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Answers***

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With spring just around the corner, your students' thoughts will soon be turning to sunshine, prom, and the NCAA basketball tournament. This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass calculations. The result is the Sweet 16 Chemistry Compound Tournament.

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culations. The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these important topics. The rules for filling out the Tournament brackets are simple: For the first round, the student determines the chemical for-

Sweet 16 Chemistry Compound Tournament SCIENTIFIC

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2005 Flinn Scientific Catalog/Reference Manual, you may have noticed that we have updated all our chemical names, labels, and MSDS to the more familiar Stock system of nomenclature. This change was a result of many chemistry teachers asking us to change to the Stock system.

“Your Safer Source for Science Supplies”

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Potassium phosphate First Round Second Round Semifinals Finals Winner Rules 1. First Round —
Write formula, most number of atoms in formula unit wins 2. Second Round — Greater number of
ions in the formula unit wins 3. Semis — Higher charge on anion wins 4. Finals — The larger molar
mass wins Sweet 16 Chemistry Compound Tournament

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