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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

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(c)Elevation of boiling point of 1 M KCl solution is nearly double than that of 1 M sugar solution because the van't Hoff factor for 1M KCl is 2 i.e.  $i = 2$  and that of sugar is 1. Q 12. An element,  $X$  (At. mass = 40 g mol<sup>-1</sup>) having f.c.c. structure, has unit cell edge length of 400 pm.

### **CHEMISTRY (Theory) SOLUTION - [d24cdstip7q8pz.cloudfront.net](http://d24cdstip7q8pz.cloudfront.net)**

Theory of conductivity What is conductivity? Conductivity is the ability of a solution, a metal or a gas - in brief all materials - to pass an electric current. In solutions the current is carried by cations and anions whereas in metals it is carried by electrons. How well a solution conducts electricity depends on a number of factors ...

### **Conductivity Theory and Practice - University of Crete**

Chemistry is the natural science involved with elements and compounds composed of atoms, molecules and ions: their composition, structure, properties, behavior and the changes they undergo during a reaction with other substances.. In the scope of its subject, chemistry occupies an intermediate position between physics and biology. It is sometimes called the central science because it provides ...

### **Chemistry - Wikipedia**

In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

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