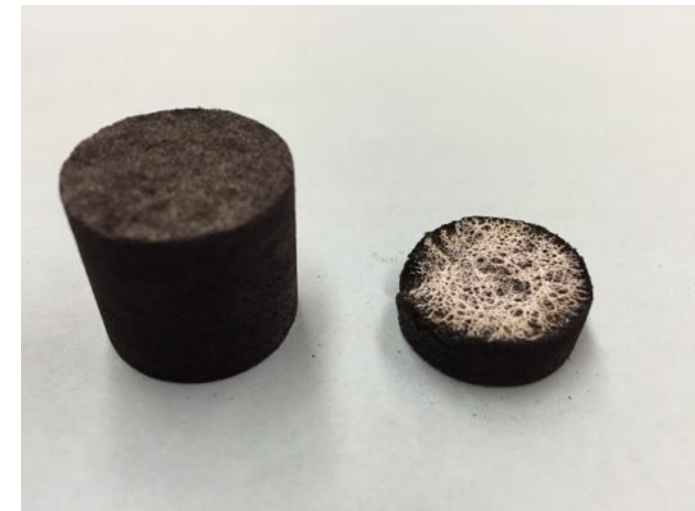
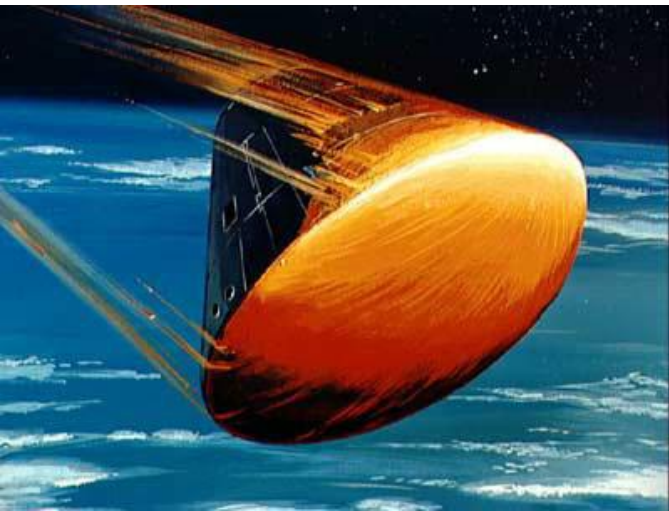




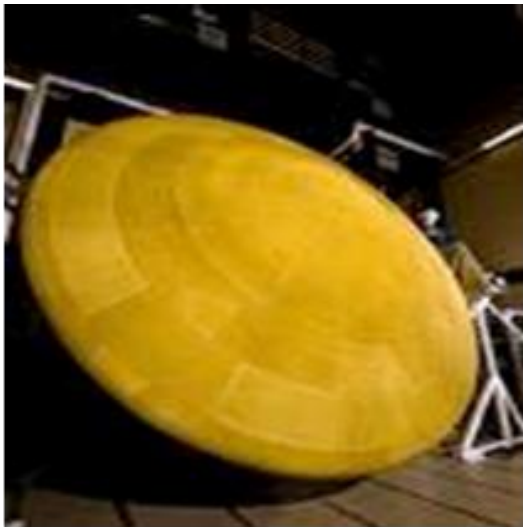
## Collecting Experimental Data for the Modeling of Reactive Gas and Pyrolysis Interactions with Hot Carbon Chars.

David Thorne: REU Program, SRI International

8/15/2016



Ablative thermal protection system (TPS) materials are required for the extreme heating encountered during entry into planetary atmospheres.



Mars Science Laboratory  
(MSL), NASA



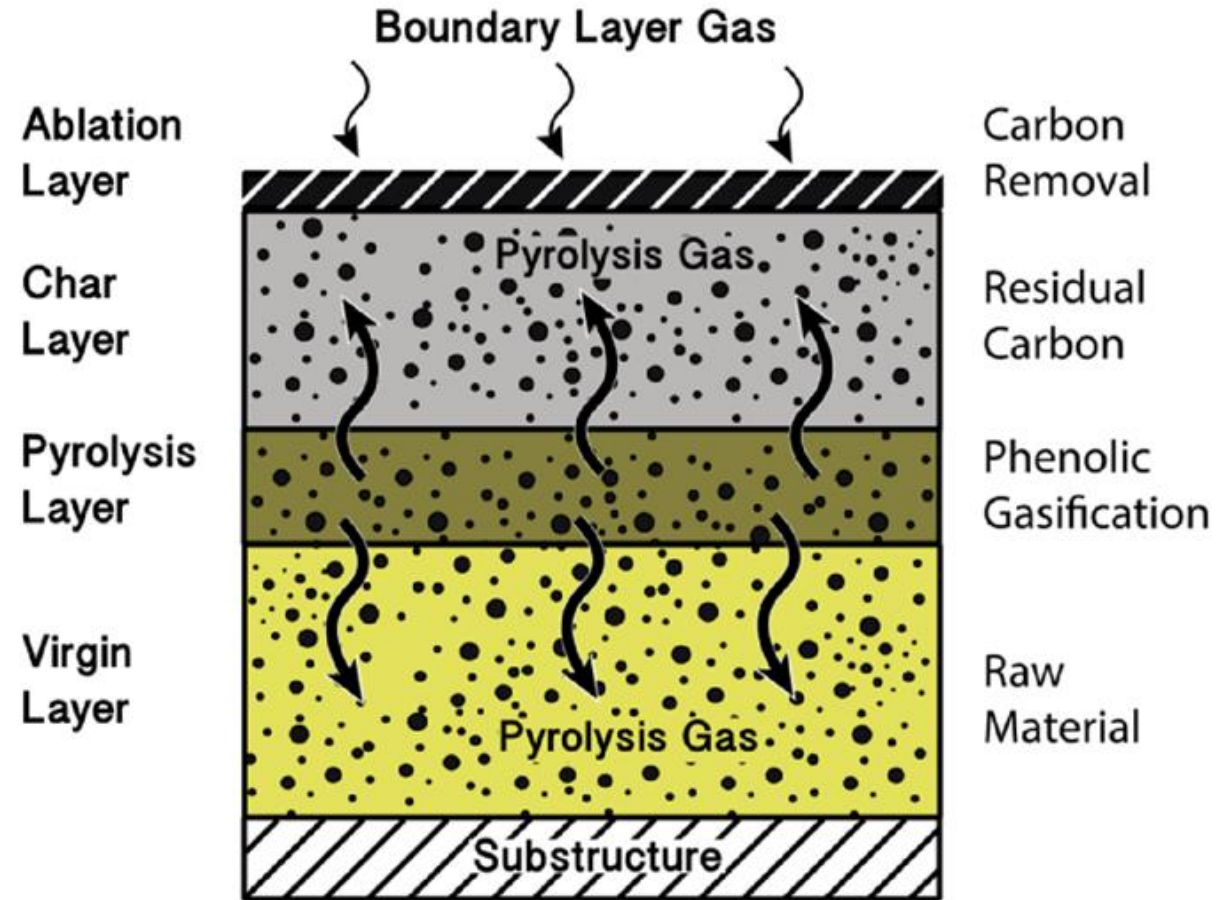
Stardust, NASA



Dragon Capsule, SpaceX

# The Ablation Process

- A boundary layer is formed from gaseous reaction products within the heat shield material and provides protection against convective heat.
- The outer surface of the material chars, while the bulk of the TPS material undergoes pyrolysis and expels product gases.

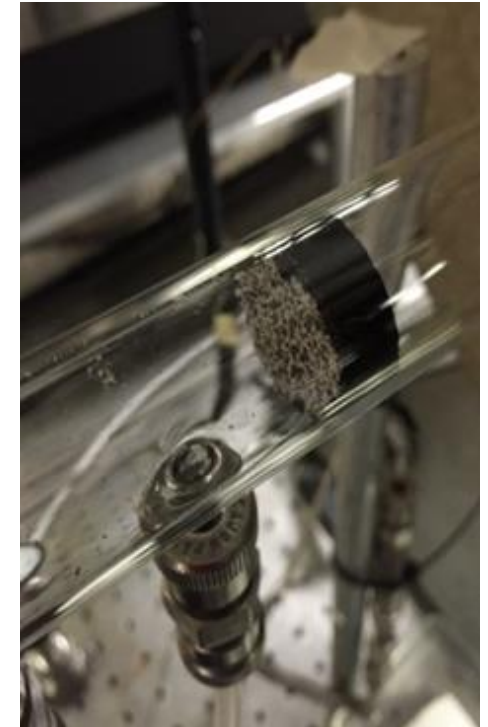
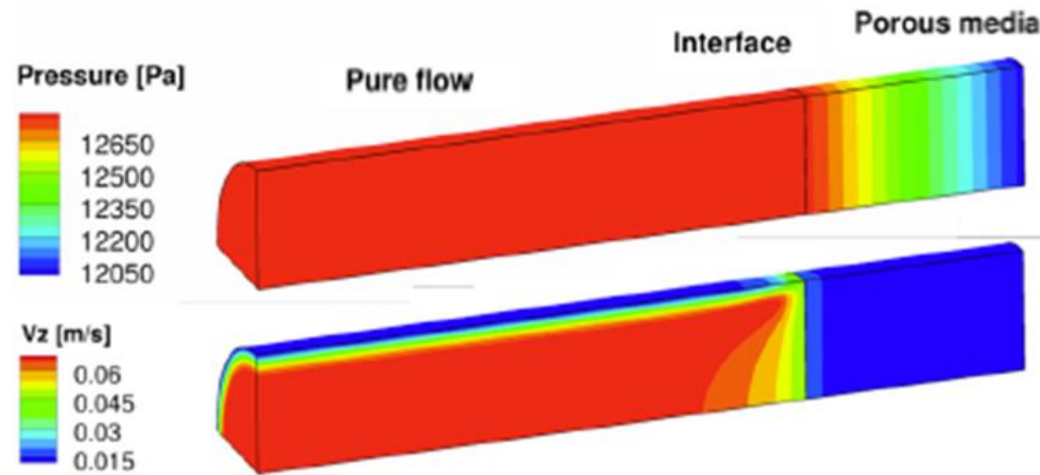


# Why is Research Involving Ablative Materials Desired?

- Modern design of heatshields is dependent on experiments performed in the 1960's;
- The physics and chemistry of ablation models are dependent on a limited set of measurements that were based on outdated techniques.
- Tend to overestimate ablation rates, calling for more material than necessary.
- The ablation community has expressed the need for modern data on the interactions between atmospheric gases, virgin carbon material, pyrolysis gases, and hot carbon char.
- Research pertaining to the ablative materials used for TPSs, remains one of the main focal points of NASA's technology development.
- Project Purpose: We are collecting modern experimental data, used to evaluate models currently used in ablation simulations, and develop new ones

# Type of Data Collected

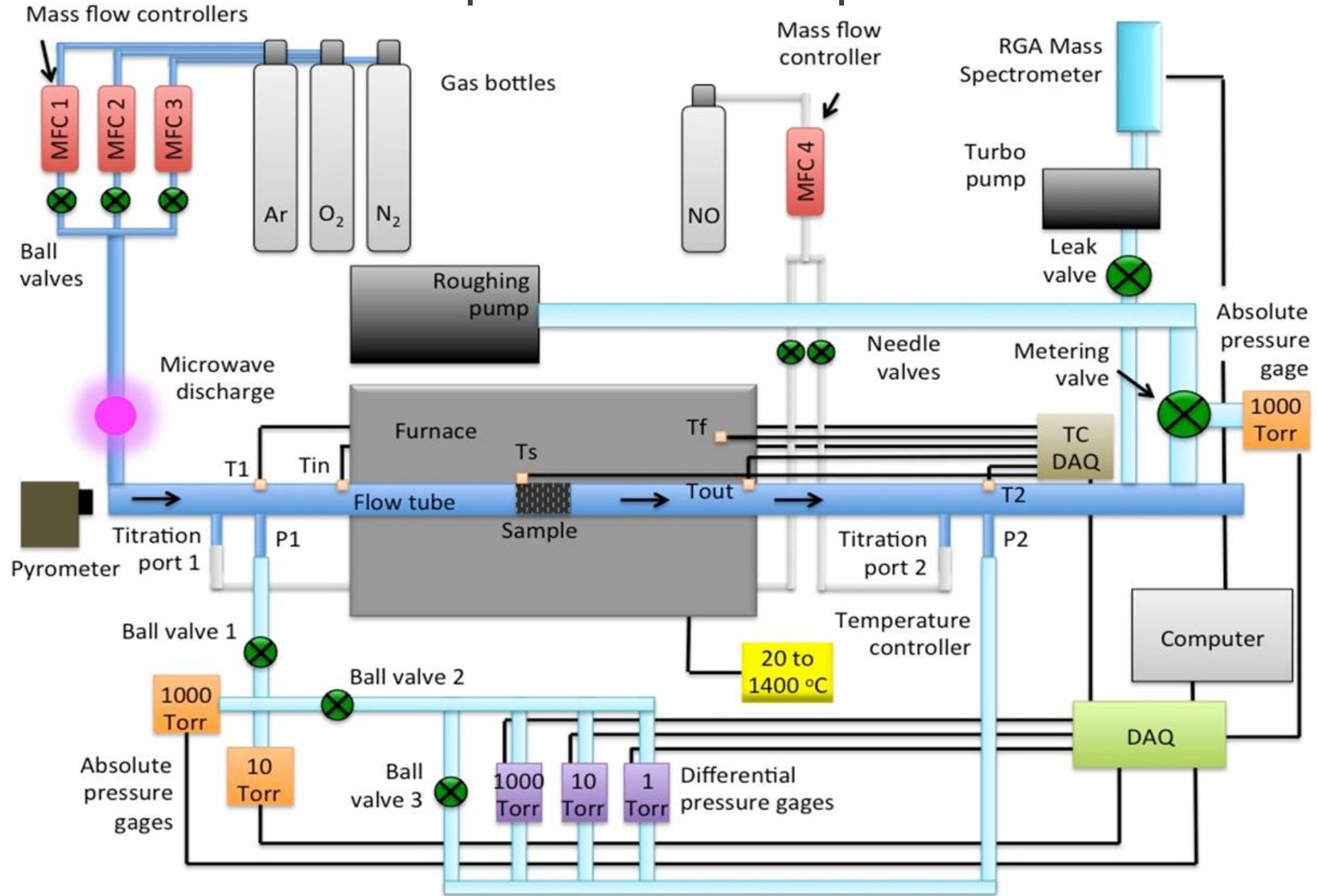
- Very simple experimental setup that can be easily simulated using fluid dynamics and material response codes
- Collect temperature, pressure, mass flow, and reaction product data



- KATS, Kentucky Aerothermodynamics and Thermal Response System, Combines fluid dynamics and material response codes in order to simulate chemical interactions with porous hot carbon chars.



# Experimental Setup



# Lab Equipment

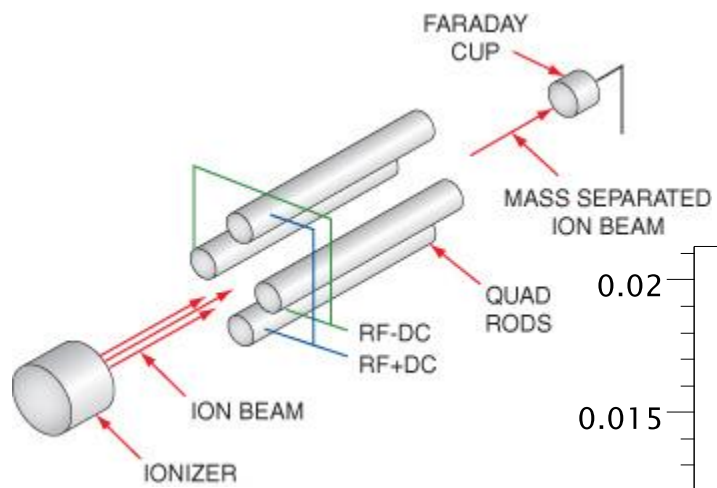
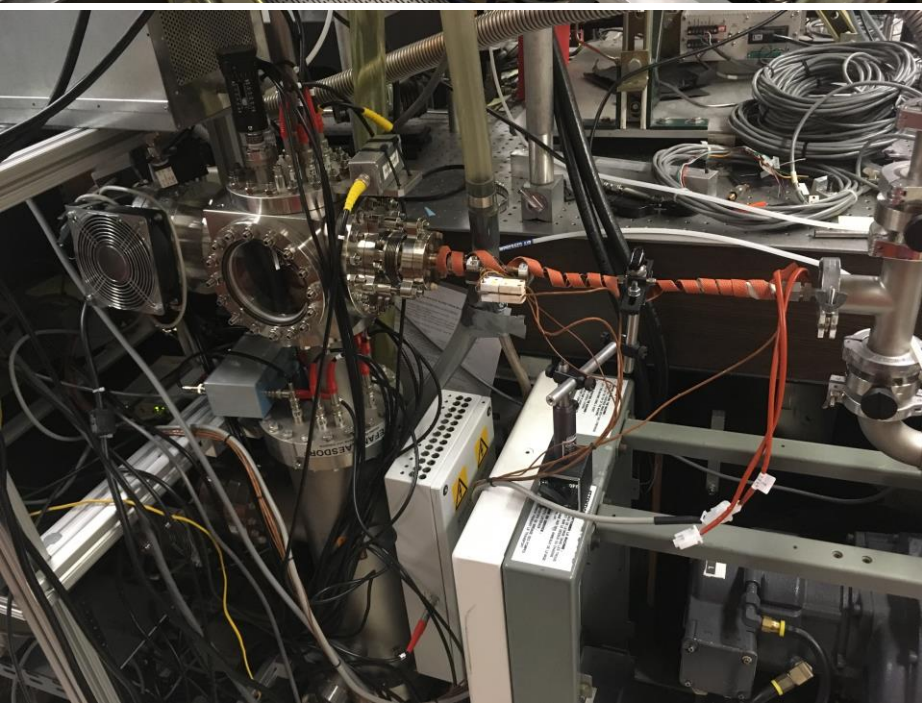
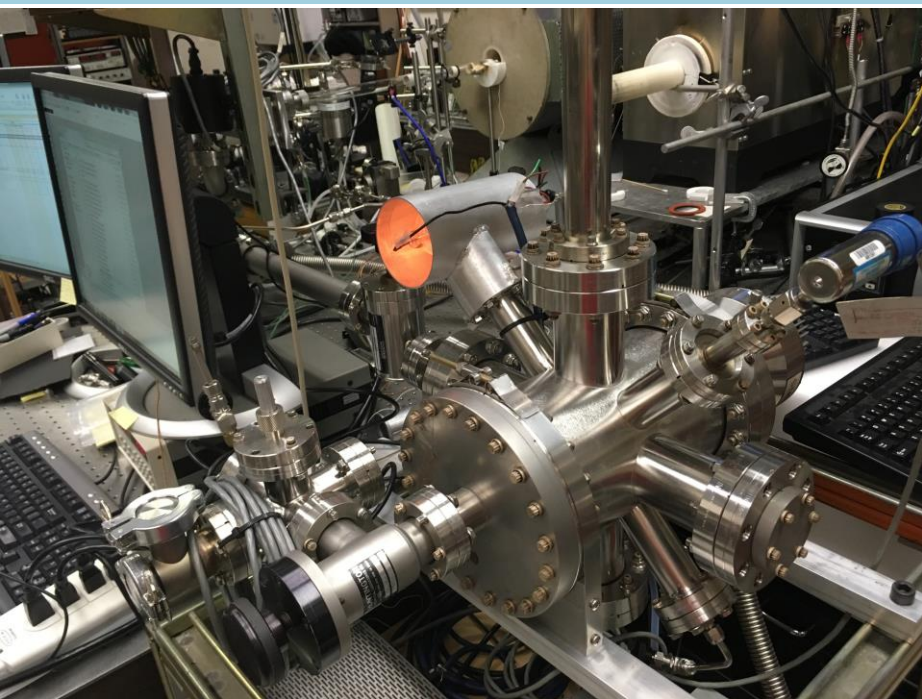


FiberForm, manufactured by Fiber Materials, Inc.

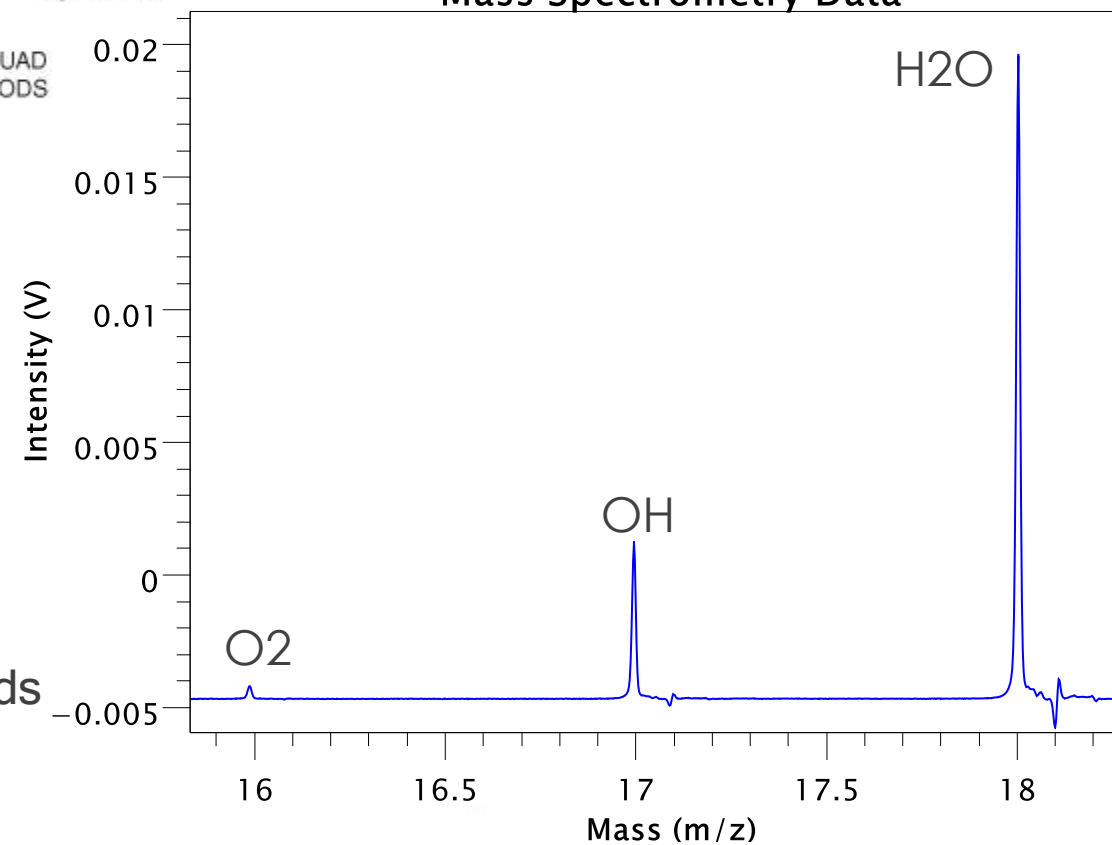
- A rigid carbon fiber composite used as preform for the carbon/phenolic ablators PICA



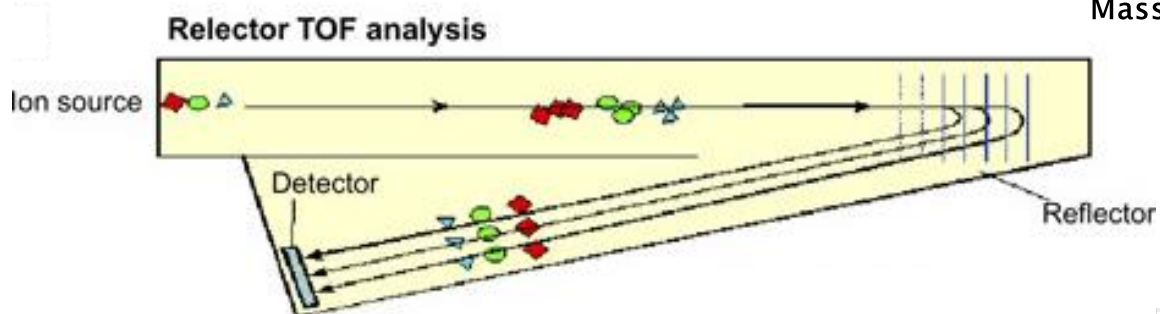




## Mass Spectrometry Data

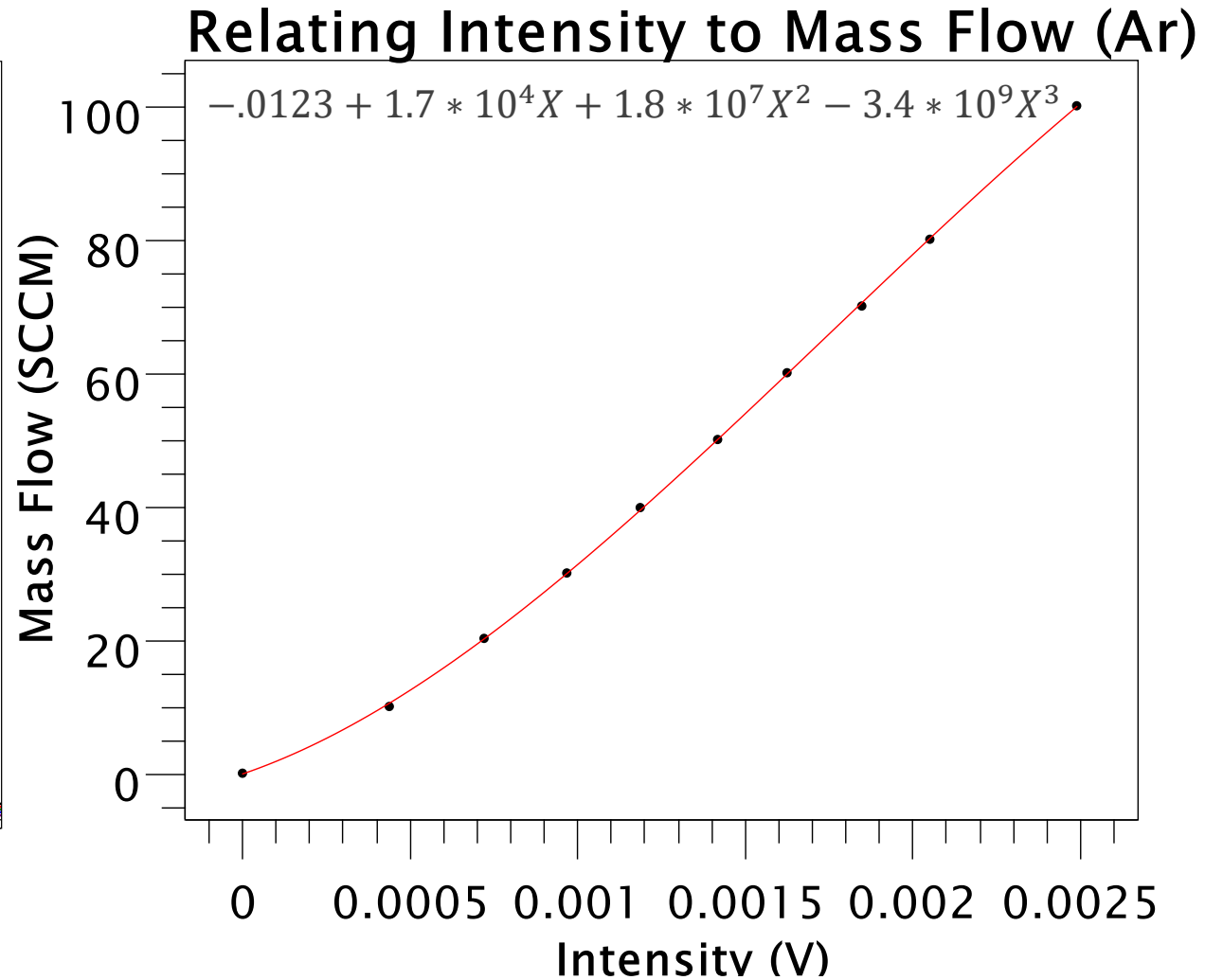
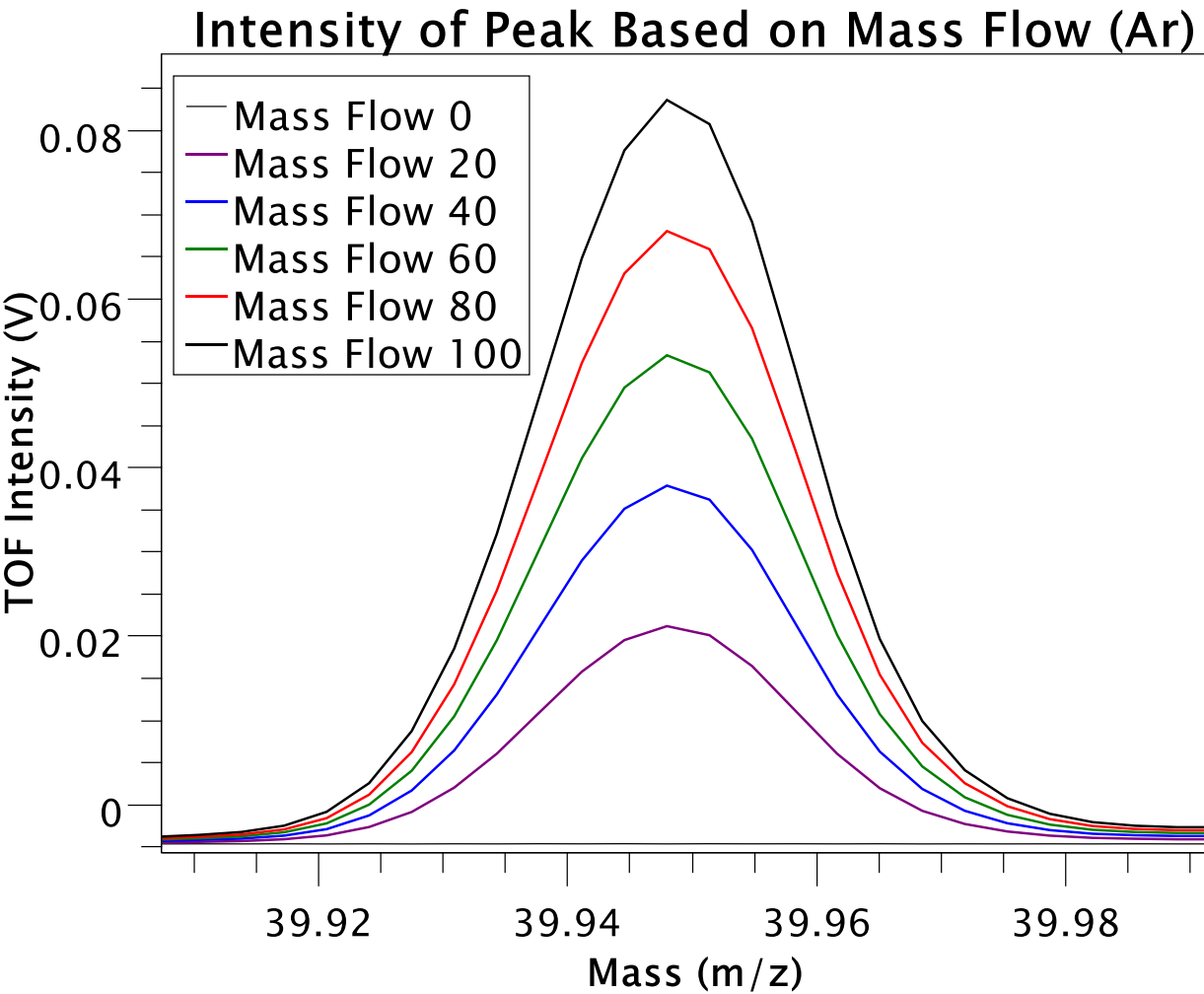


- Long flight path yields high resolution





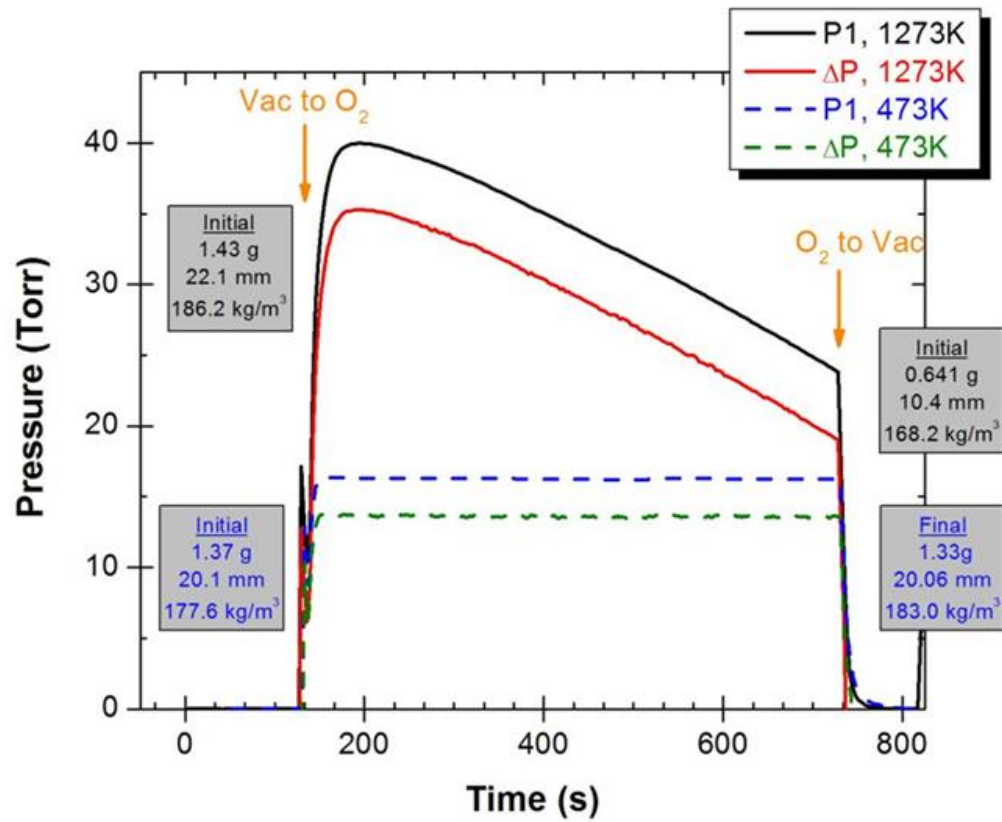
## Quantification



# Overview of Experiments

- Argon and Helium experiments were performed in order to obtain mass loss not associated with reactions.
- O<sub>2</sub> is very reactive with carbon, easy to model, and common in planetary atmospheres. O<sub>2</sub> and pyrolysis gasses are the core of the experiments.
- CO and CO<sub>2</sub> are products of the carbon-oxygen reaction, and O is present in O<sub>2</sub>, so individual experiments are required for each.
- A titration is required to produce O atoms ( $\text{N} + \text{NO} \rightarrow \text{N}_2 + \text{O}$ ). So N and NO must all have individual experiments.

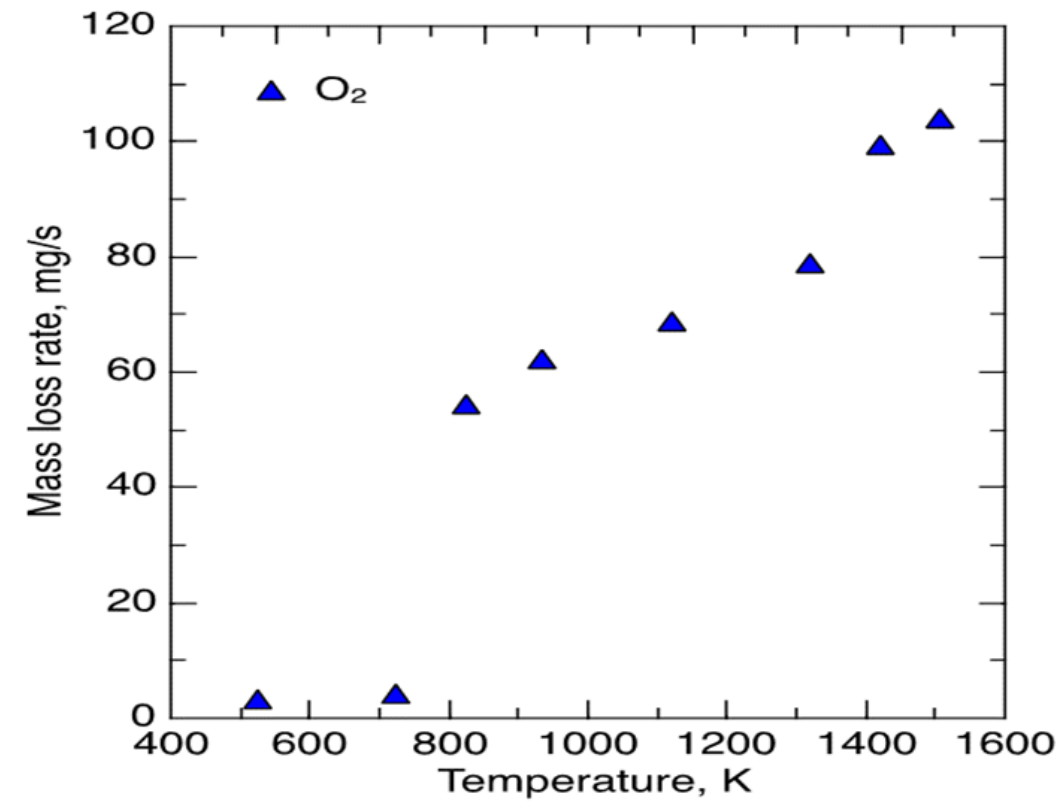
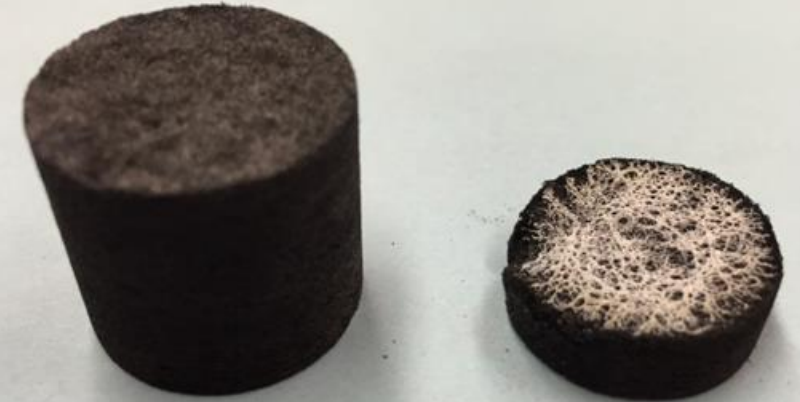
## O<sub>2</sub> Experiments



Upstream Pressure drop  
due to recession of plug

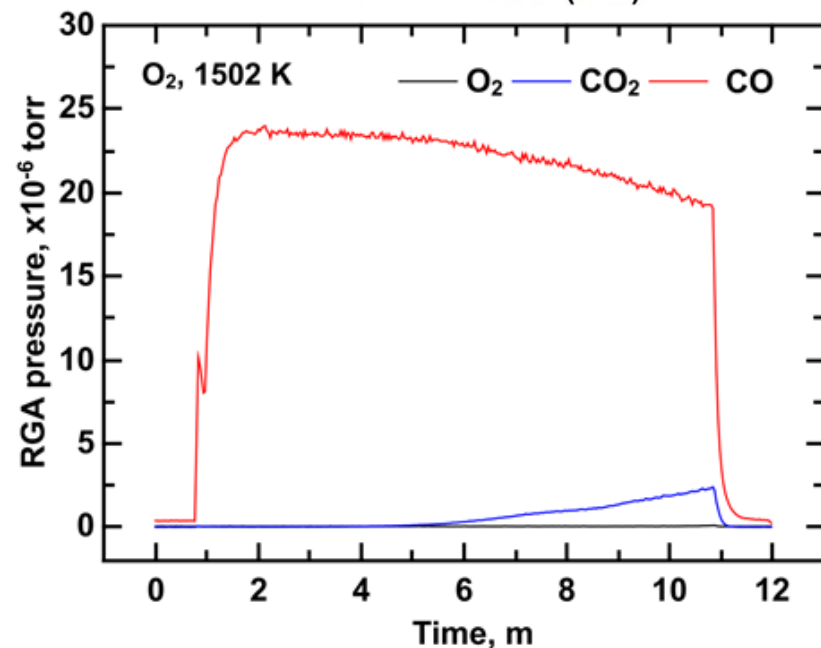
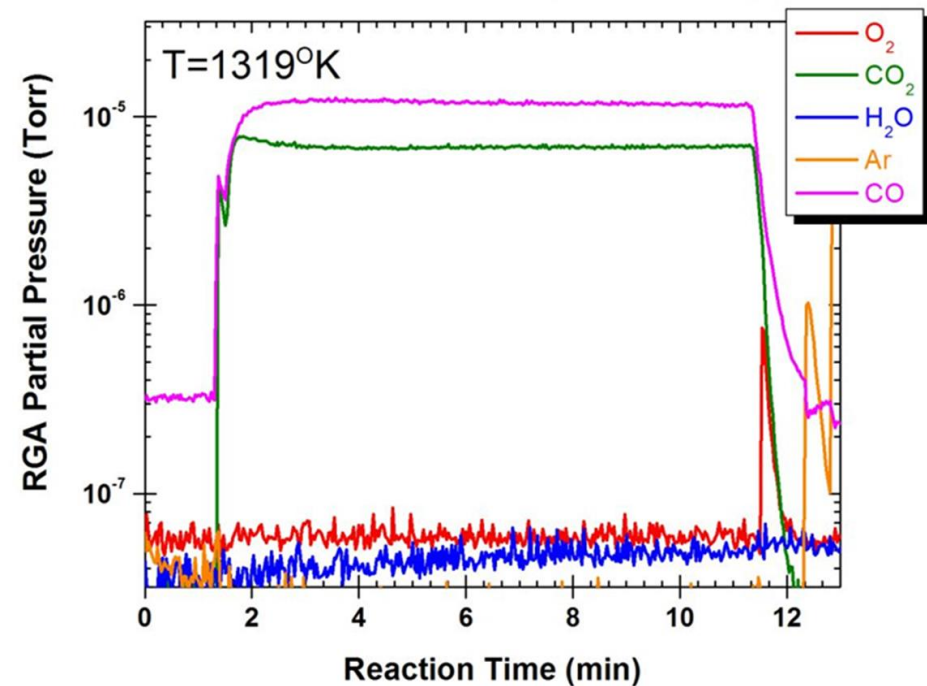
Mass loss is temperature dependent

Whitish material attributed to Ca residue



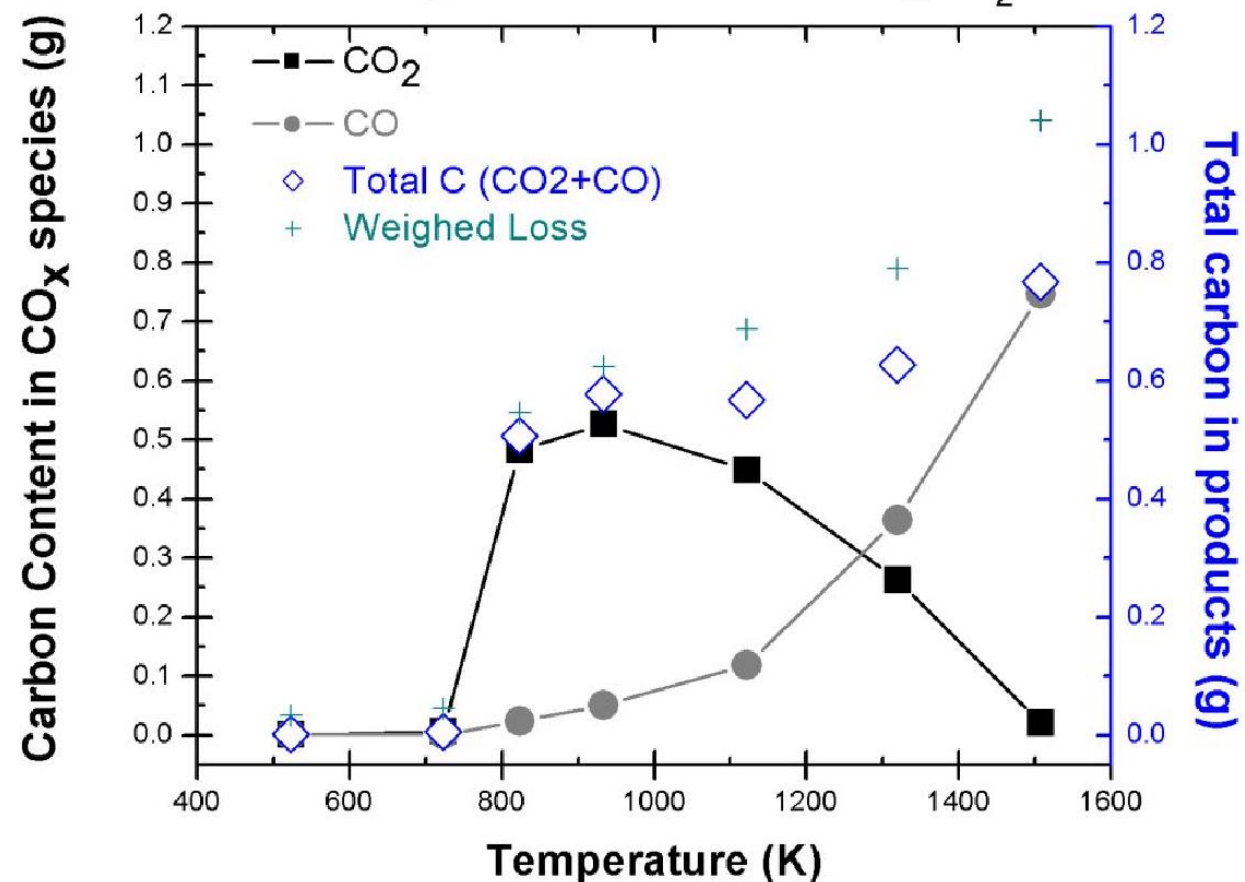


## FiberForm Oxidation, Run OxO2\_09



## O<sub>2</sub> Experiments

### CO and CO<sub>2</sub> Generation during O<sub>2</sub> oxidation



*Boudouard Reaction*  
 $2\text{CO} \rightleftharpoons \text{CO}_2 + \text{C}$

## NO Experiments

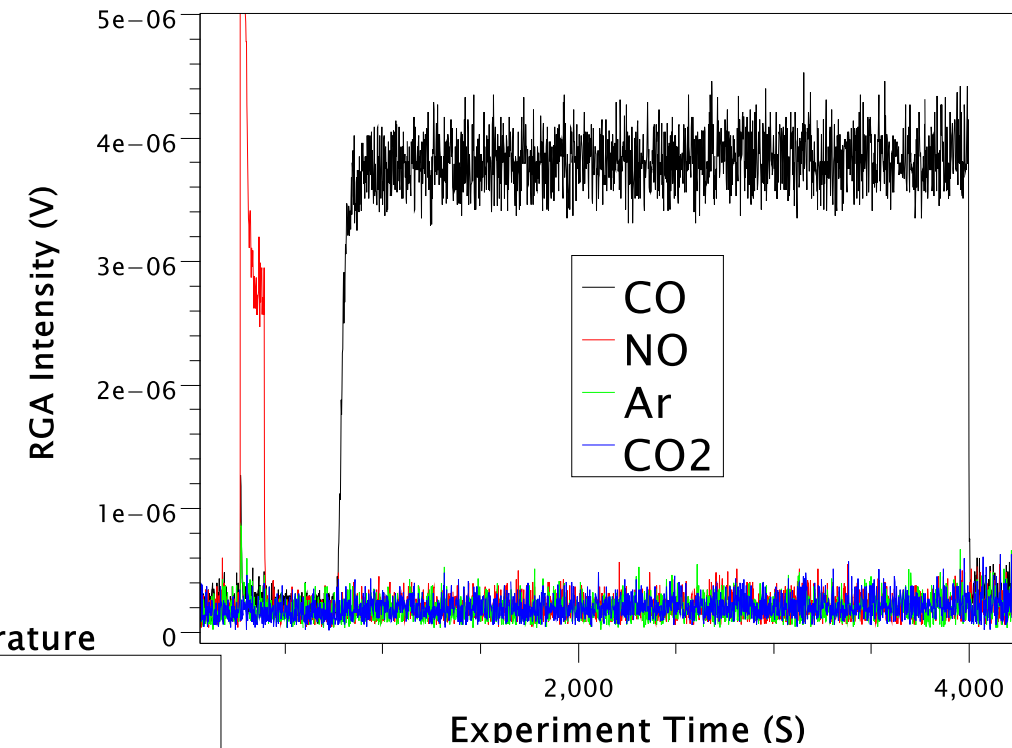
- Titration:  $\text{N} + \text{NO} \rightarrow \text{N}_2 + \text{O}$
- Products due to O must be distinguished from those due to NO;  
NO's chemical interaction with the plug must be determined
- Initial experiments were the standard 10 minutes long
- Mass loss was very small, increase sensitivity to measurement by increasing experiment time

	Temperature (C)	Mass Loss (%)
NO	1250	2.8
O2	1200	58.5
Ar/N2	1200	~1.9

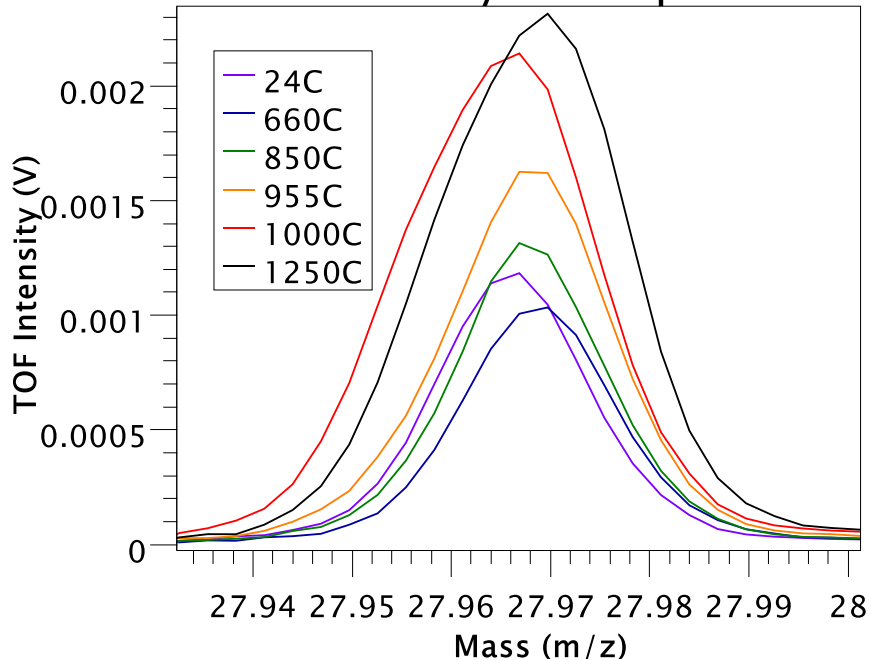
# NO hour series

- Received 12% mass loss at highest temperature
- Products were not detected until above 800C, mostly CO, and a small amount of CO<sub>2</sub>
- $\text{C} + \text{NO} \rightarrow \text{CO} + \frac{1}{2}\text{N}_2$

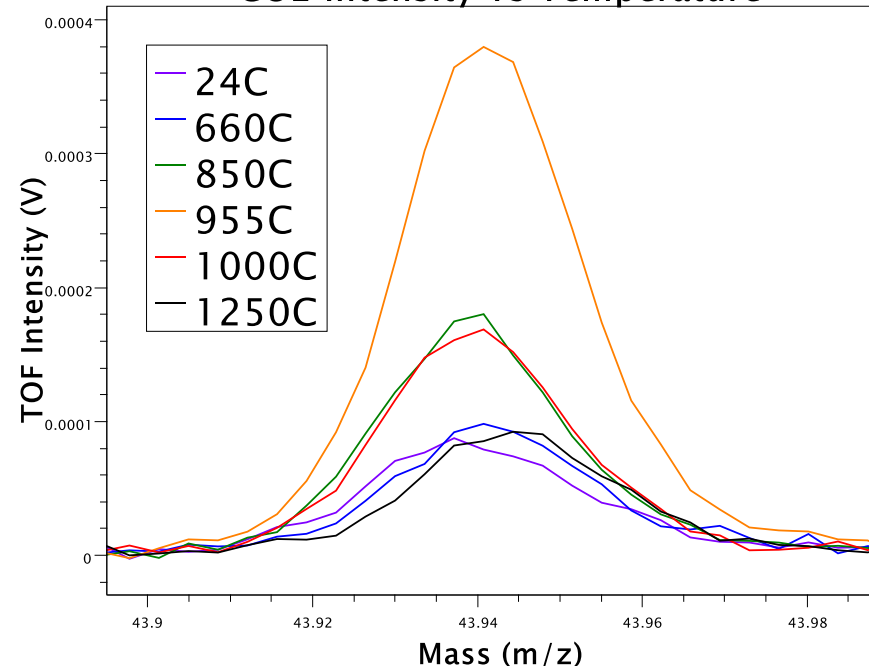
NO+Ar at 1250C



CO Intensity Vs Temperature



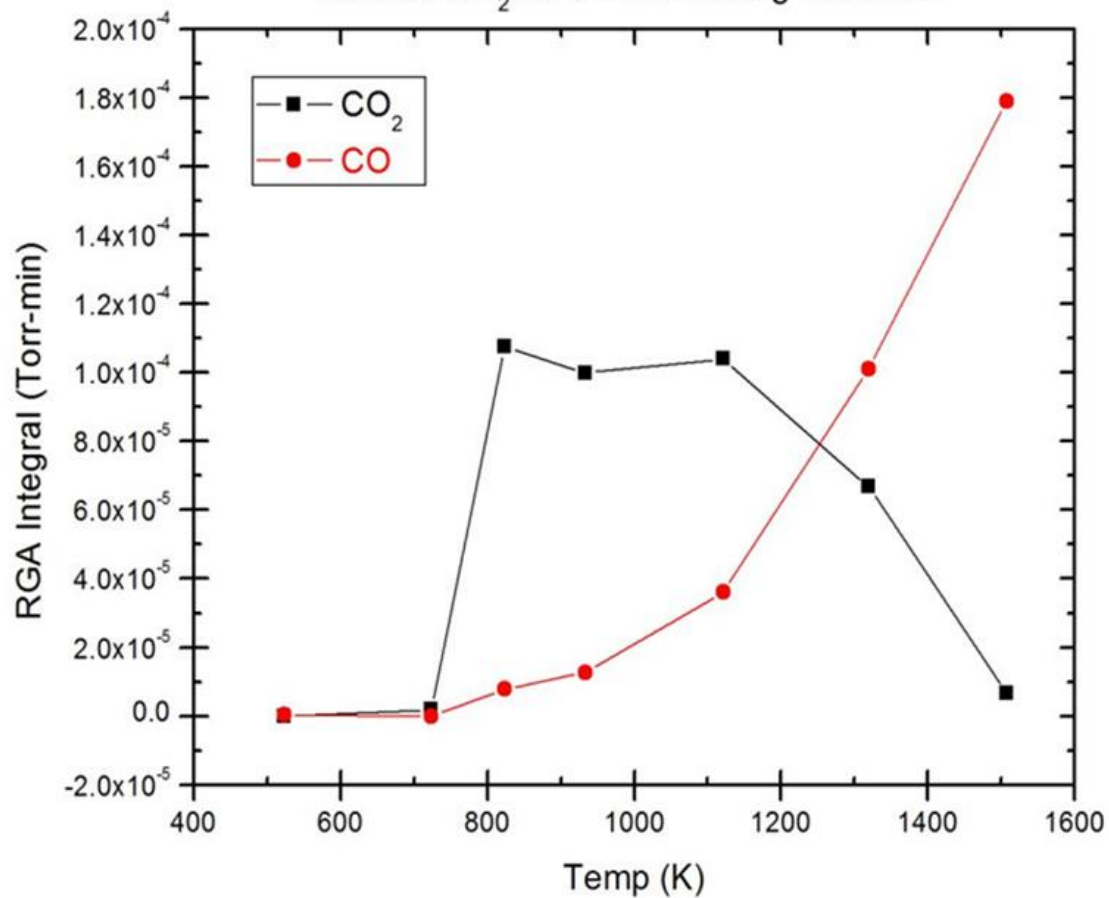
CO<sub>2</sub> Intensity Vs Temperature



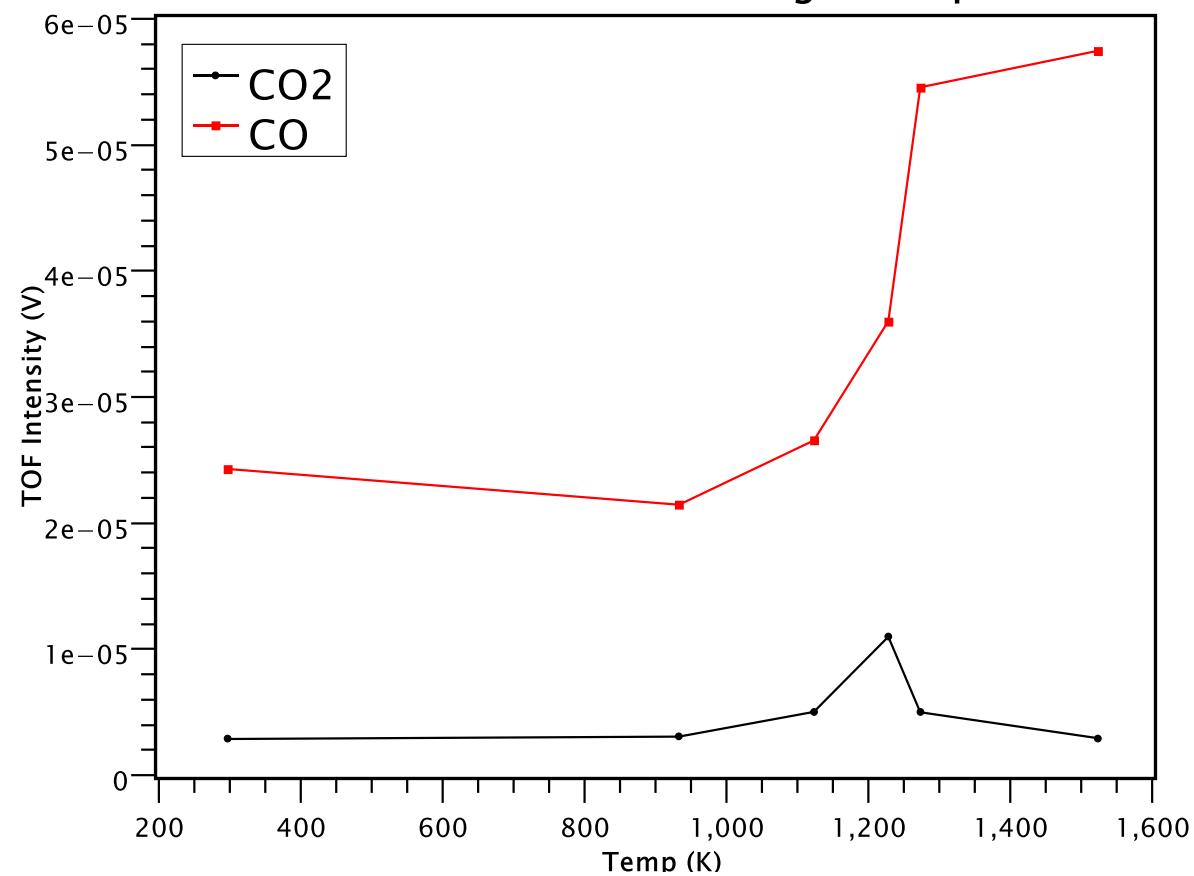


## NO hour series

CO and CO<sub>2</sub> Generation during Oxidation



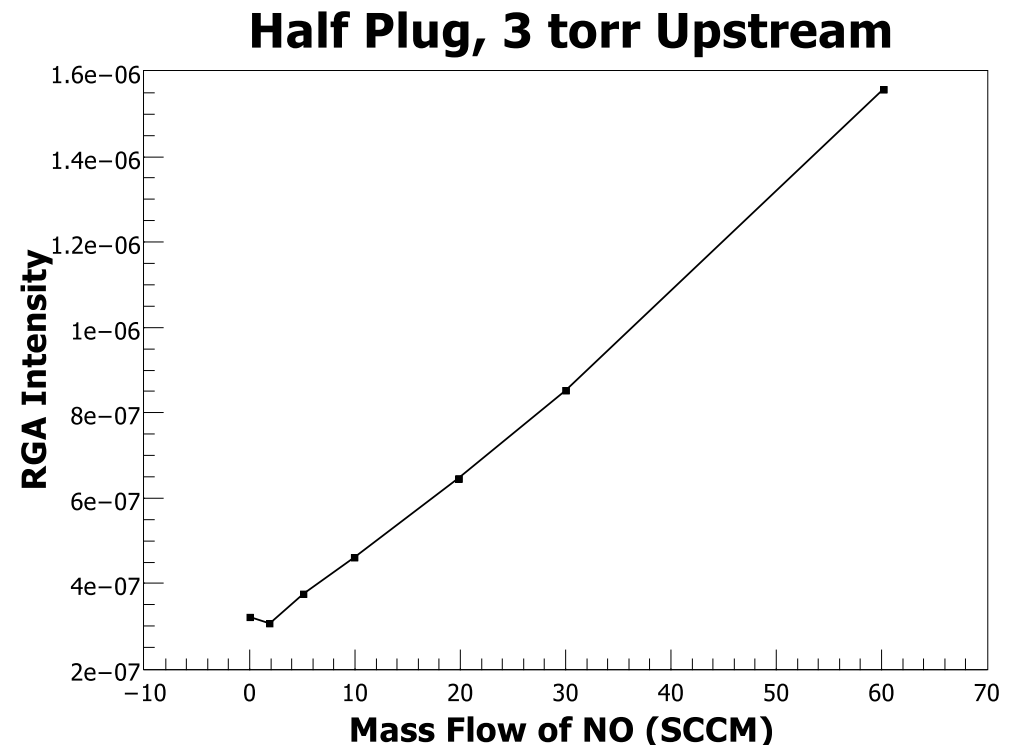
CO and CO<sub>2</sub> Generation During NO Experiments



## O Experiments

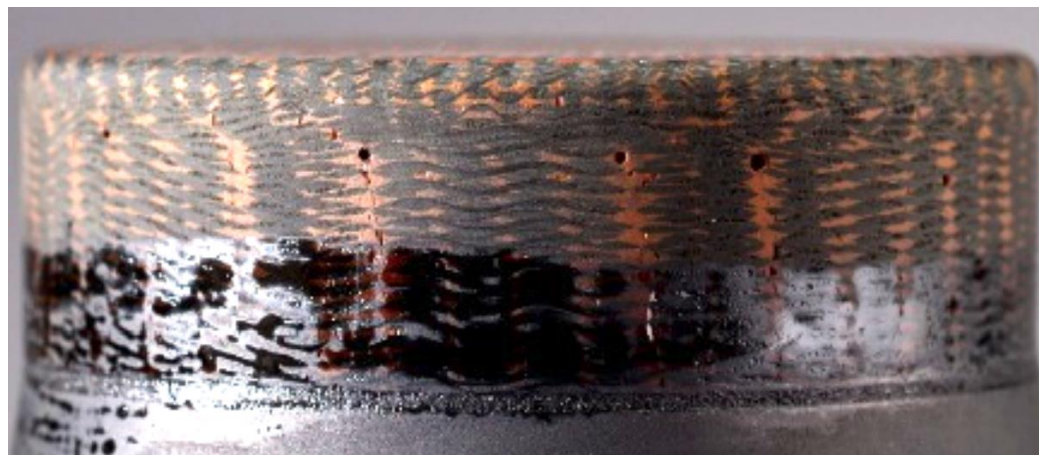
- No reaction with the plug was observed during discharge experiments
- High pressures may cause recombination O atoms before reaching plug surface
- Upstream pressure was lowered using a half plug/in-plane plug, still no reaction

- It was shown that at 10 torr, no plug, NO intensity remained constant during titration
- At 3 torr, no plug, N was shown to combine with NO (no increase in NO signal) during titration
- Titration still showed no reaction at 3 torr.



# Future Experiments

- Improve O experiments
- Experiment with pyrolysis gases (phenol, benzene, alcohols, hydrogen, were all produced during pyrolysis experiments of carbon/phenolic material here at SRI)
- Perform experiments with PICA instead of Fiberform, as well as with conformal ablative materials.



← Woven Ablative Material





## Acknowledgments

REU Mentor: Jason White

REU Program Manager: Sanhita Dixit

Partners in the lab: Thomas Cochell, UK  
and Francesco Panerai, NASA

Thank you

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