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Physical Chemistry Fundamentals: Figure 3.6 1. Reversible isothermal expansion from A to B at T_h ; the entropy change is q_h/T_h , where q_h is the energy supplied to the system as heat from the hot source. 2. Reversible adiabatic expansion from B to C. No energy leaves the system as heat, so the change in entropy is zero.

Atkins/ de Paula: Physical Chemistry - Tunghai University

vii This book provides solutions to nearly of the exercises and problems in Mathematics for Physical Chemistry, fourth edition, by Robert G. Mortimer. This edition is a revision of a third edition published by Elsevier/Academic Press in

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