NB: The graded, first version of the report must be returned if you hand in a second time!

HA2: Shape of aluminium nanoparticles

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Task Nº ॒	Points	Avail. points
Σ		

Introduction

The properties of nanoparticles differ immensely from the equivalents in bulk materials. However, the properties are not only dependent on the size of the particles; their shape turns out to play an important role as well. Usually the shape can experimentally be controlled by the adjustment of various parameters during the synthesis. In that way non-equilibrated (from a thermodynamic point of view) particles can be made by e.g. applying stabilizing organic molecules onto the surface. However, to predict the thermodynamic shape (corresponding to the one with the lowest energy for a given number of particles) the so-called Wulff construction may be used. In this construction the surface energies of the nanoparticle's various facets are essential[1]. This study computationally investigates such Wulff constructions for large aluminium nanoparticles by applying density functional theory (DFT), with a restriction to the (111) and the (100) facets. The goal is to calculate the most stable shape of these nanoparticles and examine how external conditions (such as molecules in gas phase) affect the shape.

Computational method

DFT is in in principle exact. However, the exact form of the exchange-correlation potential is not known and hence various approximations are made to solve specific problems. The approximations can become increasingly intricate and elaborate, often with accompanying increase in the computational complexity and cost. A commonly used approximation is to treat the exchange-correlation potential as a second degree Taylor potential in the electron density. !! This violates properties of the XC potential which are known to be true. How and why can we use this then!!

There are many different GGA functionals an no single one is optimal for all applications. For metals, the PBE, or simply the LSDA functional will capture the relvant physical properties of the system as the density does not vary rapidly [].

Atop of xc-functionals, an appropriate basis set needs to be specified for the problem at hand. Commonly used basis sets are linnear combination of atomic oribitals (LCAO), a set of plane waves, and real space grids. A difficulty with LCAO is that one need sufficient information about the system at hand to make an educated guess about the atomci orbitals. However, the computational process is easy to parallelize and is generally easily computed since the solution for the orbitals can be extrapolated in space. The real space method is generally computationally expensive (however often exact). Plane waves have neat physical properties, such as that the energy is only dependent on the wave-vector and that solutions become more exact when one increases the number of plane waves. A problem with the plane waves basis set is that it utalizes fourier transforms which are generally tedious to parallellize. We did not investigate the atomic atorbitals of aluminum enough to make an educated guess and we tried to avoid unreasonably heavy computational burden. On top of this, we really enjoyed that the plane waves converges to an accurate solution as the basis set expanded. Thus, we decided to use the plane wave basis set in the simulation of the aluminum nano particles.

Another thing to consider is the discrete k spacing for the energy bands. The k-space domain need too be sampled at fine enough distances to capture most parts of the band structure. If k values are sampled too far apart, the energy can differ from the actual energy, both an increase and a decrease. To get a sufficient amount of k-points is often not the most computationally heavy step and therefore can be exaggerated a bit. To find a reasonable value for the spacing, simulations were made with plane wave sets with increasing energy, and the spacing for which the total energy had stabilized was used in further simulations.

To get a good estimate of the surface energy one has to include atoms of the bulk since they effect the interatomic distance near the surface, and hence the surface energy. The surface approaches an stable energy at about XXXX atoms, see figure XXXXX.

The simulation was made both in vacuum and with CO molecules attached to the surface. The CO molecules was attached on top of aluminium atoms in both the 111 and 100 surface. The occupacy of atoms was set to 1/4 by adsorbing one CO molecule to a 2x2 surface. Is it ok to do it this way?

information about the previous density when navigating in the energy landscape. (The

information about the direction towards the minimal energy should not be lost in each iteration step). This is implemented with a so called mixer. The new density is usually updated according to the following rule:

$$n^{new} = \beta n^{old} + (1 - \beta)n^{new} \tag{1}$$

For metals, the recommended value for β is about 0.1 ??. GPAW has a build in mixer, which takes in three arguments, β , number of previous generations used 'nmaxold' and a weight. For the two latter parameters, the default value was used, 5 generations weight 50.

Results

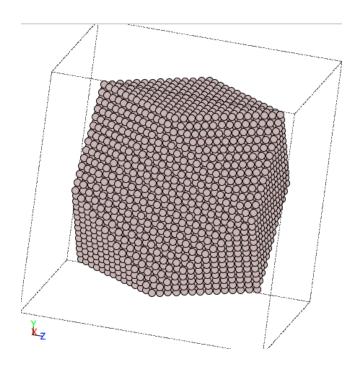


Figure 1: Wulff construction of aluminum in vacuum

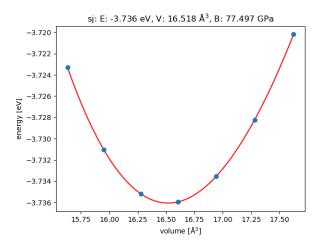


Figure 2: Total lattice energy for varying lattice spacings

Different shapes of wulf construction

Discussion	
(Try to bring order in your results, rationalize the results to the best of your ability)	
Conclution	
References	
[1] Magnus Rahm, <i>Task and background</i> . Pingpong - TIF035 Computational Materials Physics VT18, 2018.	
https://pingpong.chalmers.se/courseId/9094/content.do?id=4260146. Retrieved 2018-02-02.	

A Source Code

A.1 Main simulation: /HA2_simulation.py

```
#!/usr/bin/env python
    # coding=utf-8
    # # # imports # # #
    import numpy as np
    from gpaw import GPAW, Mixer, PW
    from ase.build import
     from ase.io import read
    from ase.units import kJ, J, m
10
    from ase.eos import EquationOfState
    from ase.build import fccll1, fccl00, add_adsorbate from ase.cluster.wulff import wulff_construction
11
     from ase.visualize import view
    from ase.parallel import rank
15
    from ase import Atoms
16
    # # # Constants and parameters # # # a\_al = 4.05     # Lattice constant for Al, experimentally determined
17
18
    N_lattice_spacings = 7 \, # Number of lattice constants to loop over to find \leftrightarrow
19
         equilibrium
    E_al = 84.67567 # Ionization energy for hardest bound core electron
    theta = 1 / 4.0 # Monolayer coverage
21
    # # # Create Al bulk and initialize calculator parameters # # #
al = bulk('Al', 'fcc', a=a_al, cubic=False) # Create Al bulk
    mixer = Mixer(beta=0.1, nmaxold=5, weight=50.0) # Recommended values for small←
          systems
26
    E_cut = [50, E_al, 100, 200, 300, 400, 500] # cut-off energy
27
    for energy in [500]: # Change to E_cut to loop and check convergence
28
        calc = GPAW(mode=PW(energy), # use the LCAO basis mode
    h=0.18, # grid spacing, recommended value in this course
    xc='PBE', # Exchange-correlation functional
29
30
31
32
                      mixer=mixer,
                      33
                      txt='out.txt') # name of GPAW output text file
34
35
         al.set_calculator(calc)
36
37
         # # # Find lattice constant with lowest energy # # #
         for eps in np.linspace(-0.02, 0.02, N_lattice_spacings):
    al.cell = (1 + eps) * cell_0 # Adjust lattice constant of unit cell
38
39
40
             al.get_potential_energy() # Calculate the potential energy for the Al \leftrightarrow
         confs = read('out.txt@0:' + str(N_lattice_spacings)) # Read the \hookleftarrow
43
               c o n gurations
44
45
         # Extract volumes and energies:
         volumes = [atoms.get_volume() for atoms in confs]
46
47
         energies = [atoms.get_potential_energy() for atoms in confs]
         # if rank == 0:
48
49
               print energies, shape(energies)
50
         # Plot energies as a function of unit cell volume (directly related to latt.\hookleftarrow
              const.)
         eos = EquationOfState(volumes, energies)
53
         v0, E_bulk, B = eos.fit()
         54
55
              this correct?
57
         N_x = 1
58
         N_y = 1
59
         N_z = 6
60
         61
62
         surface111.center(axis=2)
63
64
         surface100.center(axis=2)
65
         # Initialize new calculator that only considers k-space in xy-plane,
66
         # since we're only looking at the surface
67
         calc2 = GPAW(mode=PW(energy), # use the LCAO basis mode
68
                       h=0.18, # grid spacing
xc='PBE', # XC-functional
70
71
                       mixer=mixer
                       kpts=(12, 12, 1), # k-point grid
txt='out2.txt') # name of GPAW output text file
72
73
         surface111.set_calculator(calc2)
         surface100.set_calculator(calc2)
```

```
78
            cell111 = surface111.get_cell() # Unit cell object of the Al FCC 111
 79
            area111 = np.linalg.norm(np.cross(cell111[0], cell111[1])) \quad \textit{\# Calc. surface} \; \hookleftarrow
                  area
            surfEn111 = surface111.get_potential_energy()  # Calc pot. energy of FCC 111
cell100 = surface100.get_cell()  # Unit cell object of the Al FCC 100
 80
 81
            area100 = np.linalg.norm(np.cross(cell100[0], cell100[1])) # Calc. surface \leftarrow
83
            surfEn100 = surface100.get_potential_energy() # Calc pot. energy of FCC 100
84
            # Calc. surf. energy per area (sigma) for FCC 111 and 100 sigma111 = (1 / (2.0 * area111)) * (surfEn111 - N_x * N_y * E_bulk) sigma100 = (1 / (2.0 * area100)) * (surfEn100 - N_x * N_y * E_bulk)
85
 86
88
89
            # # # Add CO adsorbate to Al surface # # #
90
            d_CO = 1.128 # CO bondlength in [
91
 92
            CO = Atoms('CO') # Create CO molecule object
            add_adsorbate(slab=surface111, adsorbate=CO, height=4.5, position='ontop') add_adsorbate(slab=surface100, adsorbate=CO, height=4.5, position='ontop') # height above based on values for CO in ASE doc. Future: We could also
 93
 94
95
            # perform equilibrium scan by looping over various heights
 96
97
 98
            CO.set_cell([10, 10, 10])
 99
            CO.center()
100
            CO.set_calculator(calc)
101
102
            energy_C0 = C0.get_potential_energy()
103
104
            surfEn111 ads = surface111.get potential energy()
105
            surfEn100_ads = surface100.get_potential_energy()
107
            sigma100\_ads = sigma100 + theta * (surfEn100\_ads - surfEn100 - energy\_CO) / 
                  area100
            sigma111\_ads = sigma111 + theta * (surfEn111\_ads - surfEn111 - energy\_CO) / \hookleftarrow
108
                  area111
109
            file = open('sigma_A1.txt', 'w')
file.write(str(sigma111) + '\t' + str(sigma100))
110
111
112
            file.close()
113
            file = open('sigma_ads.txt', 'w')
file.write(str(sigma111_ads) + '\t' + str(sigma100_ads))
114
115
            file.close()
117
118
            if rank == 0:
                 a = 0
119
                 print area111, surfEn111, E_bulk, sigma111
120
```

A.2 Data processing and wulff construction: HA2_data_processing.py

```
#from ase.io import read
     from ase.build import fcc111, fcc100
     from ase.cluster.wulff import wulff_construction
     from ase.visualize import view
     import matplotlib.pyplot as plt
     import numpy as np
    # input files
filename1 = 'sigmas.txt'
filename2 = 'sigmas_ads.txt'
10
11
     # import and manage data
     sigmas = np.array(np.loadtxt(filename1))
15
     sigma111 = sigmas[0]
16
     sigma100 = sigmas[1]
    sigmas_ads = np.array(np.loadtxt(filename2))
sigma111_ads = sigmas_ads[0]
17
18
     sigma100_ads = sigmas_ads[1]
     theta = 1 / 4 # Monolayer coverage
20
    print(sigma111, sigma100)
    print(sigma111_ads, sigma100_ads)
23
    Al = wulff_construction('Al',
                                surfaces=[(1, 0, 0),
                                           (1, 1, 1)],
27
                                energies=[sigma100, sigma111],
                                size=10000,
structure='fcc'
28
29
                                rounding='below') # What does this one do?
30
31
    Al.center(vacuum=10)
     view(Al)
33
34
     sigma100\_ads = sigma100 + theta * sigma100\_ads
    sigmal11_ads = sigma111 + theta * sigma111_ads
```

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