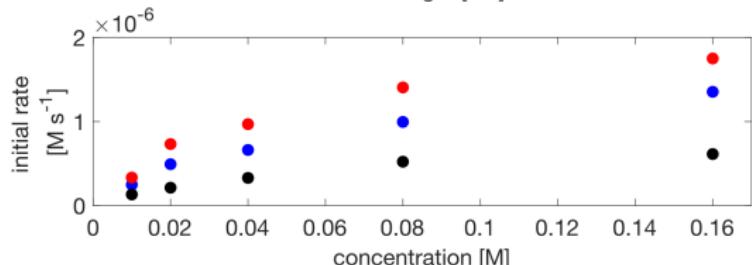
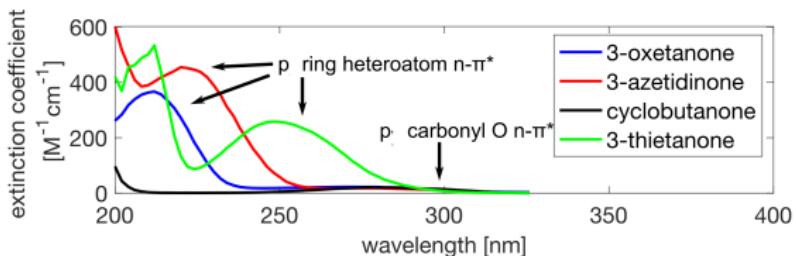


CO evolution

Convenient measure of reaction kinetics.

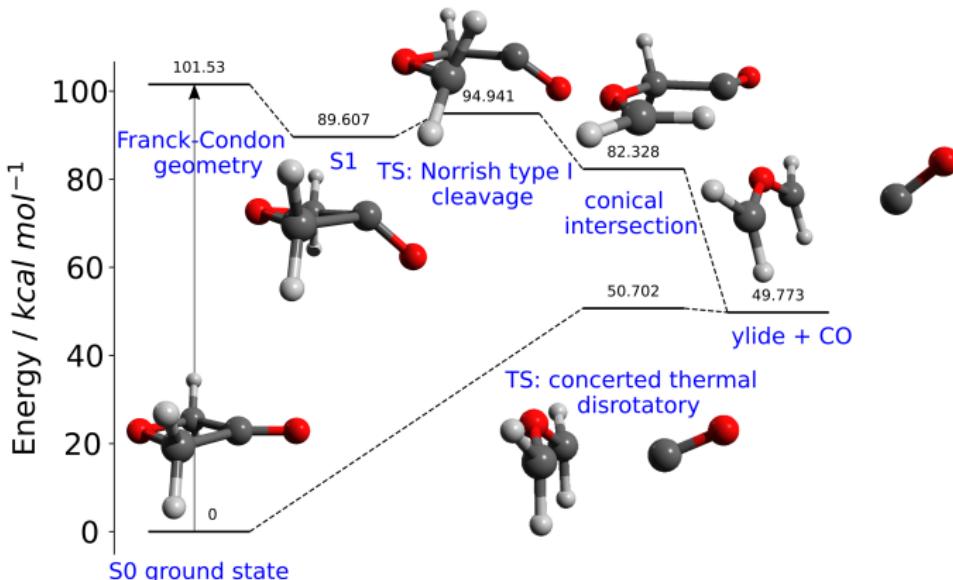


CO evolution



Model

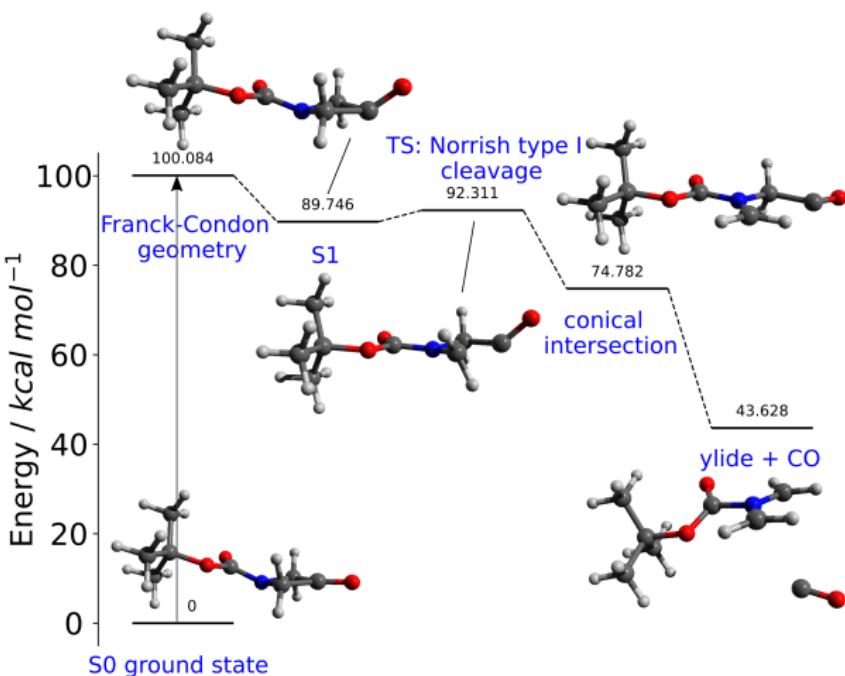
Carbonyl ylide formation



DFT for ground states B3LYP/6-311+G(2d,p)
TD-DFT for excited state energies; spin-flip for conical intersection

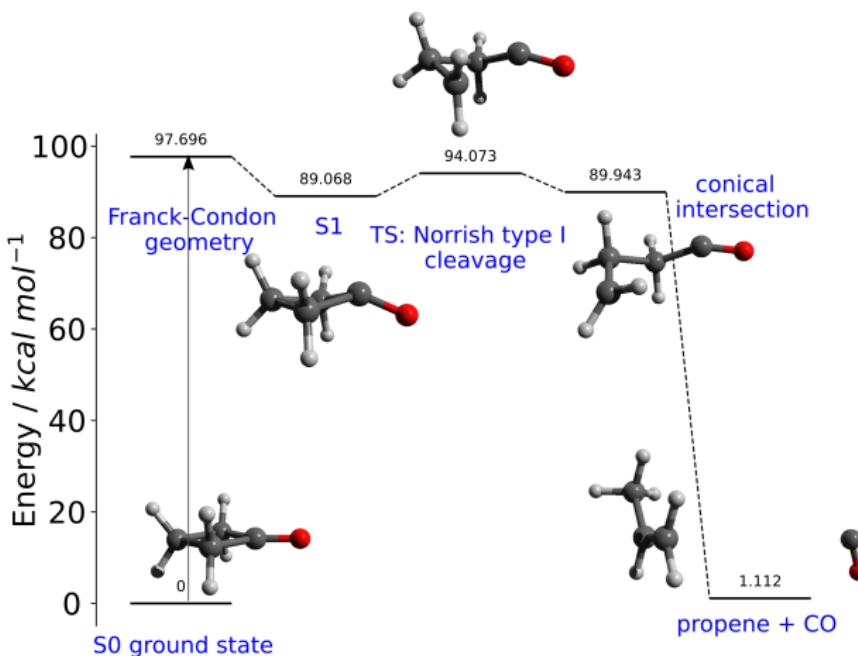
Model

Azomethine ylide formation

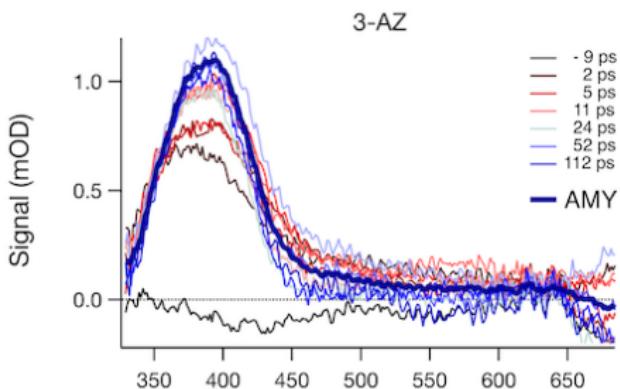
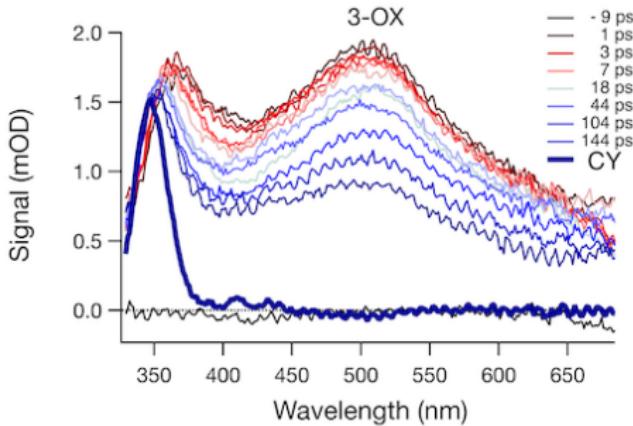


Model

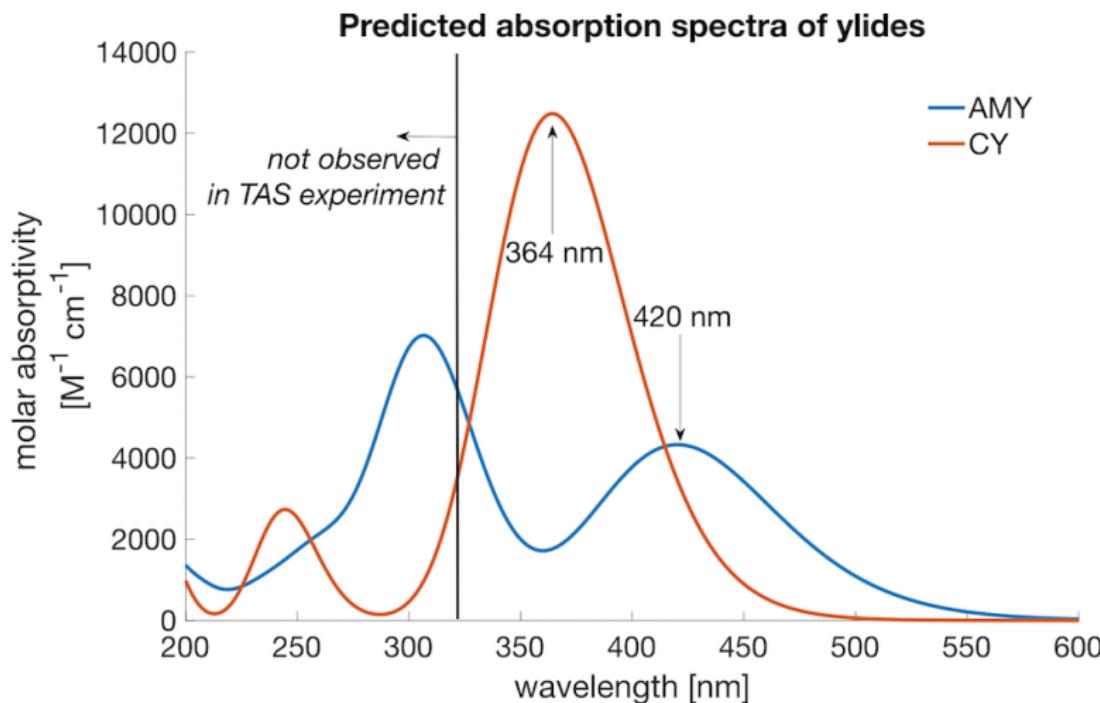
Cyclobutanone opening



Transient absorption spectroscopy

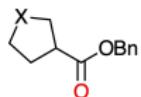
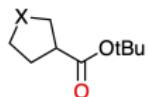
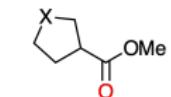


Transient absorption spectroscopy

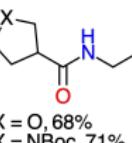
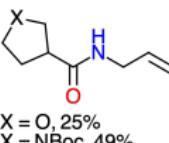
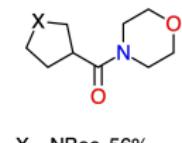
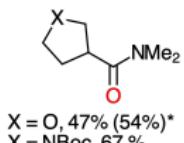


Scope

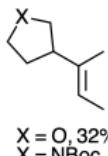
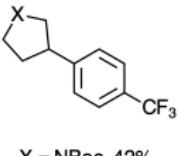
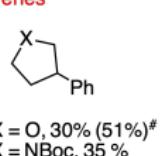
acrylates



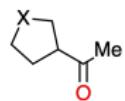
acrylamides



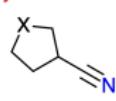
styrenes



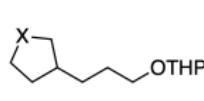
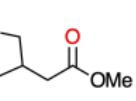
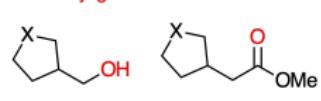
vinyl ketones



acrylonitriles

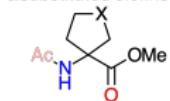


non-conjugated olefins

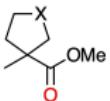


Scope

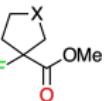
disubstituted olefins



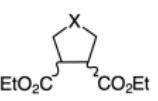
X = O, 48%
X = NBoc, 76%



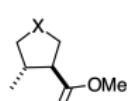
X = O, 48% (51%)*
X = NBoc, 75%



X = O, 40%
X = NBoc, 64%

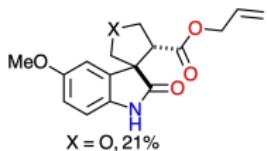


X = O, 53%
(trans:cis = 1:3)
X = NBoc, 75%
(trans:cis = 1:1)



X = O, 44%
X = NBoc, 79%

trisubstituted olefins

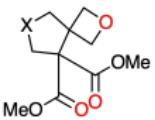


X = O, 21%



X = O, 34%

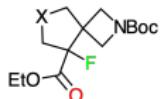
tetrasubstituted olefins



X = O, 42%
X = NBoc, 57%



X = O, 38% (49%)*
X = NBoc, 54% (61%)*

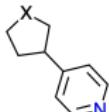


X = O, 65%
X = NBoc, 60%
(45%)*

heterocyclic olefins

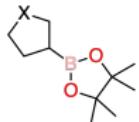


X = O, 20%
X = NBoc, 41%

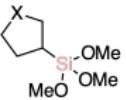


X = O, 20%
X = NH, 41%

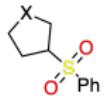
vinyl heteroatoms



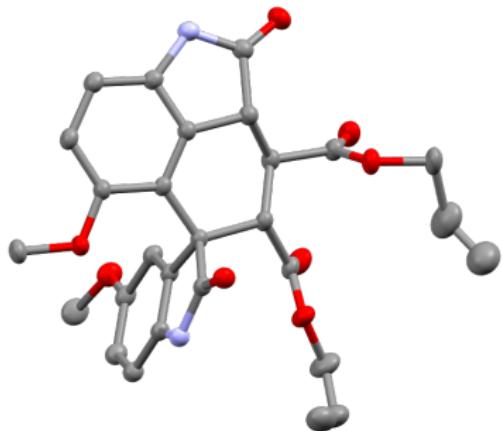
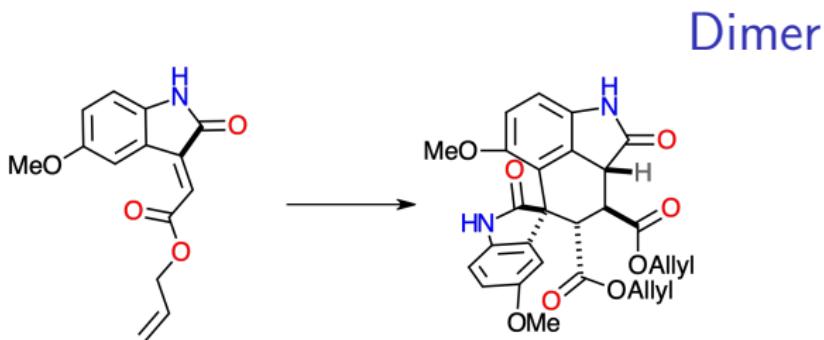
X = O, 50% (71%)*
X = Nboc, 66%



X = O, 44%
X = Nboc, 47%

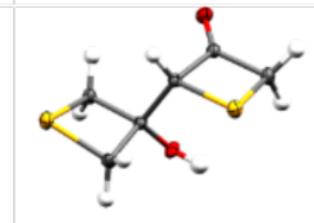
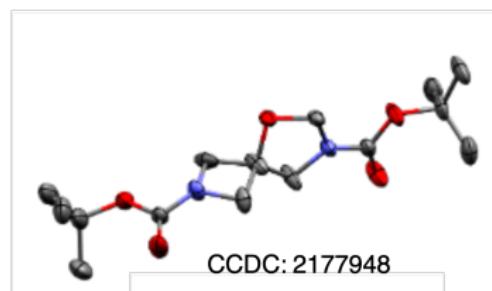
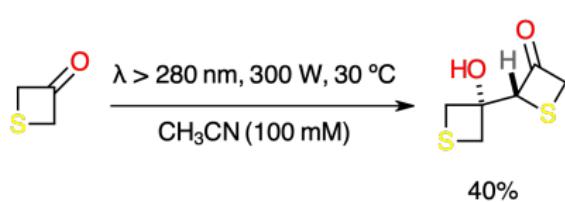
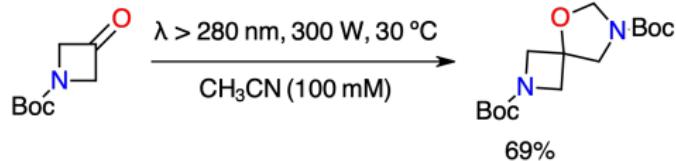


X = O, 58%
X = Nboc, 72%



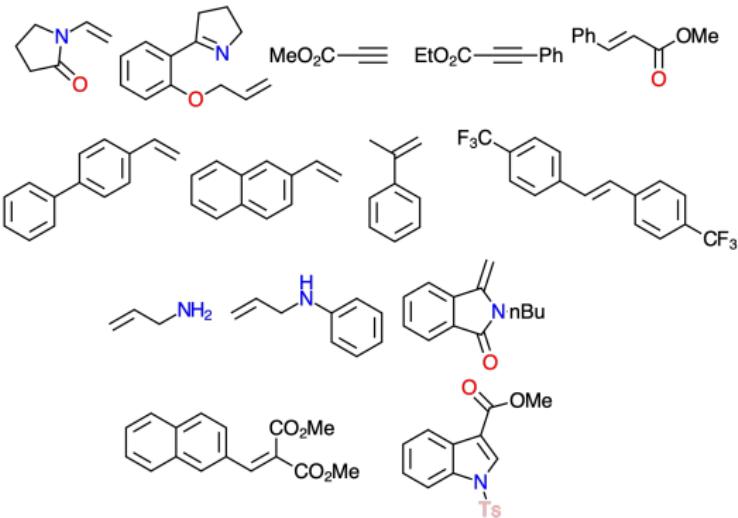
CCDC: 2193412

Homo-coupling

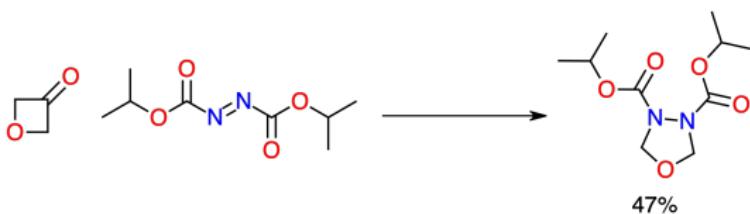
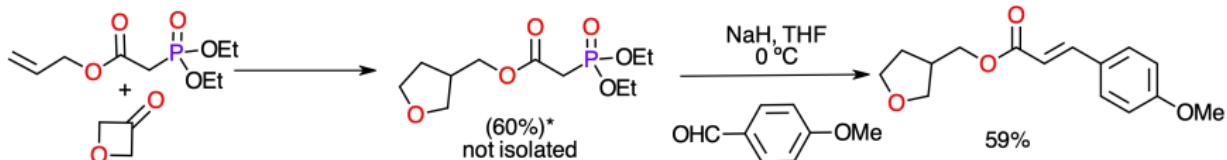
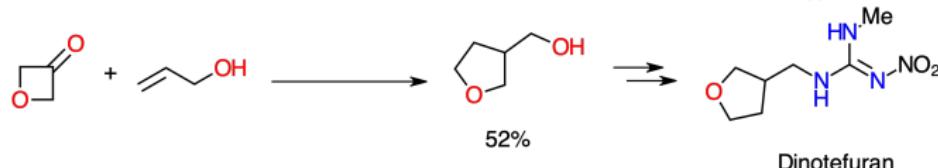
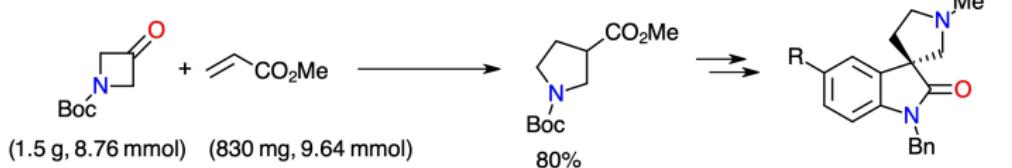


Anti-scope

Substrates that did not give the expected products.

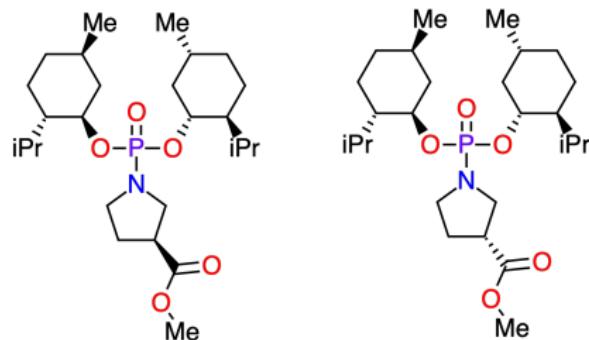
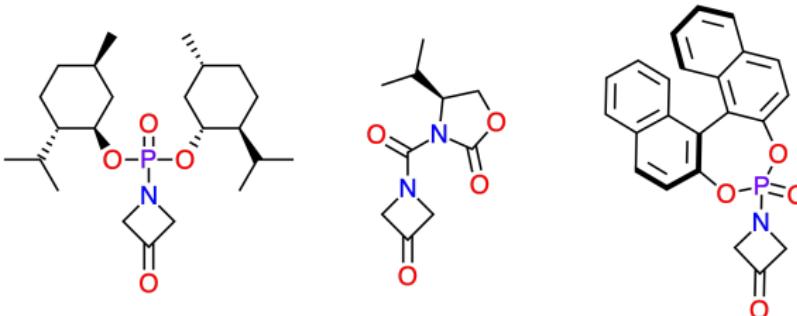


Use



Chiral auxiliary

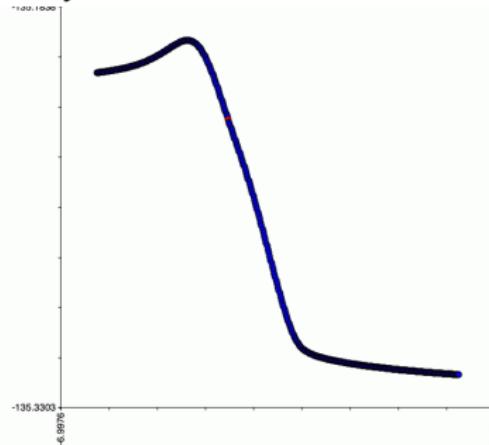
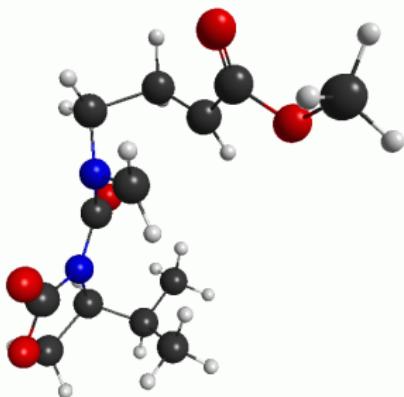
Diastereoselective β -proline synthesis.



Can we achieve facial selectivity?

Early transition state.

About 7 kcal/mol activation energy from ylide.



Lab goals

- Experimentally access new chemical space through the “upside-down” world of photochemistry
- Leverage measurements and quantitative models to better understand and improve synthetic chemistry
- Connect structures of newly made molecules to their biological effects in complex living systems

Credits

Current group:

Manvendra Singh

Graduate student

Bryce Gaskins

Undergraduate student

Mauricio Bahena Garcia

Graduate student

Victor Fadare

Graduate student

Alhamza Hamza

PharmD student

Elizabeth Miller

Undergraduate student

Collaborators:

Elles lab KU (Transient absorption spectroscopy)

Funding:

Chemical Biology of Infectious Disease COBRE pilot project, General Research Fund (KU), KU Cancer Center pilot project, XSEDE/ACCESS NSF computational resource.

Former group members:

Zach Pearson (UG), Sri Kolluru (PDF), Pawan Dhote (PDF), Nathan Garza (Tech), Amar Kumar (GRA), Koki Takemoto (GRA), Matt McCurry (GRA), Vishva Shah (UG), Cybelle Arrey (PharmD), Ambrosee Wilkinson (UG).