

# Module 8: Cellular Respiration

## Keys to Success & Study Guide

### Learning Objectives

By the end of this module, you should be able to:

1. **Map** the stages of cellular respiration to their locations in the cell.
2. **Calculate** the inputs and outputs (ATP, NADH, FADH<sub>2</sub>) for each stage.
3. **Explain** the function of the Electron Transport Chain and chemiosmosis.
4. **Compare** aerobic respiration, anaerobic respiration, and fermentation.

### Key Terminology Checklist

*Define these terms in your own words to ensure mastery.*

- [ ] **Aerobic vs. Anaerobic:** Requiring oxygen vs. not requiring oxygen.
- [ ] **Oxidation:** Loss of electrons (glucose is oxidized during respiration).
- [ ] **Reduction:** Gain of electrons (oxygen is reduced to water).
- [ ] **Chemiosmosis:** ATP synthesis driven by the flow of H<sup>+</sup> ions down their electrochemical gradient through ATP synthase.
- [ ] **ATP Synthase:** The enzyme complex that synthesizes ATP from ADP and P<sub>i</sub>.
- [ ] **Cristae:** Inner mitochondrial membrane folds that increase surface area for the ETC.

### Concept Check

#### 1. The Electron Transport Chain

- **Question:** What is the role of the ETC?
- **Key Answer:** The ETC transfers electrons through protein complexes, using the released energy to pump H<sup>+</sup> ions across the inner membrane. The resulting proton gradient drives ATP synthesis via ATP synthase (oxidative phosphorylation).

## 2. Oxygen's Role

- **Question:** Why is oxygen essential for aerobic respiration?
- **Key Answer:** Oxygen is the final electron acceptor in the ETC. Without it, electrons cannot flow, the proton gradient collapses, and ATP production ceases. Oxygen is reduced to water.

## 3. Fermentation

- **Question:** Why do cells perform fermentation?
- **Key Answer:** Fermentation regenerates  $\text{NAD}^+$  from NADH, allowing glycolysis to continue producing 2 ATP per glucose in the absence of oxygen. The carbon in pyruvate is converted to ethanol (yeast) or lactate (muscle) rather than fully oxidized.