

Overview

Group Item (bold text)

Spectrum Item

Index | number of items in group

Tree Widget

Index | number of data points; spacing (x_{\min}, x_{\max})

- spacing (x difference among points):
 - r – probably regular spacing among all datapoints
 - i – probably irregular spacing among all datapoints
 - value calculated as average for all datapoints

Plot Widget

- running on PyQtGraph (modified)

<http://www.pyqtgraph.org/>

Console

- IPython console running
- runs on qtconsole

<https://ipython.org/ipython-doc/3/interactive/qtconsole.html>

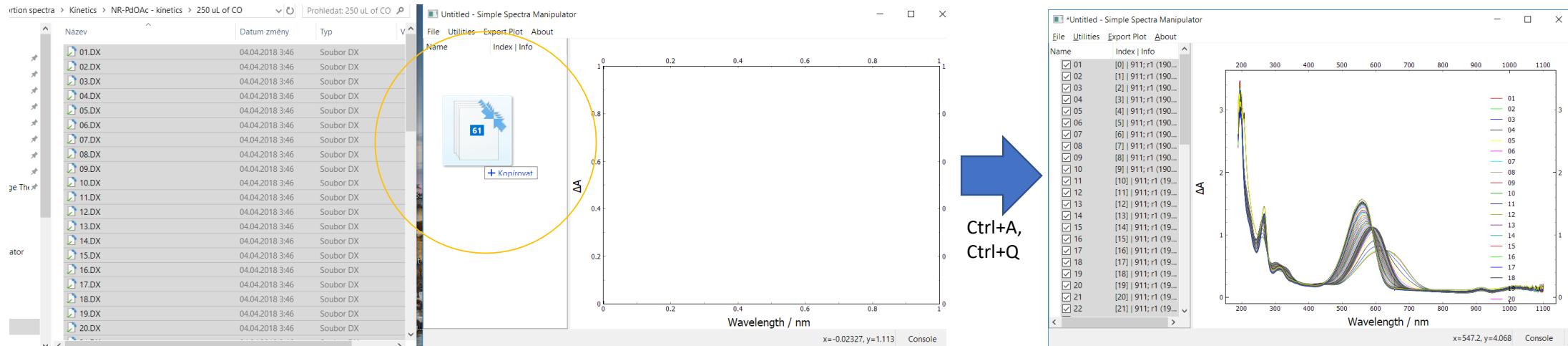
The screenshot illustrates a software application interface. On the left is a **Tree Widget** containing a list of spectrum items. A specific item, "integrals-corr-norm + 1", is expanded, showing its sub-items. Labels point to "Group Item (bold text)" and "Spectrum Item". To the right is a **Plot Widget** displaying multiple curves over time (0 to 30,000 seconds). A label points to "Index | number of data points; spacing (x_{\min}, x_{\max})". Below the plot is an **IPython Console** window showing an IPython session. A label points to "Index | number of items in group". At the bottom, a status bar shows cursor coordinates ("x=-2770, y=44.31") and a "Console" button.

Status bar

Cursor coordinates in Plot Widget

Importing files and Plot Widget interaction

- To import some Spectra, just **drag** the files from file explorer and **drop** them in the Tree Widget window (you can also from **File->Import Files**), then to select all of the imported spectra, press **Ctrl + A** and then **Ctrl + Q** which is the shortcut for checking the selected spectra. And so on.... The program is intuitive I hope...



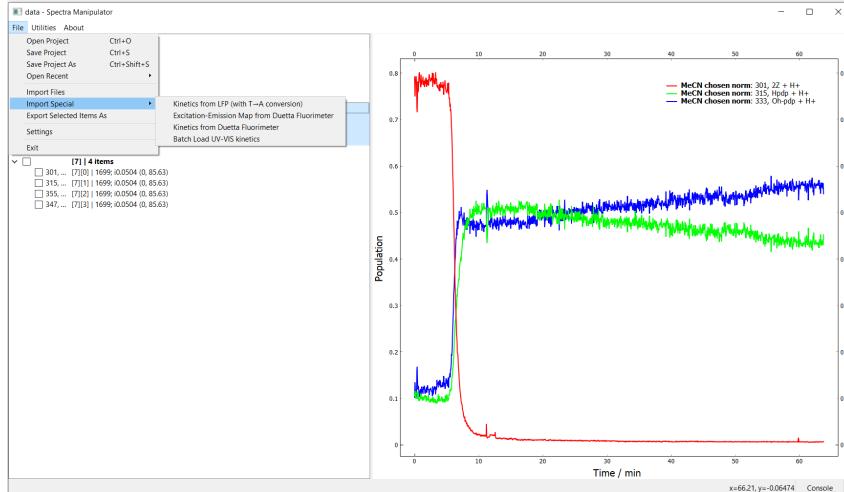
- Some DX files contains also uncertainties of measurement, in this way, the two items will be imported for one DX file, you can get rid of the uncertainties by deleting these items only (use select every n-th item function – *vide infra*)
- To put spectra in group, just select the spectra you want to put in the group and right click on the selection and choose **Move Items To New Group** or **Copy Items to New Group**
- Saving to project is made by the same way as in other programs, **Ctrl + S** for Save or **Ctrl + Shift + S** for Save As, the extension for project files is **.smpj**, you can load the project from the **File->Open project** or drag-drop the **smpj** file to Tree Widget window like for spectra, or just double click on smpj file (*.smpj extension has to be associated in Windows)

Plot Widget interaction

- Default mouse mode is strongly preffered (3 button mode, you will get use to it), that is, the **scaling** is done by right mouse button dragging, left mouse button dragging **pans** the scene, **zooming** is done by mouse wheel as usual
- If you get lost in a plot, just right click in the Plot Widget and click **View All**
- For more information, see the pyqtgraph documentation https://pyqtgraph.readthedocs.io/en/latest/mouse_interaction.html

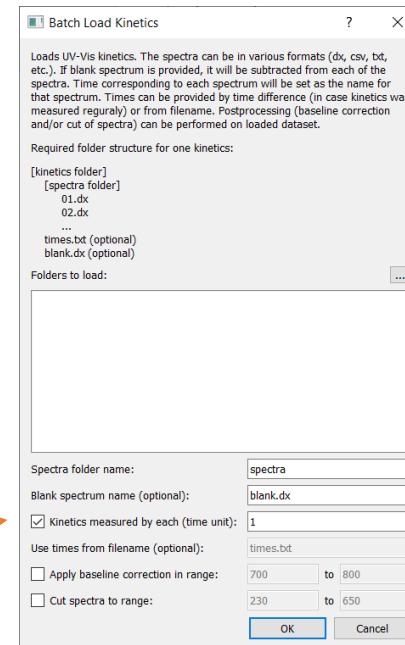
Importing special data: Nano kinetics, UV-vis kinetics, EEM and kinetics from Duetta fluorimeter

- In File->Import Special->
- for kinetic from nano, data are automatically transformed to change of absorbance (transmittance to absorbance conversion)



• Batch load UV-Vis kinetics

- For batch loading a processing (baseline correction, cut to region) of UV-vis kinetics
- Special folder structure is required



- Time that corresponds to each spectrum from kinetics will be loaded from filename if unchecked and this file is provided,
- If checked, names will be setup as follows: name = index * time_unit, index starts from 0

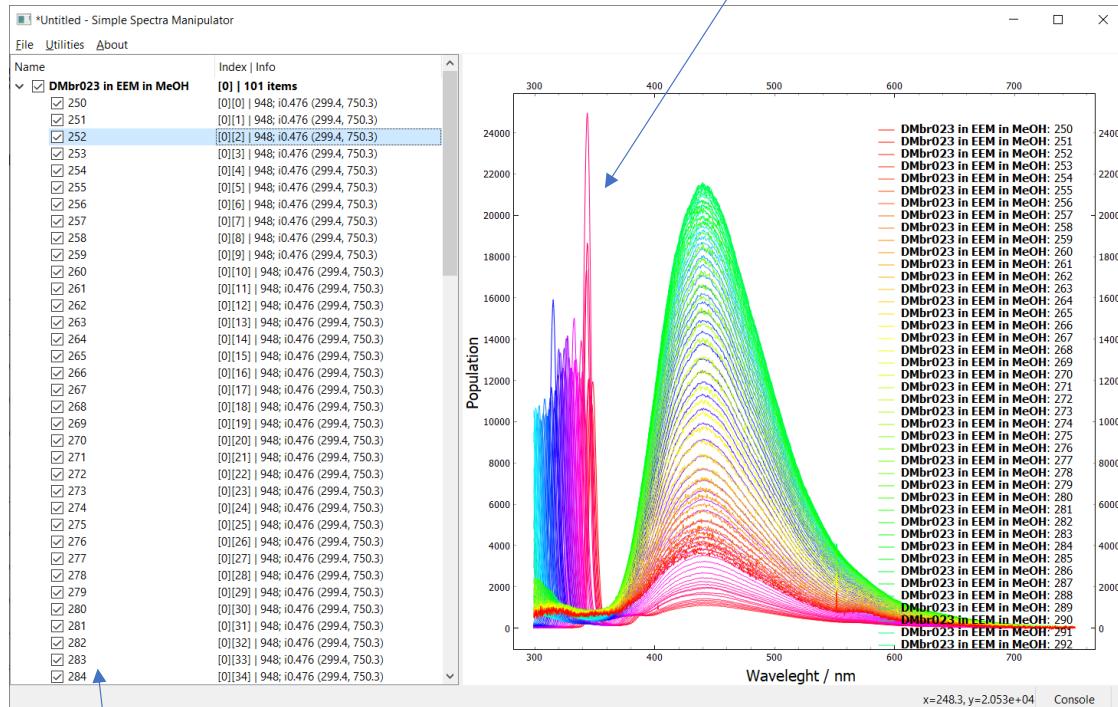


Wavelength ranges to use the correction in



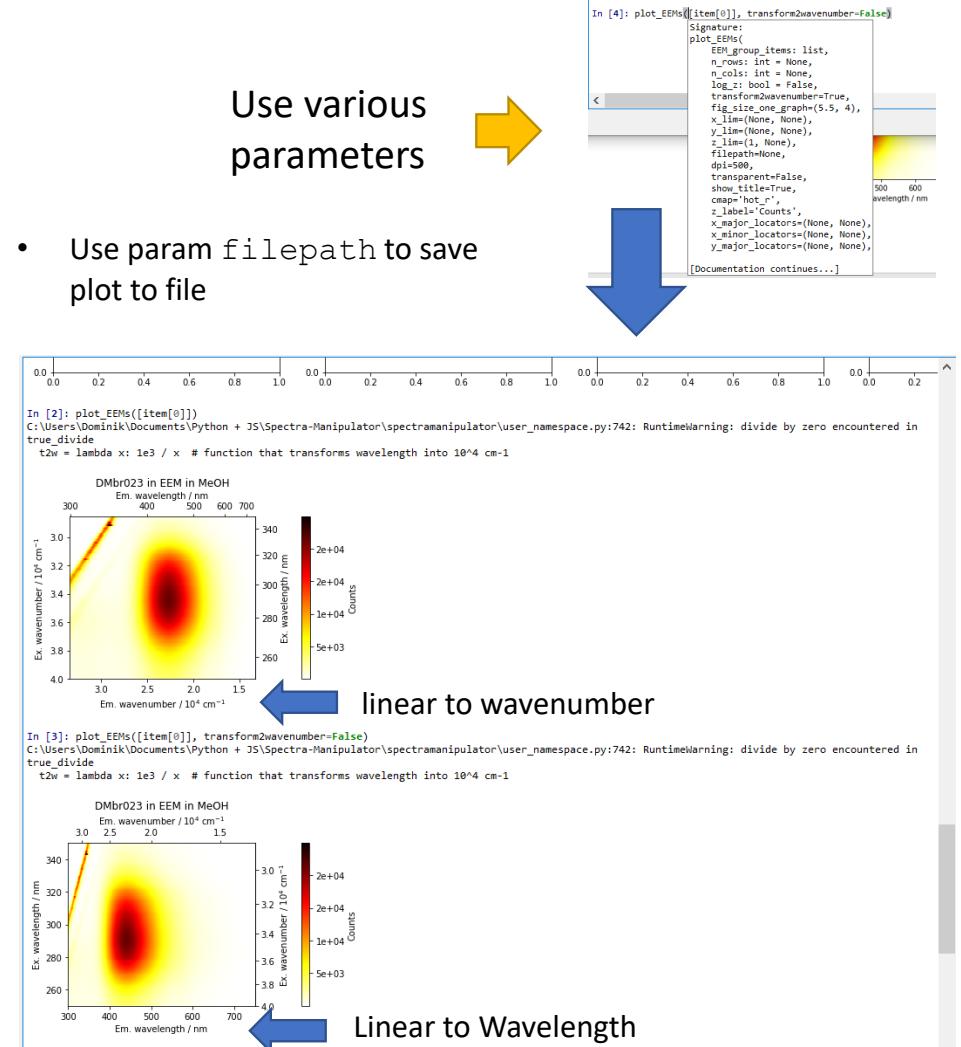
Importing special data: EEM from Duetta fluorimeter

- Importing of Excitation-Emission map (EEM)



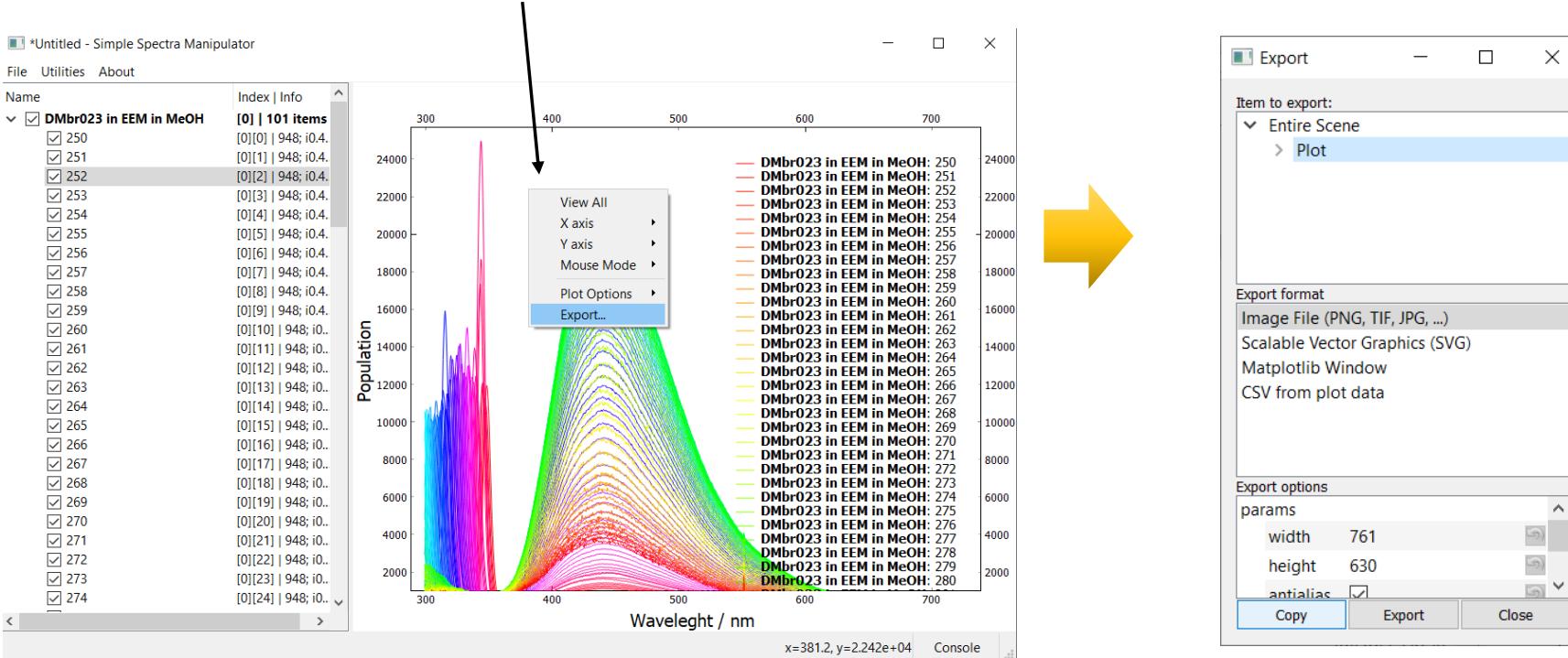
Rayleigh scattering

- Plotting of EEM heatmap from console:



Copy plot to clipboard as image or save it to file

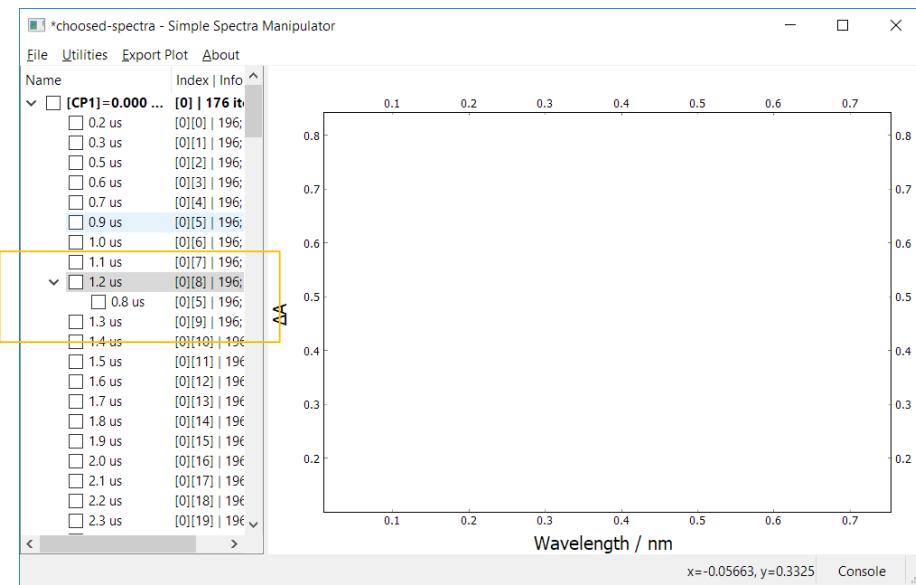
- Right click on Plot Widget and click **Export**



- Press **Copy** to copy to clipboard or **Export** to file

Moving and copying the items/groups

- You can freely move the individual and multiple spectra in the Tree Widget, as well as groups of spectra. For moving, just use left mouse button. The **copying** within the list works the same, but you have to **hold Ctrl** when releasing the items in the desired location. The interaction is fast for small number of items, but for large lists, >1000 items, it became slower. This is a bug in PyQt/Qt package and cannot be fixed. It may be solved in PyQt6, however qtconsole is not compatible with PyQt6 yet...
- The list is designed to have only 2 levels (spectra in group cannot have children). But, during some fast manipulation, the similar situation that is shown below can happen:

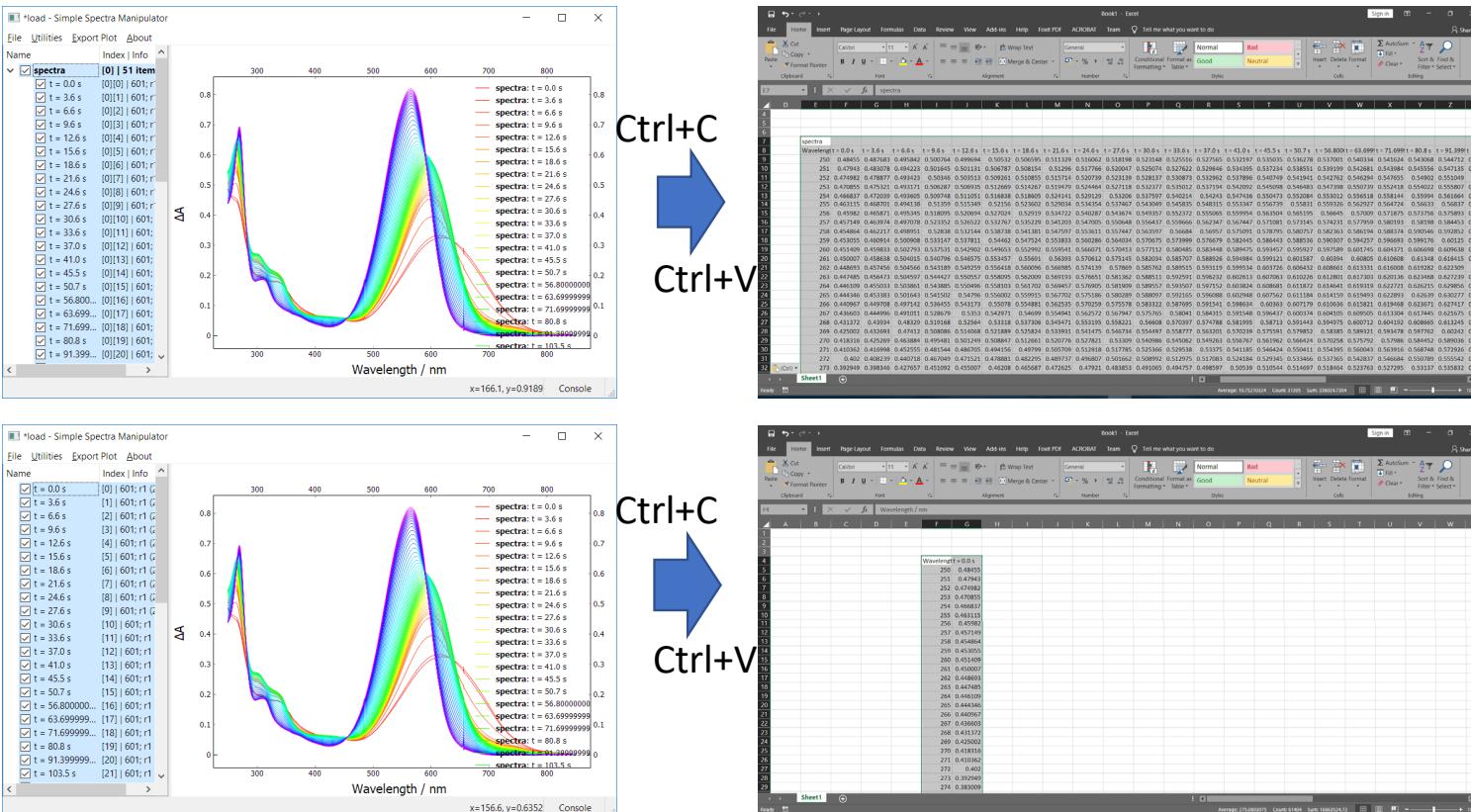


- If that happens, you have to **manually move it** to the grandparent or root node, otherwise, the program will crash sooner or later during some operations, its just a bug that can happen a you have to count with that

- You can also open another SM program and move the items/groups between different program instances, in this way, the data are always copied regardless of holding Ctrl or not

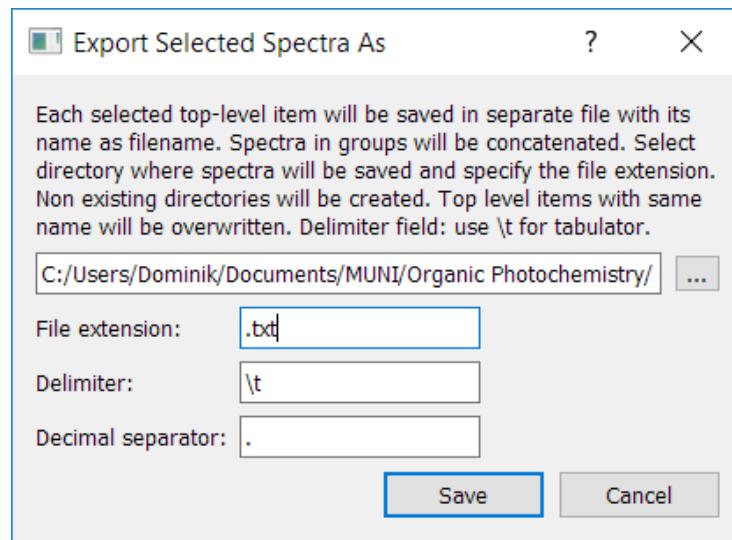
Copying to Excel/Origin

- To copy the spectra, just select the spectra you would like to copy and press Ctrl+C (or the right click, With Selected Items->Copy items to clipboard). The results are shown below. If the spectra are **in group**, they will be **concatenated** to a matrix (this is usually what we want). In this case, all spectra in the group must have the same number of points and same x range, otherwise, error will be raised. If the spectra are **placed in the root** (top level items), they will be **stuck to one another** (only first spectrum seen in Excel). Please, look at Setting Export tab, to change appropriate settings. If you want to retain this compatibility, the delimiter in Clipboard must be **tabulator \t**. Only with this settings, data can be easily transferred to Excel/Origin by Windows Clipboard. You can also **import data** from clipboard to the program and also from Excel. Check Settings Import Tab: Clipboard for more info.



Exporting spectra to files

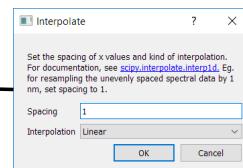
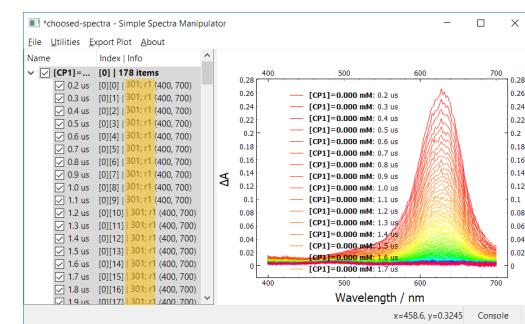
- The exporting of spectra to files works the same way as with copying. Spectra in group will be concatenated. Spectra in root (top level items) will be saved as individual files. Just select the root items you want to export as files and go to **File->Export Selected Spectra As** and this dialog below should appear:



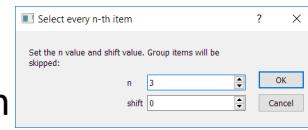
files are saved with **utf-8 encoding**

- Just select the **directory** where you want your spectra to be saved and set the file extension (you can omit the dot, so eg. just write 'txt' or 'csv'). Delimiter is the character that separates the columns, then click Save
- The name of the file is determined by the name of the top level item** (if the top level item has no name, 'Untitled' with index is used as name)
- Files with the same name in the saved directory will be overwritten** without notification, beware of that
- For header options, look at settings Export tab

Operations Overview



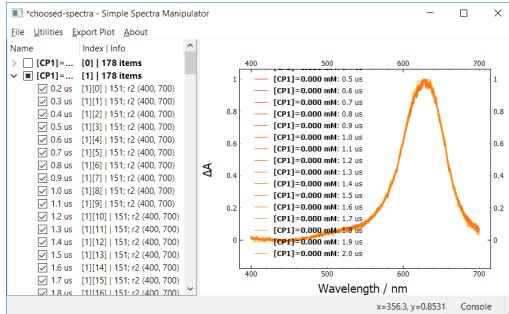
Select Every n-th Item



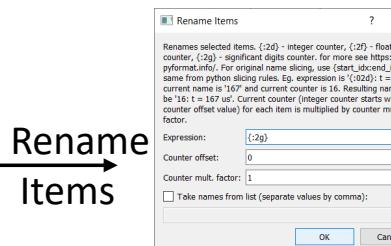
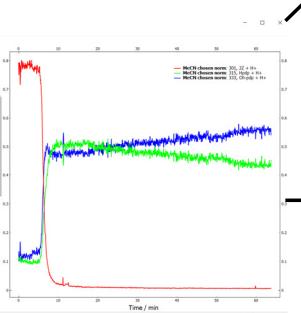
n=3, shift=0

Interpolate

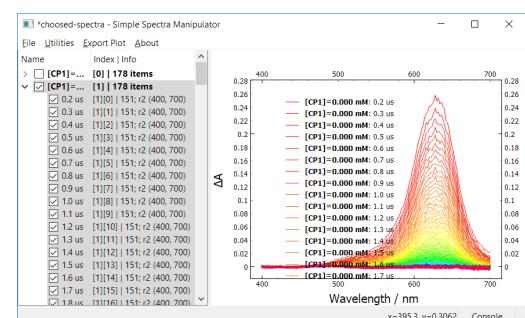
- used for example to change irregular spacing to regular here - change the spacing from 2 nm to 1 nm



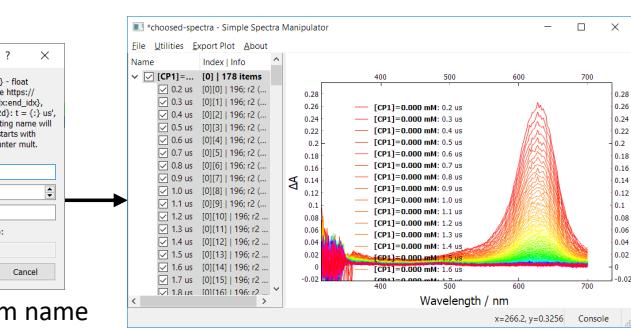
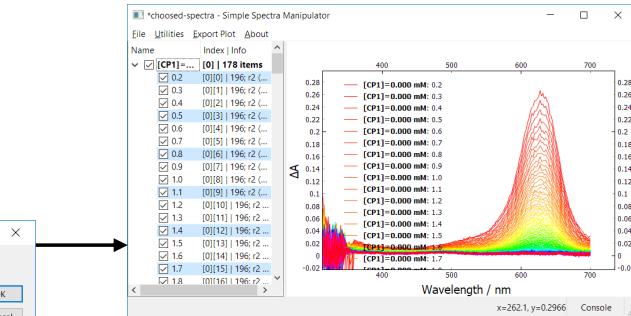
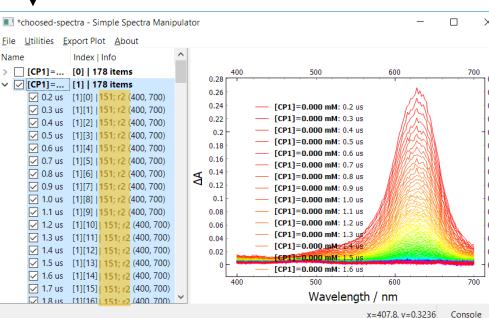
Normalize



added 'us' after each spectrum name



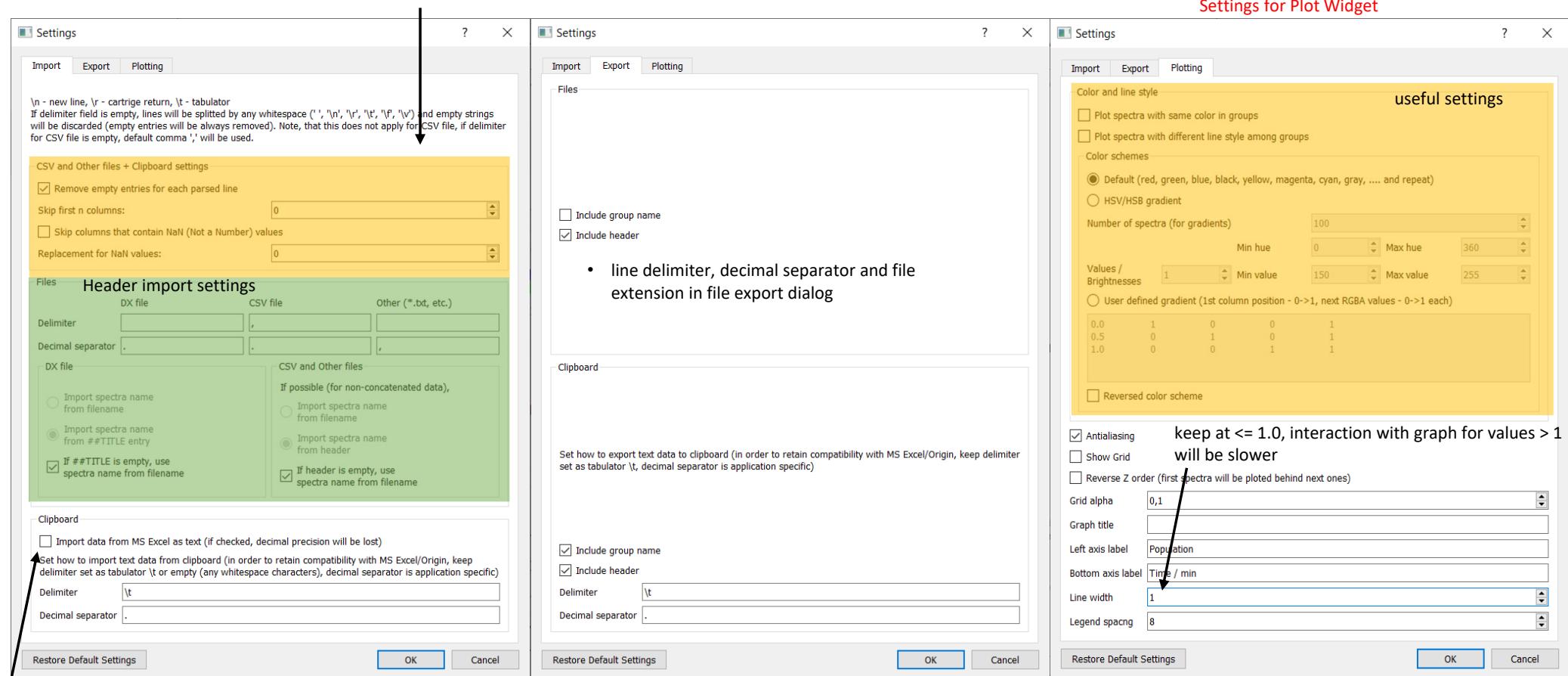
Baseline correct



- Extend by zeros only extends the spectra by zeros for selected x range, while keeping the spacing the same as original spectrum

Settings

- most important part, sets how to import files (sets the **line delimiter** and **decimal separator**) for different kind of files, for files, that has for example empty entries (only delimiter eg. two commas) at the beginning of the file, sets the number of columns that will be skipped, or there is an option that will remove all empty entries from each line, please, consider these options based on the file you are trying to import
- the CSV file is parsed using the csv module for python
- delimiter for dx file is space - " " (not visible)



When copying data from Excel, Excel saves data into clipboard in many formats, text and XML raw data format among others. Data in text are saved as they are seen in Excel (when rounding is on, values may be saved without full decimal precision). To import data copied from Excel with full decimal precision, raw Excel XML data has to be parsed. Therefore, for full decimal precision, keep this checkbox unchecked.

Console and simple computations

- The main reason for having a console in the program is to provide interface to fast arithmetic operations with spectra items and all groups of items and show program messages and reports. Regular IPython console is running in the window (qtconsole widget – lot of thanks to programmers for this package), so also for a low level interaction with the running program (for experienced users, see documentation).
- When performing arithmetic operation between Spectrum – Spectrum, the **number of points** and same x range (**start and end value**) of both spectra have to be the same, otherwise, error is raised
- The most important is variable `item`, which is internally a root item in Tree Widget and provides an access to all members. Just write `item` and then in square bracket rewrite the index, that is in the info column for each Item/Group. When performing any arithmetic operation (addition, subtraction, multiplication, division, negation, power) among number/spectrum/group, the new Spectrum or SpectrumList object is always returned as a result of computation. So, nothing is happening with the original data. The y values are used in calculations. To add the result to the Tree Widget, you have to call function `add_to_list(result)` which takes result of computation as an argument. Examples are given below:
- Also spectrum & array operations are possible (array can be list, tuple or ndarray) and the array must have the same number of elements as the spectrum
- Similarly, group & array operation works as well (each spectrum and each number from the array are used for operation)

In [1]: `type(item)`

Out[1]: `treeview.item.GenericItem`

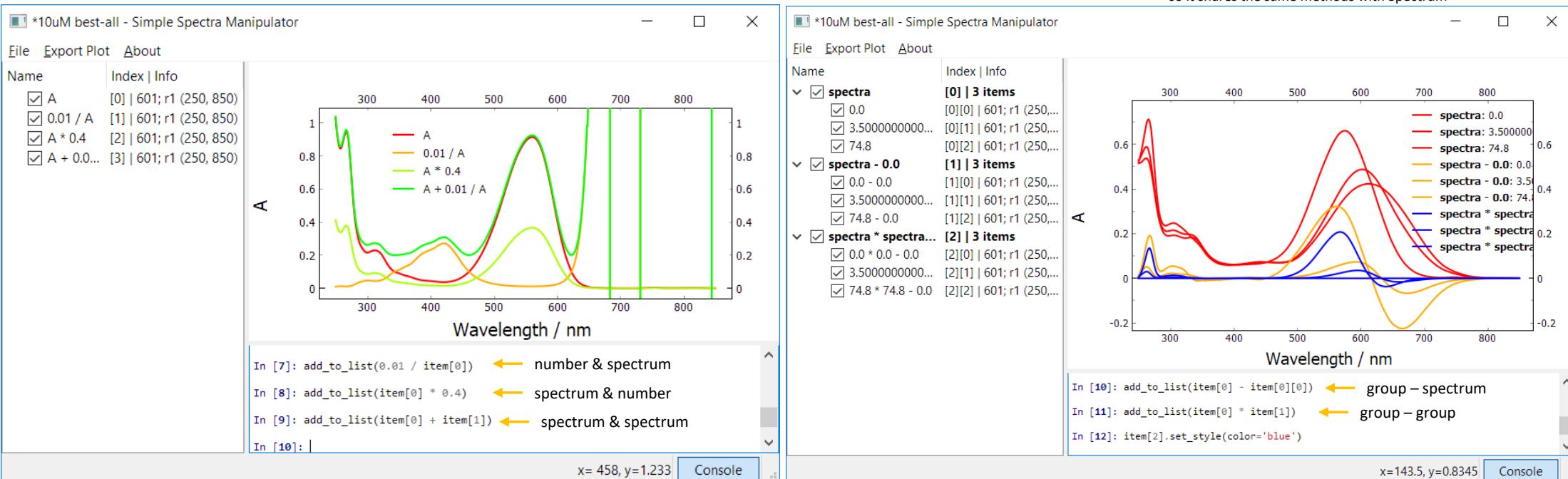
In [2]: `type(item[0])`

Out[2]: `treeview.item.SpectrumItemGroup`

In [3]: `type(item[0][0])`

Out[3]: `treeview.item.SpectrumItem`

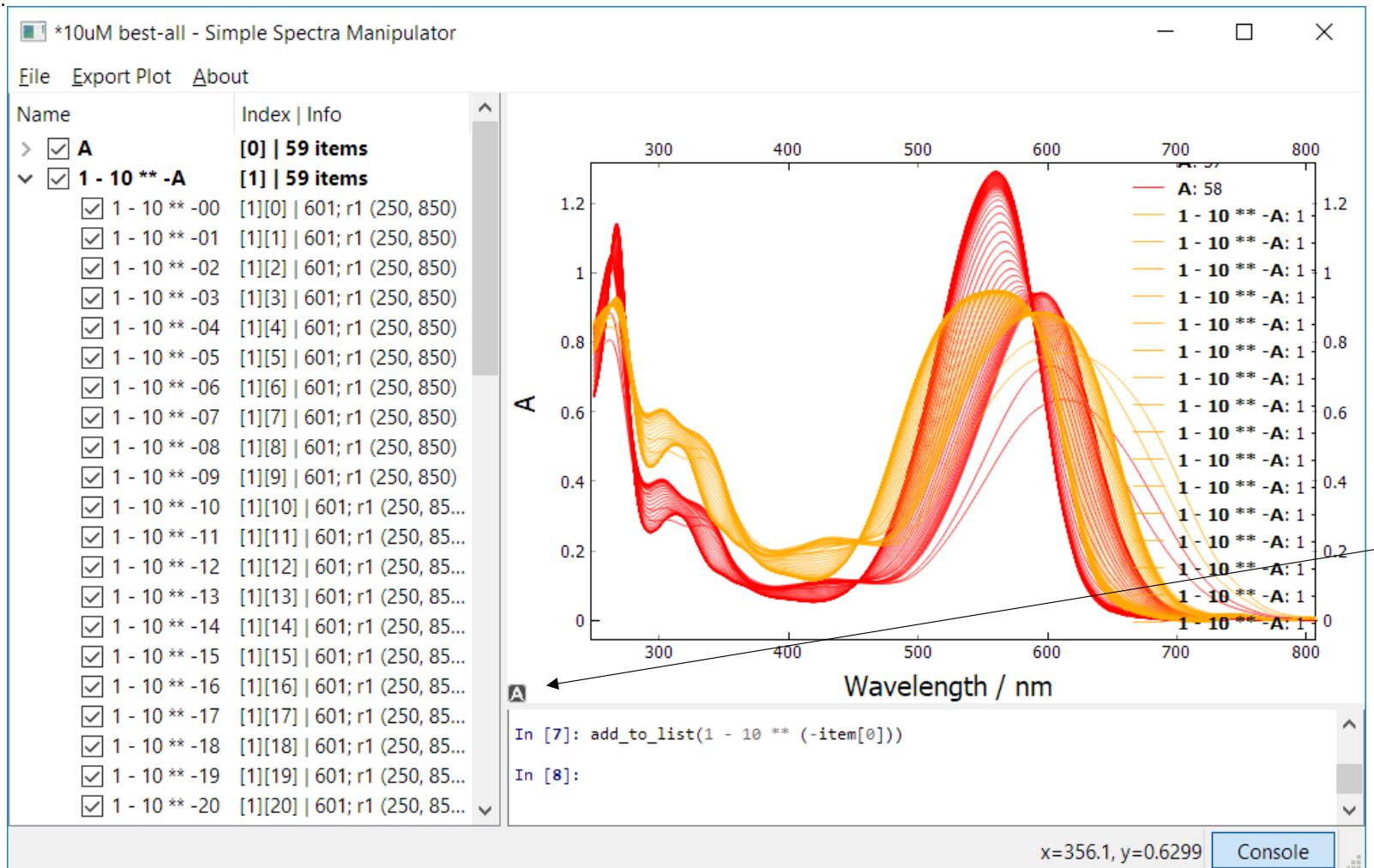
SpectrumItemGroup inherits from **SpectrumList**,
so it shares the same methods with SpectrumList
SpectrumItem inherits from **Spectrum**,
so it shares the same methods with Spectrum



- The main power of the Console comes with the use of defined functions in classes Spectrum / SpectrumList (see documentation) like integration, differentiation, savitsky-golay filter, fourier transform and others for the groups of spectra.

Console and simple computations

- More complicated transformations, eg. $1 - 10^{-A}$ (fraction of absorbed light), can be performed as easily as simple ones. $**$ is operator for power in Python. Here, the degree of abstraction is pretty significant. We are performing operation on a group of spectra. The operation $1 - 10^{-A}$ is performed for every y value of each spectrum in the group, in total $601 * 59$ calculations (actually many more, because this operation is splitted internally into 3 separate operations – negation, then power and lastly subtraction). In Excel, we would have to make a matrix of 601×59 absorbances and then perform operation on each cell.



Common console commands

`add_to_list(result)`

- Takes one parameters and adds result to Tree Widget, it can be Spectrum or list of spectra

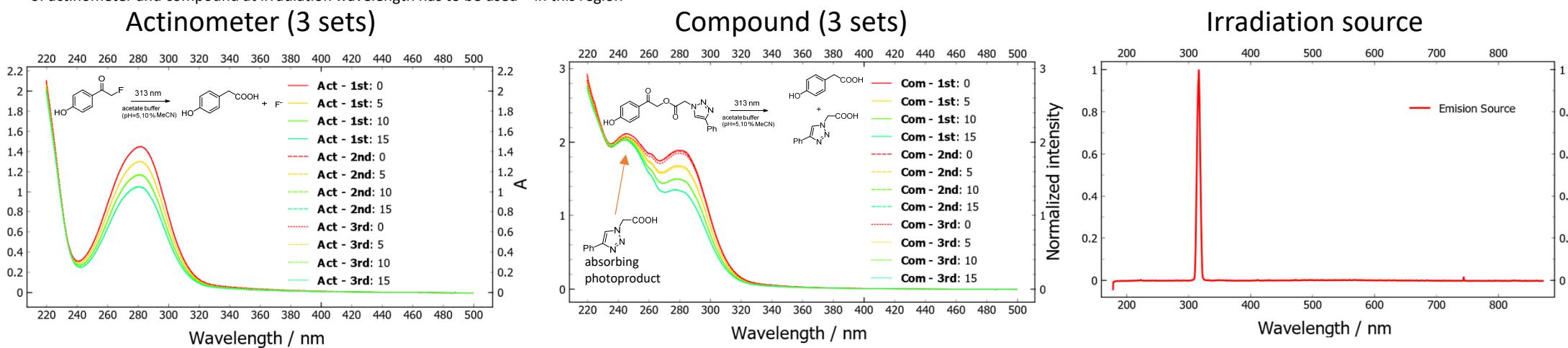
Operations on group

`.getTranspose(max_entries=1000)`

- Returns the group in which x and y coordinates were switched, names of individual items must be numbers corresponding to the y coordinate
- If former x dimension will exceed `max_entries` parameter, error will be risen (this is to prevent the program to became not responding in case the former x dimensions is too big)

Example: Photodecomposition QY calculation

- In this example, I show how to use this program to calculate the quantum yield of photodecomposition of measured compound based on the known photodecomposition quantum yield of an actinometer
- First, the light source was established, it was Hg(Xe) arc lamp with suitable filter (here $\sim 320 \pm 20$ nm) and emission spectrum was measured using OceanOptics
- Then, the actinometer solution in cuvette was irradiated for 5, 10 and 15 minutes and UV-VIS spectra were taken, this was then repeated 2 times which gave in total 3 sets of measured spectra, the red curves are spectra before irradiation and yellow, green and cyan are after 5, 10 and 15 minutes of irradiation, respectively
- The same procedure was repeated for compound whose photodecomposition quantum yield was not known
- The volumes of the cuvette were kept the same for all measurements, the initial absorbance should be similar and high (~2), here, the actinometer absorbance was a bit lower (the absorbance at irradiation wavelength is actually very small, but we could not follow the reaction by absorption spectroscopy if we wanted the absorbance at irradiation wavelength to be high)
- The showed spectra of actinometer and compound are already processed by baseline correction and cut to (220 to 500 nm) range, the emission source spectrum was baseline corrected and normalized
- The goal is to calculate the photodecomposition quantum yield of measured compound with the use of proper statistics and with the use of irradiation source spectrum, the correction to **unequal absorbances** of actinometer and compound at irradiation wavelength has to be used - in this region



Example: Photodecomposition QY calculation - theory

- The decrease of absorbance (concentration of compound) during photoirradiation at constant intensity is proportional to the photodecomposition quantum yield, light intensity and the fraction of the light $X_{abs}(t)$, that was absorbed by the solution at that time. The differential equation is similar to the **photobleaching model (irradiated and plotted wavelength are the same)**, found in Documentation under the fit models.

$$\frac{dc(t)}{dt} = -\frac{1}{V} I_0 \Phi \times X_{abs}(t)$$

$$X_{abs}(t) = 1 - 10^{-A(t, \lambda_{irr})}$$

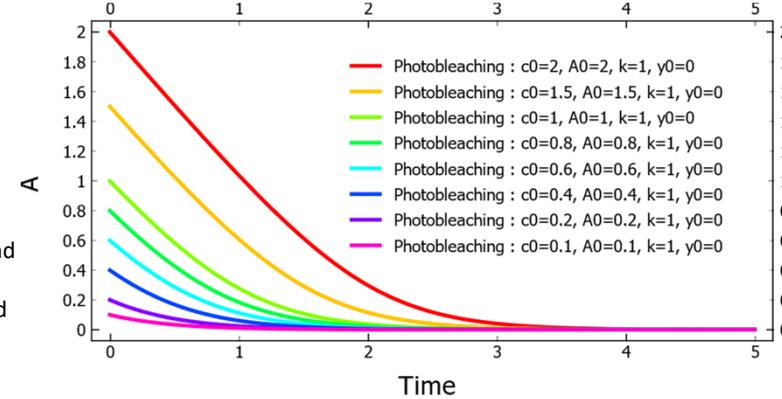
$$X_{abs}(t) \propto \int_{\lambda_0}^{\lambda_1} I_{source}(\lambda) (1 - 10^{-A(t, \lambda)}) d\lambda$$

➤ irradiation at single wavelength

➤ irradiation with broad emission intensity**

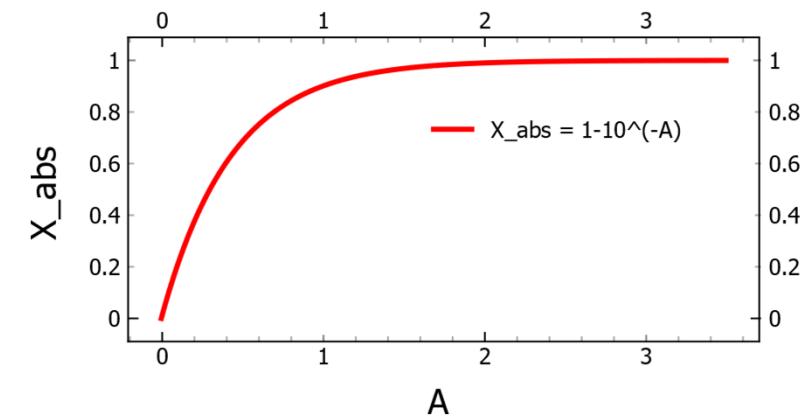
- If we would solve the differential equation for the simple case (irradiation at single wavelength, photobleaching model), it is obvious that at higher irradiated initial absorbances ($A > 2$), the fraction of absorbed light is close to 1 and the value is 'constant' with changing of absorbance, as a result, the rate of decrease is almost constant, because all light is absorbed, the result is linear decrease
- However, with lower absorbances, the fraction of absorbed light decreases and the rate of decay slows down, which causes changing of the **linear decay into first order decay** (in the limit that initial absorbance goes to zero)

- So we can approximate the differential equation by changing the infinitesimal changes to discrete changes
- By combining the equations, we get rid of the constant light intensity and volume of the solution, that was kept the same for both actinometer and compound
- The quantum yield for actinometer is known, and the only unknown in the equation is now the photodecomposition quantum yield of compound (concentration can be calculated from molar abs. coefficients or from initial concentration, which are known)



compound

Dependence of the fraction of absorbed light on the absorbance of the solution



**The part of absorbed light must be a number in the interval of (0, 1), so it has to be normalized, however, because we actually need to calculate ratio of these fractions for actinometer and measured compound, normalization factor would cancel out, so we don't have to compute this factor...

$$X_{abs}(t) = \frac{1}{\int_{\lambda_0}^{\lambda_1} I_{source}(\lambda) d\lambda} \int_{\lambda_0}^{\lambda_1} I_{source}(\lambda) (1 - 10^{-A(t, \lambda)}) d\lambda$$

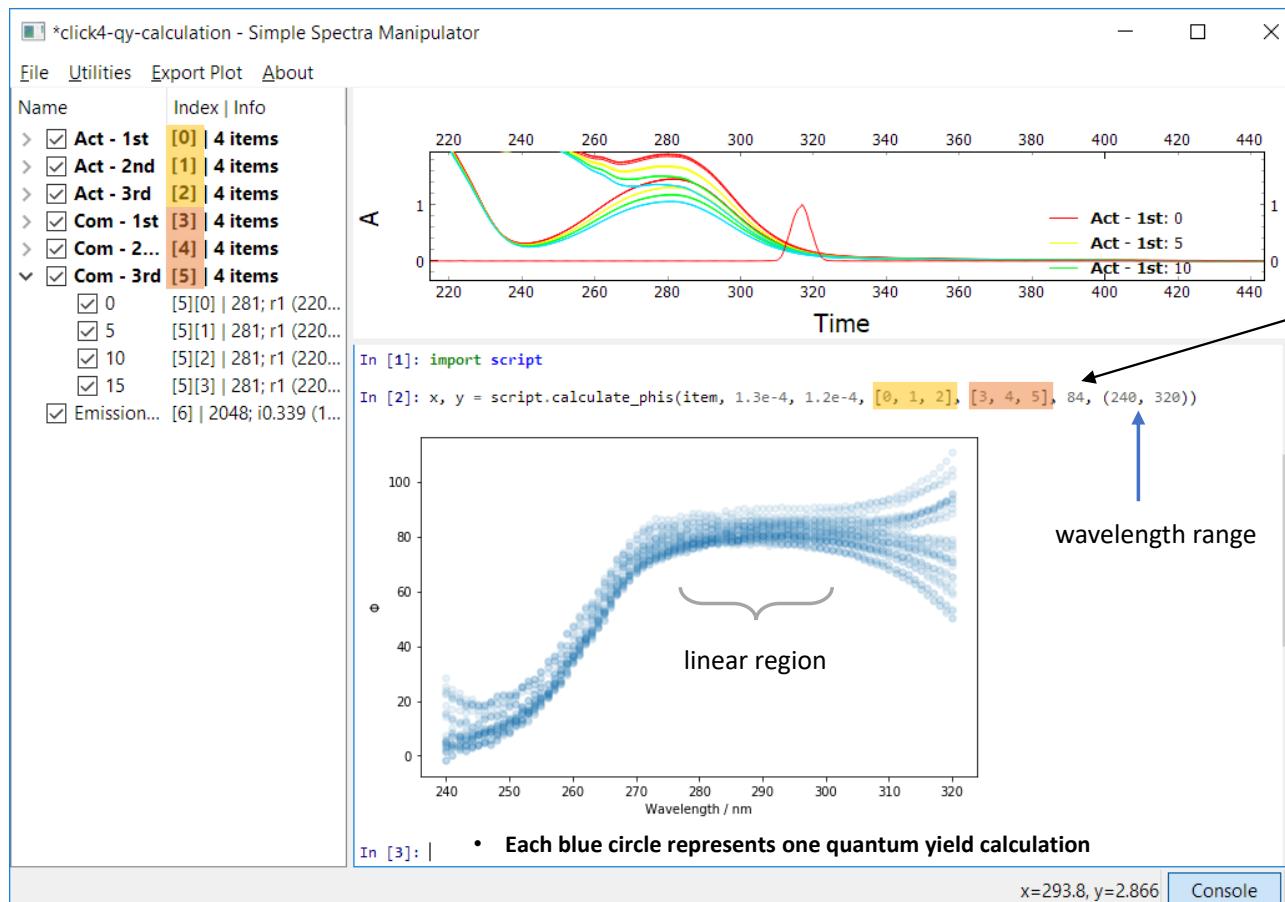
$$\frac{\Delta c_{act}}{\Delta t_{act}} \approx -\frac{1}{V} I_0 \Phi_{act} \times X_{abs,act}(t_{act})$$

$$\frac{\Delta c_{com}}{\Delta t_{com}} \approx -\frac{1}{V} I_0 \Phi_{com} \times X_{abs,com}(t_{com})$$

$$\frac{\Delta c_{act} \Delta t_{com}}{\Delta c_{com} \Delta t_{act}} = \frac{\Phi_{act}}{\Phi_{com}} \times \frac{X_{abs,act}(t_{act})}{X_{abs,com}(t_{com})} \quad \rightarrow \quad \Phi_{com} = \Phi_{act} \times \frac{\Delta c_{com} \Delta t_{act}}{\Delta c_{act} \Delta t_{com}} \times \frac{X_{abs,act}(t_{act})}{X_{abs,com}(t_{com})}$$

Example: Photodecomposition QY calculation - uncorrected

- There is number of ways how to perform the calculation, but we will focus on the easier one, only with the use of **closest combinations**. It means, that we will combine first difference of actinometer with first difference of the compound, the second difference of actinometer with the second difference of the compound and so on... If we take two sets (each has 4 measurement), we only make calculation 3 times. We will however combine all sets, so that is in total 9 combinations (3x3). So, for one wavelength, we get in total $9 \times 3 = 27$ results. We can perform this calculation for every wavelength and plot the 2D map. Then, we can pick the quantum yields in a range of wavelengths, where it looks linear and perform statistics and error estimation from this range.
- Now, we have to write python code that will perform the calculation, lets first try it without correction. The code was written to file `cript.py` and was placed into folder of SSM.



$$c_{act} = 1.3 \times 10^{-4} \text{ M}$$

$$c_{com} = 1.2 \times 10^{-4} \text{ M}$$

$$\Phi_{act} = 84 \%$$

- The changes in concentrations are computed from changes in absorbances according to Beer-Lamber law, A_0 is absorbance before irradiation

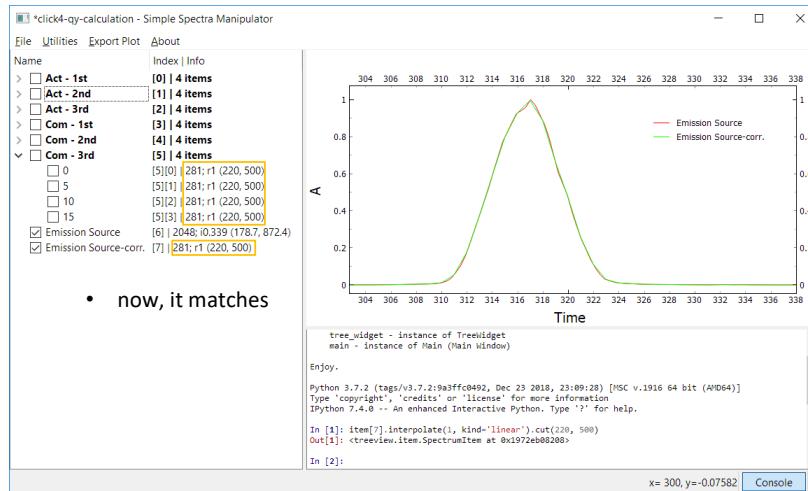
$$\Delta c = \frac{\Delta A}{\epsilon l} \quad \epsilon l = \frac{A_0}{c_0}$$

$$\Delta c = \Delta A \times \frac{c_0}{A_0}$$

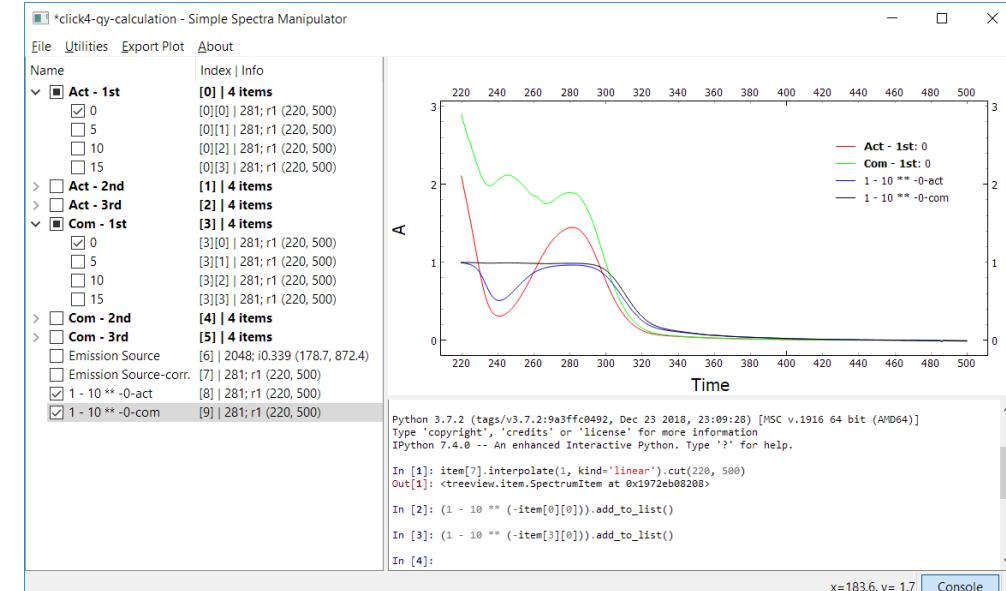
- Now, we should incorporate the correction to unequal absorbances

Example: Photodecomposition QY calculation - correction

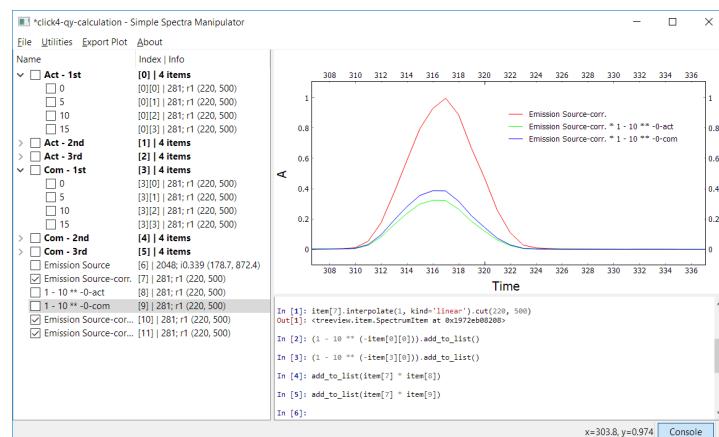
- First, we have to further process the emission source spectrum, because the points are unevenly spaced and the x range does not match with the spectra of actinometer and compound, so we would not be able to calculate any integrals... So we can do it manually, or by just one line of code in the console. I made a copy of the emission spectrum, renamed it and performed the computation. The result is apparent.



- Now, we can try to calculate how much light is gonna be absorbed by for example first stock solutions of compound and actinometer, the black and blue spectra are the result after performing $1 - 10^{-A}$ transformations for compound and actinometer, respectively



- We can then try to calculate the actual light intensity, that would be absorbed by the solution (blue and green spectra), and we can see, that less than half of the light was absorbed



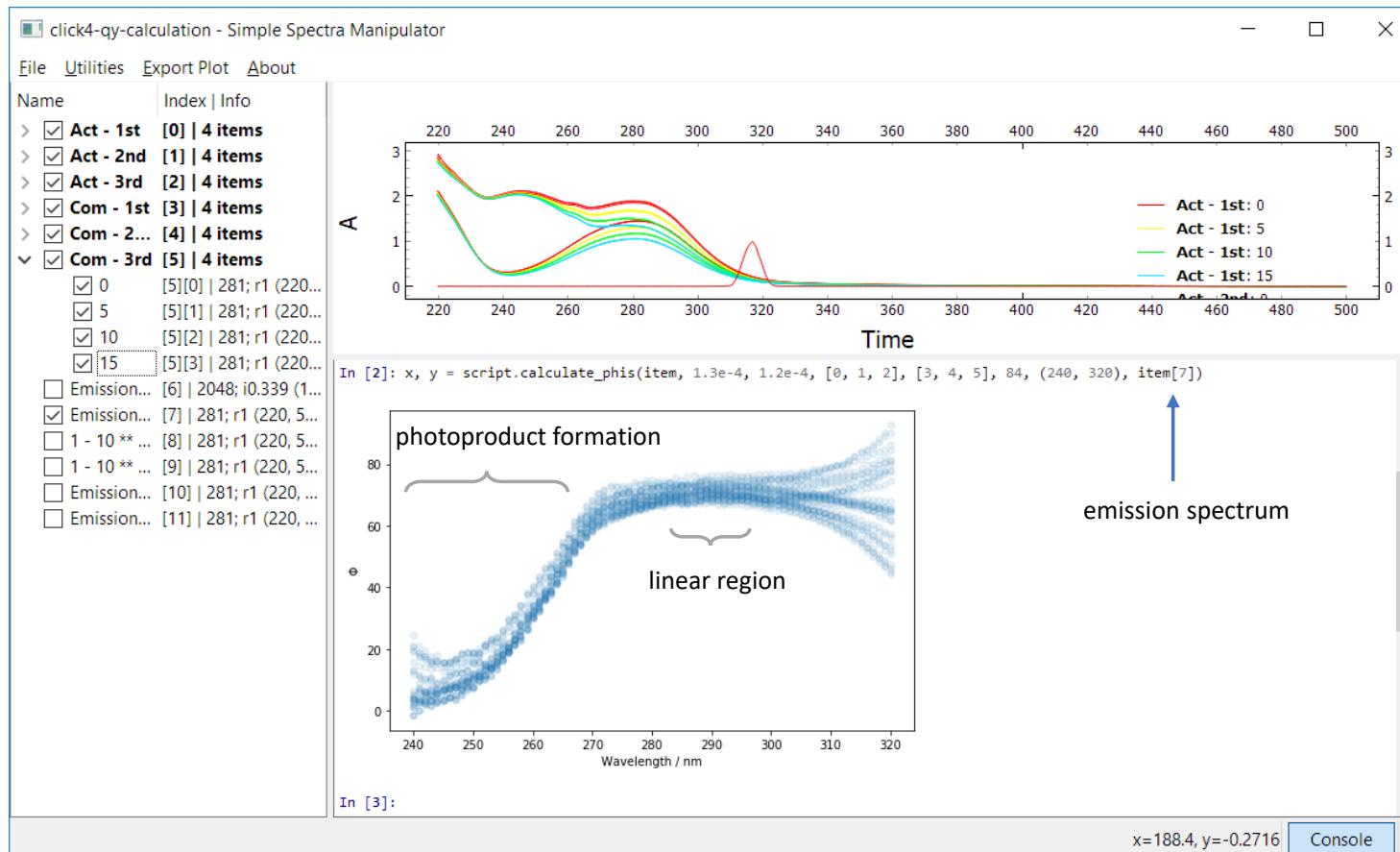
- We can calculate it precisely by integrating the resulting spectrum and normalizing by normalization factor (integral of emission source)
- So, by calling `integral()` function without any arguments, the whole spectrum is integrated (here from 220 to 500 nm) by trapezoid integration method. The actinometer and compound actually absorbed 33 % and 39 % light from emission source, respectively

```
In [4]: add_to_list(item[7] * item[8])
In [5]: add_to_list(item[7] * item[9])
In [6]: norm_factor = item[7].integral()
In [7]: X_abs_act, X_abx_com = item[10].integral()/norm_factor, item[11].integral()/norm_factor
In [8]: X_abs_act, X_abx_com
Out[8]: (0.3284490163431788, 0.3920187209579002)
In [9]: |
```

$$X_{abs}(t) = \frac{1}{\int_{\lambda_0}^{\lambda_1} I_{source}(\lambda) d\lambda} \int_{\lambda_0}^{\lambda_1} I_{source}(\lambda) (1 - 10^{-A(t,\lambda)}) d\lambda$$

Example: Photodecomposition QY calculation - corrected

- Now, we can automate this and add this to our script and perform the calculation again, but now with the correction. We can see, that the linear region part was decreased a little bit (from cca 80 to 70). This make sense, because the amount of absorbed light by **compound** was a bit **higher** than that of actinometer, so more of it was bleached. If the changes of concentrations are the same, quantum yield of compound photobleaching must be lower when compared to that of uncorrected result.



- The linear region seems to be around 285 to 295 nm.
- Before this region ($\lambda < 270 \text{ nm}$), there is a formation of **photoproduct** from the measured compound, therefore, the quantum yield values are lower.
- After this region ($\lambda > 300 \text{ nm}$), both actinometer and spectrum are not absorbing anymore, therefore, very **large error** is obvious in the the result.

Example: Photodecomposition QY calculation – statistics on 285-295 nm region

- Now, we can pick the linear region, average the values in it and calculate the standard deviation. For that, we crop x and y variables that we got from the calculation. The x variable represent the wavelengths and y value the quantum yields at these wavelengths. Both objects are NumPy 2D arrays (matrices).

```
In [12]: x[0]
Out[12]:
array([240., 241., 242., 243., 244., 245., 246., 247., 248., 249., 250.,
       251., 252., 253., 254., 255., 256., 257., 258., 259., 260., 261.,
       262., 263., 264., 265., 266., 267., 268., 269., 270., 271., 272.,
       273., 274., 275., 276., 277., 278., 279., 280., 281., 282., 283.,
       284., 285., 286., 287., 288., 289., 290., 291., 292., 293., 294.,
       295., 296., 297., 298., 299., 300., 301., 302., 303., 304., 305.,
       306., 307., 308., 309., 310., 311., 312., 313., 314., 315., 316.,
       317., 318., 319., 320.])
```

```
In [13]: start_idx, end_idx = 285 - 240, 295 - 240
```

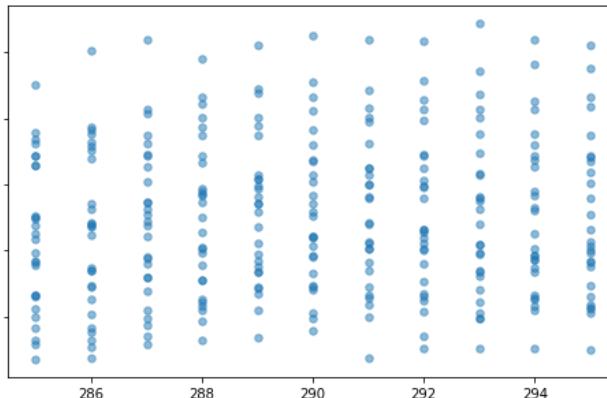
```
In [14]: start_idx, end_idx
Out[14]: (45, 55)
```

```
In [15]: x = x[:, start_idx:end_idx+1] ← crop matrices
```

```
In [16]: y = y[:, start_idx:end_idx+1]
```

```
In [17]: y.shape → statistics from 27*11 = 297 values
Out[17]: (27, 11)
```

```
In [18]: plt.scatter(x, y, s=30, alpha=0.5)
Out[18]: <matplotlib.collections.PathCollection at 0x16e1286a438>
```

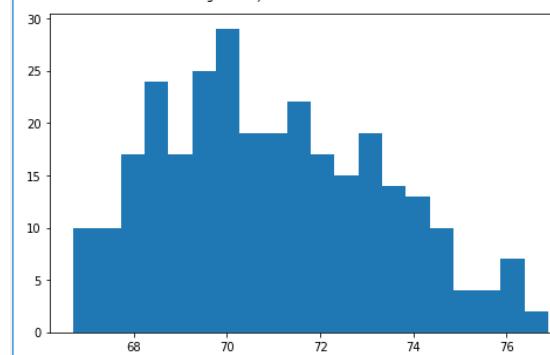


```
In [19]: avg, stdev = np.average(y), np.std(y)
```

```
In [20]: avg, stdev
Out[20]: (70.98725947331127, 2.3565530300937145)
```

- So, we can conlude, that the photodecomposition quantum yield is $\Phi_{\text{com}} = (71 \pm 2)\%$
- Also, we can plot histogram from these 297 values to see, whether we can use standard deviation estimation, because it should be used only if the data are normally distributed. Well, it seems more or less OK, but it would be better to have more results for proper estimation.

```
In [27]: plt.hist(y.flatten(), bins=20)
Out[27]:
(array([10., 10., 17., 24., 17., 25., 29., 19., 19., 22., 17., 15., 19.,
       14., 13., 10., 4., 4., 7., 2.]),
array([66.70384466, 67.21353199, 67.72321932, 68.23290666, 68.74259399,
       69.25228132, 69.76196865, 70.27165599, 70.78134332, 71.29103065,
       71.80071799, 72.31040532, 72.82009265, 73.32977998, 73.83946732,
       74.34915465, 74.85884198, 75.36852931, 75.87821665, 76.38790398,
       76.89759131]),
<a list of 20 Patch objects>)
```



- Stdev is calculated by function `np.std()` and the result is sigma, so the error is given in the statistical level of **one sigma**

$$\sigma = \sqrt{\frac{1}{N} \sum_{i=1}^N (x_i - \mu)^2}, \text{ where } \mu = \frac{1}{N} \sum_{i=1}^N x_i$$

<https://docs.scipy.org/doc/numpy/reference/generated/numpy.std.html>

- Some comment regarding the script design is needed. Here, the QY values were calculated by using closest combination, however, we could combine the two sets differently, eg. by combining each difference in one set with each difference in other set. We would get more points. Or we could combine all possible differences from one set with all possible differences with other set. If the set would contain 4 spectra, we could choose at max 6 differences. If we combined two sets, we would get $6*6=36$ results (instead of 3 as in this simple case).
- Also, the correction to unequal absorbances here is only approximation (it is OK if we keep the absorbance changes small), it can be done better, but the code would be much more complicated (some numerical integration should probably be involved)
- Possible code modifications are left as an excercise for the reader

Example: Photodecomposition QY calculation

- Source code from `script.py`

```

1.import numpy as np
2.import matplotlib.pyplot as plt
3.from spectrum import Spectrum
4.
5.
6.def calculate_phis(item, c_0_act, c_0_com, act_idxs, com_idxs, phi_act, wl_range=(240, 330), em_source=None):
7.    sp = item[act_idxs[0]][0]
8.
9.    idx_0 = Spectrum.find_nearest_idx(sp.data[:, 0], wl_range[0])
10.   idx_1 = Spectrum.find_nearest_idx(sp.data[:, 0], wl_range[1]) + 1
11.
12. # hardcoded number of solutions, this is not general, for other use, please change this number or rewrite the code
13. solutions = 27
14. wls = idx_1 - idx_0 # number of wavelengths
15. wls_mat = np.tile(sp.data[idx_0:idx_1, 0], (solutions, 1)) # define matrix of wavelengths (x values for plotting)
16. matrix = np.zeros((solutions, wls)) # define matrix of results, dimensions: 27 x [number of wavelengths]
17.
18. # for each wavelength in defined range
19. for i in range(wls):
20.     # get the current wavelength
21.     wl = sp.data[idx_0 + i, 0]
22.
23.     # calculate the 27 values at this wavelength
24.     phis = calculate_phi(item, c_0_act, c_0_com, act_idxs, com_idxs, phi_act, wl=wl, em_source=em_source)
25.
26.     # fill the i-th column of the matrix with the solution
27.     matrix[:, i] = np.asarray(phis)
28.
29. # plot the matrix, parameter s is size of the circle, alpha sets transparency
30. plt.rcParams['figure.figsize'] = [8, 5]
31. plt.scatter(wls_mat, matrix, s=30, alpha=0.1)
32. plt.ylabel('$\Phi$')
33. plt.xlabel('Wavelength / nm')
34. plt.show()
35.
36. return wls_mat, matrix
37.
38.
39.def calculate_phi(item, c_0_act, c_0_com, act_idxs, com_idxs, phi_act, wl=285, em_source=None):
40. # define list of results
41. phis = []
42.
43. # for every actinometer set
44. for act in act_idxs:
45.     # for every compound set
46.     for com in com_idxs:
47.         # perform calculation between current sets and add the results into list
48.         phis += calculate_set(item[act], item[com], c_0_act, c_0_com, phi_act, wl, em_source)
49.
50. return phis

```

- Presented code is definitely not perfect and could be optimized in many ways, it is, however, sufficient for demonstration purposes

```

53. def calculate_set(act_set, com_set, c_0_act, c_0_com, phi_act, wl=285, em_source=None):
54.     # assuming that the number spectra in actinometer and compound set is the same
55.     # thus, len(act_set) == len(com_set), ** both act_set and com_set are instances of SpectrumList!
56.
57.     # number of items in set
58.     length = len(act_set)
59.
60.     # find index in data array of actinometer and compound that corresponds to the choosed wavelength
61.     idx_act = Spectrum.find_nearest_idx(act_set[0].data[:, 0], wl)
62.     idx_com = Spectrum.find_nearest_idx(com_set[0].data[:, 0], wl)
63.
64.     # set absorbances at time zero for both actinometer and compound at wavelength wl
65.     A0_act = act_set[0].data[idx_act, 1]
66.     A0_com = com_set[0].data[idx_com, 1]
67.
68.     # calculate the ratio of c0 and A0 for actinometer and compound
69.     c0A0_act = c_0_act / A0_act
70.     c0A0_com = c_0_com / A0_com
71.
72.     # define list of computed quantum yields
73.     phis = []
74.
75.     for i in range(1, length):
76.         # calculate change of absorbance between two measurements for act and com
77.         dA_act = act_set[i - 1].data[idx_act, 1] - act_set[i].data[idx_act, 1]
78.         dA_com = com_set[i - 1].data[idx_com, 1] - com_set[i].data[idx_com, 1]
79.
80.         # calculate corresponding change in concentrations
81.         dc_act = c0A0_act * dA_act
82.         dc_com = c0A0_com * dA_com
83.
84.         # calculate the changes in times, use the value in name of the spectrum
85.         dt_act = float(act_set[i - 1].name) - float(act_set[i].name)
86.         dt_com = float(com_set[i - 1].name) - float(com_set[i].name)
87.
88.         # calculate the quantum yield
89.         phi_com = phi_act * dc_com * dt_act / (dc_act * dt_com)
90.
91.         # if user provided emission source, calculate the correction
92.         if em_source is not None:
93.             # calculate the absorbed light intensity (we use absorption spectrum
94.             # at the begining of the interval - i-1
95.             abs_int_act = em_source * (1 - 10 ** (-act_set[i - 1]))
96.             abs_int_com = em_source * (1 - 10 ** (-com_set[i - 1]))
97.
98.             # integrate those spectra
99.             x_abs_act = abs_int_act.integral()
100.            x_abs_com = abs_int_com.integral()
101.
102.             # apply the correction
103.             phi_com *= x_abs_act / x_abs_com
104.
105.             # add the result to the list
106.             phis.append(phi_com)
107.
108.     return phis

```

QY calculation →

correction →