### Watching paint dry

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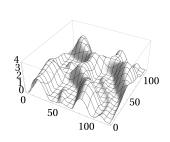




## "Wrinkling paint" vs simulation results



Wrinkling paint



Simulation results with dichloromethane



## Paint: Viscosity changes can affect paint stability

- 1. Paint is a highly viscous medium
- 2. Paint has molecules of different shapes and sizes and invariably components of different **concentrations**
- 3. The viscosity of the paint matters a lot because:
  - Highly viscous paint would stay put on a vertical surface (falling films)
  - ► Highly viscous paint is used to create "impasto" art <sup>4</sup>
  - ► In high viscosity paint, the pigment components don't move about too fast as they are impeded
  - Low viscosity paint behaves in the opposite manner
  - ► The viscosity of the paint, determines the span of time required for the coat of paint to harden (to polymerize)
  - ► So if the viscous time scale is **larger** than the Thermo/solutocapillarity time scales, there is greater chance of severe wrinkling



### Impasto Art



Impasto Art: Art that sticks out

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## Addition of a solvent to paint

- 1. Turpentine is added to paint to make it's viscosity "manageable".
- 2. There is such a thing as too much turpentine. It causes "wrinkling".
- 3. Wrinkling <sup>6</sup> can be caused because: non-uniform evaporative mass flux (of the turp.), + non-uniform vapor recoil = thermocapillary stresses on the surface = migration of paint polymer particles.
- 4. The non-uniform distribution of polymeric paint molecules innately sets up solutocapillarity.



## Smudged paint





Oil paint "wrinkling" as a result of dilution with solvent

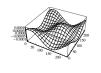
## Controlling thermocapillary (TC) wrinkling

- Based on our numerical observation, TC wrinkling can be controlled to produce desired patterns with the proper application of perturbations to drying paint \*
- 2. Our observations suggest:
  - 2.1 The initial condition has an effect on the final structure.
  - 2.2 Given the physical characteristics of the paint (thickness of coat, strength of thermocapillarity, strength of gravity), desired patterns (wavelengths) can be produced.
  - 2.3 This should be valid for polymeric substances like paint.
- \* Our numerical evidence is based on the assumption that a layer of paint is susceptible to long wave instabilities.



#### Numerical evidence:

#### Smooth sine-like initial conditions



$$L=\lambda_{\max}$$
,  $\cos$ 



$$L = \lambda_{\mathsf{max}}, \, \mathsf{cos}^2$$

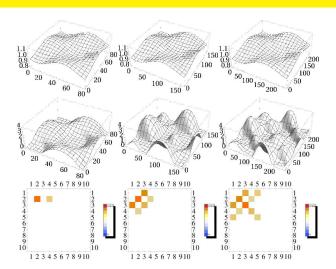


$$\mathit{L} = \lambda_{\mathsf{max}}$$
,  $\cos \sin$ 





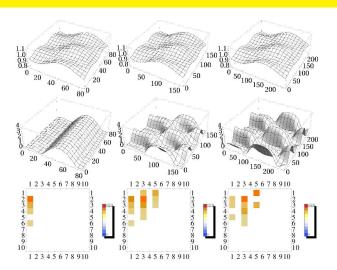
## Effect of initial conditions with $L = n\lambda_{\text{max}}$ , where n = 1, 2, 3 E=0.0001, G=0.0







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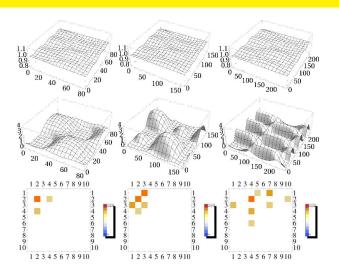






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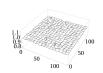
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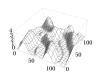
## Numerical evidence: Jagged white noise (Using Dichloromethane(DCM) in zero gravity as an example\*)



Initial condition



DFT, Initial condition



Rupture

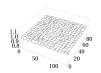


DFT, rupture

Random perturbations, 2.35 mm DCM, g=0.0  $m/s^2$  in a domain size where  $L=n\lambda_{\max}$ , n=non whole number

\* "Thick" DCM in zero gravity would behave like "thin" DCM in regular gravity

# Numerical evidence: noisy initial condition (Using DCM in zero gravity as an example)





Initial condition



DFT, Initial condition



DFT, rupture

Rupture



Random perturbations, 2.35 mm DCM,  $g=0.0 \ m/s^2$  in a domain size with rectangular domain

#### Conclusions

- ► Thermocapillarity and solutocapillarity cause wrinkling in paint.
- ► If paint were subject to long wave instabilities, we can "create" art by controlling the perturbation that a coat of paint is subject to.
- ► The nature of the perturbation changes mean curvature which in turn has an effect on strength of various terms.





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