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20.1 Oxidation States and Oxidation-Reduction Reactions Review section 4.4 of the textbook, especially the Oxidation numbers rules on page 137. Chemistry is an experimental science where measurements are made.

20.1 Oxidation States and Oxidation-Reduction Reactions ...

Reactions occur to produce a more stable arrangement of electrons in an atom, ion, or molecule. The arrangement of electrons in an atom determine whether it will bond with other atoms and which atoms it will bond with. □ Full sets of valance electrons are needed. □ Heat is often a factor in initiating reactions.

9-1 The Nature of Chemical Reactions

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the orientation and the energy of the colliding particles.,for the reaction to occur the particles must be oriented in a good position that lets the bonds to break and the atoms to rearrange otherwise the colliding molecules would just bounce off each other. Sometimes the collision doesn't occur because the colliding particles need enough energy...

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