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quantitative chemical analysis method involving the separation of an analyte from a sample by a physical or chemical process and subsequent mass measurements of the analyte, reaction product, and/or sample indicator substance added to the sample in a titration analysis to permit visual detection of the end point quantitative analysis

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For example, quantitative analysis could determine blood cholesterol levels or the ratio of lipoproteins in plasma or the amount of protein excreted in urine. Here again, quantitative analysis complements qualitative analysis, since the latter identifies the nature of a chemical while the former tells you how much there is.

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