

P2 Review Midterm - study

Physics 2 (Trường Đại học Quốc tế, Đại học Quốc gia Thành phố Hồ Chí Minh)



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PHYSICS 2: FLUID MECHANICS AND THERMODYNAMICS

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Midterm Review

Chapter 1: Fluid Mechanics

density:
$$\rho = \frac{m}{V}$$

pressure:
$$p = \frac{F}{A}$$

Pascal's law:

$$\frac{F_i}{A_i} = \frac{F_0}{A_0}$$

Archimede's principal:

$$F_b = \rho_{fluid} gV$$
: buoyant force

Equation of continuity:

$$A_1 V_1 = A_2 V_2 = constant$$

Bernoulli's equation:

$$p_1 + \frac{1}{2}\rho v_1^2 + \rho g y_1 = p_2 + \frac{1}{2}\rho v_2^2 + \rho g y_2 = constant$$

Chapter 2: Heat, Temperature and the First Law of Thermodynamics

Temperature:

$$T(K) = T(C^0) + 273.15$$

Thermal expansion:

Linear expansion: (solids)

$$\Delta L = L \alpha \Delta T$$

Area expansion: (solids)

$$\Delta A = A\alpha_A \Delta T; \alpha_A = 2\alpha$$

Volume expansion: (solids and liquids)

$$\Delta V = V \beta \Delta T; \beta = 3\alpha$$

Heat capacity:

$$Q = C \Delta T = C (T_f - T_i)$$

Specific capacity:

$$Q = cm \Delta T = cm (T_f - T_i)$$

Latent heat:

$$Q = Lm$$

The first law of thermodynamics:

$$\Delta E_{int} = E_{int,f} - E_{int,i} = Q - W$$

Three special cases:

1. Adiabatic processes:

$$Q = 0 \Rightarrow \Delta E_{int} = -W$$

2. Constant-volume (isochoric) processes:

$$W = 0 \Rightarrow \Delta E_{int} = Q$$

3. Cyclical processes:

$$\Delta E_{int} = 0 \Rightarrow Q = W$$

Work done by the gas:

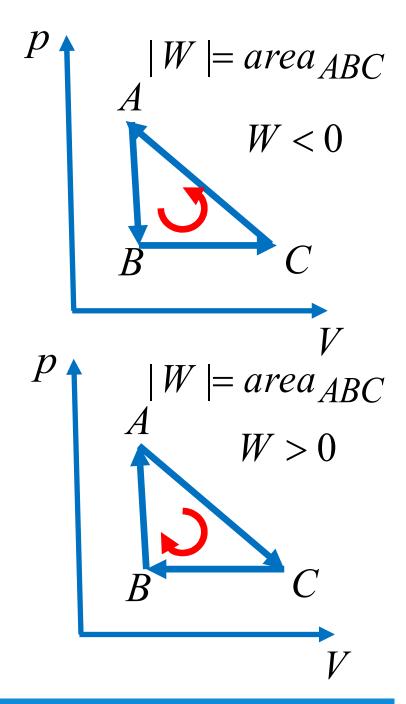
-Expansion: W > 0

-Compression: W < 0

Energy transferred as heat Q:

-Heat transferred to the gas (receiving energy as heat):

-Heat transferred from the gas (releasing energy as heat):



Heat Transfer Mechanisms:

Conduction:

$$P_{\text{cond}} = \frac{Q}{t} = kA \frac{T_{\text{H}} - T_{\text{C}}}{L} \qquad \text{(Unit: W = J/s)}$$

Steady-state:

$$P_{\text{cond}} = \frac{k_2 A(T_H - T_X)}{L_2} = \frac{k_1 A(T_X - T_C)}{L_1}$$

If the slab consists of n materials:

$$P_{\text{cond}} = \frac{A(T_{\text{H}} - T_{\text{C}})}{\sum_{i=1}^{n} (L_i/k_i)}$$