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PEARSON CHEMISTRY: The law of conservation of mass is explained in Lesson 2.4 (p. 50). Students are introduced to balancing chemical equations in Lesson 11.1 (pp. 346- 354). Students learn to write and balance the five different types of chemical reactions in Lesson 11.2 (pp. 356-367).

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Representative Particle Chemical Formula Representative Particles in 1.00 mol Atomic oxygen O Oxygen gas molecule Sodium ion Sodium chloride false Use the chemical formula to find the number of atoms in one molecule and multiply this number by Avogadro's number, the number of particles in one mole. atom $6.02 \cdot 10^{23}$ O $2 \cdot 6.02 \cdot 10^{23}$ ion ...

Chemical Quantities - Mr. Mutic's Chemistry Biology AP Biology

84 Guided Reading and Study Workbook SECTION 9.4 NAMING AND WRITING FORMULAS FOR ACIDS AND BASES (pages 271–273) This section explains the three rules for naming acids and shows how these rules can also be used to write the formulas for acids.

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