

## *Exploring Equilibrium Mini Lab Answers*

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### Exploring Equilibrium Mini Lab Answers

Exploring Equilibrium Mini-Lab Lab Write-up: Answers to questions, graphs, data table. Group. PART A. Login to the computer and open a web browser. Go to . Click "Play with Sims," then find the "Chemistry" section, and choose "Reactions and Rates." When the simulation is open, click on the "Many Collisions" tab. Look at the screen and observe everything you can find out about ...

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b) Describe the nature of dynamic equilibrium when small numbers of particles (such as 50, as compared to  $6.022 \times 10^{23}$ ) are present. 4) Predict what will happen when the temperature is raised so it is NOT above the activation energy max but IS above the energy level of the products. 5) Raise the temperature as described.

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6) Predict what will happen when the temperature is raised so it is above the activation energy max. 7) Raise the temperature as described. Again, was your prediction correct? Describe and explain any differences. 8) What did you notice about the rate at which reactants/products fluctuated between the three different temperatures? If you didn't notice anything, hit "reset all" and test ...

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### Exploring Equilibrium Mini Lab Part A

b) Describe the nature of dynamic equilibrium when small numbers of particles (such as 50, as compared to  $6.022 \times 10^{23}$ ) are present. 4) Predict what will happen when the temperature is raised so it is NOT above the activation energy max but IS above the energy level of the products. 5) Raise the temperature as described.

### Exploring Equilibrium Mini-Lab - Las Lomas Science

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### Exploring Equilibrium Pre Lab Answers - pdfsdocuments2.com

I used "Reactions and Rates" in my regular chem class in this brief activity, which also includes a non-PhET, hands-on activity. My use of the sim was for conceptual purposes only; I was hoping to attack some misconceptions about equilibrium, activation energy, and temperature's affect on

equilibrium position.

### **Exploring Equilibrium - PhET Contribution**

Exploring Equilibrium Lab Pre-Lab Questions 1. False, The reactions continue, but there is an equal balance of opposing reaction rates 2. False, the concentration of reactants and products remain constant with time. They don't necessarily have to be equal. 3. (a) When the humidity is low, the paper will be blue.

### **Color lab - Exploring Equilibrium Lab Pre-Lab Questions 1 ...**

What is equilibrium? How is it reached? Many students assume that the concentrations or amounts of reactants and products—rather than rates—must be equal at equilibrium. This short lab activity helps to dispel that notion.

### **Equilibrium Straw Activity | Carolina.com**

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### **Exploring Equilibrium Mini Lab Part B**

Explain what effects whether the equilibrium position favors the products or the reactants. Predict how addition of a reactant or product will affect the forward and reverse reaction rates, and once this new system reaches equilibrium how the reactant and product concentrations will compare to the original system at equilibrium.

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