

## *Exploring Equilibrium Post Lab Answers*

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**Exploring Equilibrium Post Lab Answers**

Exploring Equilibrium Lab Pre-Lab Questions 1. False, The reactions continue, but there is an equal balance of opposing reaction rates 2. False, the concentration of reactants and products remain constant with time. They don't necessarily have to be equal. 3. (a) When the humidity is low, the paper will be blue.

**Color lab - Exploring Equilibrium Lab Pre-Lab Questions 1 ...**

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Equilibrium is reached again with the precipitation of more NaCl to compensate for the added Cl<sup>-</sup> ions by shifting to the left Over-concentrated KSCN + Fe<sub>2</sub>(NO<sub>3</sub>)<sub>3</sub> solution Immersed in boiling water, the solution slowly turned blue Fe<sup>3+</sup>(aq) + SCN<sup>-</sup>(aq) FeSCN<sup>2+</sup>(aq) Removing heat

**Equilibrium and Le Chatelier's Principle Lab by on Prezi**

Lab Furniture Lab Techniques ... Exploring Equilibrium: It Works Both Ways—Student Laboratory Kit ... Teacher Notes with sample data and answers to questions and all necessary chemicals and consumable supplies. Complete for 30 students working in pairs. Specifications. Materials Included in Kit: Bromocresol green solution, 0.04%, 35 mL

**Exploring Equilibrium: It Works Both Ways—Student ...**

I used "Reactions and Rates" in my regular chem class in this brief activity, which also includes a non-PhET, hands-on activity. My use of the sim was for conceptual purposes only; I was hoping to attack some misconceptions about equilibrium, activation energy, and temperature's affect on equilibrium position.

**Exploring Equilibrium - PhET Contribution**

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b) Describe the nature of dynamic equilibrium when small numbers of particles (such as 50, as compared to  $6.022 \times 10^{23}$ ) are present. 4) Predict what will happen when the temperature is raised so it is NOT above the activation energy max but IS above the energy level of the products. 5) Raise the temperature as described.

**Exploring Equilibrium Mini-Lab - Las Lomas Science**

This short activity is done after Le Chatelier's Principle is covered in class. Students have already done the first lab, "Exploring Equilibrium," where they investigated temperature changes a little bit; here, they test their ideas and predictions about temperature changes in light of Le Chatelier's Principle. Subject Chemistry

**Exploring Equilibrium 2 (Le Chatelier's Principle) - PhET ...**

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Equilibrium POST-LAB QUESTIONS] 1. Write the chemical equation for the complex-ion equilibrium that results when excess chloride ion is added to an aqueous solution of cobalt chloride. Note the observed color of each complex ion underneath its chemical formula.

### Chemistry Post-Lab Questions | Chemical Equilibrium (7.6K ...

Chemistry Equilibrium Test. STUDY. PLAY. Reaction Rate. The change in concentration of a reactant or product per unit time, generally calculated and expressed in moles per liter per second. Reversible Reaction. A chemical reaction that can occur in both the forward and the reverse directions.

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