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Solutions to problems Solutions to numerical problems Solving for  $n$  from the perfect gas equation  $[1.8]$  yields  $n = pV/RT$  and  $n = m/M$ , hence  $p m/ V = Mp/RT$ . Rearrangement yields the desired relation, that is  $p p M$  or  $p/p$  Draw up the followino table and then plot  $p/ p$  versus  $p$  to find the zero pressure limit of  $p/ p$  where all gases behave ideally.

### CHEM 401—Physical Chemistry Chapter 1 Homework Solutions

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### Chem 332: Physical Chemistry II (Spring 2013)

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