

Quantum Machine Learning

Anatole von Lilienfeld

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Many of the most relevant chemical properties of matter depend explicitly on atomistic and electronic details, rendering a first principles approach to chemistry mandatory. Alas, even when using high-performance computers, brute force high-throughput screening of compounds is beyond any capacity for all but the simplest systems and properties due to the combinatorial nature of chemical space, i.e. all compositional, constitutional, and conformational isomers. Consequently, efficient exploration algorithms need to exploit all implicit redundancies present in chemical space. I will discuss recently developed statistical learning approaches for interpolating quantum mechanical observables in compositional and constitutional space. Results for our models indicate remarkable performance in terms of accuracy, speed, universality, and size scalability.

"Machine Learning, Quantum Mechanics, and Chemical Compound Space"

By Ramakrishnan and von Lilienfeld

Reviews in Computational Chemistry, edited by Abby L. Parrill and Kenny B. Lipkowitz

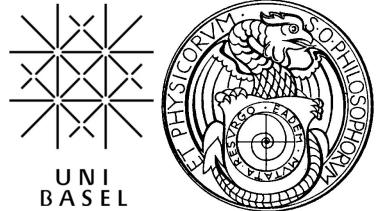
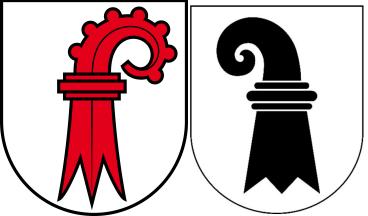
Volume **30**, Chapter 5, pages 225-256 (2017)

[www.arxiv.org/abs/1510.07512](https://arxiv.org/abs/1510.07512)

Essay: "Quantum Machine Learning in Chemical Compound Space"

Just accepted, *Angew. Chem. Int. Ed.* (2018)

<http://rdcu.be/AJNU/>



Departement
Chemie



U.S. DEPARTMENT OF ENERGY
INCITE
LEADERSHIP COMPUTING



FONDS NATIONAL SUISSE
SCHWEIZERISCHER NATIONALFONDS
FONDO NAZIONALE SVIZZERO
SWISS NATIONAL SCIENCE FOUNDATION



MARVEL



NATIONAL CENTRE OF COMPETENCE IN RESEARCH

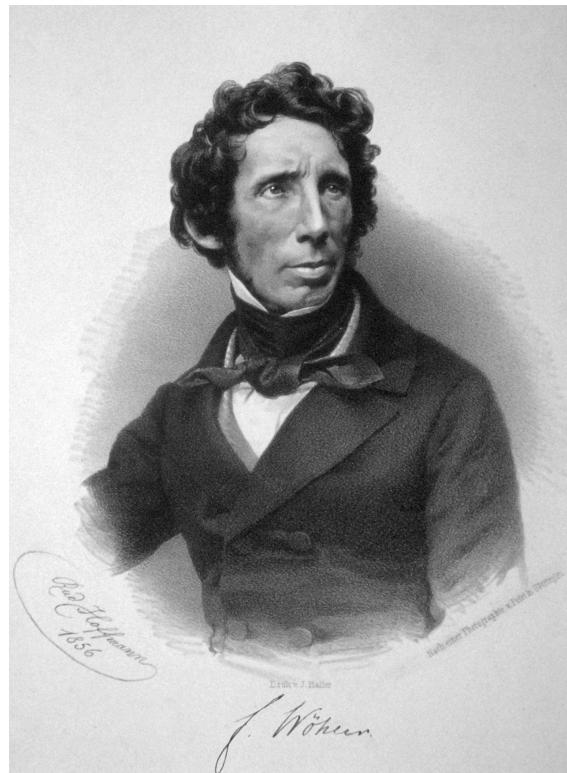
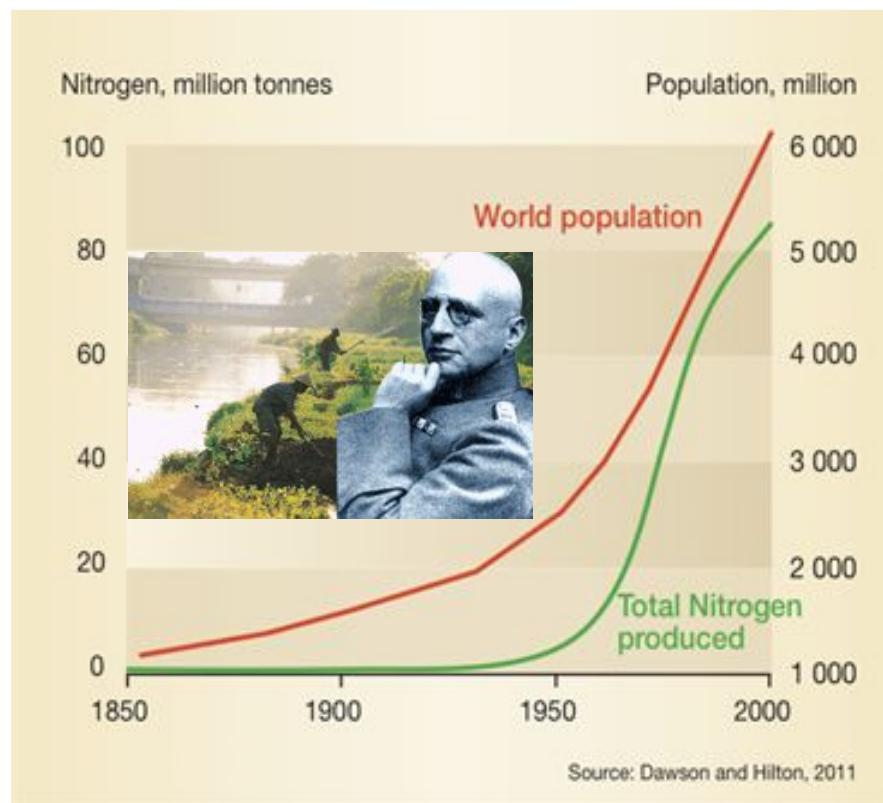
Research at Google

Chemistry?

Millennium Essay by V. Smil

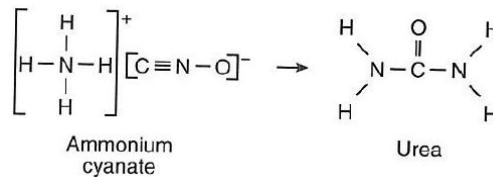
Nature 400, 415 (29 July 1999) | doi:10.1038/22672

“What is the most important invention of the twentieth century? Aeroplanes, nuclear energy, space flight, television and computers will be the most common answers ... Yet none of these can match the synthesis of ammonia from its elements.”

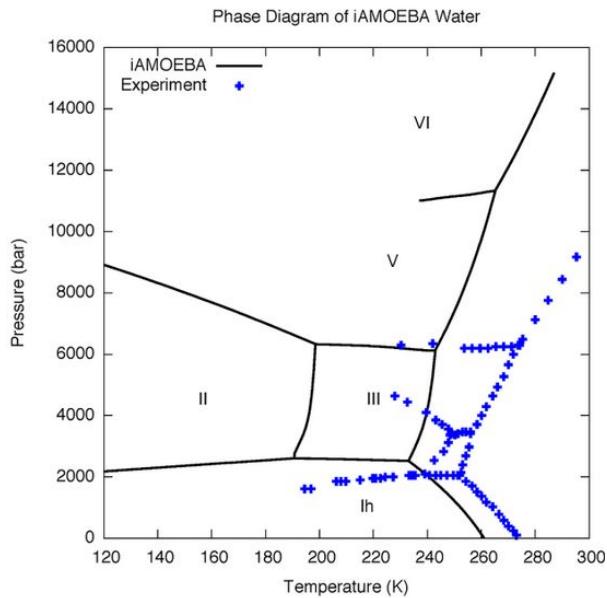


Friedrich Wöhler (1800-1882)

1828: Urea from ammonia and cyanate (in solution+heat)



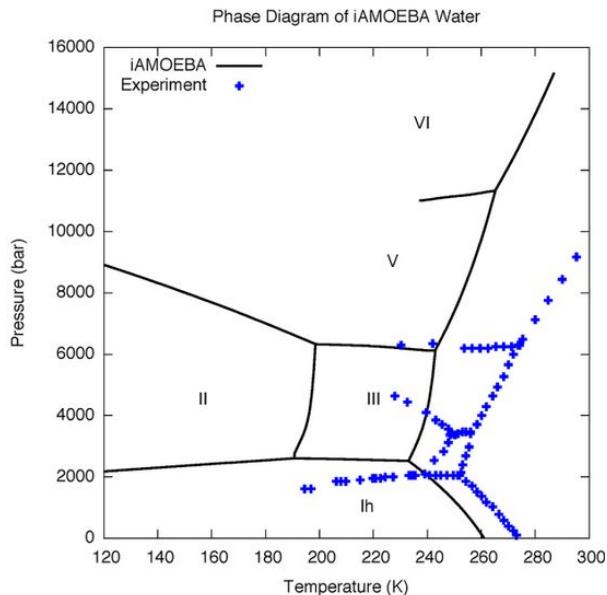
Chemistry on a computer? Predict outcomes!



Structure

Pande et al, *J. Phys. Chem B* (2013)

Chemistry on a computer? Predict outcomes!

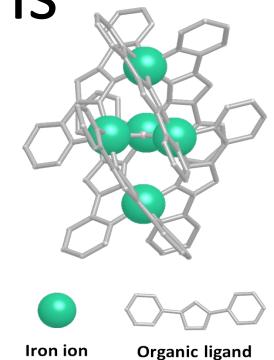
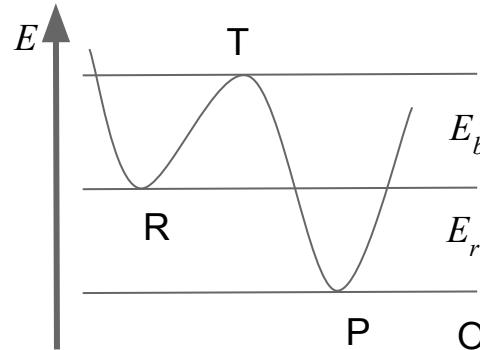


Pande et al, *J. Phys. Chem B* (2013)

Structure

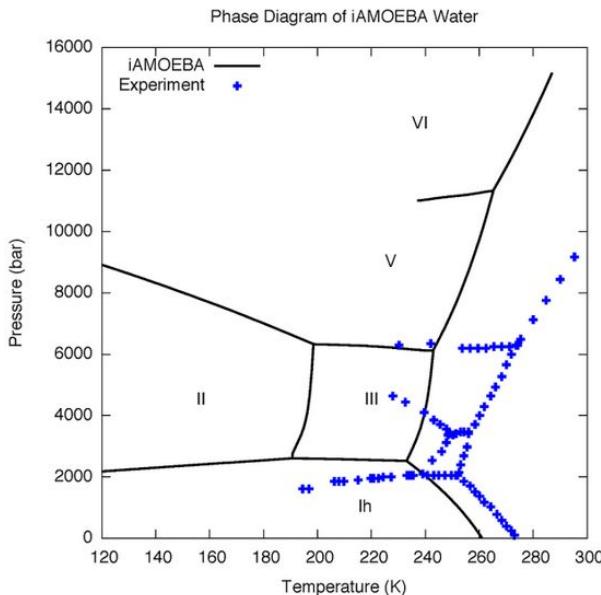
Reactions

Fe based catalyst for water oxidation



Okamura et al, *Nature* (2016)

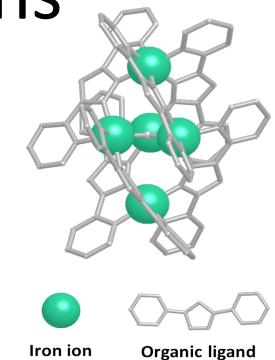
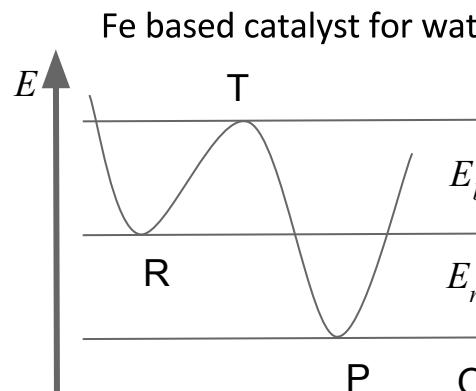
Chemistry on a computer? Predict outcomes!



Pande et al, *J. Phys. Chem B* (2013)

Structure

Reactions

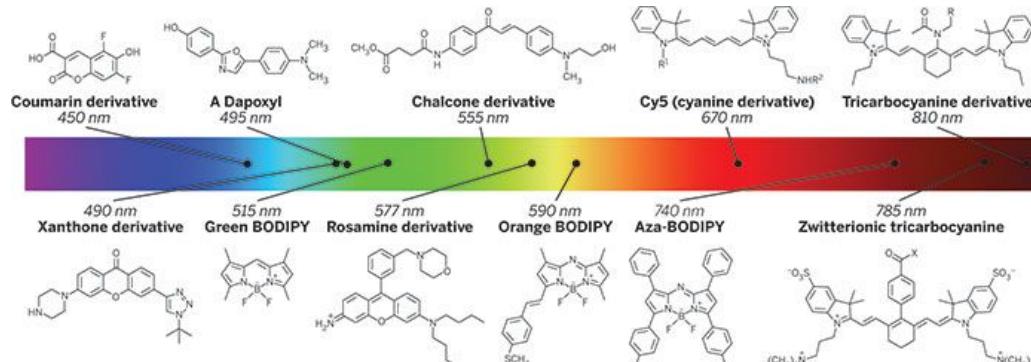


Okamura et al, *Nature* (2016)

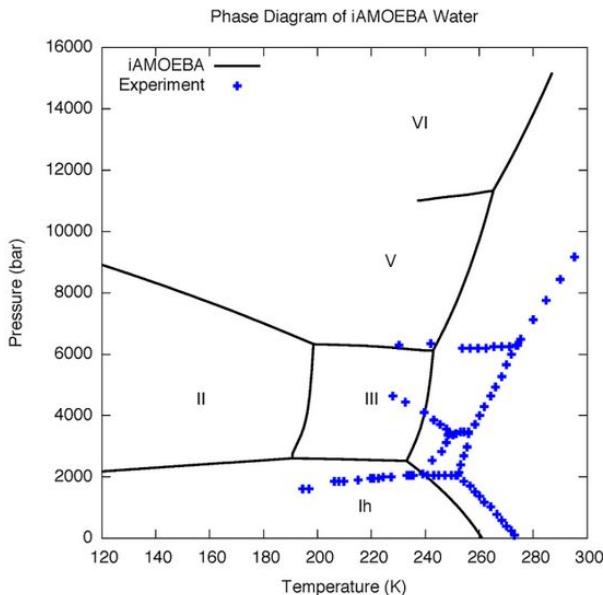
Properties



OLEDs



Chemistry on a computer? Predict outcomes!

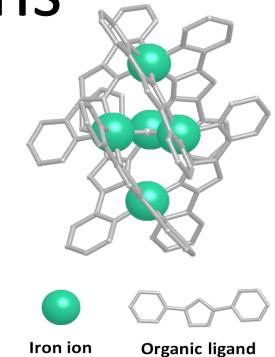
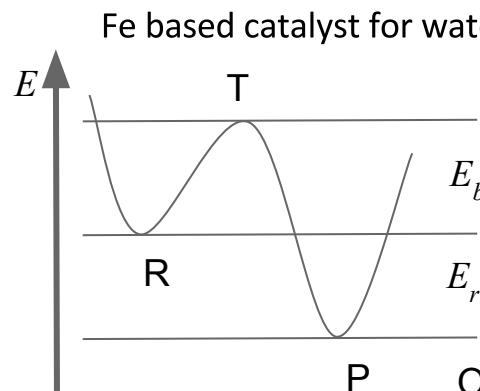


Pande et al, *J. Phys. Chem B* (2013)

We suck!

Structure

Reactions

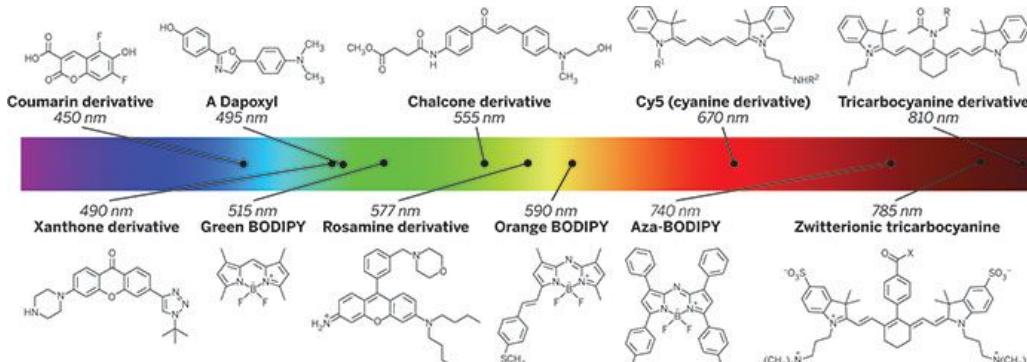


Okamura et al, *Nature* (2016)

Properties

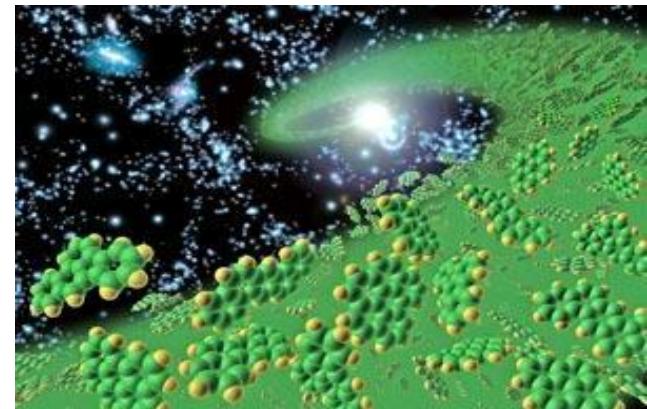
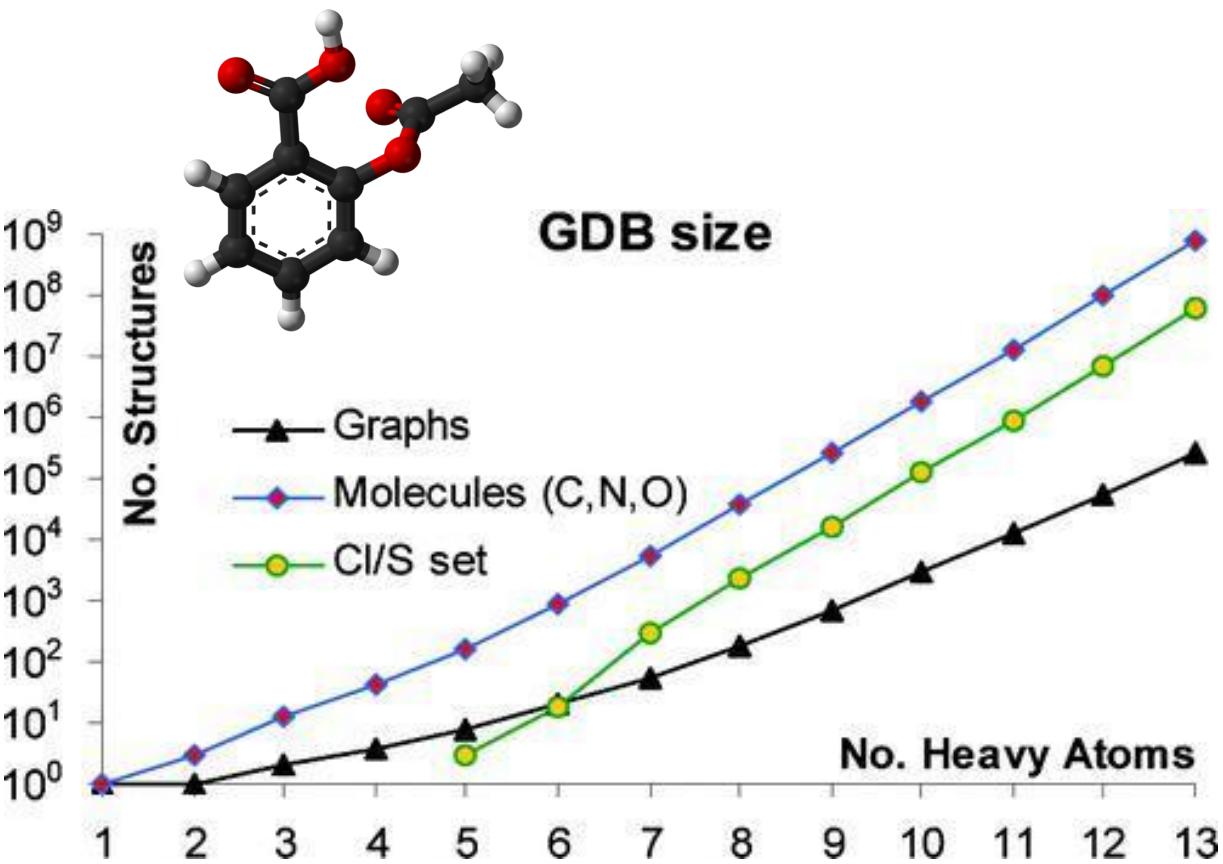


OLEDs



“The greatest shortcoming of the human race is our inability to understand the exponential function”

Al Bartlett, U of Colorado Boulder

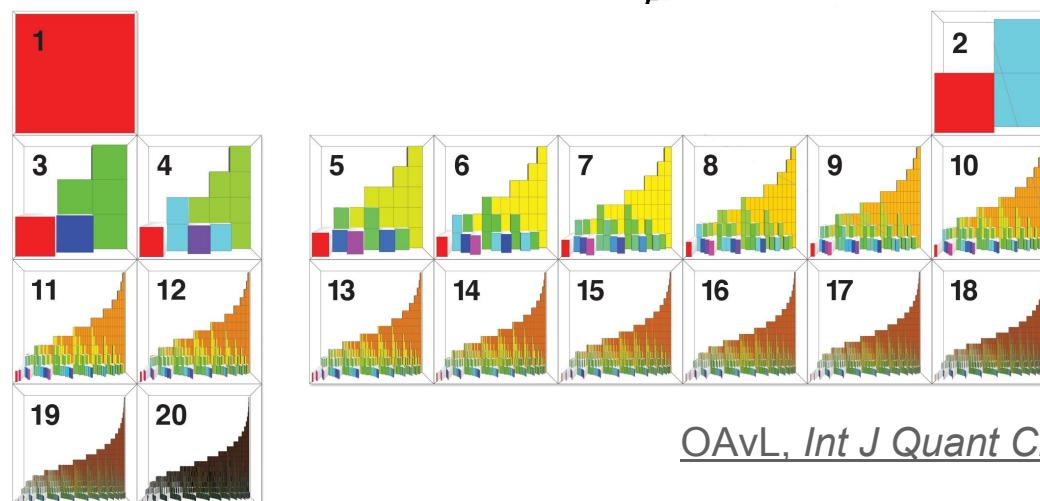
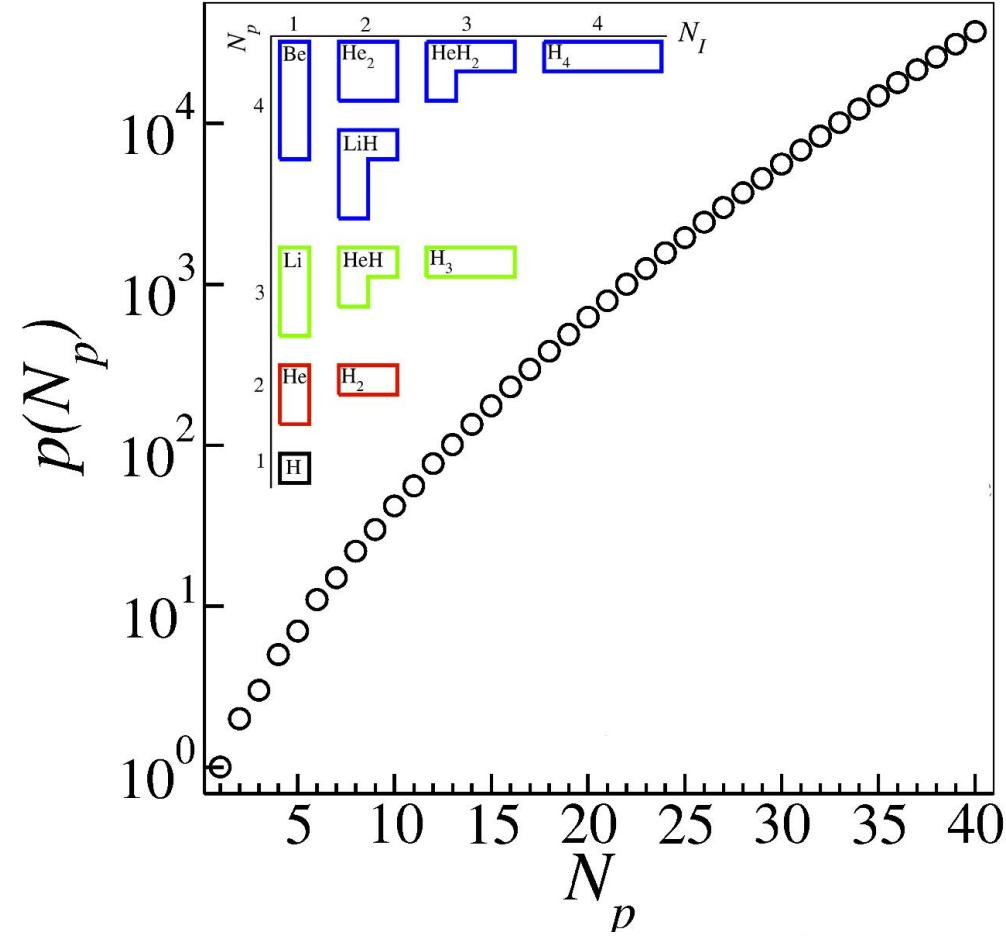


© by NASA

J.-L. Reymond and coworkers, *J Am Chem Soc* (2009) and ff

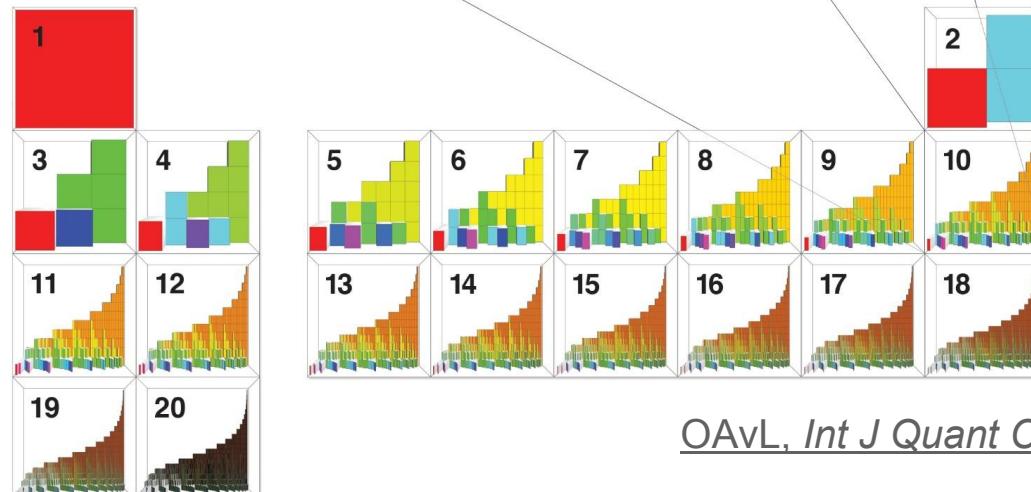
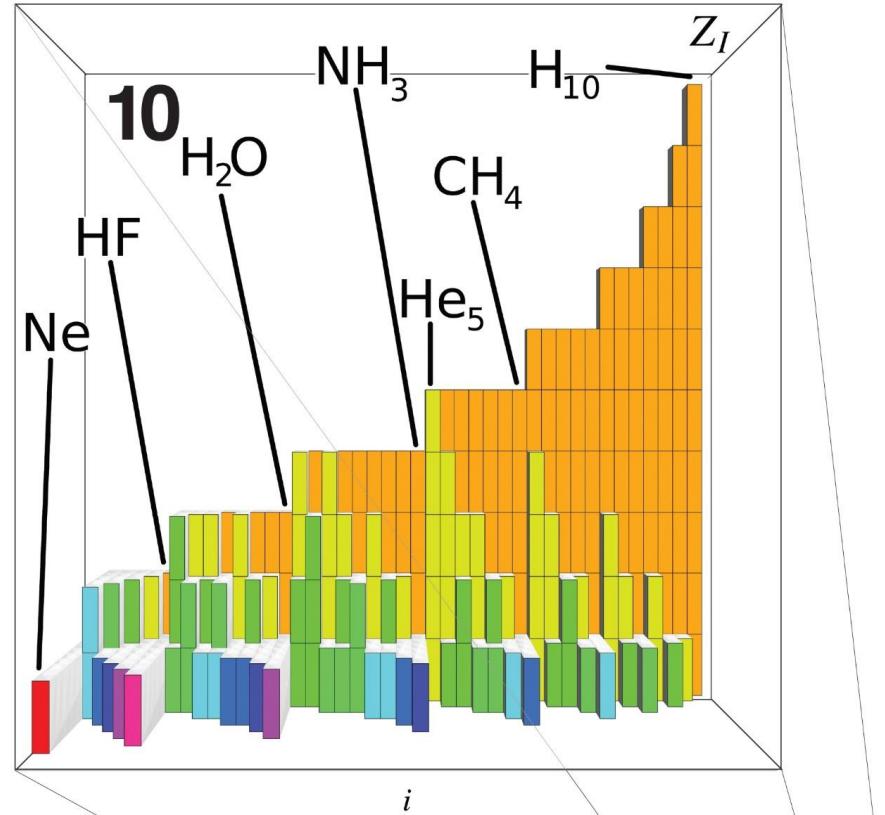
Composition

Young-Ferrers



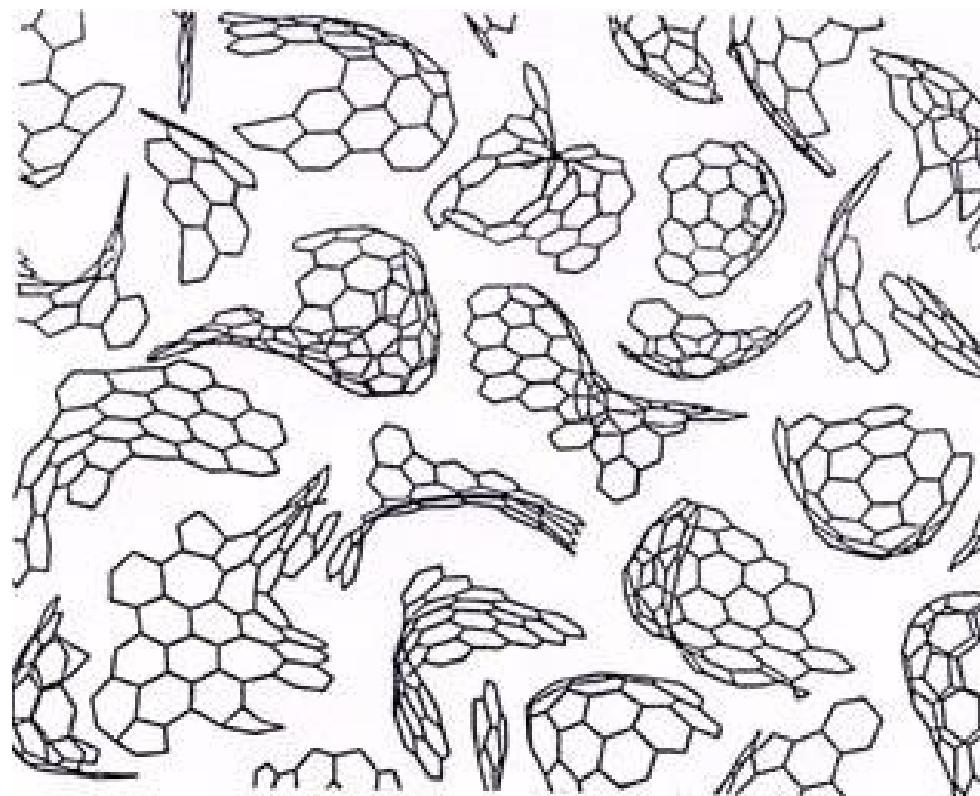
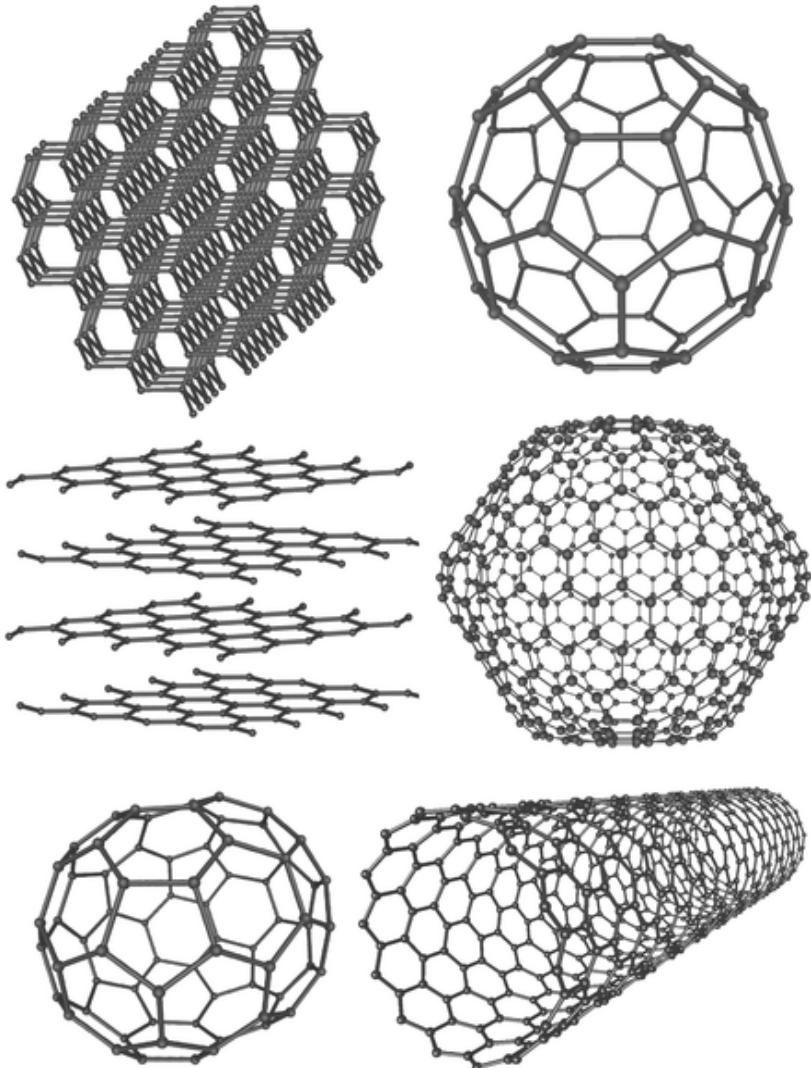
Composition

10 protons



Spatial configuration

Carbon allotropes

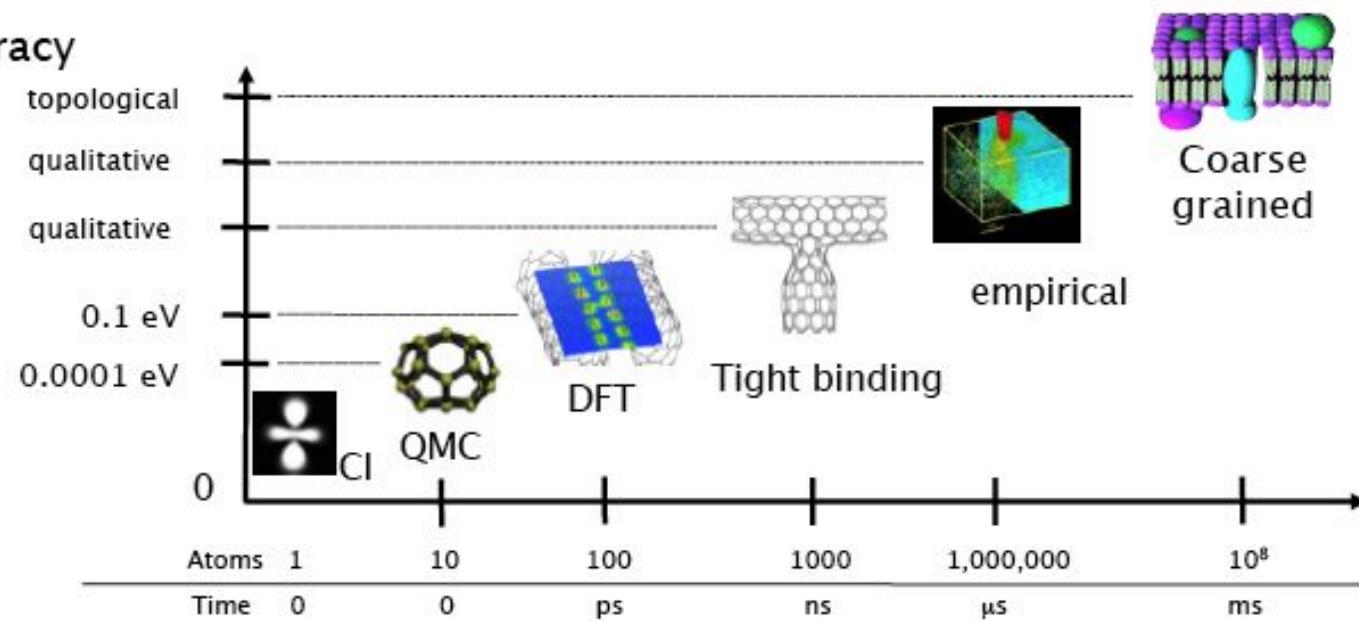


$\text{CCS} \gg 10^{60}$

“Chemical Space”, by
Kirkpatrick and Ellis,
Nature (2004)

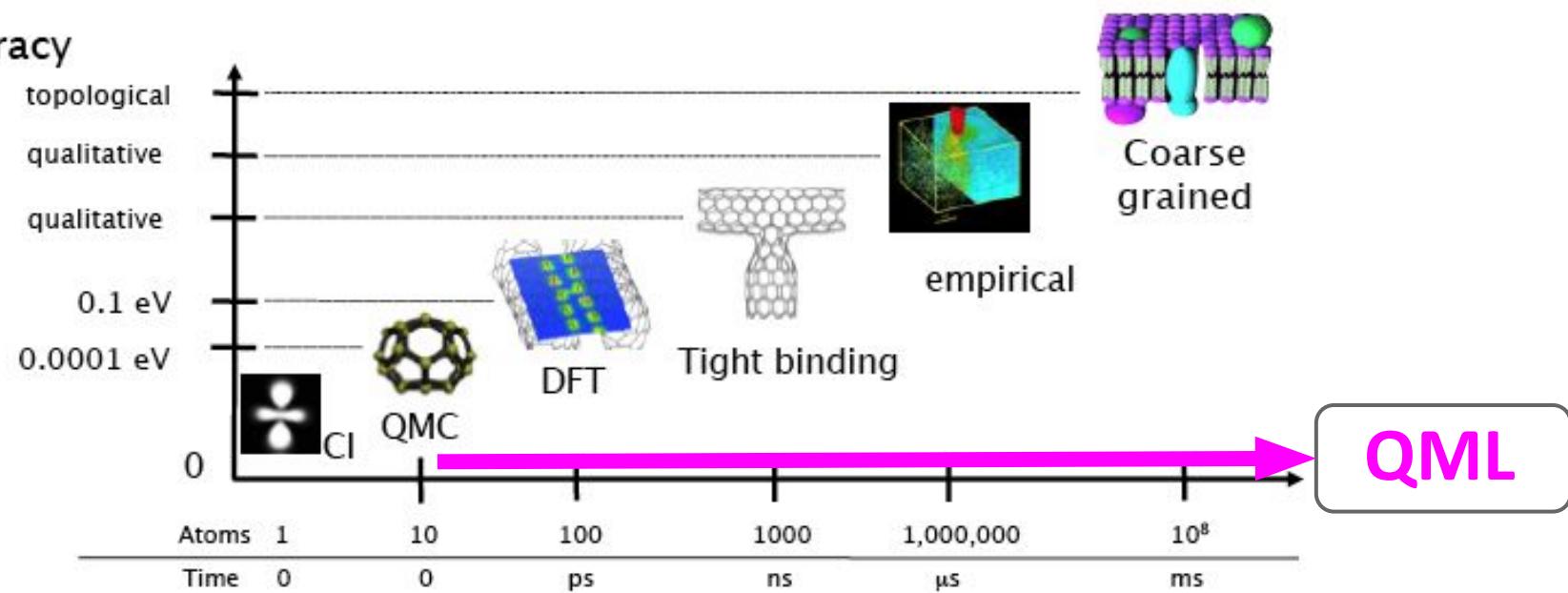


Accuracy



Picture from
Gabor Csanyi

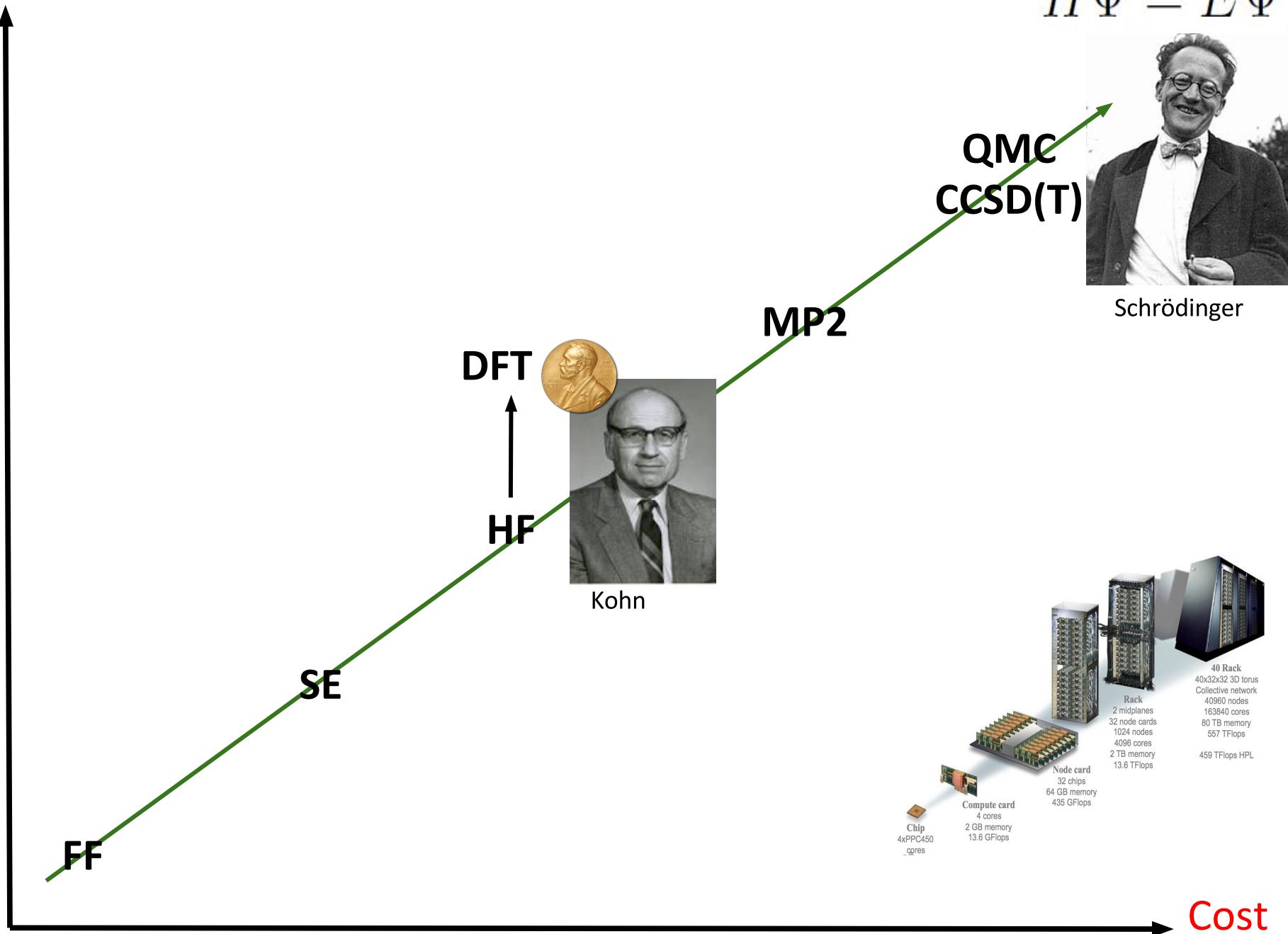
Accuracy



Picture from
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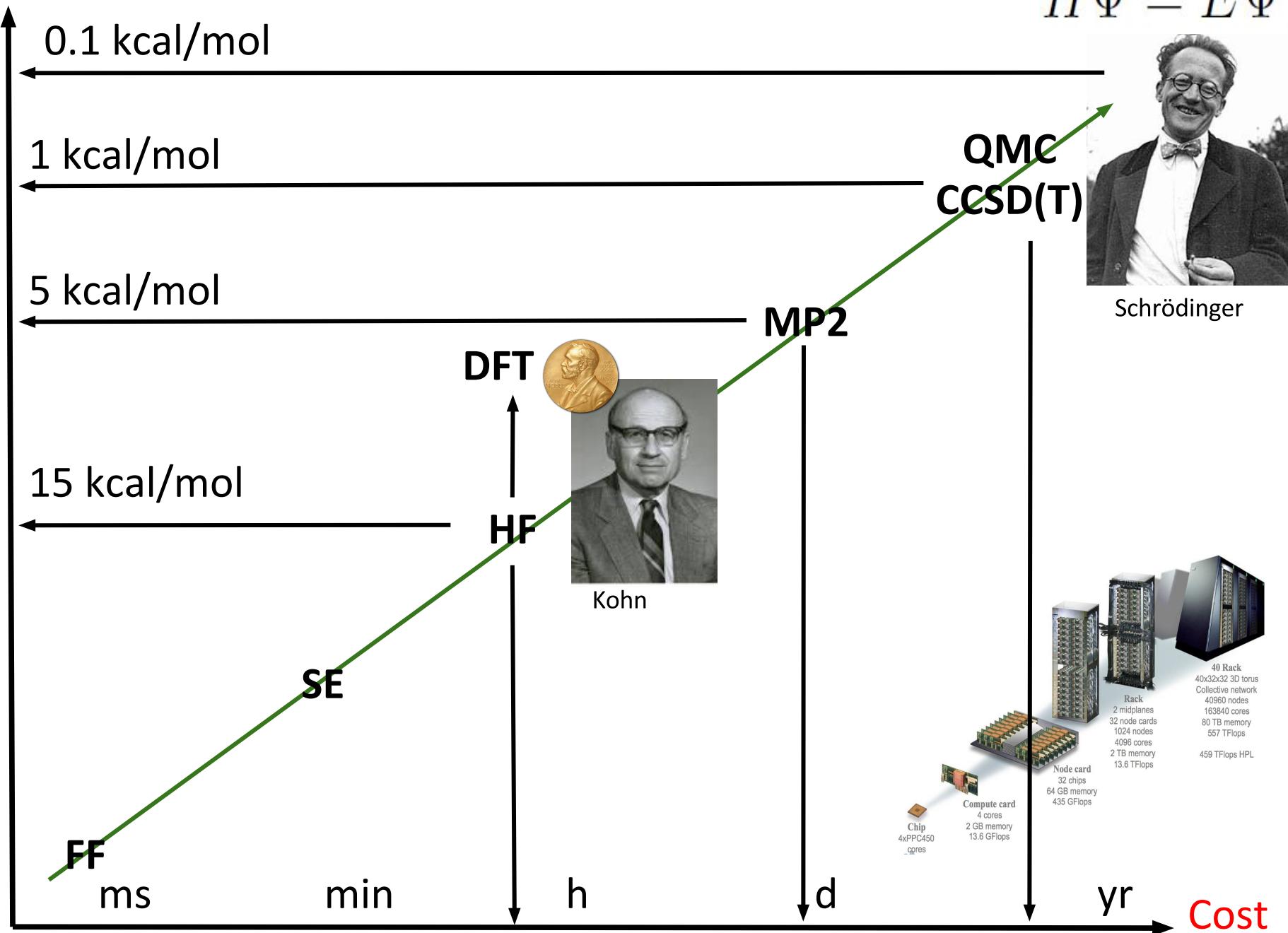
Accuracy & Universality

$$H\Psi = E\Psi$$



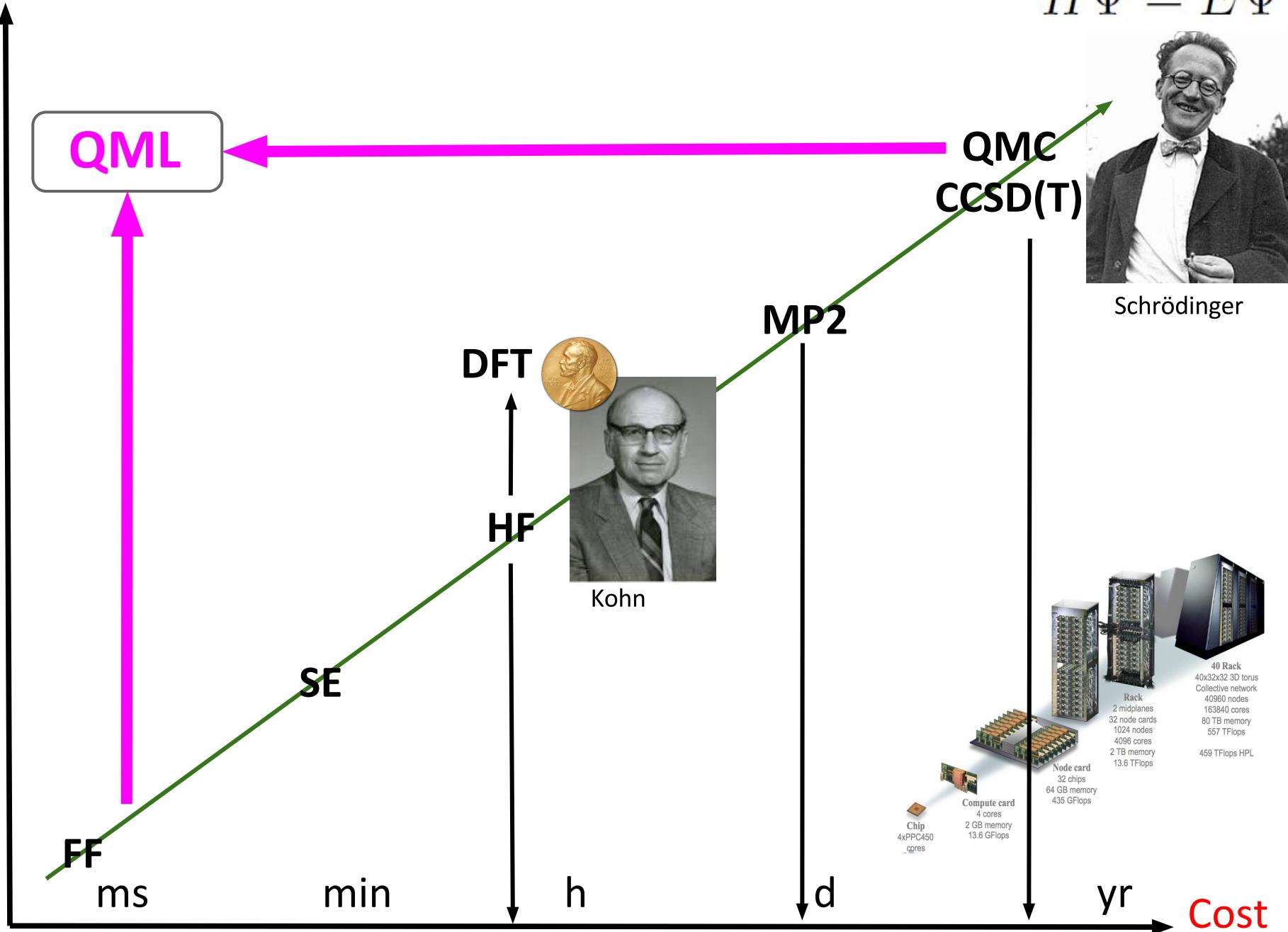
Accuracy & Universality

$$H\Psi = E\Psi$$

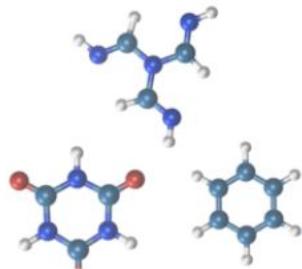


Accuracy & Universality

$$H\Psi = E\Psi$$



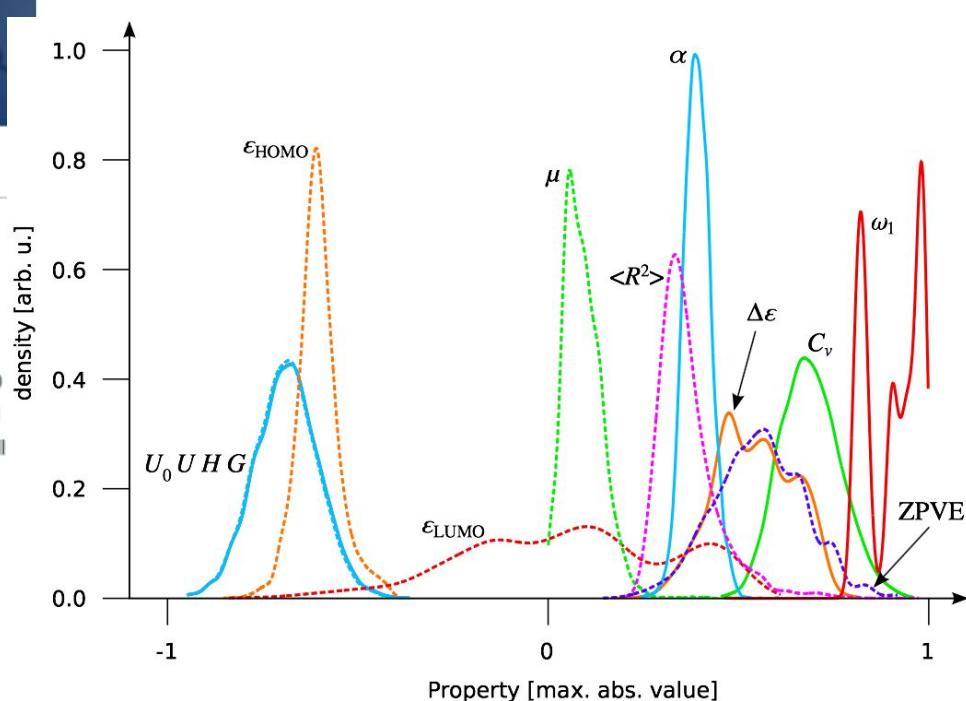
Featured Data Descriptor

Quantum chemistry structures and properties of
134 kilo molecules

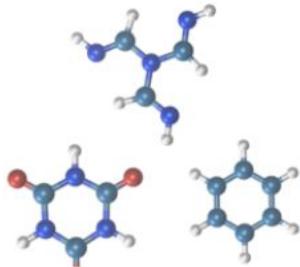
Ramakrishnan et al.

Data Descriptor | 05 August 2014

The authors calculate quantum properties for 134,000 small organic molecules, helping map a vast chemical space that includes important molecules such as small amino acids, nucleobases and various



Featured Data Descriptor



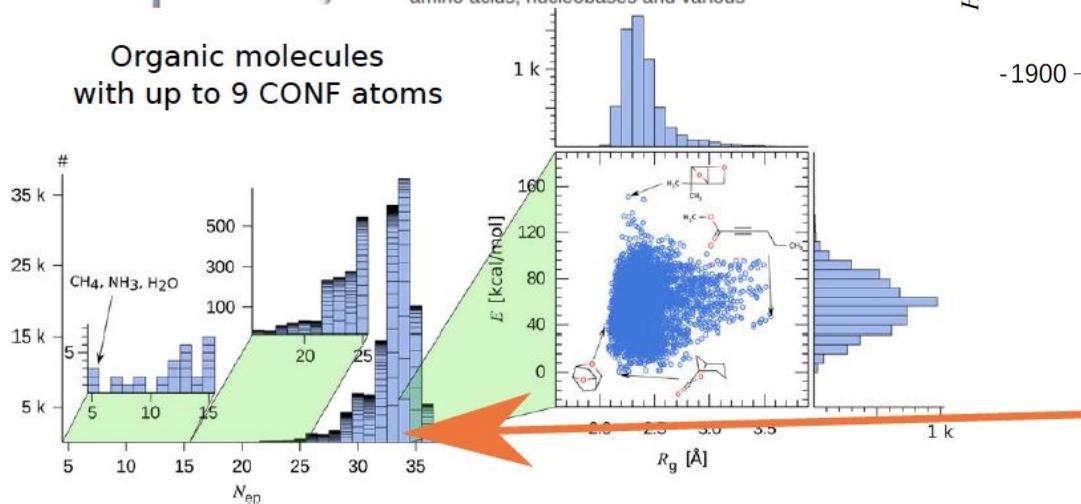
Quantum chemistry structures and properties of 134 kilo molecules

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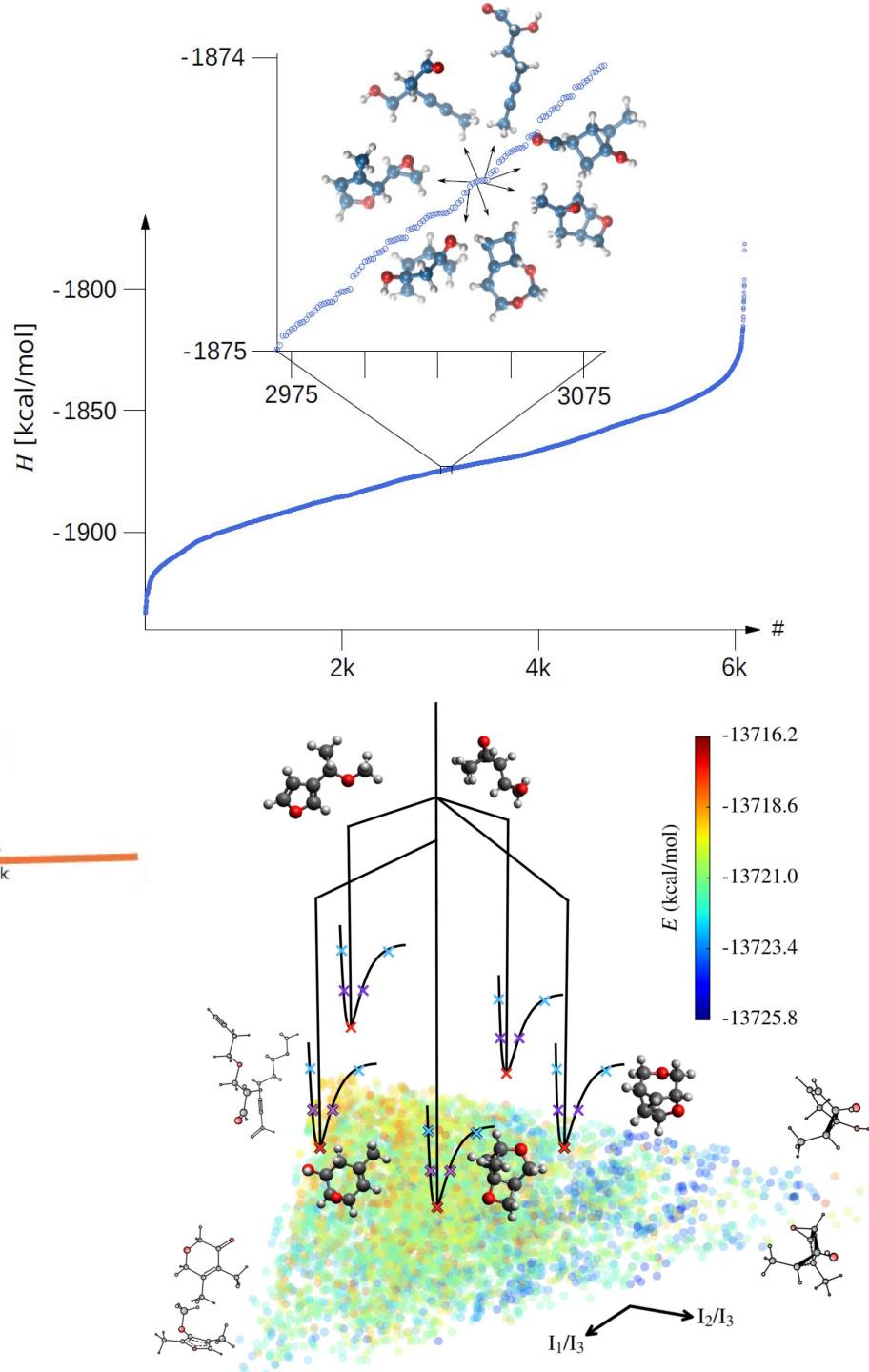
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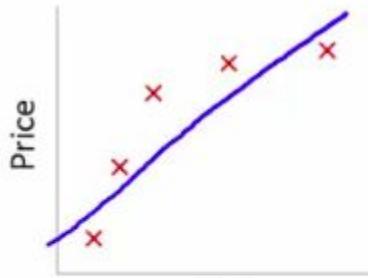
Organic molecules with up to 9 CONF atoms



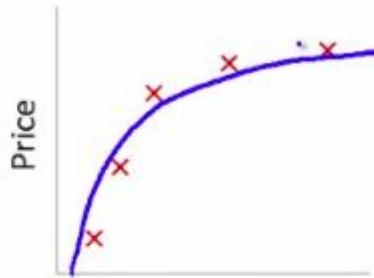
“Enumeration surpasses human imagination”
J.-L. Reymond

Ramakrishnan et al, *Scientific Data* (2014)

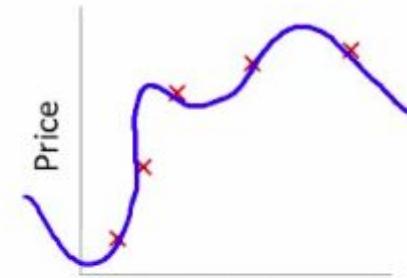
Overfitting?



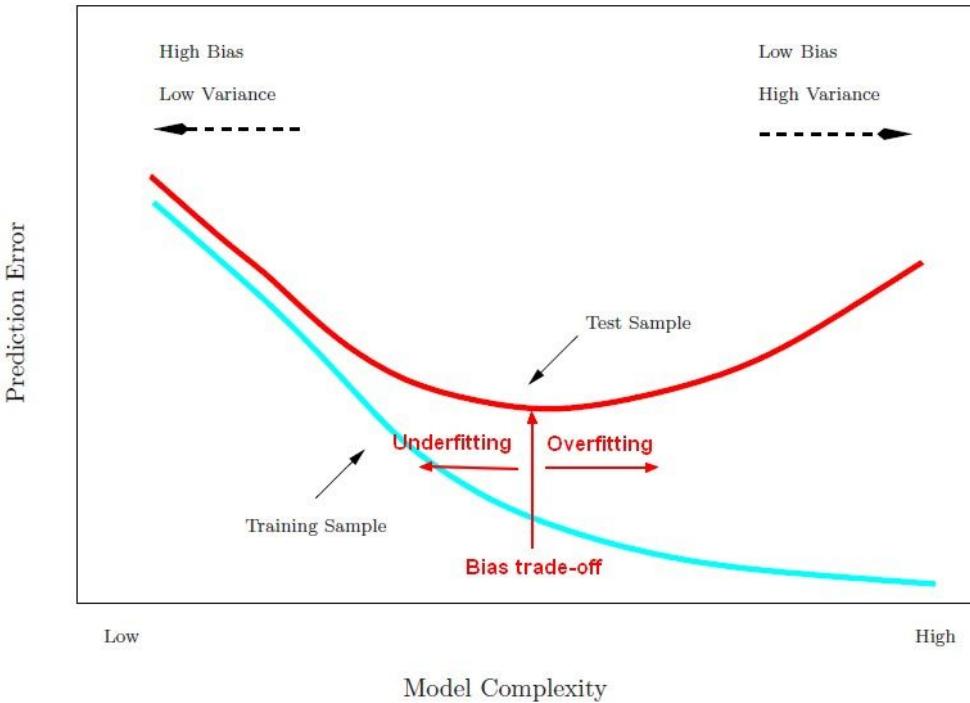
High bias
(underfit)

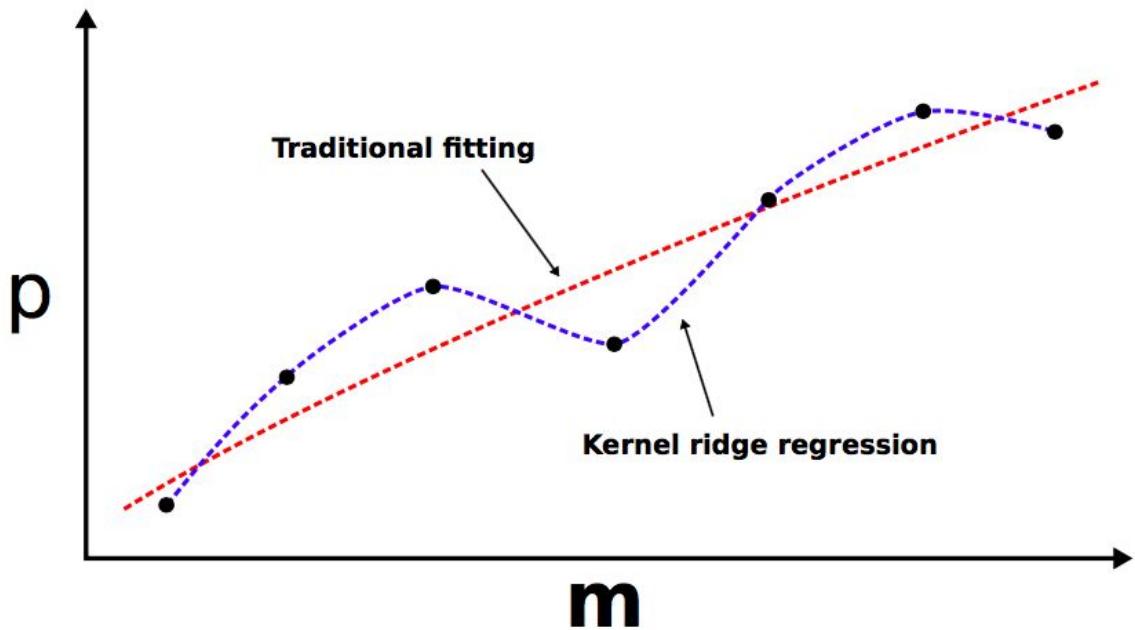


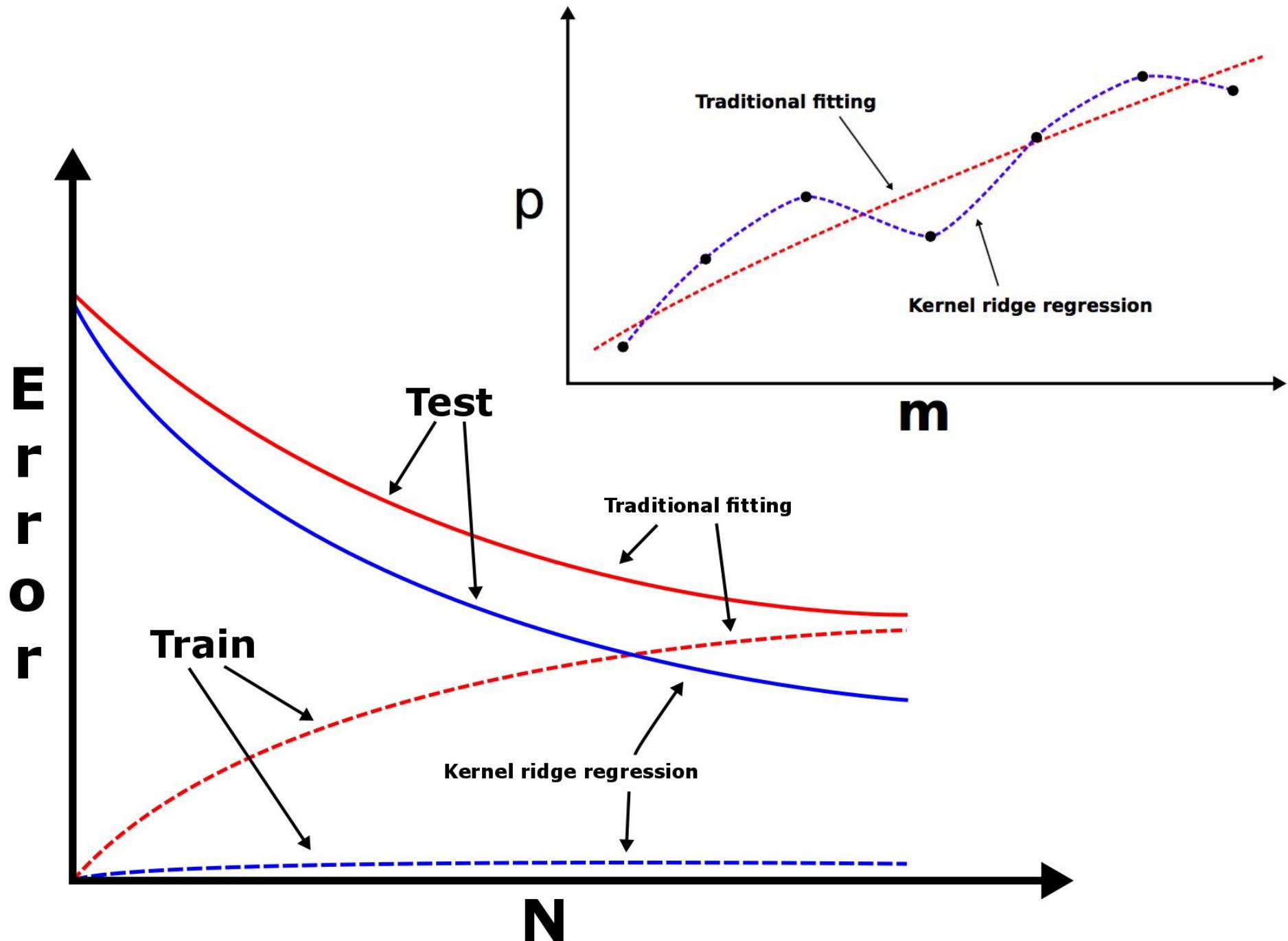
"Just right"

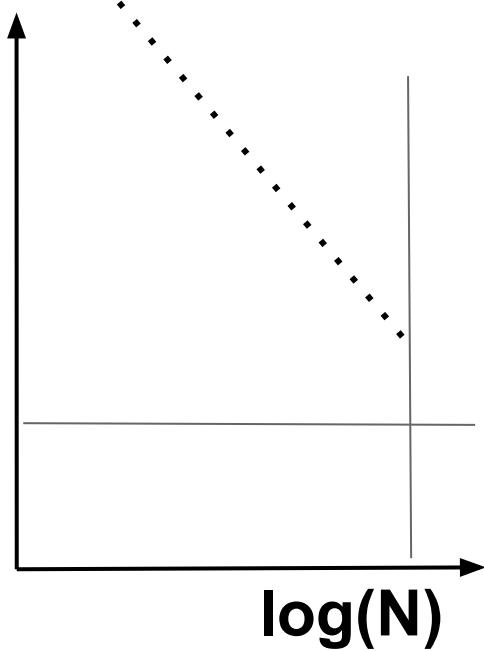


High variance
(overfit)









The bigger the data the better ...

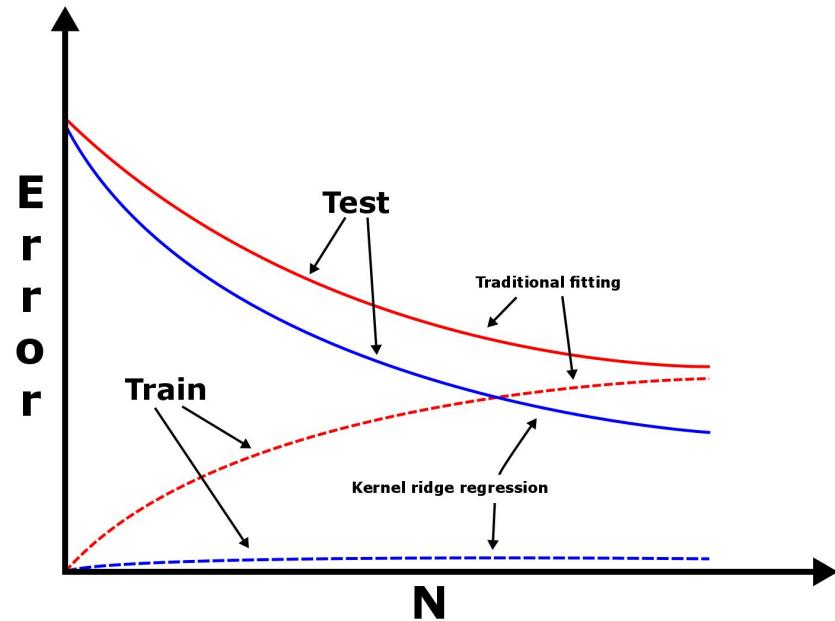
Vapnik, V., *The Nature of Statistical Learning Theory*, Springer (1995)

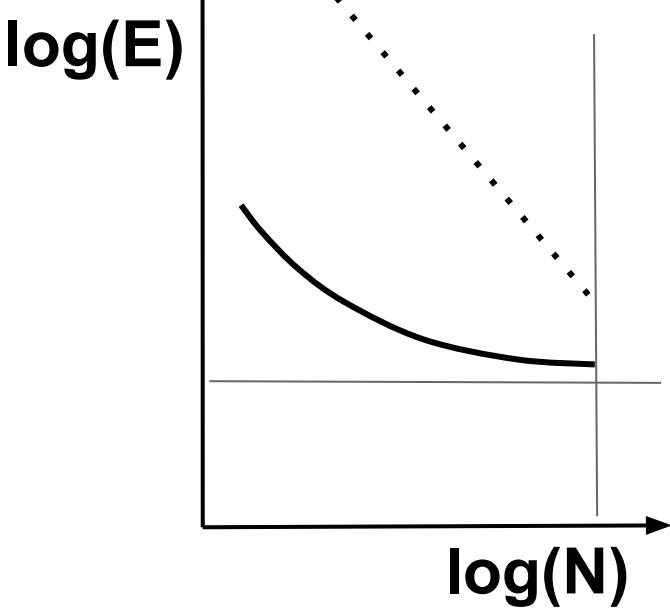
$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

$$\vec{\alpha} = \mathbf{K}^{-1} \vec{P}^{\text{ref}}$$

Error $\sim a/N^b$

$$\rightarrow \log(\text{Error}) = \log(a) - b \log(N)$$





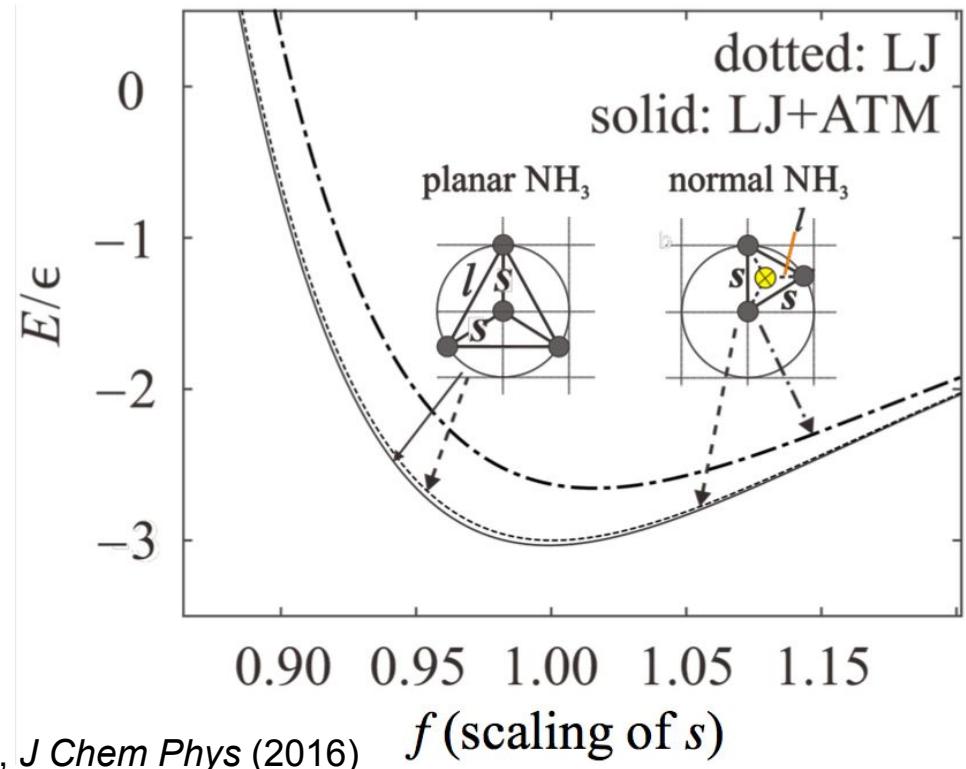
$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

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Error $\sim a/N^b$

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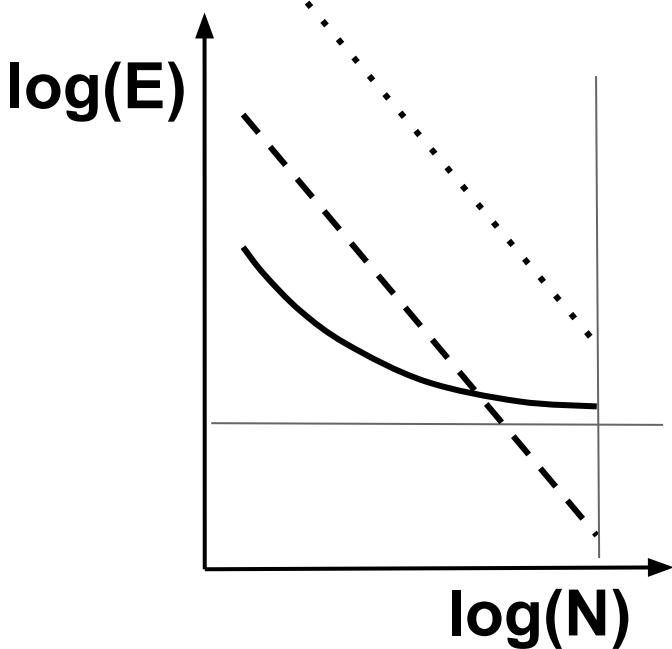
Vapnik, V., *The Nature of Statistical Learning Theory*, Springer (1995)



Huang, OAvL, *J Chem Phys* (2016)

LJ: Lennard-Jones 2-body vdW potential

ATM: Axilrod-Teller-Muto 3-body vdW potential



The bigger the data the better ...

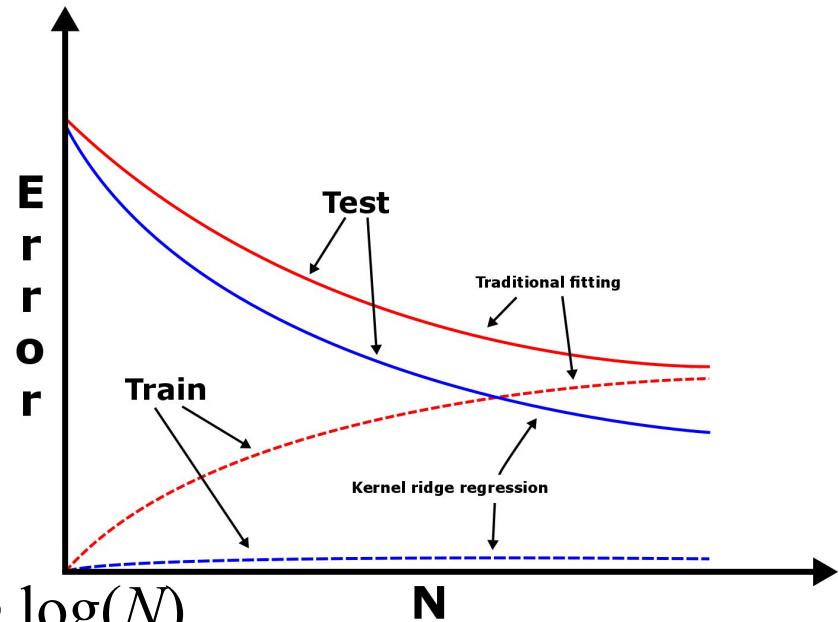
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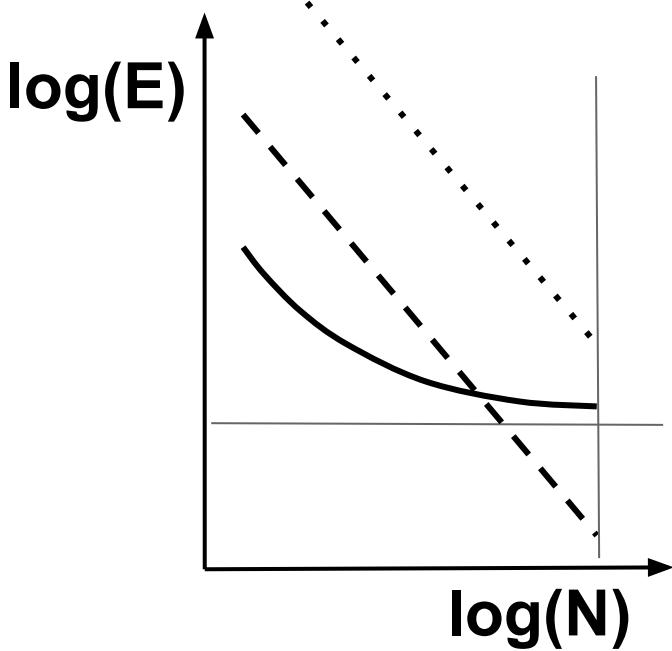
$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

$$\vec{\alpha} = \mathbf{K}^{-1} \vec{P}^{\text{ref}}$$

Error $\sim a/(N')^b$, e.g. $N' = N c$

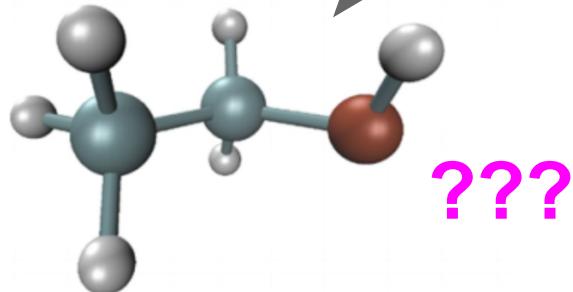
$$\rightarrow \log(\text{Error}) = \log(a) - b \log(c) - b \log(N)$$





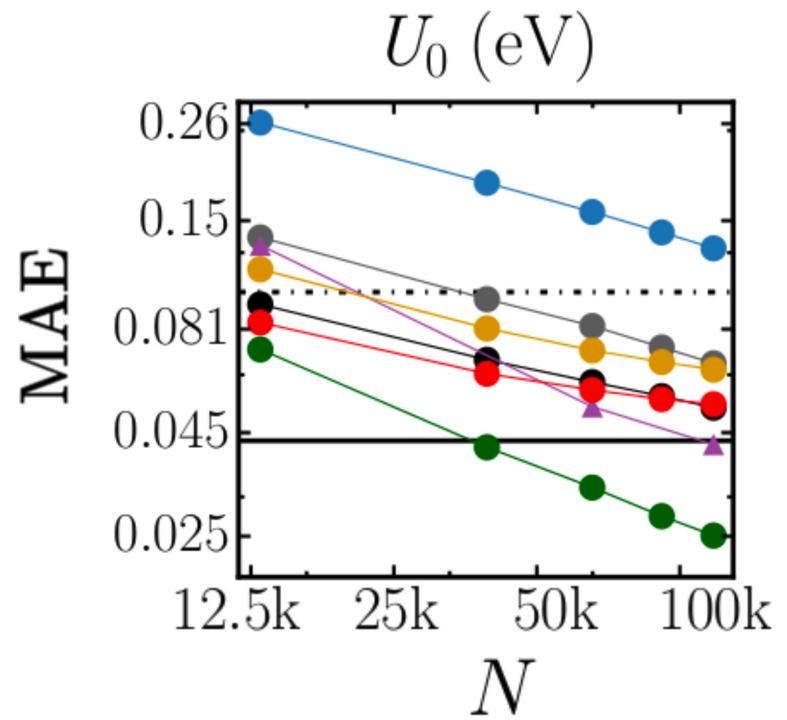
$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

$$\vec{\alpha} = \mathbf{K}^{-1} \vec{P}^{\text{ref}}$$



The bigger the data the better ...

Vapnik, V., *The Nature of Statistical Learning Theory*, Springer (1995)

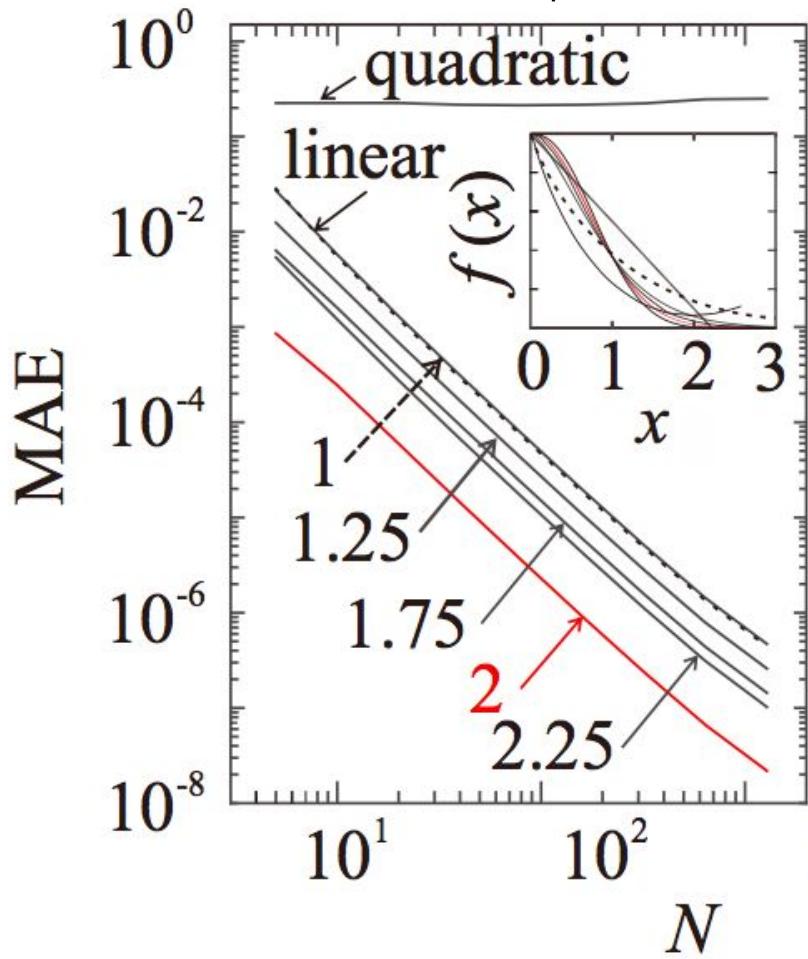


Representation

$$\log(\text{Error}) = a - b \log(N)$$

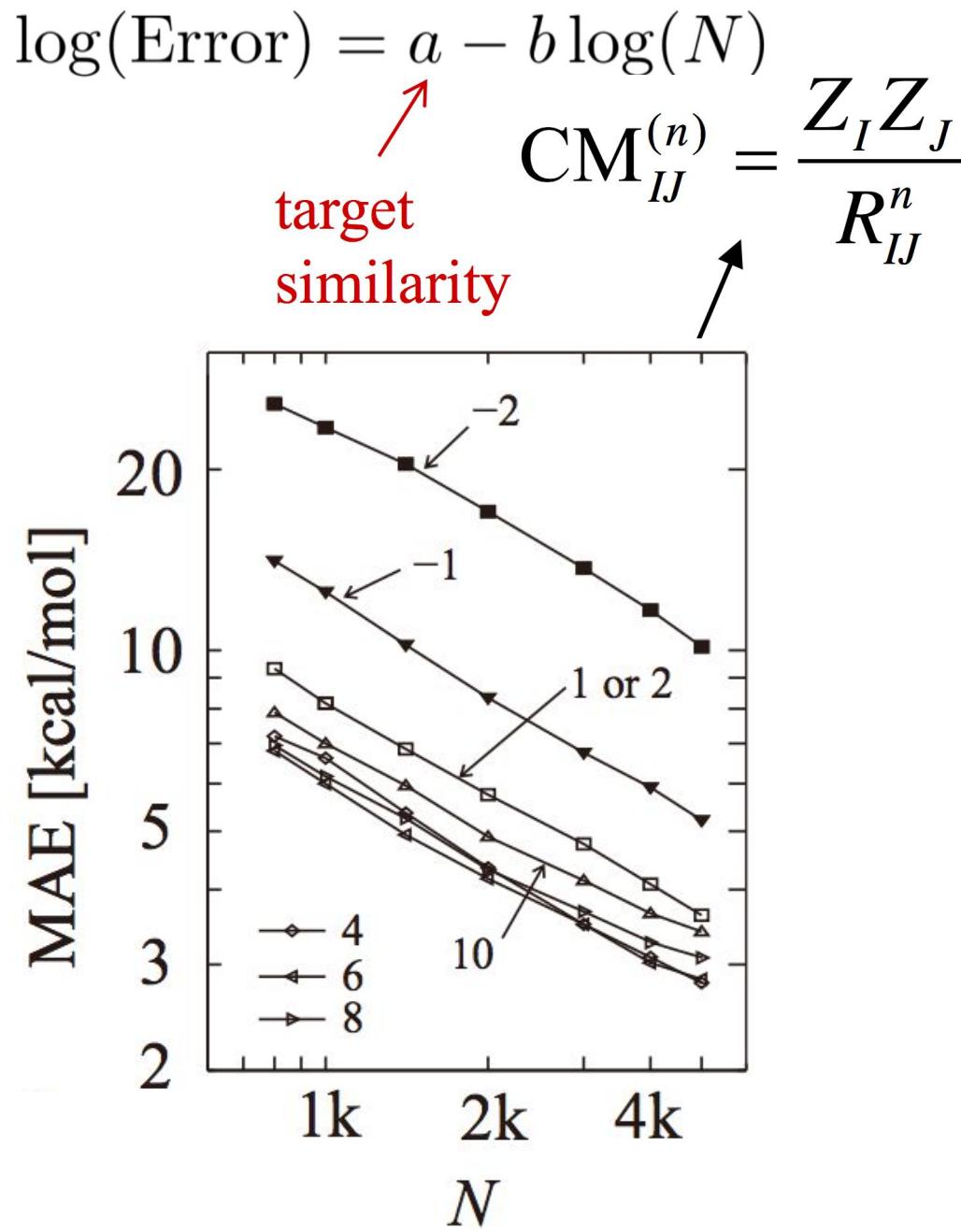
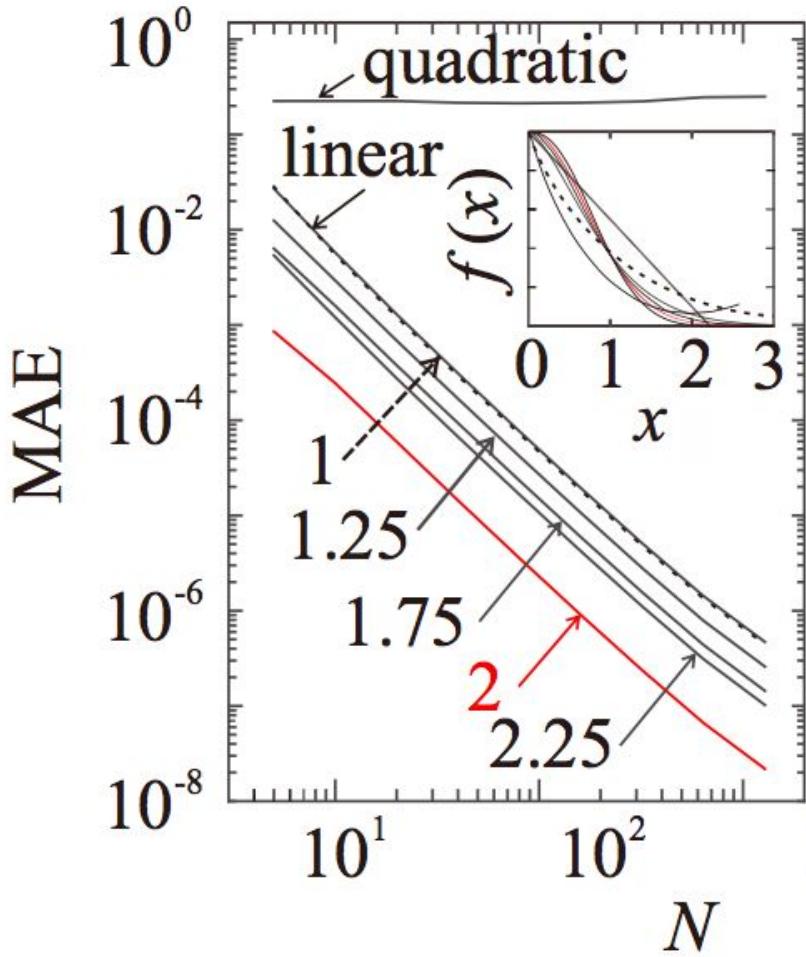
$$f^{\text{est}}(x) = \sum \alpha_i k(\underbrace{ax_i + b}_{M_i}, \underbrace{ax + b}_M)$$

target
similarity



Representation

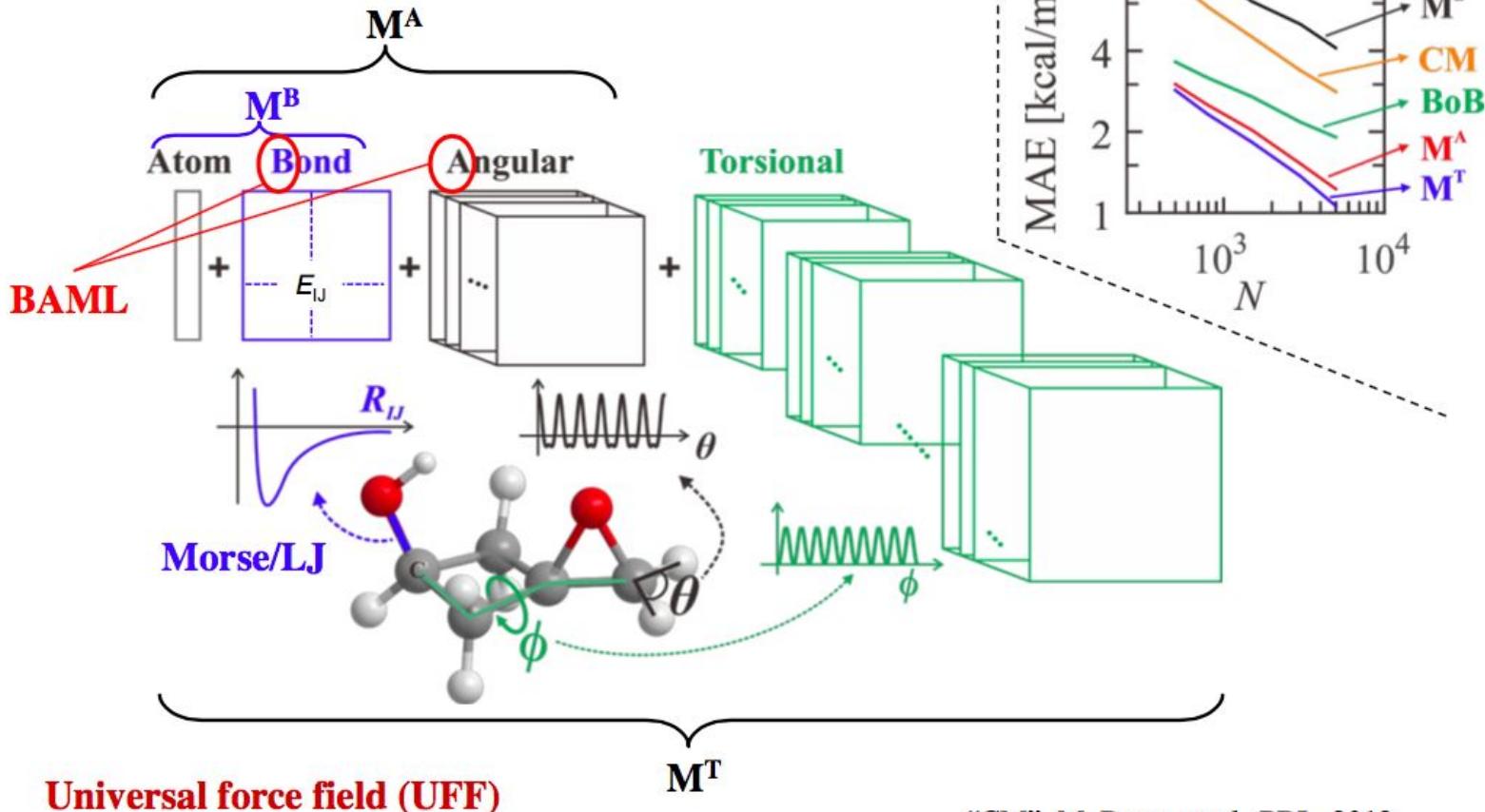
$$f^{\text{est}}(x) = \sum \alpha_i k(\underbrace{ax_i + b}_{M_i}, \underbrace{ax + b}_M)$$



BAML

Approach: best M is unique AND good model

bags of UFF contributions

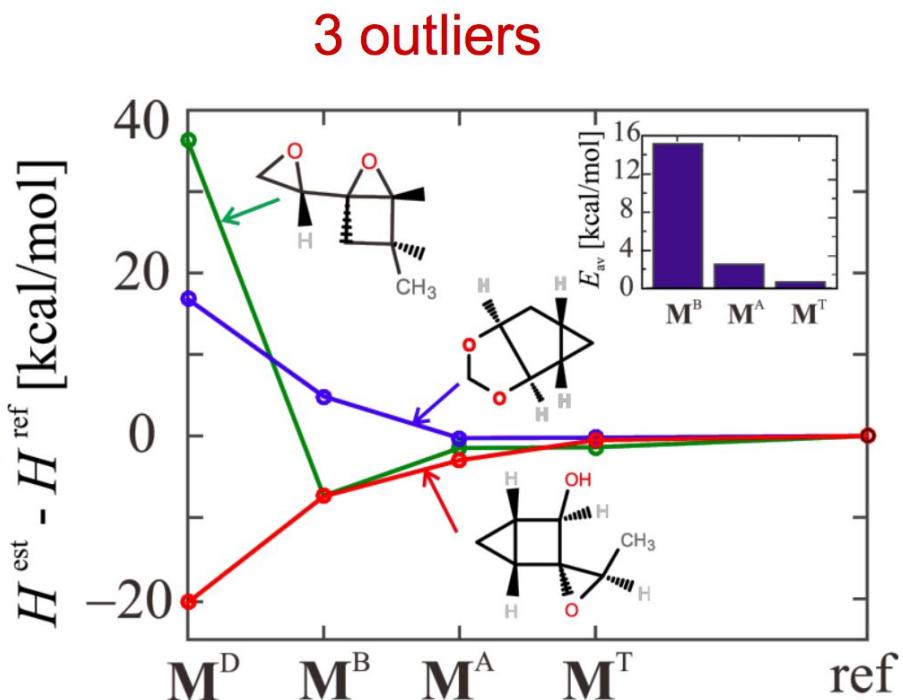
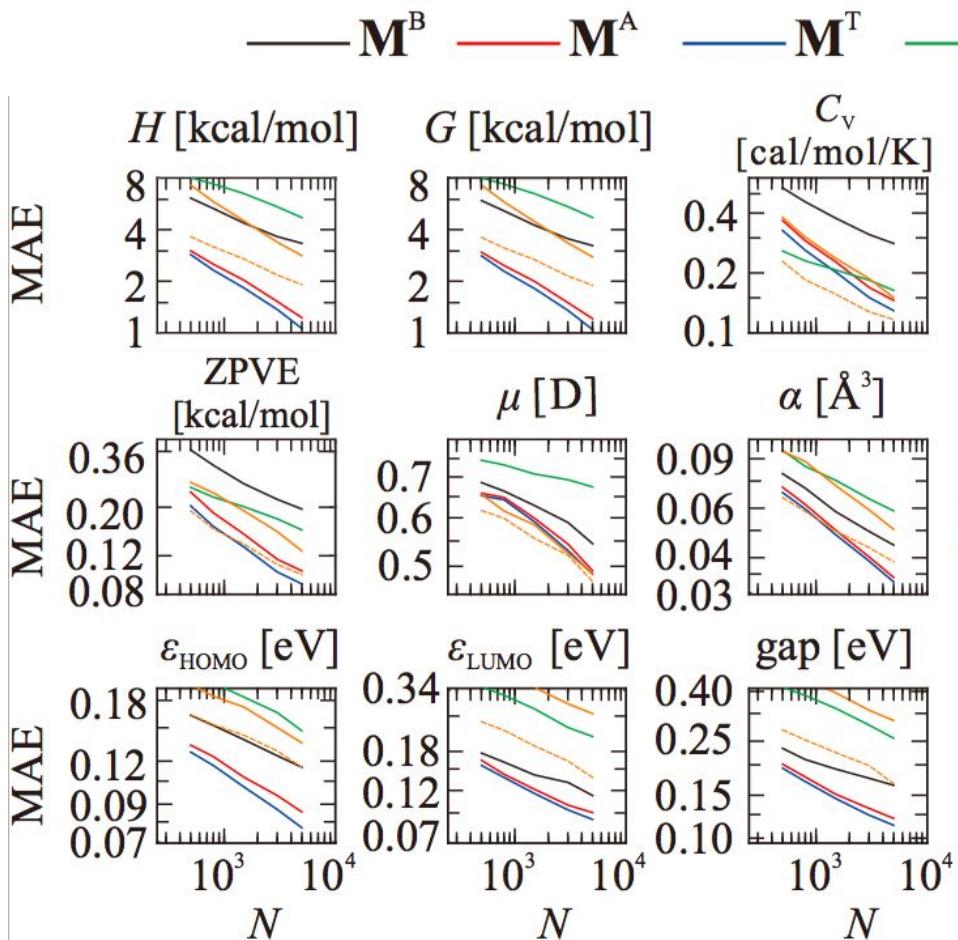


Universal force field (UFF)

A. K. Rappe, *et al.*, JACS, 1992

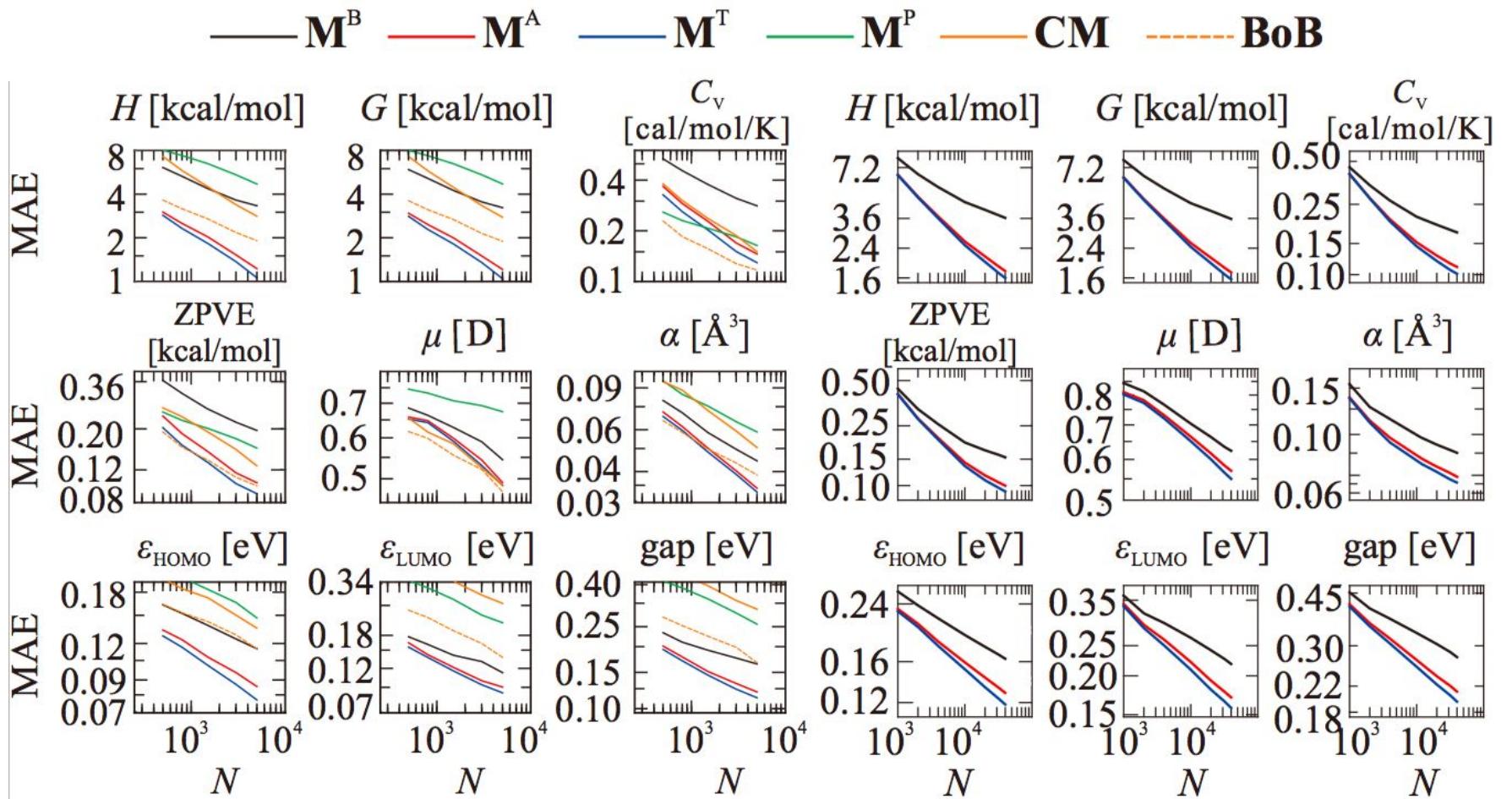
"CM", M. Rupp, *et al.*, PRL, 2012
"BoB", K. Hansen, *et al.*, JPCL, 2015

BAML



6k constitutional isomers of $\text{C}_7\text{O}_2\text{H}_{10}$

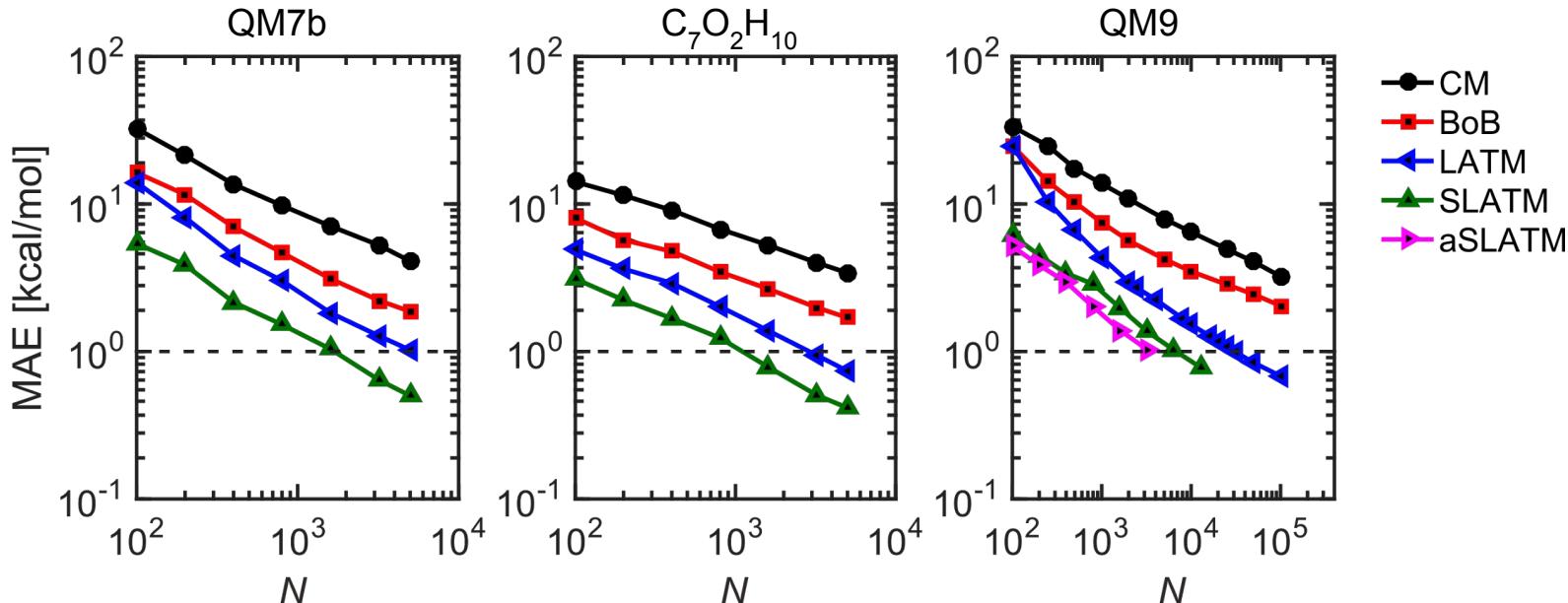
BAML



6k constitutional isomers of $\text{C}_7\text{O}_2\text{H}_{10}$

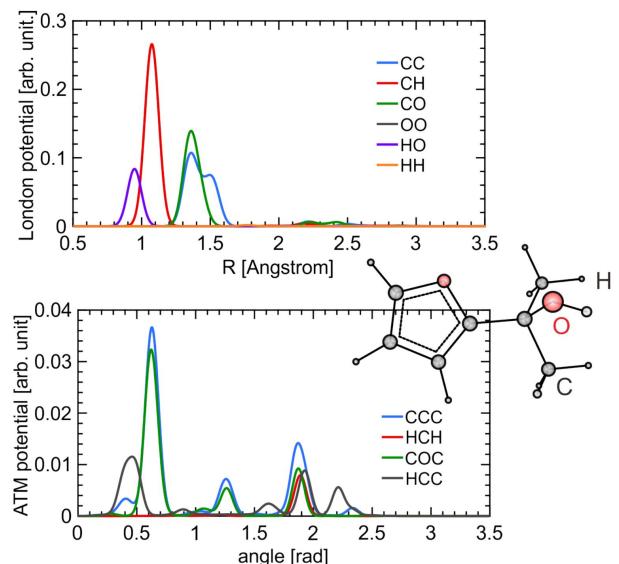
QM9 (134k molecules)

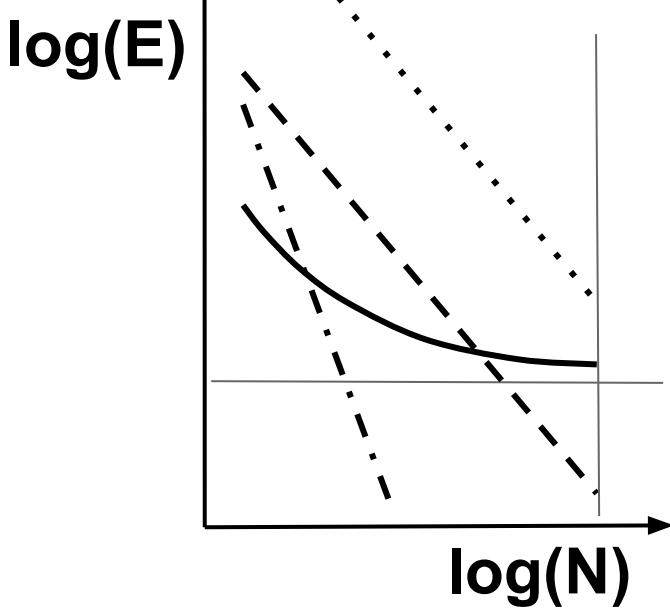
Atoms + London + Axilrod-Teller-Muto (LATM)



$$E^{(2)}(\mathbf{R}_I, \mathbf{R}_J) = -\frac{C_{6IJ}}{R_{IJ}^6}$$

$$E^{(3)}(\mathbf{R}_I, \mathbf{R}_J, \mathbf{R}_K) = C_{9_{IJK}} \frac{3 \cos[\phi_I] \cos[\phi_J] \cos[\phi_K] + 1}{R_{IJ}^3 R_{IK}^3 R_{JK}^3}$$



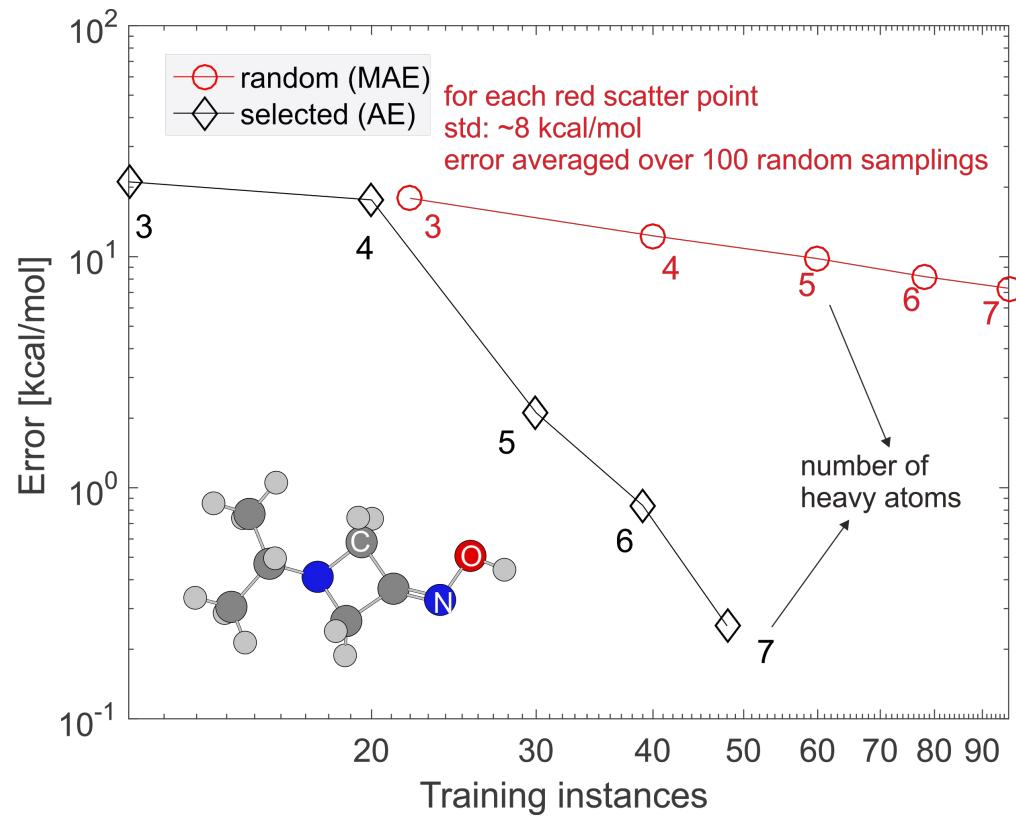


$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

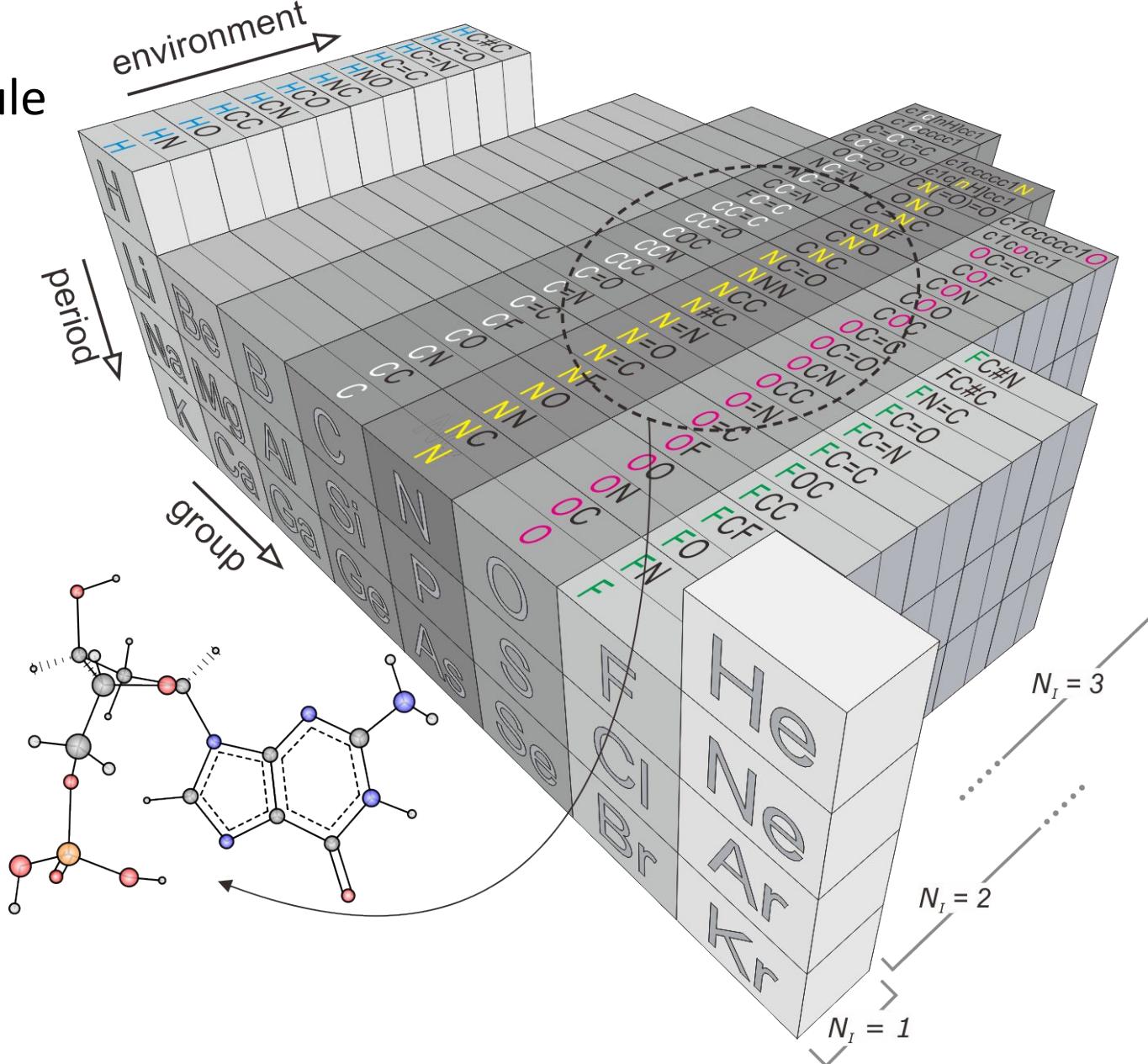
$$\vec{\alpha} = \mathbf{K}^{-1} \vec{P}^{\text{ref}}$$

The bigger the data the better ...

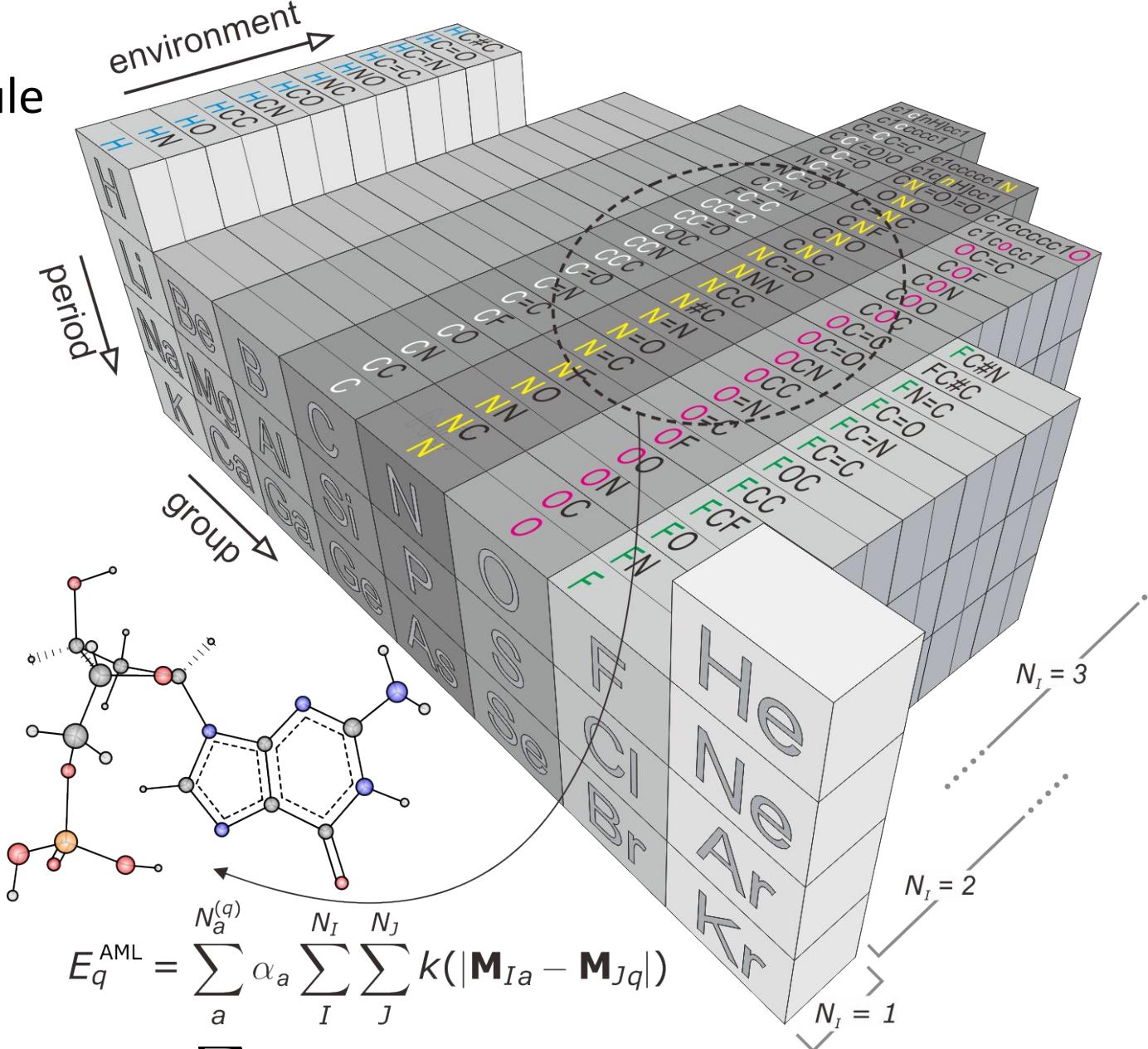
Vapnik, V., *The Nature of Statistical Learning Theory*, Springer (1995)



Atom in a Molecule “AM-on”

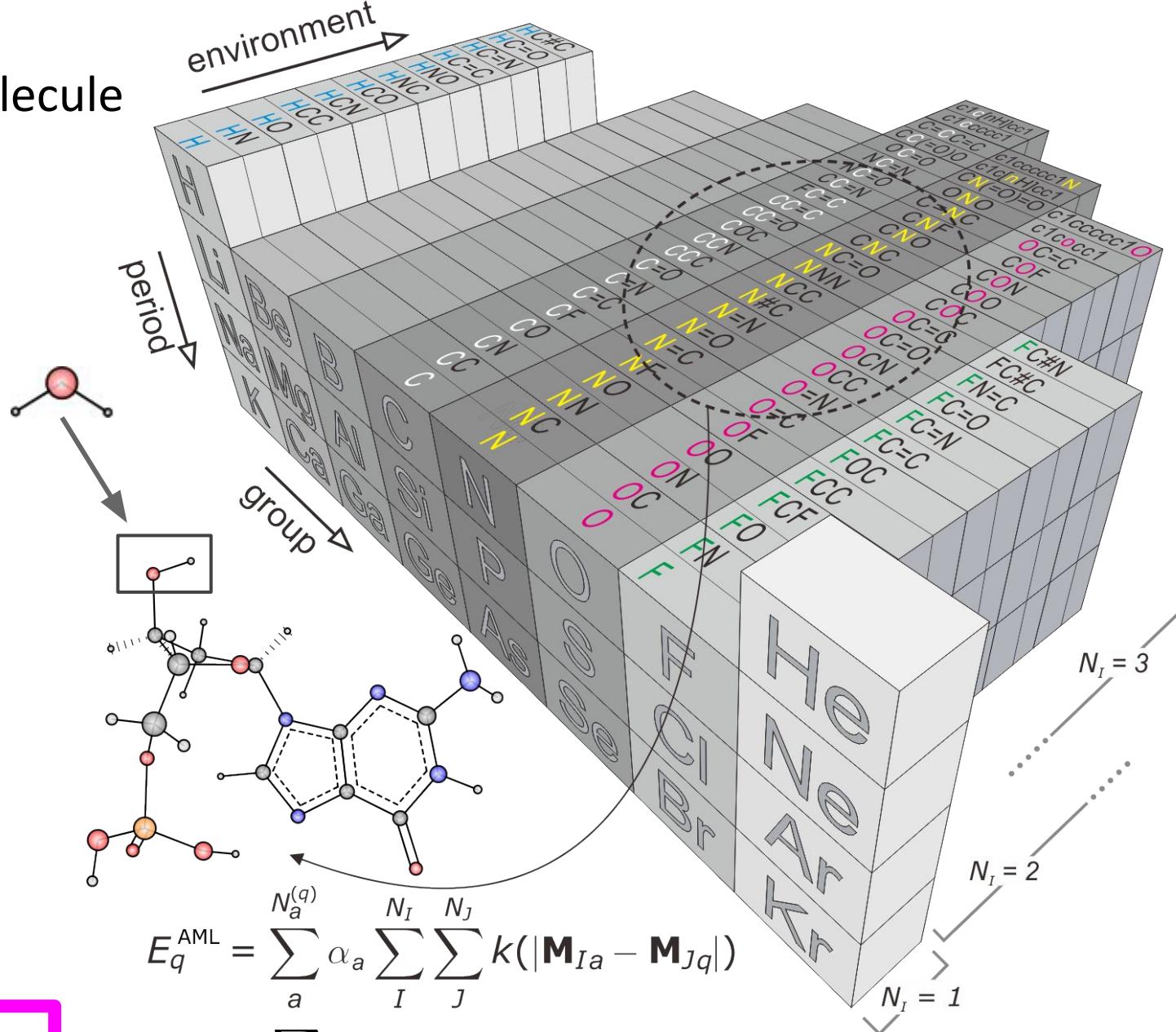


Atom in a Molecule “AM-on”



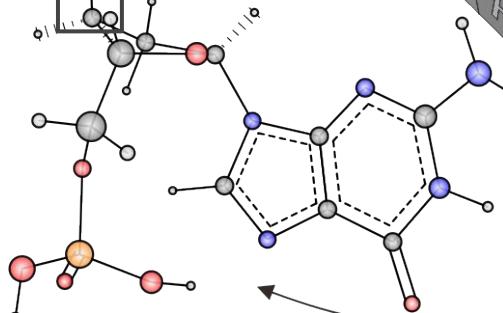
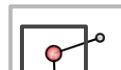
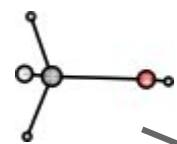
$$N_I < N_J$$

Atom in a Molecule “AM-on”



$$N_I < N_J$$

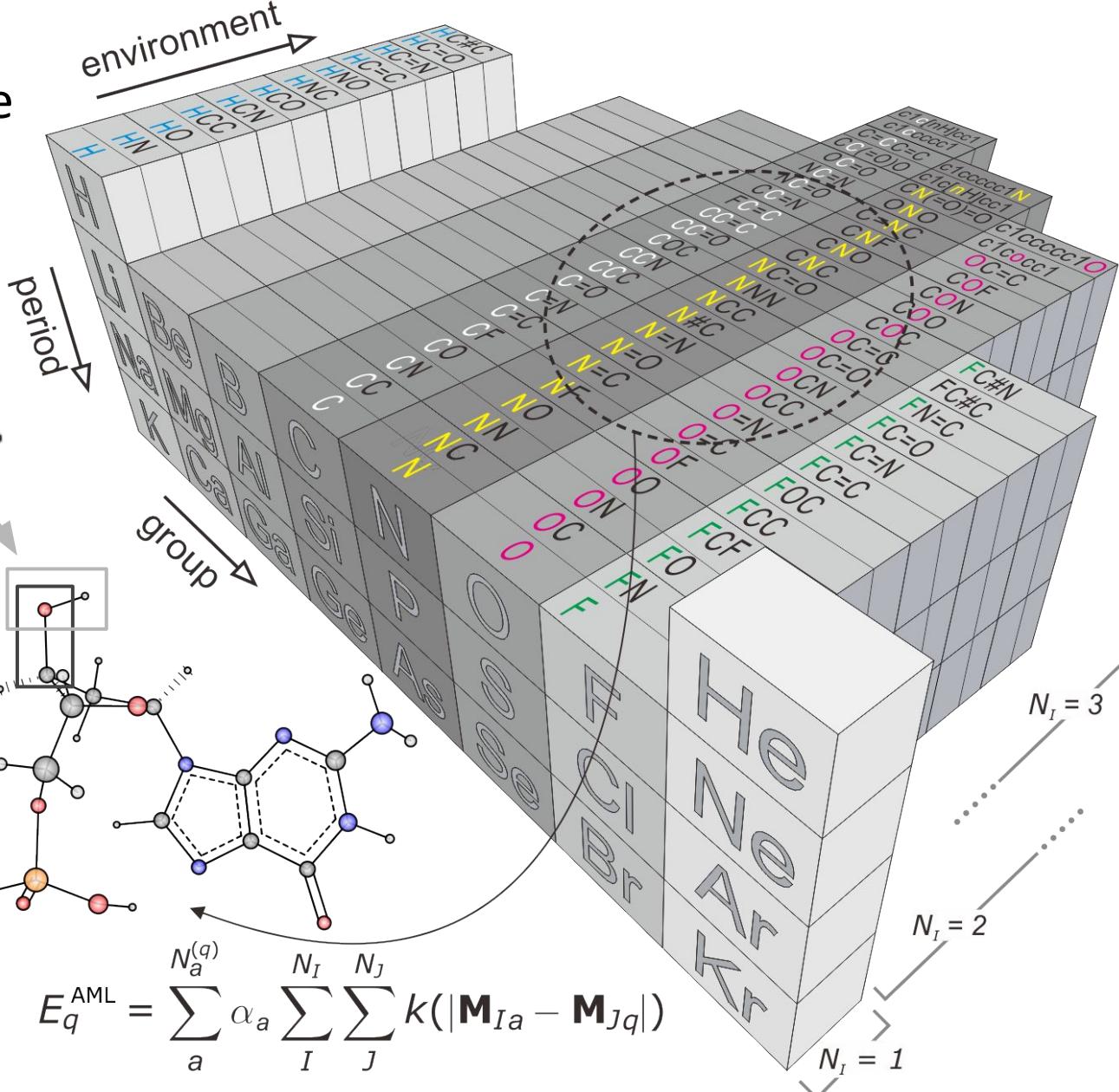
Atom in a Molecule “AM-on”



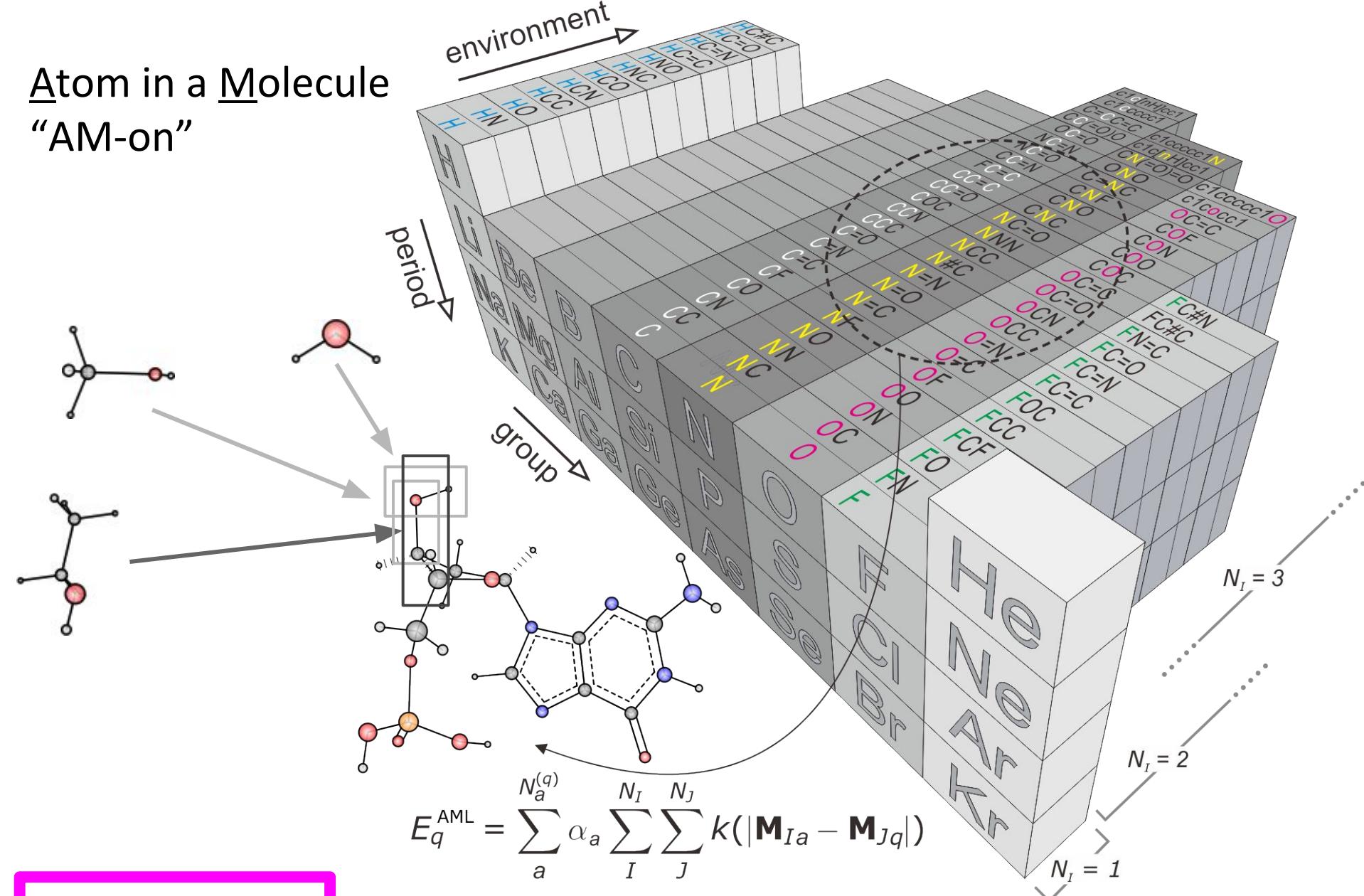
$$E_q^{\text{AML}} = \sum_a N_a^{(q)} \sum_I N_I \sum_J N_J k(|\mathbf{M}_{Ia} - \mathbf{M}_{Jq}|)$$

$$= \sum_J E_q^J$$

$$N_I < N_J$$

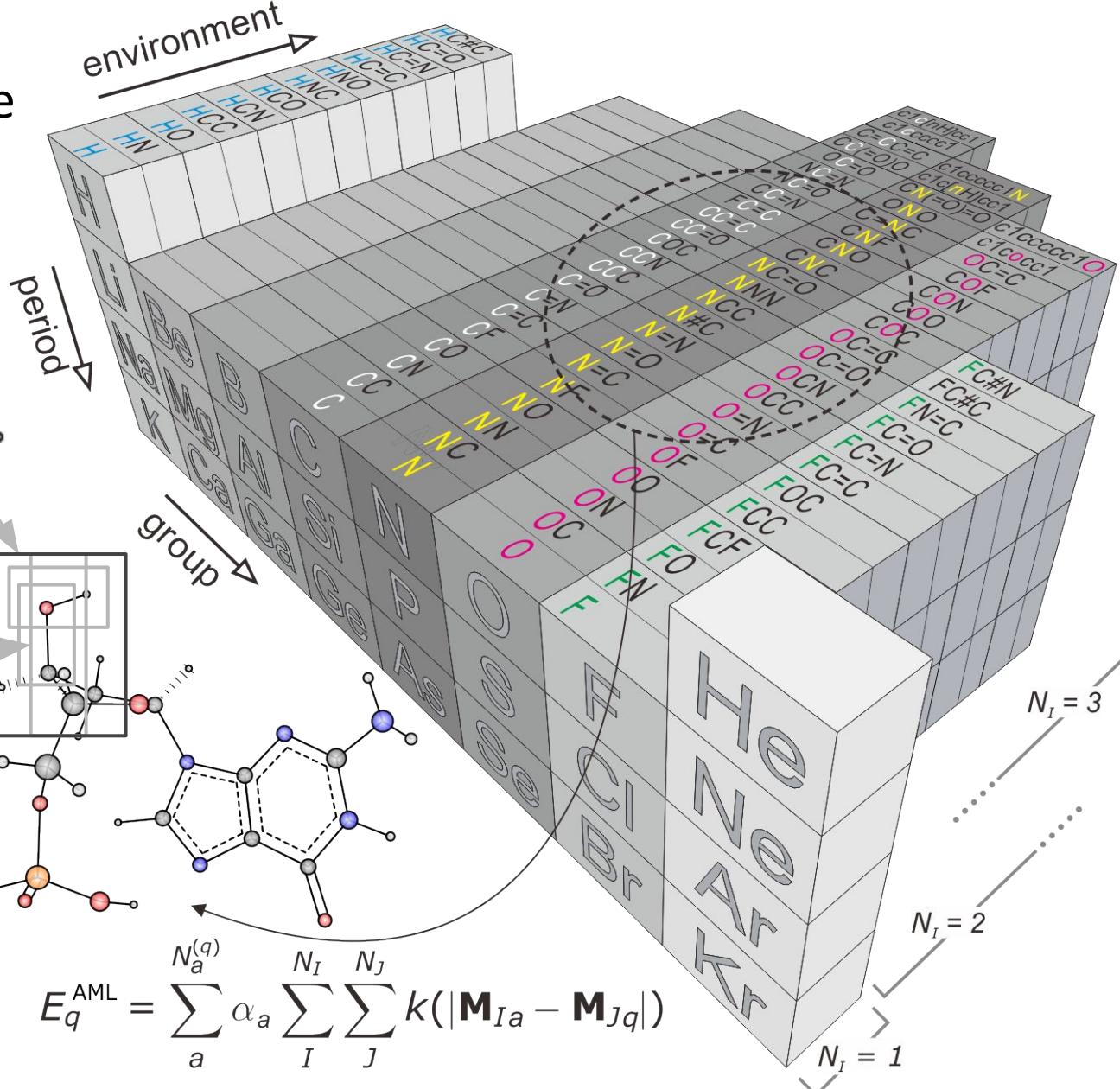
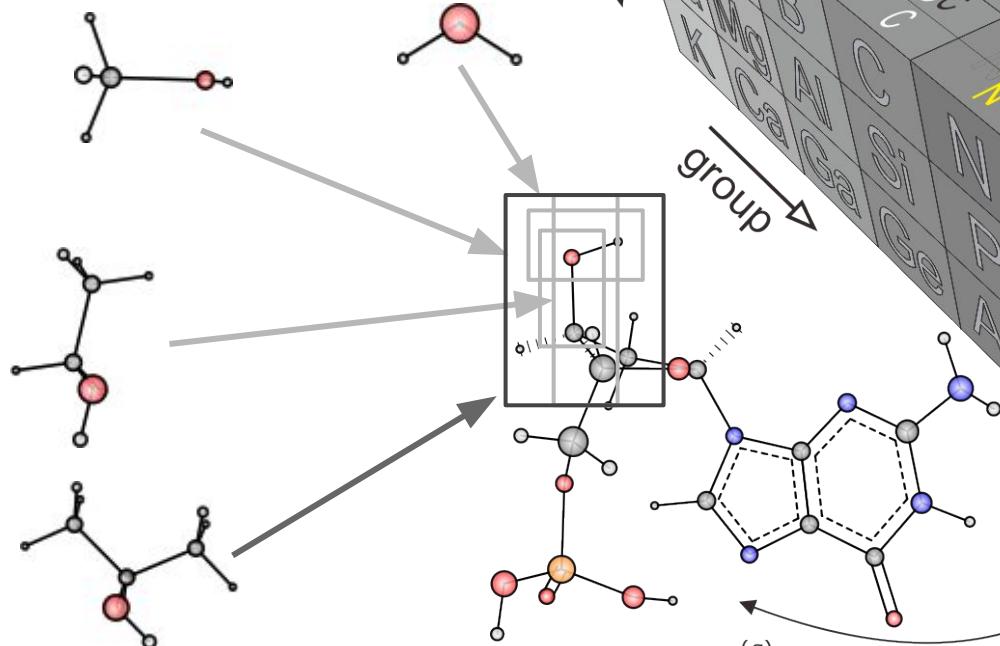


Atom in a Molecule “AM-on”



$$N_I < N_J$$

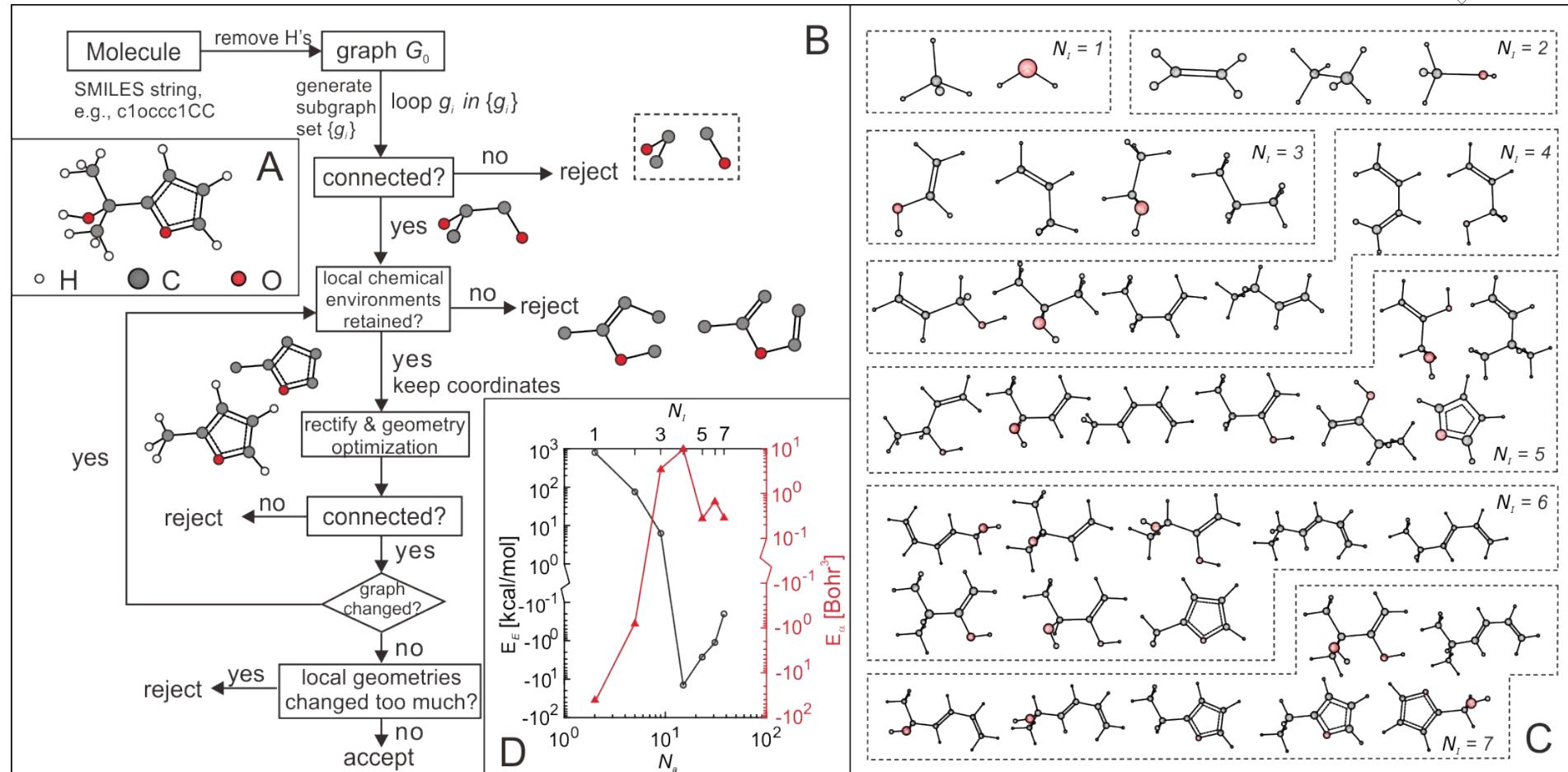
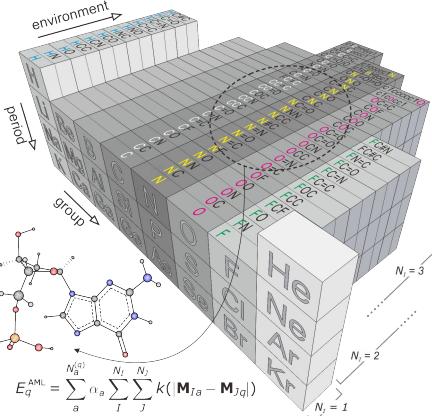
Atom in a Molecule “AM-on”

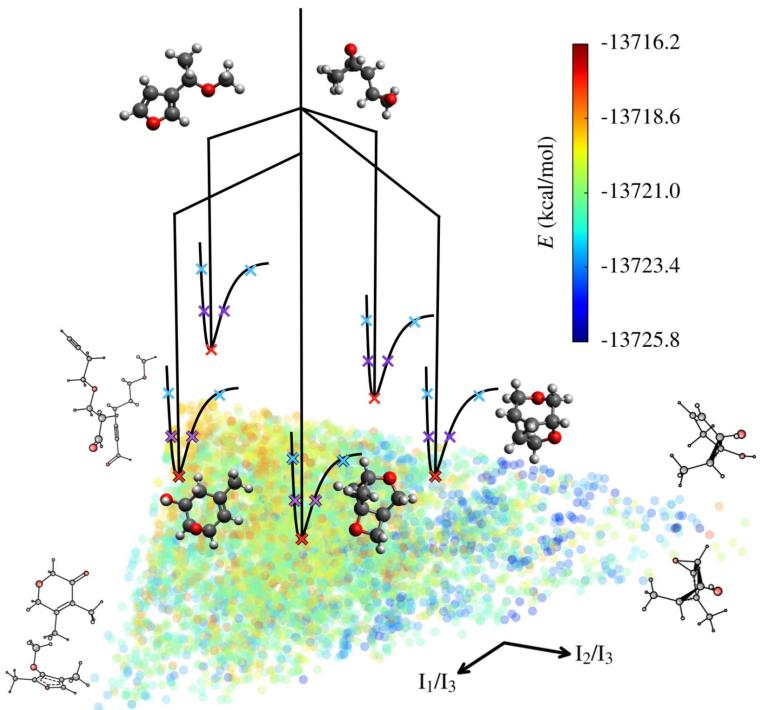
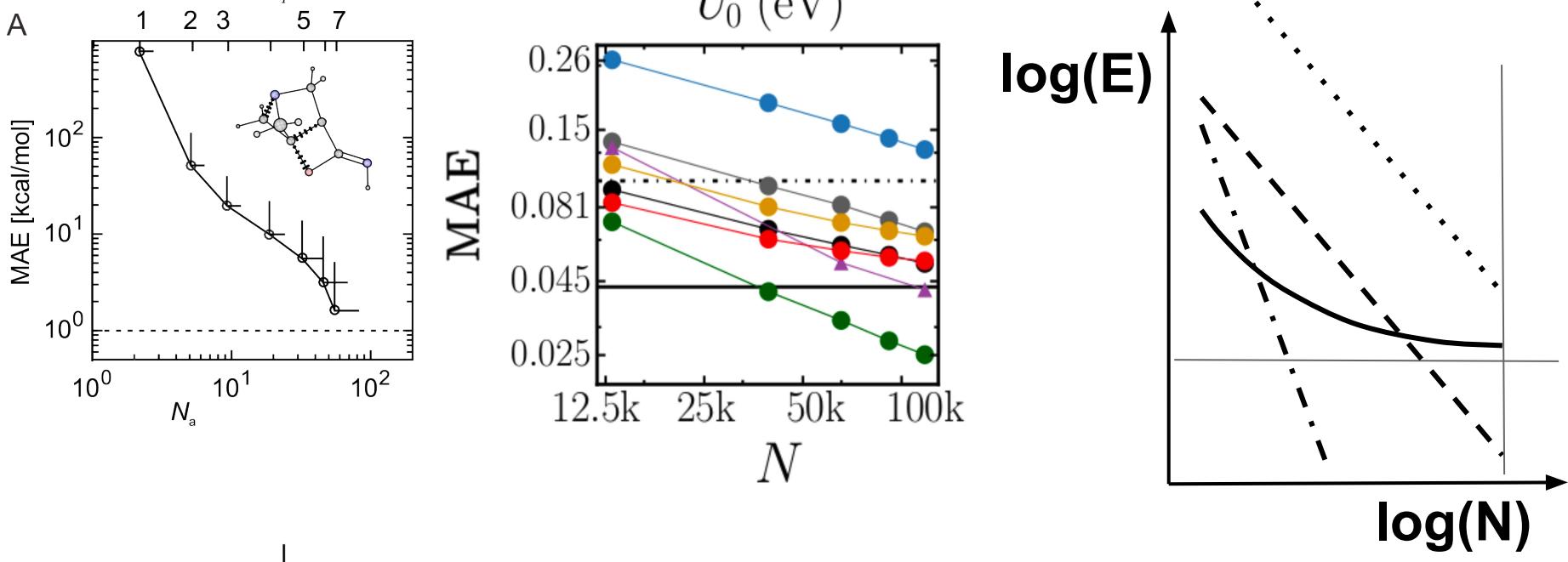


$$E_q^{\text{AML}} = \sum_a \alpha_a \sum_I \sum_J k(|\mathbf{M}_{Ia} - \mathbf{M}_{Jq}|)$$

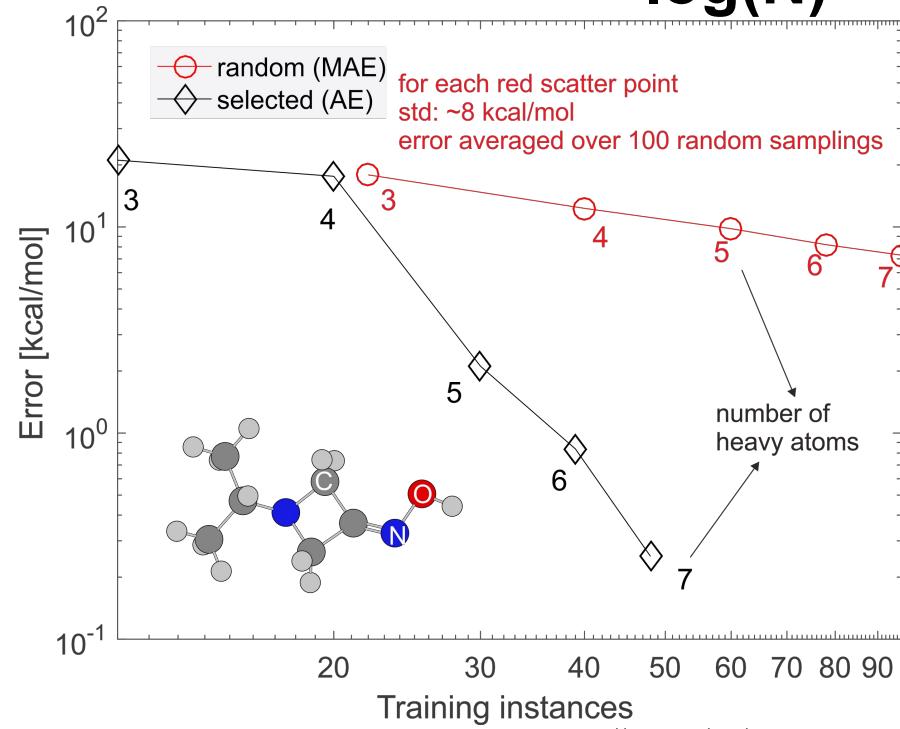
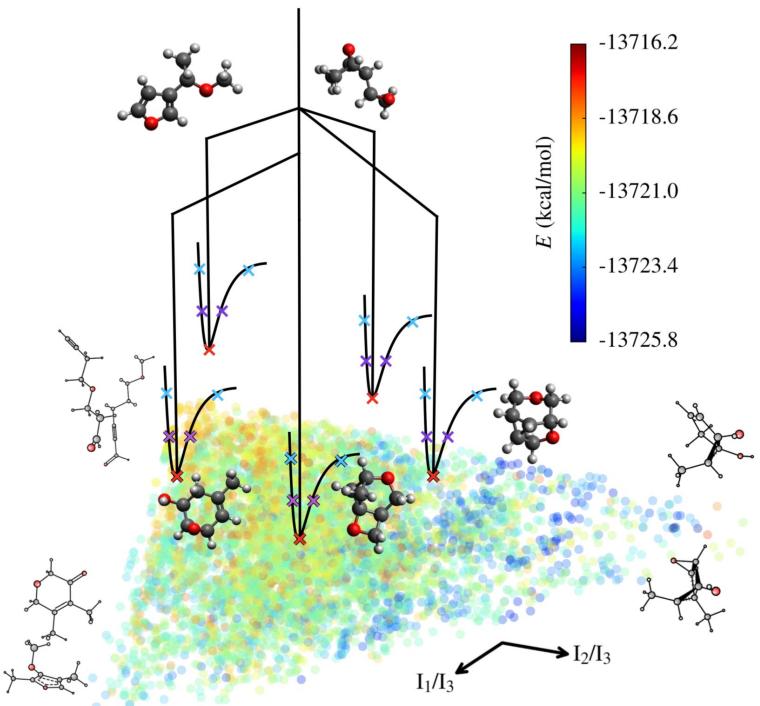
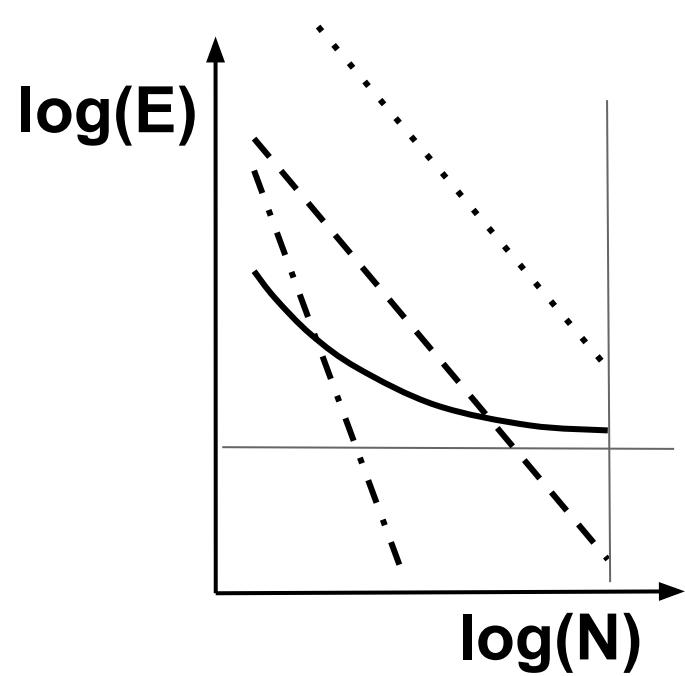
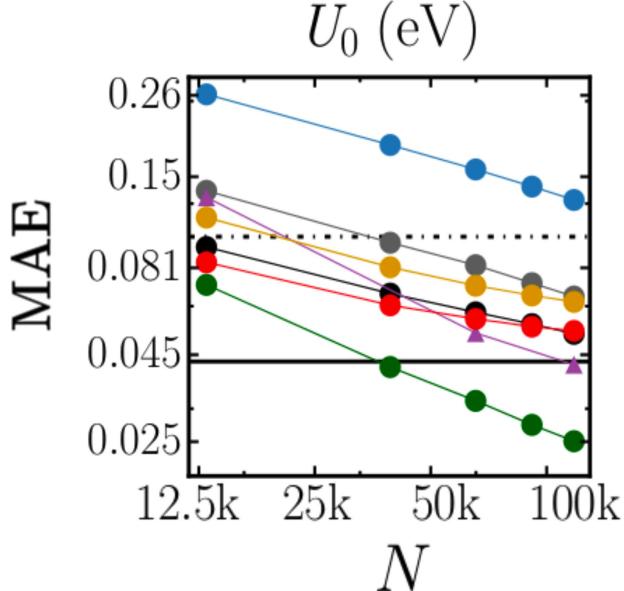
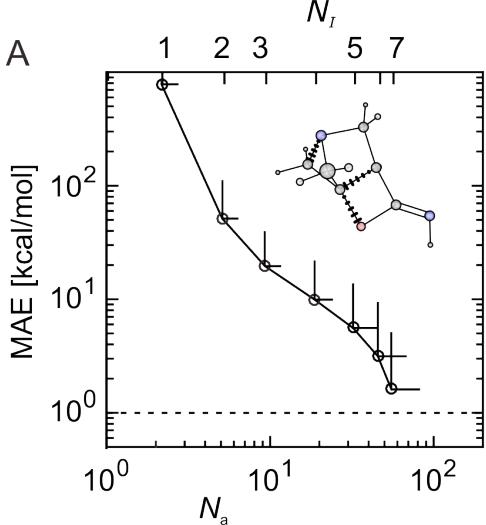
$$= \sum_J E_q^J$$

an atom in a molecule: “AM-on”

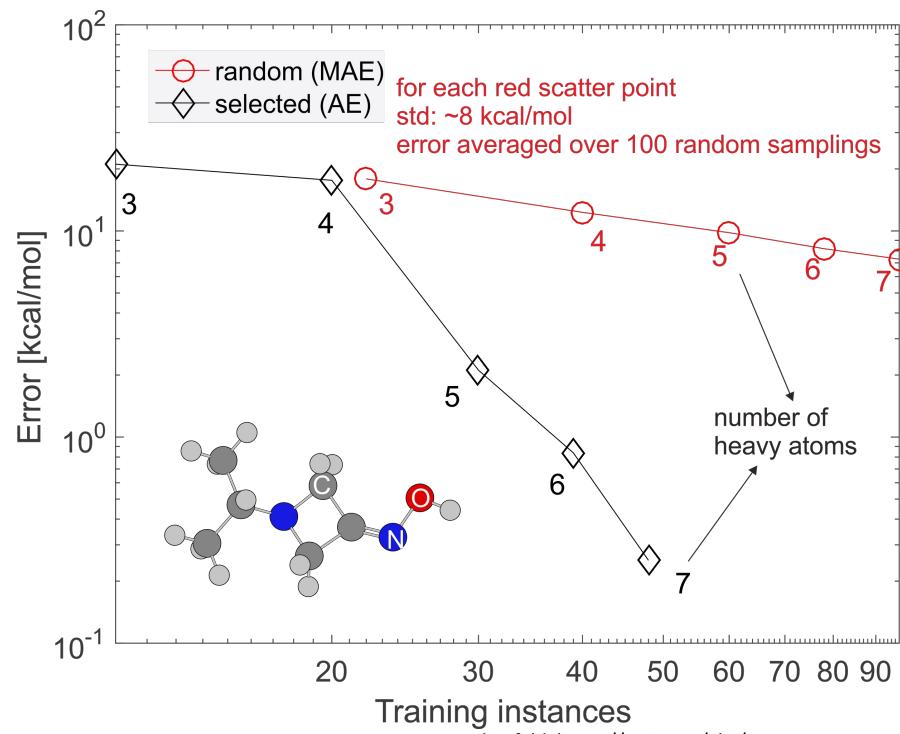
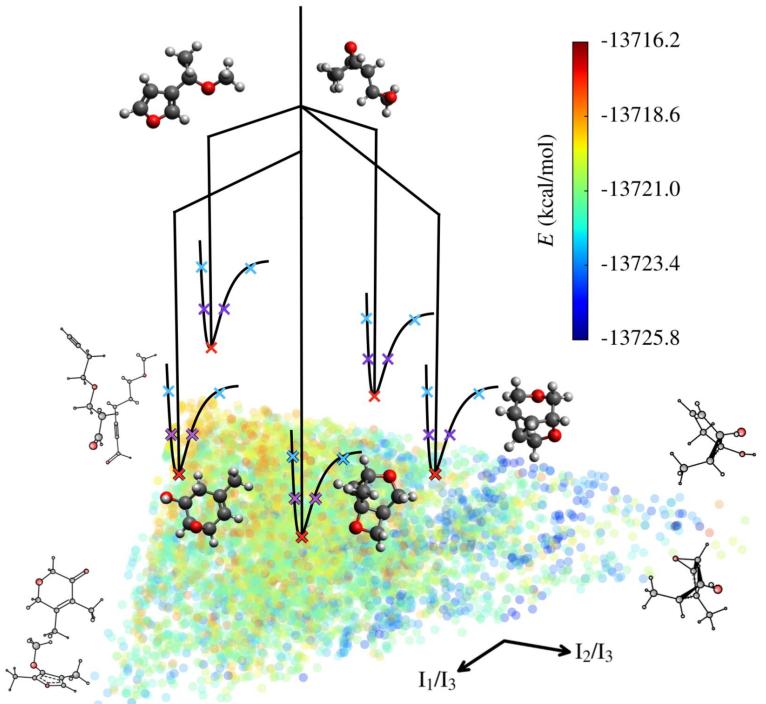
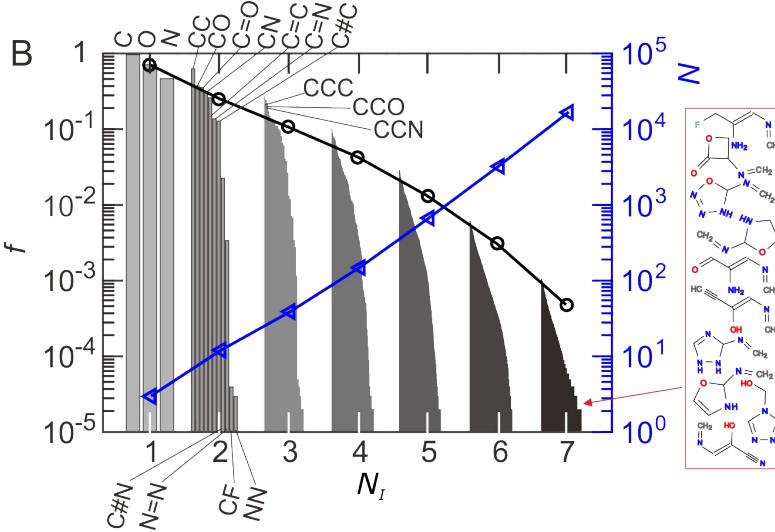
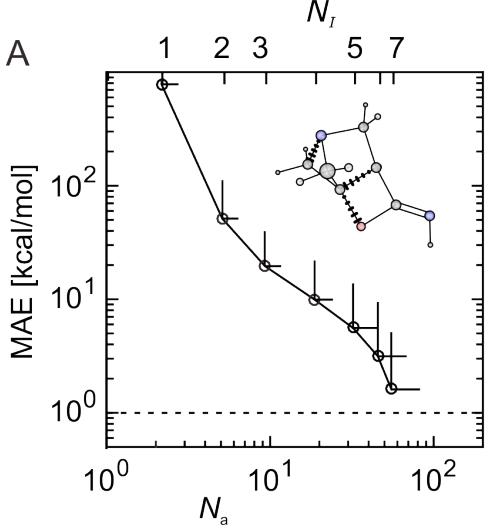




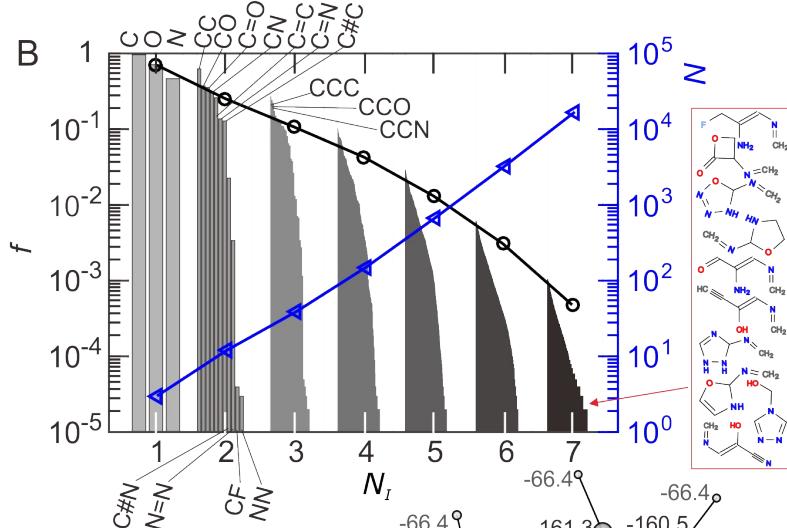
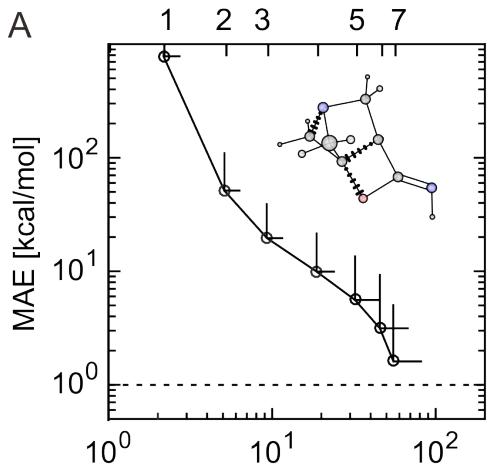
A



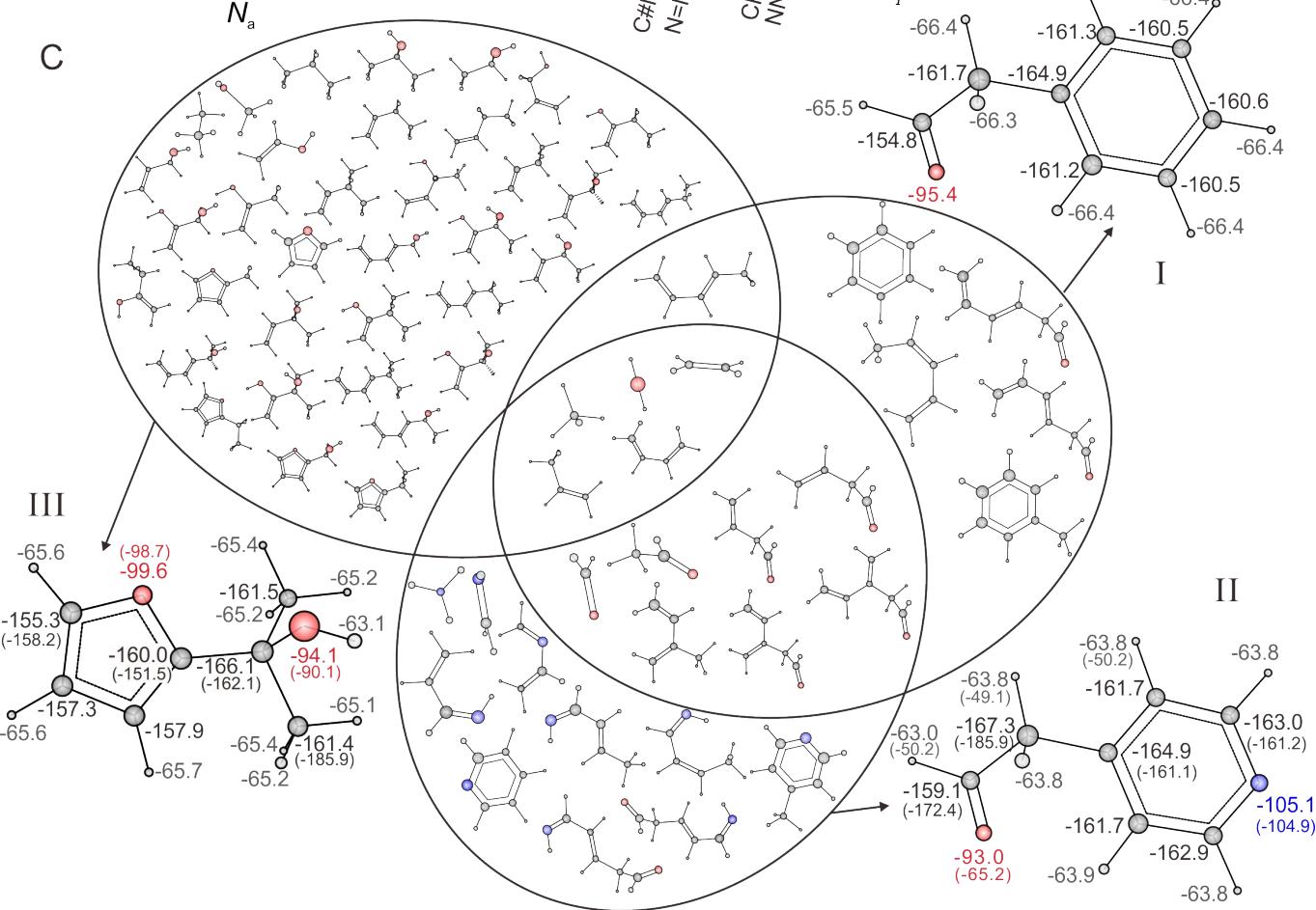
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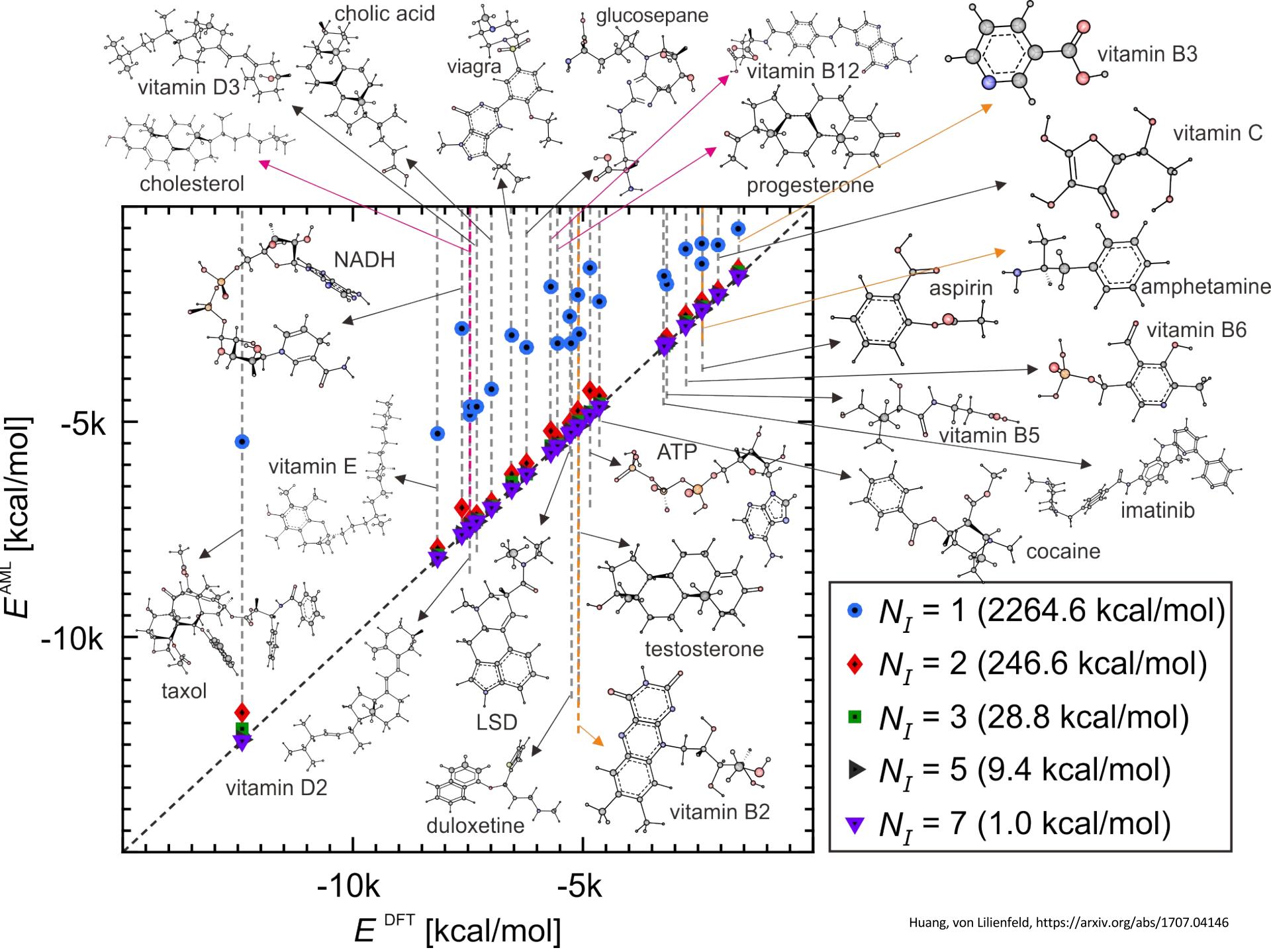


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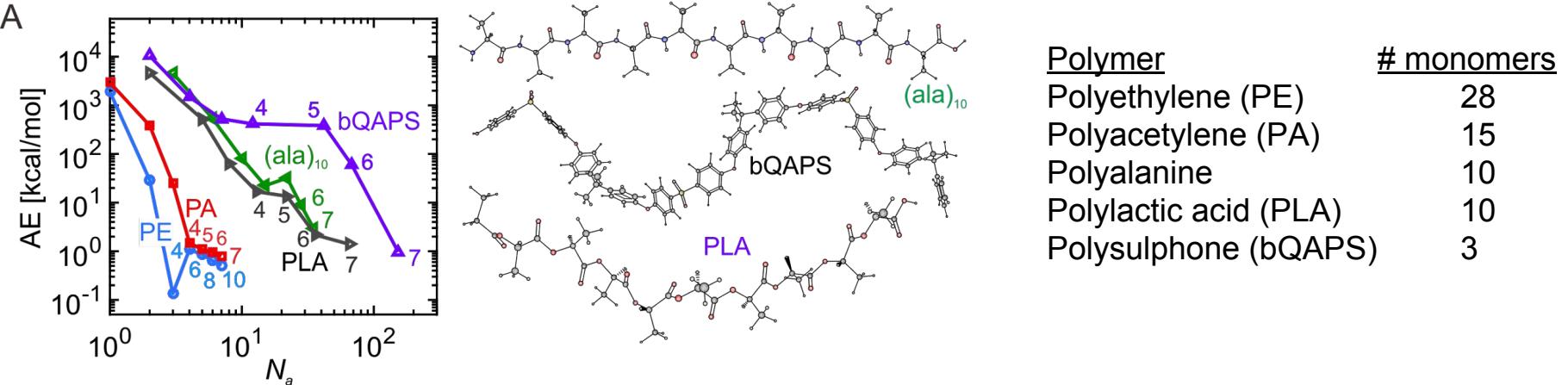


C

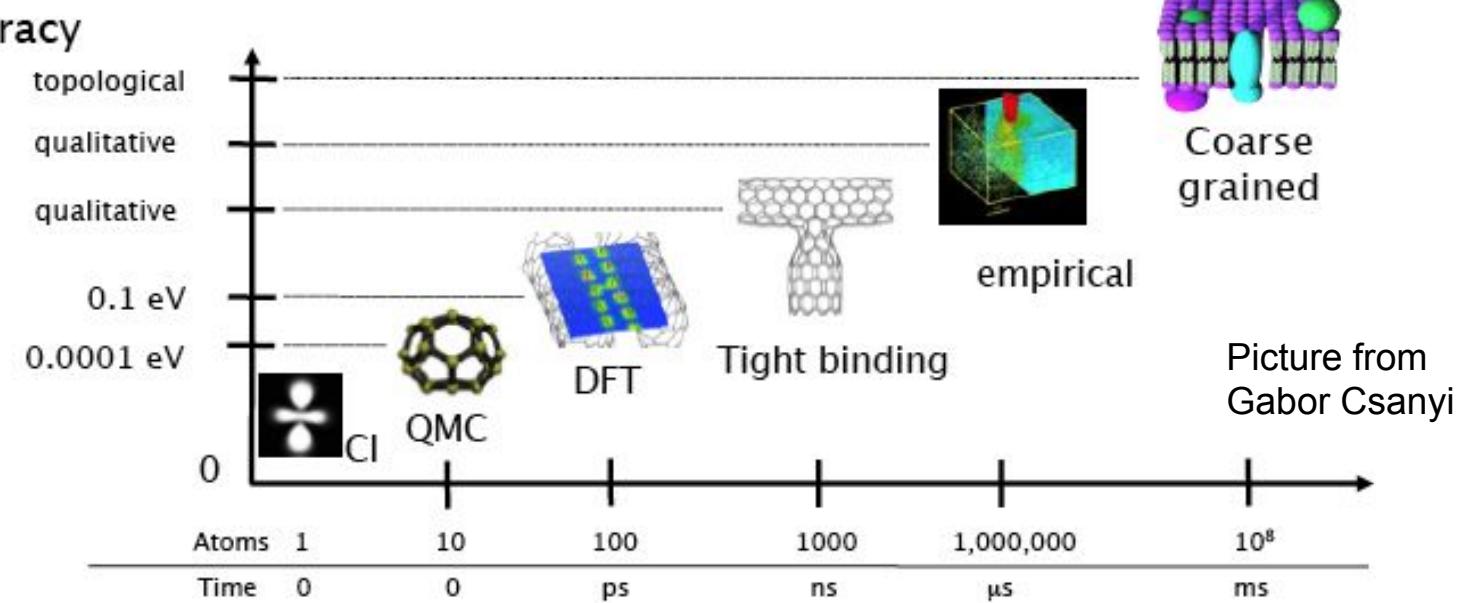


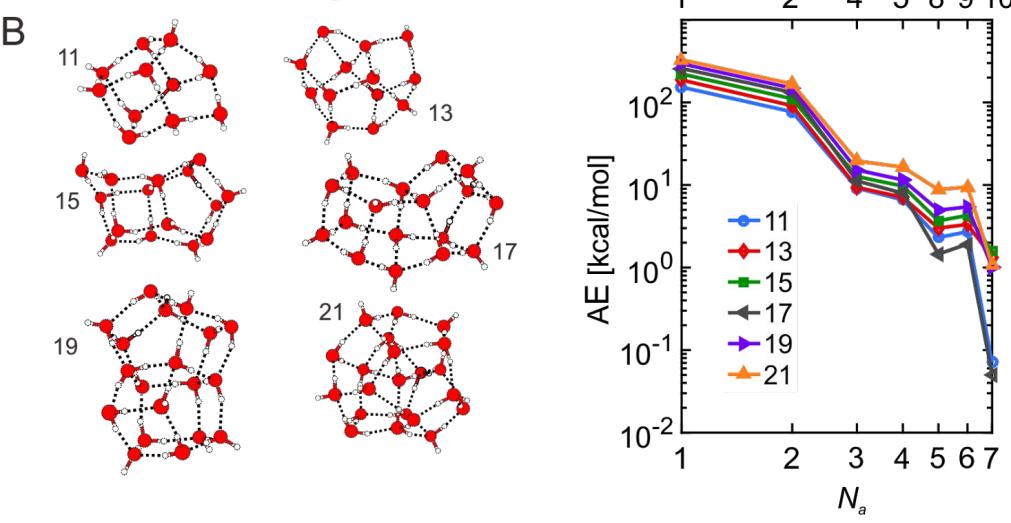
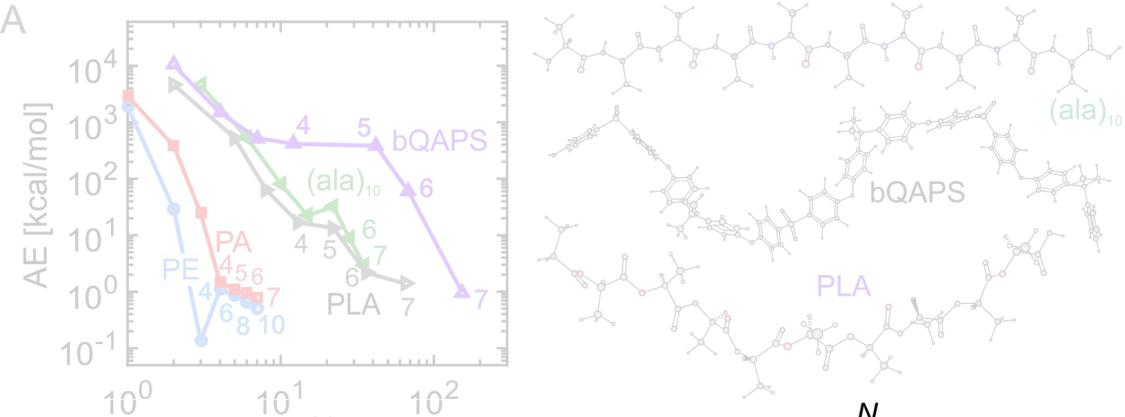


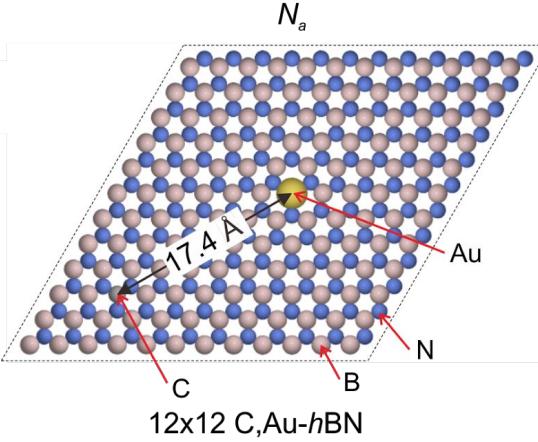
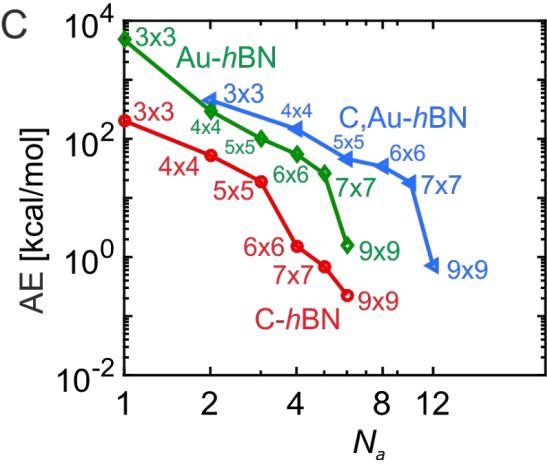
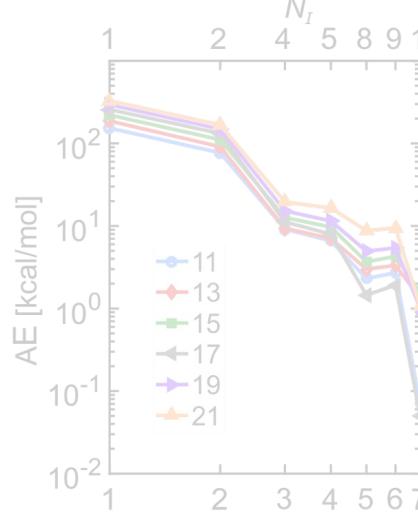
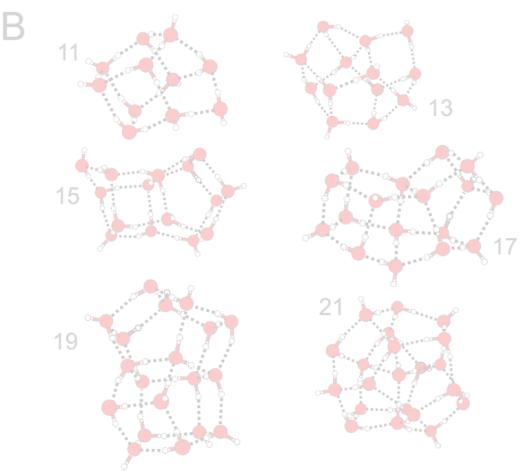
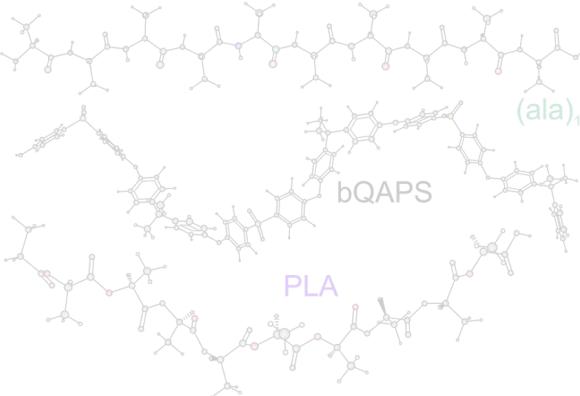
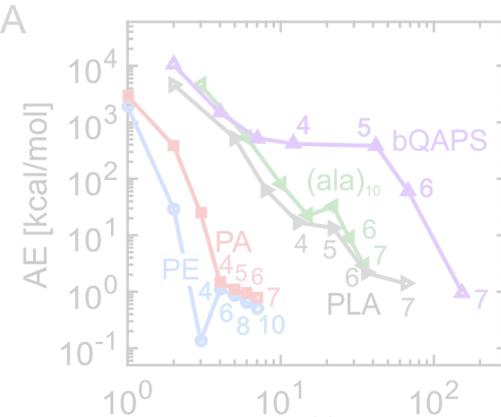
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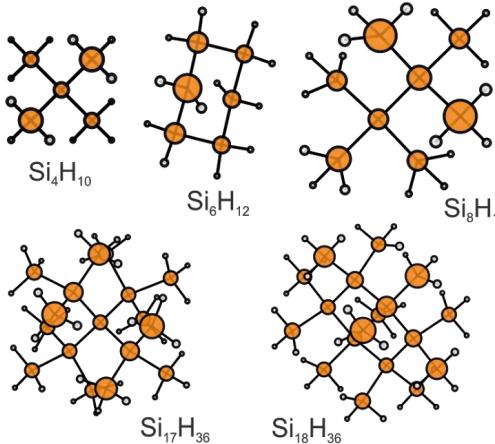
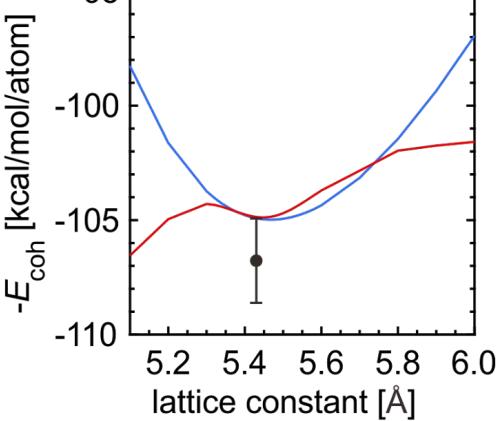
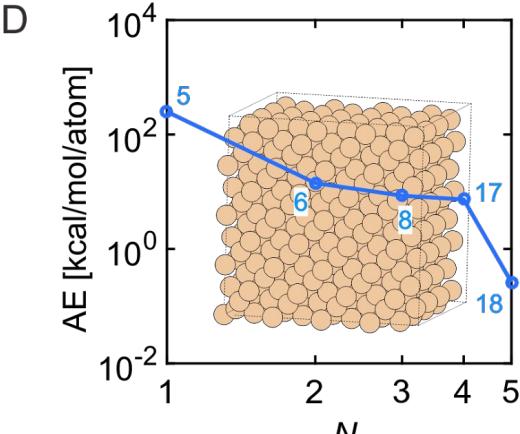
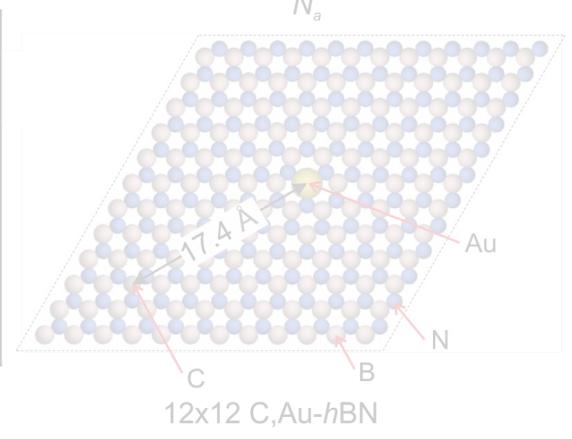
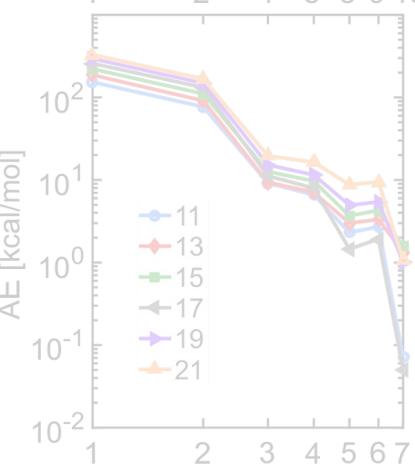
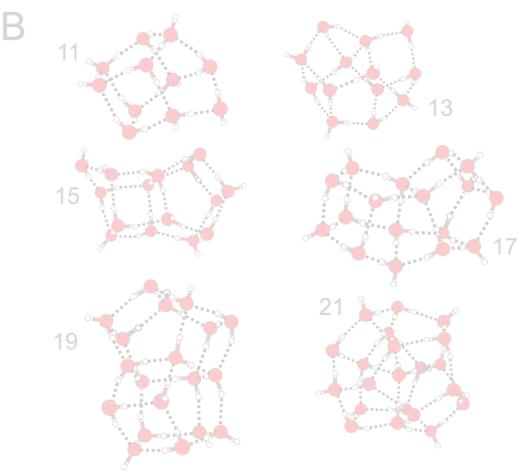
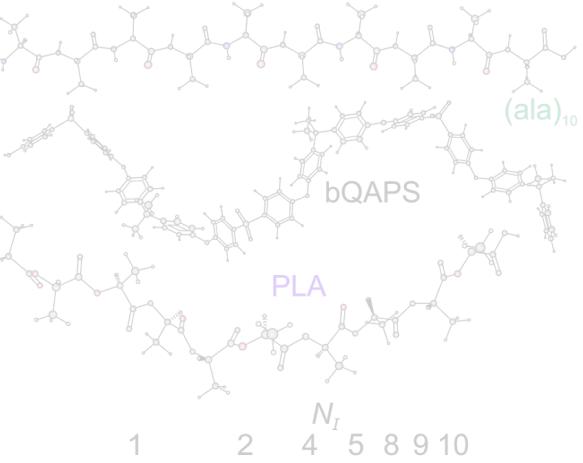
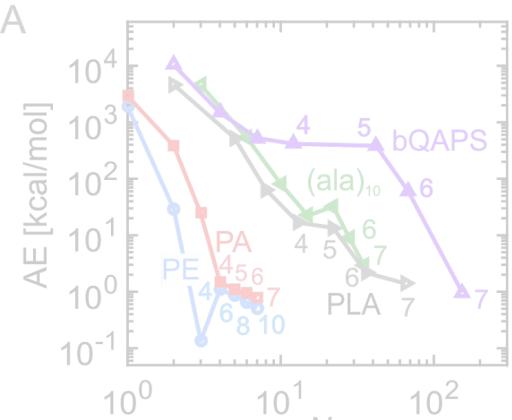


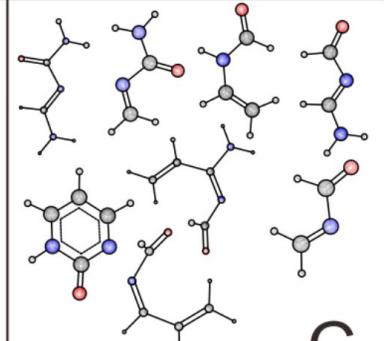
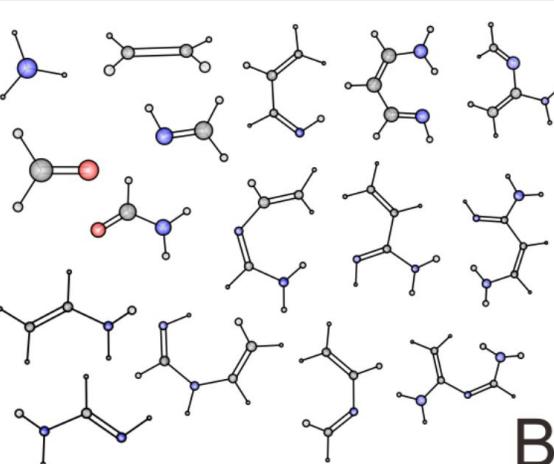
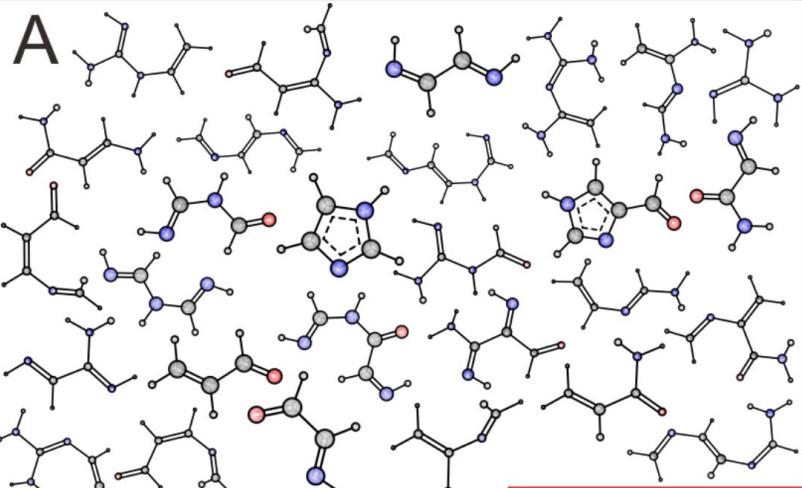
Accuracy



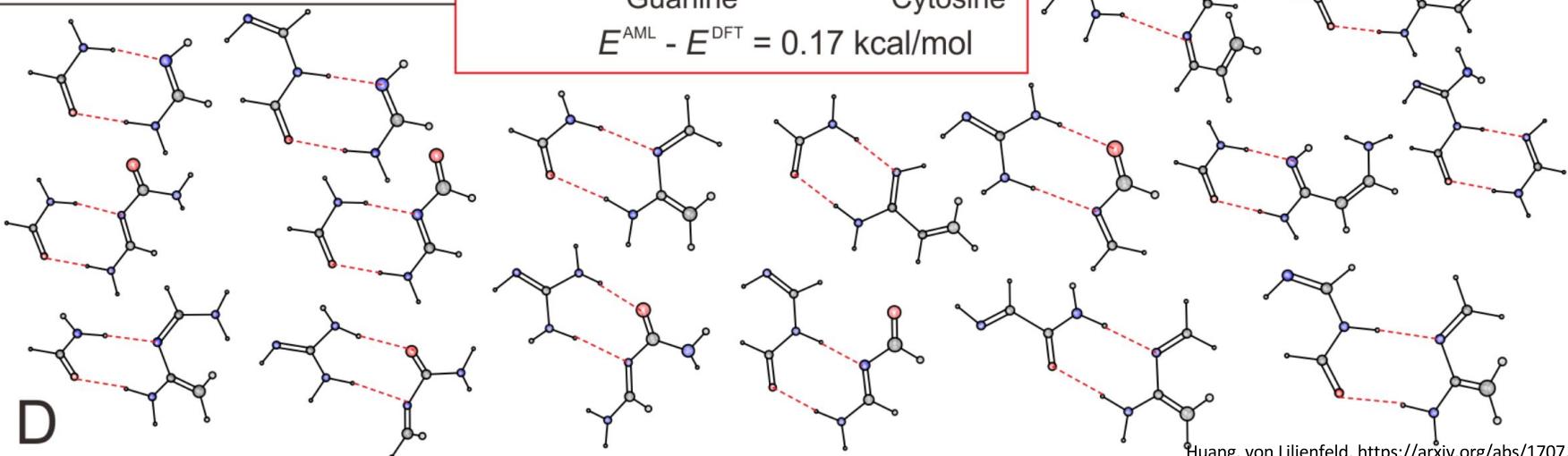
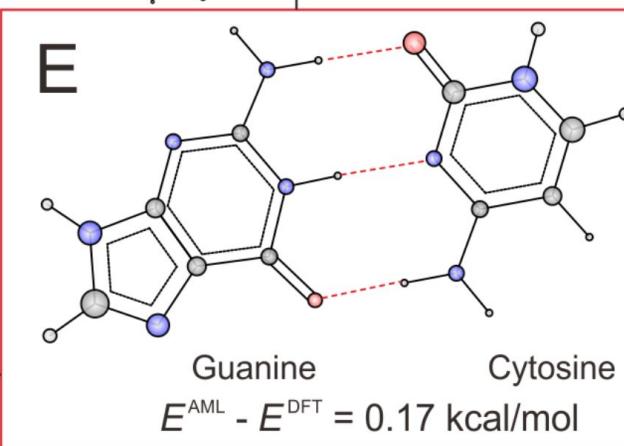




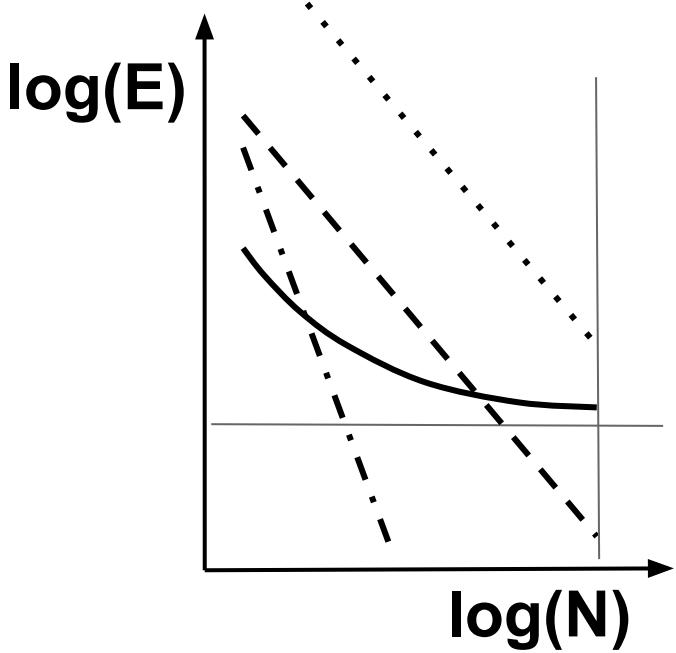




B



D



$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

$$\vec{\alpha} = \mathbf{K}^{-1} \vec{P}^{\text{ref}}$$

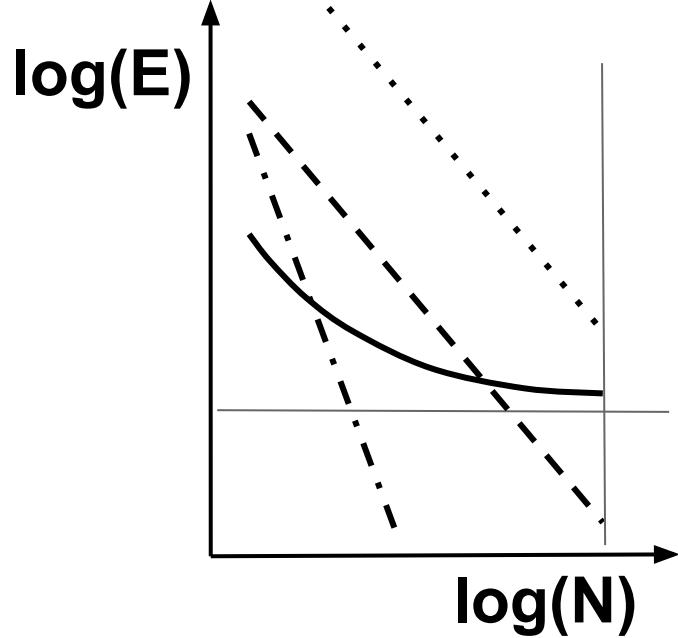
$$E_q = \langle \Psi_q | \hat{H} | \Psi_q \rangle$$

$$O_q = \langle \Psi_q | \hat{O} | \Psi_q \rangle$$

$$\kappa \sim \Psi$$

$$\alpha \sim \hat{O}$$

Ramakrishnan, OAvL, CHIMIA (2015)



$$P^{\text{est}}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

$$\vec{\alpha} = \mathbf{K}^{-1} \vec{P}^{\text{ref}}$$

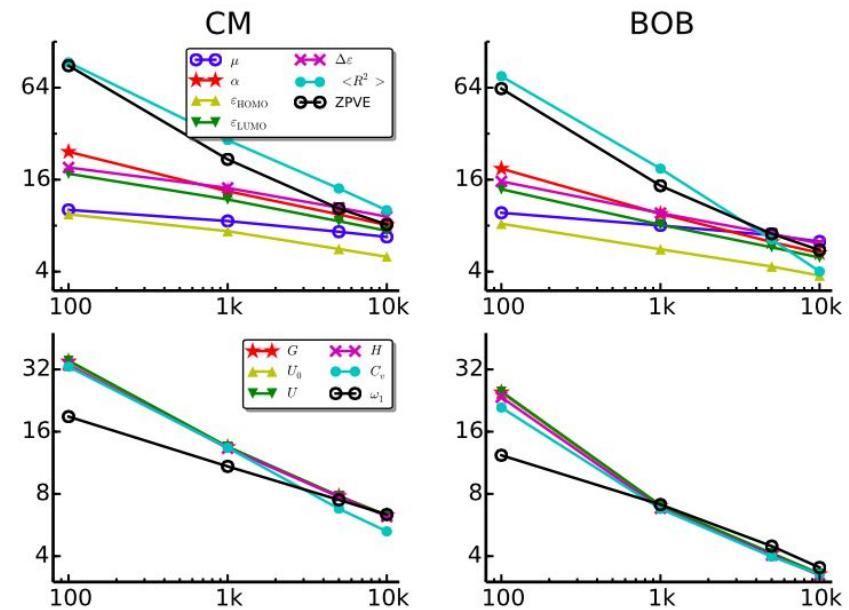
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$$\kappa \sim \Psi$$

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Ramakrishnan, OAvL, CHIMIA (2015)



Ramakrishnan, OAvL, CHIMIA (2015)

Final remarks

correlations (inductive) vs. **laws** (deductive)

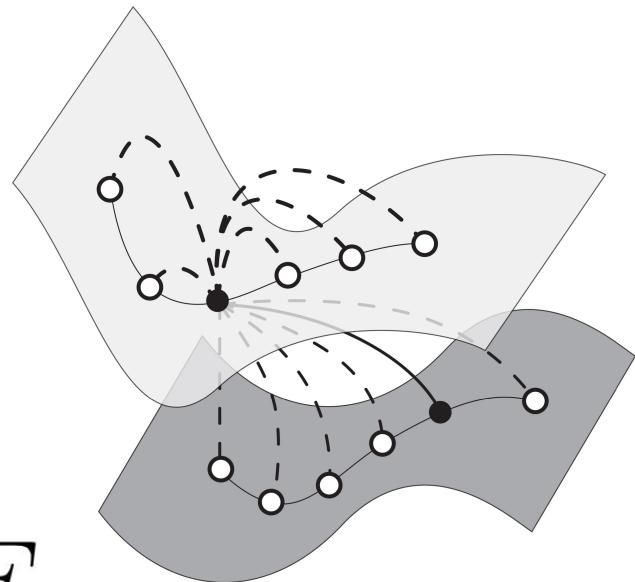
$$H(\{Z_I, \mathbf{R}_I\})\Psi(\mathbf{r}) = E\Psi(\mathbf{r})$$

Erwin



$$H(\{Z_I, \mathbf{R}_I\}) \xrightarrow{\Psi} E$$

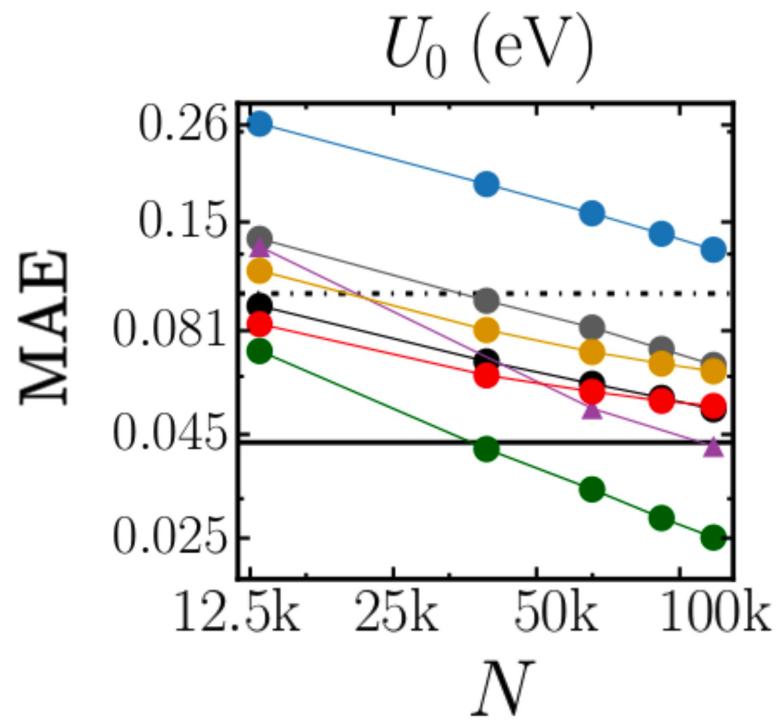
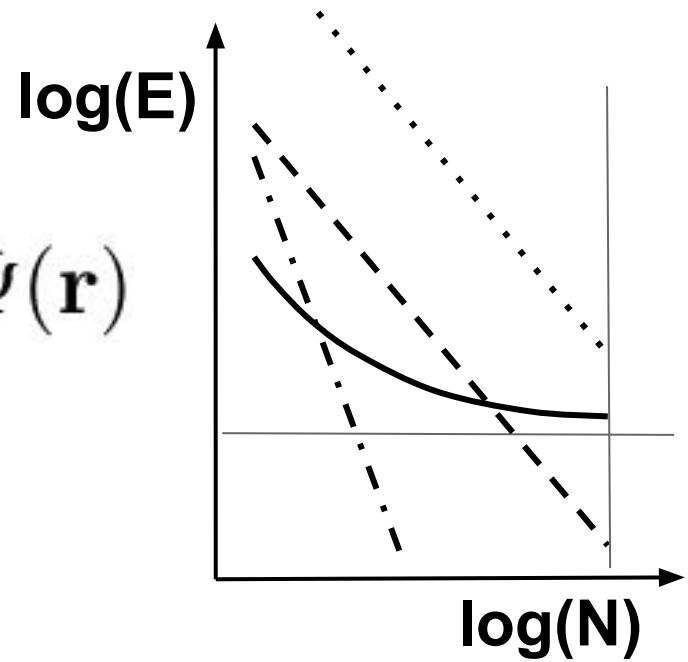
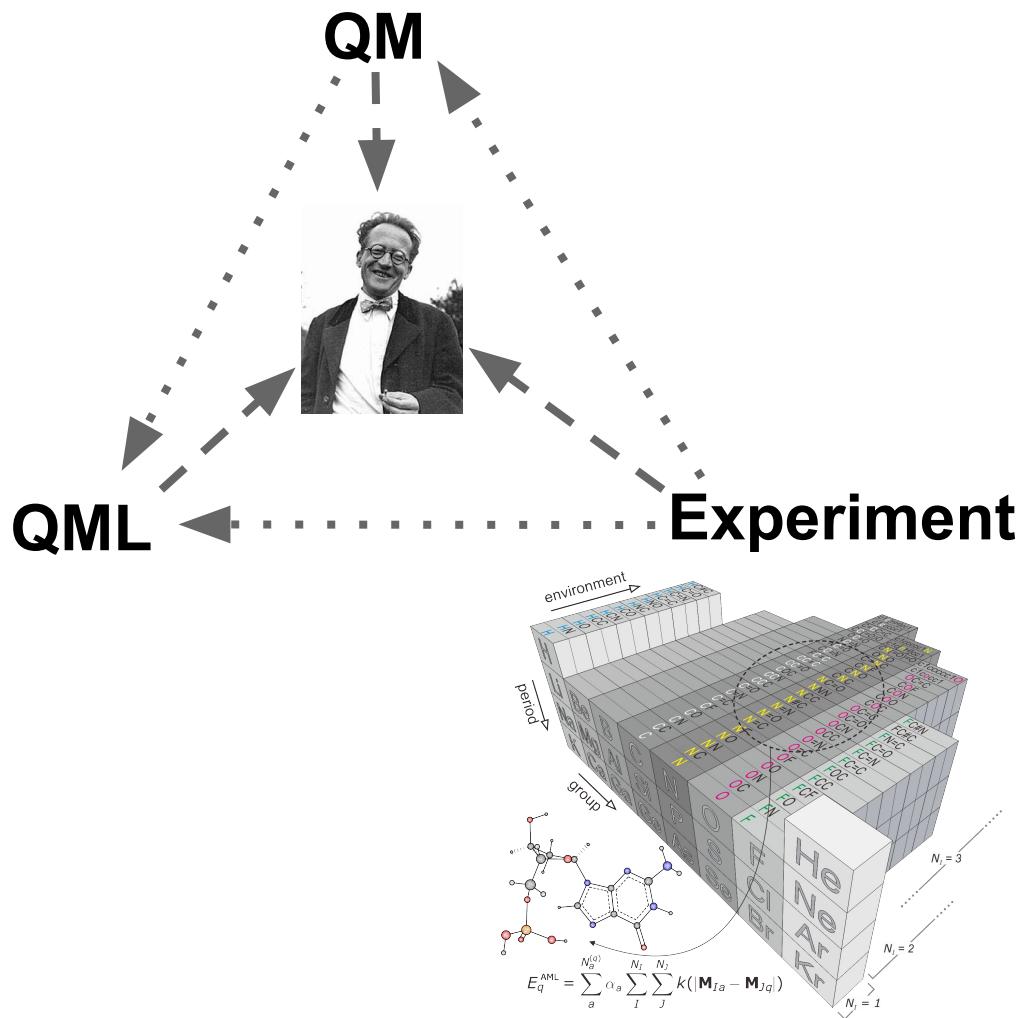
$$\{Z_I, \mathbf{R}_I\} \xrightarrow{\text{ML}} E$$



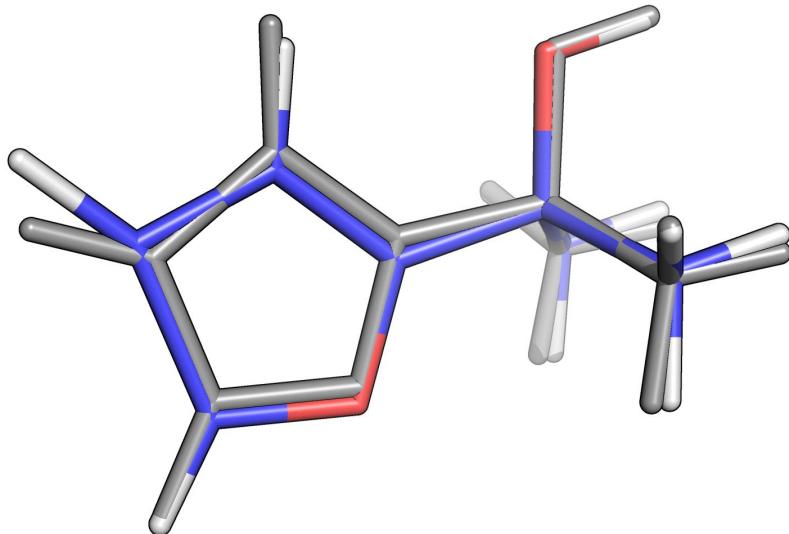
Chang, von Lilienfeld, CHIMIA (2014)

Final remarks

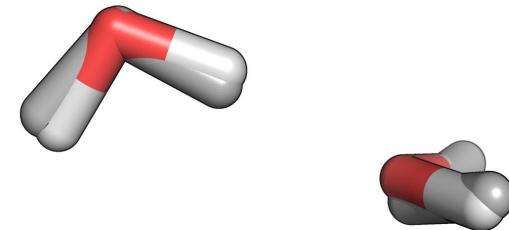
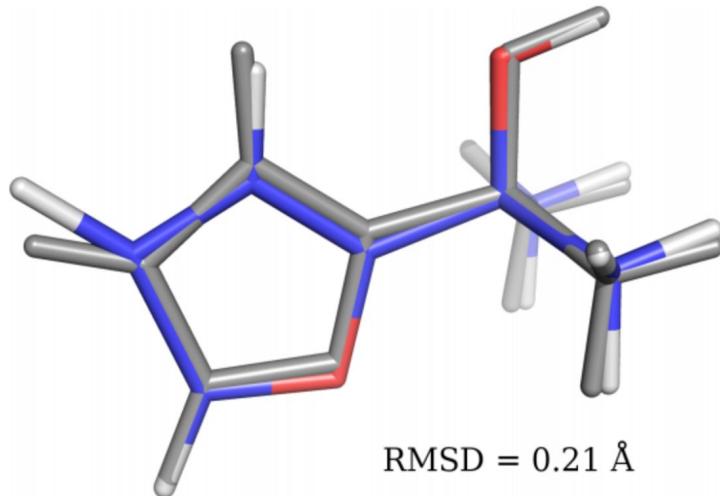
$$H(\{Z_I, \mathbf{R}_I\})\Psi(\mathbf{r}) = E\Psi(\mathbf{r})$$



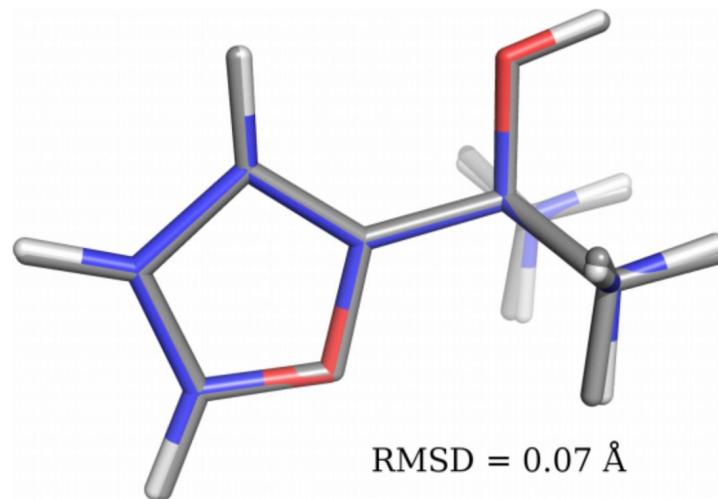
Outlook: Forces



Initial



Converged





Background of group members
-Computer Science
-Mathematics
-Physics (atomistic/molecular/solid)
-Chemistry (physical/computational)

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