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lattice energy. - the energy released when an ionic solid is formed from its gaseous ions. This is the most favorable energy term in the process. Let's consider the steps involved in the formation of solid potassium chloride from its elements at standard conditions.

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A good answer should state that as the distance between a proton (nucleus) and electron increases, the attractive force decreases. This is followed by a quick class discussion to summarize the previous day's lesson and leads quickly into model 3 and 4 of the POGIL activity.

## answer key - BetterLesson

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Chem 115 POGIL Worksheet - Week 11 - Solutions Chemical Bonds Key Questions. 1. Using the electronegativity values in the table above, calculate  $\delta\chi$  for the pairs of elements forming bonds in each of the following species, and describe the type of bonding as either pure covalent, polar covalent, or ionic: H2, HCl, PbBr2, NaCl, NCl3.

#### Chem 115 POGIL Worksheet - Week 11 - Solutions Chemical Bonds

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