

# Diachrony of Spectra

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## Question

Why is it defined like this?

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Hydrogen atom

Spectral theory of elliptic equations

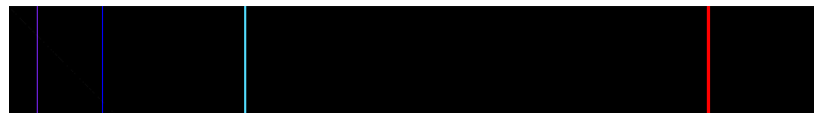
Gelfand theory

Algebraic geometry

# Hydrogen spectral series



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410.2nm

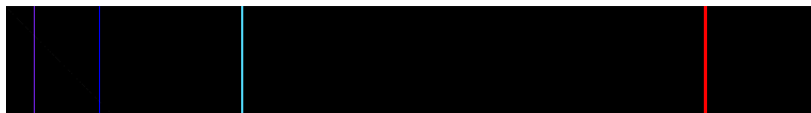
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How can we explain and compute this phenomenon?

# Rydberg's formula : Bohr model

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The constant  $h$  is called the Planck constant and  $\hbar := \frac{h}{2\pi}$ .

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From the three relations

$$mvr = n\hbar, \quad \frac{mv^2}{r} = -k \frac{(+e)(-e)}{r^2}, \quad E = K + V = \frac{1}{2}mv^2 - k \frac{e^2}{r},$$

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## Proposition (Rydberg formula)

*The wavelengths  $\lambda$  of absorbed or emitted photons from a hydrogen atom is estimated by the following formula:*

$$\frac{1}{\lambda} = R \left( \frac{1}{n_1^2} - \frac{1}{n_2^2} \right), \quad \text{for } n_1, n_2 \in \mathbb{N},$$

where  $R := \frac{k^2 e^4 m}{4\pi\hbar^3 c}$  is the Rydberg constant.

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Solving the equation, we obtain the probability distribution function  $|\Psi(t, x)|^2$  of the electron at time  $t$ !

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$$\begin{aligned} i\frac{d}{dt}\phi(t) &= E\phi(t), \\ (-\Delta + V(x))\psi(x) &= E\psi(x). \end{aligned}$$

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### Remark

The first one is not mathematically correct statement because we should resolve some technical issues on convergence.

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## The Beginning of Spectral Theory

# Rydberg's formula : Schrödinger equation

By long long calculations, we can obtain the following heuristically:

## Proposition

*The eigenvalues of  $\mathcal{H} = -\Delta - |\mathbf{x}|^{-1}$  is*



# Separation of variables

# Spectral theorem of normal matrices

# Spectral theorem of compact operators

# Spectral theorem of elliptic operators

# Banach algebras and $C^*$ -algebras

## Example 1 : Bounded operators

## Example 2 : Continuous functions

# Spectra, multiplicative homomorphisms, maximal ideals



# Gelfand-Naimark theorem

# Algebraic variety

# Coordinate ring

Maximal ideal is a point

# Problem of unified codomains

# Functoriality