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Chapter 3 Review, pages 186-191 - Pre University Courses

2 Chemistry, First Canadian Edition 2.5 (a) In CO, there is one carbon atom for every oxygen atom (or the ratio of C to O atoms is 1:1). (b) In CH

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29. (a) Answers may vary. Sample answer: Hydrogen gas has a high enthalpy of combustion, releasing about 2.5 times the quantity of energy per gram than methane but the widespread use of hydrogen as a fuel has the limitations of high production cost and difficulty in storage and transport. (b) Answers may vary. Sample answer: Possible ...

Chapter 5 Review, pages 338-339 - Weebly

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