

# Problem Set 3

Phys 715

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## 1 Sunlight

We want to find the energy density per volume for blackbody radiation for given wavelength  $\lambda$  and temperature  $T$ .

### (a) $\rho(T, \lambda)$

We can start with the expression for  $U$  in terms of an integral over  $k$ :

$$U = 2 \frac{V}{(2\pi)^3} 4\pi \hbar c \int_0^\infty dk k^3 \left( e^{\beta \hbar c k} - 1 \right)^{-1} \quad (1)$$

We want this integral in terms of  $\lambda$ . We use  $k = \frac{2\pi}{\lambda}$ , and  $dk = -\frac{2\pi}{\lambda^2} d\lambda$ , to rewrite:

$$U = 2 \frac{V}{(2\pi)^3} 4\pi \hbar c \int_{k=0}^{k=\infty} dk \left( \frac{2\pi}{\lambda} \right)^3 \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (2)$$

$$U = 2 \frac{V}{(2\pi)^3} 4\pi \hbar c \int_{\lambda=\infty}^{\lambda=0} \left( -\frac{2\pi}{\lambda^2} \right) d\lambda \left( \frac{2\pi}{\lambda} \right)^3 \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (3)$$

$$\frac{U}{V} = \frac{(2\pi)^4}{(\pi)^3} \pi \hbar c \int_{\lambda=\infty}^{\lambda=0} \left( -\frac{1}{\lambda^2} \right) d\lambda \left( \frac{1}{\lambda} \right)^3 \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (4)$$

$$\frac{U}{V} = \frac{(2\pi)^4}{\pi^2} \hbar c \int_{\lambda=\infty}^{\lambda=0} \left( -\frac{1}{\lambda^2} \right) d\lambda \left( \frac{1}{\lambda} \right)^3 \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (5)$$

$$\frac{U}{V} = -16\pi^2 \hbar c \int_{\lambda=\infty}^{\lambda=0} d\lambda \frac{1}{\lambda^5} \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (6)$$

$$\frac{U}{V} = 16\pi^2 \hbar c \int_0^\infty d\lambda \frac{1}{\lambda^5} \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (7)$$

We want the energy density per wavelength, so we can identify  $\rho$  with the integrand:

$$\rho(\lambda, \beta) = 16\pi^2 \hbar c \frac{1}{\lambda^5} \left( e^{\frac{\beta \hbar c 2\pi}{\lambda}} - 1 \right)^{-1} \quad (8)$$

This means

$$\rho(\lambda, T) = \frac{16\pi^2 \hbar c}{\lambda^5} \left( e^{\frac{2\pi \hbar c}{k_B T \lambda}} - 1 \right)^{-1} \quad (9)$$

## (b) Finding $\lambda$ to maximise $\rho$

Let's rewrite  $\rho$  a bit:

$$\rho(\lambda, T) = \frac{16\pi^2\hbar c}{\lambda^5 \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right)} \quad (10)$$

To simplify the math a bit, we can notice that maximising  $\rho$  should be equivalent to minimising the denominator. Let's do that:

$$0 = \frac{d}{d\lambda} \left( \lambda^5 \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) \right) \quad (11)$$

$$0 = \lambda^5 \frac{d}{d\lambda} \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) + \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) \frac{d}{d\lambda} \lambda^5 \quad (12)$$

Continuing on, pausing only to admire symmetry,

$$-\lambda^5 \frac{d}{d\lambda} \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) = \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) \frac{d}{d\lambda} \lambda^5 \quad (13)$$

$$-\lambda^5 e^{\frac{2\pi\hbar c}{k_B T \lambda}} \left( -\frac{2\pi\hbar c}{k_B T \lambda^2} \right) = \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) (5\lambda^4) \quad (14)$$

$$\frac{2\pi\hbar c}{k_B T} \lambda^3 e^{\frac{2\pi\hbar c}{k_B T \lambda}} = \left( e^{\frac{2\pi\hbar c}{k_B T \lambda}} - 1 \right) (5\lambda^4) \quad (15)$$

Defining  $y = \frac{2\pi\hbar c}{k_B T}$ :

$$y e^{\frac{y}{\lambda}} = 5\lambda \left( e^{\frac{y}{\lambda}} - 1 \right) \quad (16)$$

$$\frac{y}{\lambda} = 5 \left( 1 - e^{-\frac{y}{\lambda}} \right) \quad (17)$$

I solved this for  $\frac{y}{\lambda}$  in Mathematica, giving us two solutions. There's the pathological solution where  $\frac{y}{\lambda} = 0$ . This corresponds to  $\lambda$  going to infinity. Looking back at (9), we may find ourselves more interested in the other solution, which we get to be  $\frac{y}{\lambda} \approx 4.966$ . We can solve this for  $\lambda_{\max}$  for a given temperature:

$$\frac{y}{\lambda_{\max}} = 4.966 \quad (18)$$

$$\lambda_{\max} = \frac{y}{4.966} \quad (19)$$

$$\lambda_{\max} = \frac{2\pi\hbar c}{4.966 k_B T} \quad (20)$$

### (c) Solar temperature from $\lambda_{\max}$

We're given here that  $\lambda_{\max}$  for sunlight is 480 nm, and we want to find the temperature of the radiation-emitting surface of the sun. We can solve (20) for  $T$ :

$$T = \frac{2\pi\hbar c}{4.966k_B\lambda_{\max}} \quad (21)$$

I unnecessarily used Mathematica to solve for  $T$ , giving  $T = 6036$  K, which seems to be around the number I get when searching around.

### (d) Differences between real spectrum and blackbody

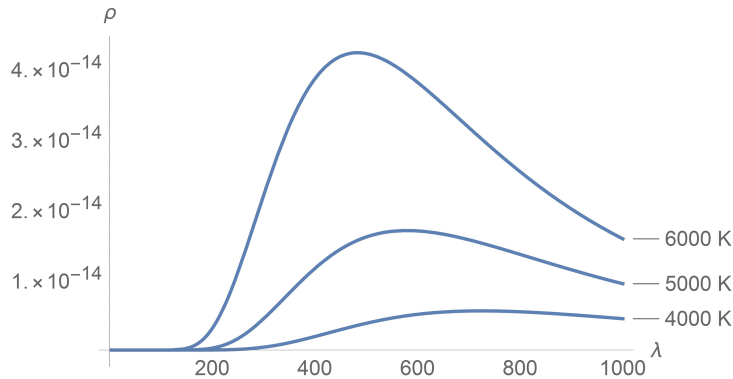


Figure 1: Spectra for 4000 K, 5000 K and 6000 K

The mention of the surface of the sun is important for the earlier part, as different layers of the sun are at different temperatures. Some brief searching suggests that the photosphere ranges from 4000 K to 6000 K. Figure 1 shows some spectra for this temperature range. The sum of these spectra (depending on the total power at each temperature for the overall scale) may probably look different than the spectrum from just one temperature.

Because of their historical importance, we also know that absorption lines are visible in the sun's spectrum. We'd expect these to show up as sharp dips in the spectrum at specific wavelengths, which would be very easily to differentiate from the blackbody spectrum.

#### Listing 1: Mathematica script

```
1 (* ::Package:: *)  
2
```

```

3 BeginPackage["PS3ProblScript`"]
4
5 CurrentDir = DirectoryName[FileNameJoin[{Directory[],
    $ScriptCommandLine[[1]]}]]
6 ImageDir = FileNameJoin[{CurrentDir, "images"}]
7 outFile = OpenWrite[FileNameJoin[{CurrentDir, "problScriptOutput.
    txt"}]]
8
9 Print["All_the_stuff_for_problem_1"]
10
11 (* Part b *)
12 sols = NSolve[ x == 5(1 - E(-x)), x]
13
14 Print[StringTemplate["Got_solutions:_`1`"] [x /. sols]]
15 WriteString[outFile, StringTemplate["Got_solutions_for_part_b:_
    `1`\n"] [x /. sols]]
16
17 (* Part c *)
18 T[\[Lambda]_] := UnitSimplify[(2 * Pi * Quantity[1, "
    ReducedPlanckConstant"] * Quantity[1, "SpeedOfLight"]) /
    (4.966 * Quantity[1, "BoltzmannConstant"] * \[Lambda])]
19 Print[StringTemplate["Temperature:_`1`"] [T[ Quantity[480, "
    Nanometers"]]]]
20 WriteString[outFile, StringTemplate["Solar_temperature_for_480_nm:
    _`1`\n"] [T[ Quantity[480, "Nanometers"]]]]
21
22 (* Part d *)
23
24 rho[\[Lambda]_, T_] := (16 * Pi2 * Quantity[1, "
    ReducedPlanckConstant"] * Quantity[1, "SpeedOfLight"]) / (\[
    Lambda]5 (E((2 * Pi * Quantity[1, "ReducedPlanckConstant"] *
    Quantity[1, "SpeedOfLight"])/( Quantity[1, "BoltzmannConstant"]
    * T * \[Lambda])) - 1))
25
26 Print["Plotting_spectra..."]
27 Export[FileNameJoin[{ImageDir, "4000And6000Spectrum.jpg"}],
28     Show[
29         Plot[rho[Quantity[1, "Nanometers"], Quantity[4000,
            "Kelvins"]], {1, 1, 1000}, AxesLabel->{\[Lambda], \[Rho]},
            PlotLabels->"4000_K",
30         Plot[rho[Quantity[1, "Nanometers"], Quantity[5000,
            "Kelvins"]], {1, 1, 1000}, AxesLabel->{\[Lambda], \[Rho]},
            PlotLabels->"5000_K",
31         Plot[rho[Quantity[1, "Nanometers"], Quantity[6000,
            "Kelvins"]], {1, 1, 1000}, AxesLabel->{\[Lambda], \[Rho]},
            PlotLabels->"6000_K",
32         PlotRange->All, PlotLabels->Automatic
33     ],
34     ImageResolution -> 1000

```

```

35 ]
36
37
38 EndPackage[]

```

Listing 2: Mathematica output

```

1 Got solutions for part b: {0., 4.96511}
2 Solar temperature for 480 nm: 6035.95 kelvins

```

## 2 Langevin Function

## 3 Molecules as Harmonic Oscillators

## 4 Phase transition of Ideal Bose System

## 5 1-D Ising Model Long-Range interactions

## 6 Spin-1 Partition functions

We're told for this problem that we have a 1-D array of  $N$  spins  $s_1, \dots, s_N$ , where  $s_i \in \{-1, 0, 1\}$ . We're also, importantly, given periodic boundary conditions  $s_{N+1} = s_1$ . Our Hamiltonian has no external field:

$$\mathcal{H} = -J \sum_{i=1}^N s_i s_{i+1} \quad (22)$$

We're asked to effectively find the transfer matrix, and show that the Helmholtz free energy  $A$  satisfies

$$A = -Nk_B T \log \lambda_+, \quad (23)$$

where  $\lambda_+$  is the largest eigenvalue of the transfer matrix  $\mathcal{T}$ .

### (a) Helmholtz free energy

We want to start by finding the partition function  $Q$ , as we know that  $A = -k_B T \log Q$ . For  $Q$ , summing over all values of each spin:

$$Q = \sum_{s_1} \dots \sum_{s_N} e^{-\beta \mathcal{H}} \quad (24)$$

$$= \sum_{s_1} \dots \sum_{s_N} e^{-\beta (-J \sum_{i=1}^N s_i s_{i+1})} \quad (25)$$

$$= \sum_{s_1} \dots \sum_{s_N} e^{\sum_{i=1}^N \beta J s_i s_{i+1}} \quad (26)$$

$$= \sum_{s_1} \dots \sum_{s_N} \prod_{i=1}^N e^{\beta J s_i s_{i+1}} \quad (27)$$

This is the point where we can define a matrix  $\mathcal{T}$  such that

$$\mathcal{T}_{ab} = e^{\beta J ab}, \quad (28)$$

where the indices run over possible spin values. Substituting this in for  $Q$  leads to:

$$Q = \sum_{s_1} \dots \sum_{s_N} \prod_{i=1}^N \mathcal{T}_{s_i s_{i+1}} \quad (29)$$

$$= \sum_{s_1} \dots \sum_{s_N} \mathcal{T}_{s_1 s_2} \mathcal{T}_{s_2 s_3} \dots \mathcal{T}_{s_{N-1} s_N} \mathcal{T}_{s_N s_1} \quad (30)$$

We recognise this sum over repeated indices as basic matrix multiplication:

$$Q = \sum_{s_1} \mathcal{T}_{s_1 s_1}^N \quad (31)$$

It's important here that we had a periodic boundary condition, or we'd have some extra work to do for the first and last spin and the next step wouldn't be possible. We have only the  $s_1$  index left to sum over, and we recognise here the trace of  $\mathcal{T}^N$ :

$$Q = \text{Tr } \mathcal{T}^N \quad (32)$$

Why's this good? We know that  $\text{Tr } \mathcal{T} = \sum \lambda_i$ , where the  $\lambda$  are the eigenvalues of  $\mathcal{T}$ , and we also know that  $\text{Tr } \mathcal{T}^N = \sum \lambda_i^N$ . As with the examples of the transfer matrix method, we can make the assumption that the three

eigenvalues for this Hamiltonian's transfer matrix have a  $T$ - and  $J$ - independent order, and name them  $\lambda_+$ ,  $\lambda_0$  and  $\lambda_-$  (where  $\lambda_+ > \lambda_0 > \lambda_-$ ). Thus we write

$$Q = \text{Tr } \mathcal{T}^N \quad (33)$$

$$= \lambda_+^N + \lambda_0^N + \lambda_-^N \quad (34)$$

$$= \lambda_+^N \left( 1 + \left( \frac{\lambda_0}{\lambda_+} \right)^N + \left( \frac{\lambda_-}{\lambda_+} \right)^N \right) \quad (35)$$

For sufficiently large  $N$ , the quantity in parentheses goes to 1 (because each parenthesised quantity therein contained goes to 0), so

$$Q = \lambda_+^N \quad (36)$$

The Helmholtz free energy is trivial:

$$A = -k_B T \log Q \quad (37)$$

$$A = -k_B T \log (\lambda_+^N) \quad (38)$$

$$A = -N k_B T \log \lambda_+ \quad (39)$$

This is what we wanted to show for this part.

### (b) The components of the transfer matrix

We can go back to (28):,  $\mathcal{T}_{ab} = e^{\beta J ab}$ , and recall that  $a$  and  $b$  ran over the values for the spins,  $\{-1, 0, 1\}$ . If either  $a$  or  $b$  is 0,  $\mathcal{T}_{ab} = 1$ , and the other components are only slightly more complicated:

$$\mathcal{T} = \begin{pmatrix} \begin{matrix} s=1 & s=0 & s=-1 \\ e^{\beta J} & 1 & e^{-\beta J} \\ 1 & 1 & 1 \\ e^{-\beta J} & 1 & e^{\beta J} \end{matrix} \end{pmatrix} \begin{matrix} s=1 \\ s=0 \\ s=-1 \end{matrix} \quad (40)$$

### (c) Equation for $\lambda_+$

We want to find the eigenvalues of  $\mathcal{T}$ . Let's solve our standard equation:

$$0 = \det(\mathcal{T} - \lambda I) \quad (41)$$

$$= \det \begin{pmatrix} e^{\beta J} - \lambda & 1 & e^{-\beta J} \\ 1 & 1 - \lambda & 1 \\ e^{-\beta J} & 1 & e^{\beta J} - \lambda \end{pmatrix} \quad (42)$$

To spare my fingers the extra typing, we'll define  $a = e^{\beta J}$  and  $b = e^{-\beta J}$ .

$$0 = \det \begin{pmatrix} a - \lambda & 1 & b \\ 1 & 1 - \lambda & 1 \\ b & 1 & a - \lambda \end{pmatrix} \quad (43)$$

$$= (a - \lambda) ((1 - \lambda)(a - \lambda) - 1) - ((a - \lambda) - b) + b(1 - (1 - \lambda)b) \quad (44)$$

$$= (a - \lambda) (a + \lambda^2 - \lambda(1 + a) - 1) - a + \lambda + b + b(1 - b + b\lambda) \quad (45)$$

$$= (a - \lambda) (\lambda^2 - \lambda(1 + a) + a - 1) - a + \lambda + b + b - b^2 + b^2\lambda \quad (46)$$

$$= (a - \lambda) (\lambda^2 - \lambda(1 + a) + a - 1) + \lambda(1 + b^2) - a + 2b - b^2 \quad (47)$$

$$= a\lambda^2 - \lambda(a + a^2) + a^2 - a - \lambda^3 + \lambda^2(1 + a) - \lambda a + \lambda + \lambda(1 + b^2) - a + 2b - b^2 \quad (48)$$

$$= -\lambda^3 + \lambda^2(2a + 1) + \lambda(-a - a^2 - a + 1 + 1 + b^2) + a^2 - a - a + 2b - b^2 \quad (49)$$

$$= -\lambda^3 + \lambda^2(2a + 1) + \lambda(b^2 - a^2 - 2a + 2) + a^2 - b^2 - 2a + 2b \quad (50)$$

This cubic equation seems rather ugly, but not unexpectedly so. I'm sure this can be simplified further by plugging in for  $a$  and  $b$  and writing some hyperbolic trig functions, but I don't think that helps, particularly given that we're not going to be solving this. It's exactly solvable though, and that's quite cool.