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Naming Acids Worksheet Key Block____ Name____ Name the following acids 1. HCL hydrochloric acid 2. HClO4 perchloric acid 3. HIO3 iodic acid 4. HI hydroiodic acid 5. H2SO4 sulfuric acid 6. H2S hydrosulfuric acid 7. HCN hydrocyanic acid 8. H2CO3 carbonic acid 9.

Name the following acids - Newton Public Schools

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Writing Formulas & Names for Acids / 2 Ternary acids are named based on the ending of the polyatomic ion they contain. Polyatomic ions ending in -ate take the suffix -ic to the root of the non-oxygen nonmetal in the polyatomic, and add the word acid after it. Polyatomic ions ending in -ite take the suffix -ous to the root of the non-oxygen nonmetal in the

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Acids and Bases How do acids and bases behave in water? Why? Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solu- ... Acids and Bases POGIL KEY ...

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Chem 116 POGIL Worksheet - Week 9 Introduction to Acid-Base Concepts Why? Acids and bases are important substances in many settings. In everyday life we encounter acids such as vinegar and lemon juice and bases such as ammonia and lye in drain cleaners. Many important processes in chemistry and biology involve reactions of acids with bases ...

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Naming Molecular Compounds Answers Pogil Activities

Write a rule for naming binary acids. All binary acids t'regin wit/9 theprefix bydro—, which precedes the name of any nonmetal other than hydrogen, in the compound. The name of this nonmetal is modified to end with the safix \sim ic. Example: hydro ic acia'. Naming Acids - 141 Model 2 — Ternary Acids (Oxyacids) Name of Acid in .

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