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is a solution in which salt is the solute and water is the solvent. Solutes and solvents can take the form of a solid, liquid, or gas. The solution takes the state of the solvent. Figure 2 lists some common solutions and the states of their respective solutes and solvents. Air, for instance, is a solution of several gases dissolved in another gas.

Section 8.1 8.1 Formation of Solutions - PC\|MAC

A solution is a homogeneous mixture of two or more substances. There are many types of solutions.

- Air — oxygen and other gases dissolved in nitrogen.
- Fish tank — oxygen in water.
- Carbonated beverages — carbon dioxide in water.
- Blood — oxygen and carbon dioxide are essential components of blood chemistry.

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8.1 Formation of Solutions. Chapter 8: Solutions, Acids, and Bases. Dissolving. Any state of matter can become part of a solution. For a solution to form, one substance must dissolve in another (definition of solution) Solution. Substances can dissolve in three ways.

8.1 Formation of Solutions by Rachel Ward on Prezi

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Chapter 16 Solutions 167 SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471-477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471-472) Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2. 1.

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5. One way to view the changes in energy associated with formation of solutions is to break the solution process down into a series of steps. The various steps are listed below. For our example of ethanol in water, circle the appropriate enthalpy change. Base your answer on the IMF's present and how these will be affected. i.

Solutions Worksheet - The Solution Process gas: liquid: solid:

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Chapter 8 Solutions, Acids, and Bases Section 8.1 Formation of Solutions (pages 228–234) This section explains the parts of a solution, the processes that occur when compounds dissolve, and how the properties of a solution compare with those of its solvent and solute.

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