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\ Prentice Hall Chemistry Chapter 19. Prentice Hall Chemistry Chapter 19. Acid dissociation constant: the ration of the concentration of the dissociated form of an acid to the undissociated form; stronger acids have larger values than weaker acids: Acidic solution:

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Chapter 19 Acids, Bases, and Salts 489 Section Review Objectives • Classify a solution as neutral, acidic, or basic, given the hydrogen-ion or hydroxide-ion concentration • Convert hydrogen-ion concentrations into values of pH and hydroxide-ion concentrations into values of pOH • Describe the purpose of pH indicators Vocabulary Key Equations

05 CTR ch19 7/12/04 8:16 AM Page 487 ACID-BASE THEORIES 19

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Chapter 19 "Acids, Bases, and Salts ... Figure 19.22, page 615. Neutralization. The solution of known concentration is called the standard solution. added by using a buret. Continue adding until the indicator . changes color.

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