Name:	
Date:	

## $\begin{array}{c} {\rm Astron~211~OOM~\#7} \\ {\rm ISM} \end{array}$

**OOM Policy:** You can work in pairs or groups of 3 if desired. You can consult class notes and books — no phones or computers. Make sure you are clear about the process you use to solve the problems: partial credit will be awarded. Always include units.

Estimate how many particles (atoms, molecules, or ions) would be in the classroom from the phases of the ISM listed below. Estimate how much total kinetic energy would be present from each of the phases.

component	T(K)	$n  ({\rm m}^{-3})$	state
molecular clouds	10-20	$10^{8-12}$	$H_2$
CNM	70	$3 \times 10^7$	Η
WNM	$10^{4}$	$10^{6}$	Η
WIM	$10^{4}$	$10^{6}$	$_{ m HII}$
HII regions	$10^{4}$	$10^{8-10}$	$_{ m HII}$

Number of molecules:

1. Molecular clouds:

2. CNM:

3. WNM:

4.	WIM:
5.	H II:
Total	energy:
1.	Molecular clouds:
2.	CNM:
3.	WNM:
4.	WIM:
5.	H II:

Compare with the relevant quantities for air at room temperature and atmospheric pressure.