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4.3 The student can collect data to answer a particular scientific question. 5.1 The student can analyze data to identify patterns or relationships. 6.1 The student can justify claims with evidence. The Thermodynamics—Enthalpy of Reaction and Hess's Law—AP Chemistry Classic Laboratory Kit is available from Flinn Scientific, Inc.

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Thermodynamics-Enthalpy of Reaction and Hess's Law - bh' Page 2 4 4 ... Pre-lab Questions (Answer on a separate sheet of paper) 1. Define  $\Delta H$ . 2. Define specific heat. ... wibt permissiopl in writing from Flinn Scientific, k. Thermodynamics-Enthalpy of Reaction and Hess's Law Page 3 3. The specific heat of a solution is  $4.18 \text{ J/(g}^\circ\text{C)}$  and its ...

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Thermodynamics - Enthalpy Of Reaction And Hess's Law. This lab demonstrates the principle of Hess's law. Hess's law states that the change in enthalpy is the same whether a reaction takes place in one step or in a series of steps.

### **Thermodynamics -Enthalpy Of Reaction And Hess's Law**

Thermodynamics: Enthalpy of Reaction and Hess's Law Judy Chen Partner: Mint Date: 13 Sept, 2011 Purpose: The purpose of this lab is verify Hess's law by finding the enthalpies of the reactions; NaOH and HCl,  $\text{NH}_2\text{Cl}$  and NaOH, and  $\text{NH}_3$  and HCl. The overall enthalpy should equals to the sum of enthalpy of the three reactions in

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3.) The specific heat of a solution is  $4.18\text{J}/(\text{g}\cdot^{\circ}\text{C})$  and its density is  $1.02\text{g/mL}$ . The solution is formed by combining  $25.0\text{mL}$  of solution A with  $25.0\text{mL}$  of solution B, with each solution initially at  $21.4^{\circ}\text{C}$ . The final temperature of the combined solutions is  $25.3^{\circ}\text{C}$ . Calculate the heat of reaction,  $q_{\text{rxn}}$ , assuming no heat loss to the calorimeter.

**Pre-lab Questions - Thermodynamics-Enthalpy of Reaction ...**

Hess's Law Lab Demonstration with NaOH and HCl (Part 2: Data & Calculation) - Julia Le - Duration: 8:39. Phuong (Julia) Le 29,325 views

**Hess Law Lab Calculations**

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