

# Phy2 test - excersice

Physics 2 (Trường Đại học Quốc tế, Đại học Quốc gia Thành phố Hồ Chí Minh)

# **Midterm Examination**

Date: 09/12/2021 Duration: 120 minutes

**Open Book and Online** 

SUBJECT: PHYSICS 2 (FLUID MECHANICS AND THERMAL PHYSICS)	
(ID: PH014IU)	
Approval by Chair of Department of Physics	Lecturer: Nguyễn Đức Diệu
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Proctor 1	Proctor 2
Signature	Signature
Full name:	Full name:
STUDENT INFO	
Student name:	
Student ID:	

## INSTRUCTIONS: the total of point is 100 (equivalent to 20% of the course)

#### 1. Purpose:

- Test your understanding of basic knowledge from the chapter 1 (Fluid Mechanics) and the chapter 2 (Heat, Temperature and the First Law of Thermal Dynamics) (CLO 3).
- Examine your skill in analysis and design a problem in science and engineering (CLO 2).
- Test your ability in applying knowledge of mathematics and physics (CLO 1).
- Evaluate your English skills in writing communication manner (CLO 4).

#### 2. Requirement:

- Read carefully each question and answer it following the requirements.
- Write the answers and draw models CLEAN and TIDY directly in the exam paper.
- Submit your exam including this cover page and the solutions of the following problems.

# The exam has five (5) questions in total

## **QUESTION 1:** (20 marks)

A large pipe with a cross-sectional area of 1 m<sup>2</sup> descends 5 m in height and narrows to 0.5 m<sup>2</sup>, where it terminates in a valve at point 1 (see Figure 1). If the assumed pressures at point 1 and 2 are both atmospheric pressure  $(P_0)$ , and the valve is opened wide and water allowed to flow freely, find the speed of the water leaving the pipe.

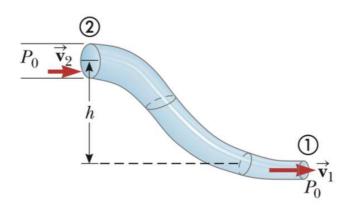


Figure 1

## **QUESTION 2:** (20 marks)

(a) A solid substance has an isotropic density  $\varrho_0$  at a temperature  $T_0$ . If its temperature is increased by an amount  $\Delta T = T - T_0$ , show that its density at the higher temperature is given by (10 marks):

$$\rho = \frac{\rho_0}{1+\beta\Delta T} = \frac{\rho_0}{1+\beta(T-T_0)}$$
, where  $\beta$  is the coefficient of volume expansion.

(b) Assuming the substance is lead, which has a density of  $\varrho = 11.3 \times 10^3 \, \text{kg/m}^3$  at 0°C and a coefficient of thermally linear expansion  $\alpha = 29 \times 10^{-6} \, ^{\circ}\text{C}^{-1}$ . Note that we are still assuming the lead density is isotropic, then the condition  $\beta = 3\alpha$  is valid. What is the density of lead at 90°C? (10 marks)

#### **QUESTION 3:** (20 marks)

- 3.1. A gas expands from I to F along the three paths indicated in Figure 2. Calculate the work done on the gas along paths (a) IAF, (b) IF, and (c) IBF. (10 marks)
- 3.2. A gas expands from I to F in Figure 2. The energy added to the gas by heat is 418 J when the gas goes from I to F along the diagonal path.
- (a) What is the change in internal energy of the gas? (5 marks)
- (b) How much energy must be added to the gas by heat for the indirect path IAF to give the same change in internal energy? (5 marks)

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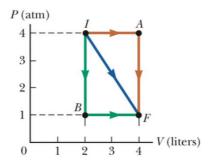


Figure 2

## **QUESTION 4:** (20 marks)

A copper slug whose mass  $m_c = 150$  g is heated in a laboratory oven to a temperature  $T = 312^{\circ}$  C. The slug is then dropped into a glass beaker containing a mass  $m_w = 110$  g of water. The heat capacity  $C_b$  of the beaker is 45 cal/K. The initial temperature  $T_i$  of the water and the beaker is  $12^{\circ}$  C. Assuming that the slug, beaker, and water are an isolated system and the water does not vaporize, find the final temperature  $T_f$  of the system at thermal equilibrium. Given the specific heat capacity of copper and water are  $c_c = 0.0923$  cal/(g.K) and  $c_w = 1.00$  cal/(g.K), respectively.

**Caution**: the heat capacity  $C_b$  of the beaker is accounted for its mass already in this problem, thus  $C_b = c_b m_b$  and its unit is given by cal/K not cal/(g.K) as of the specific heat capacity of water and copper.

#### **QUESTION 5:** (20 marks)

Figure 3 shows the cross section of a wall made of three layers. The layer thicknesses are  $L_1$ ,  $L_2$  = 0.7 $L_1$ , and  $L_3$  = 0.35 $L_1$ . The thermal conductivities are  $k_1$ ,  $k_2$  = 0.9 $k_1$ , and  $k_3$  = 0.8 $k_1$ . The temperatures at the left side and right side of the wall are  $T_H$  = 70° C and  $T_C$  = -35° C, respectively. Thermal conduction is steady.

- (a) What is the temperature difference  $\Delta T_2$  across layer 2 (between the left and right sides of the layer)? (10 marks)
- (b) If  $k_2$  were, instead, equal to  $1.1k_1$ , would the rate at which energy is conducted through the wall be greater than, less than, or the same as previously (10 marks)

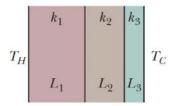


Figure 3

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