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Holt ChemFile: Problem-Solving Workbook 246 Equilibrium of Salts, K<sub>sp</sub> Equilibrium of Salts, K<sub>sp</sub> When you try to dissolve a solid substance in water, you expect the solid form to disappear, forming ions or molecules in solution. Most substances, however, are only slightly soluble in water. For example, when you stir silver chloride in water,

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Holt ChemFile: Problem-Solving Workbook 313 Electrochemistry Name Class Date Problem Solving continued You can use reduction potentials to predict the direction in which any redox reaction will be spontaneous. A spontaneous reaction occurs by itself, without outside influence. The redox reaction will proceed in the direction for which the

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