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a. one that has 5 electrons in the third energy level b. one with an electron configuration that ends

in 4s<sup>2</sup>4p<sup>5</sup> c. the Group 6A element in period 4 5. Identify the elements that have electron configurations that end as follows. a. 2s<sup>2</sup>2p<sup>4</sup> b. 4s<sup>2</sup> c. 3d<sup>10</sup>4s<sup>2</sup> 6. What is the common characteristic of the electron configurations of the elements Ne and Ar?

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