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3. Solutions are said to be in a ____ phase even though the components may have been in different phases before the solution was formed. 4. Pure gold is 24 carat. 14-carat gold contains 14 parts gold and 10 parts other metals. 14-carat gold is said to be a(n) ____, which is a type of ____ solution. An example of a gaseous solution is

Worksheet: Solutions Introduction Name K

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Worksheet: pH Calculations KName EY

7. How many liters of solution can be produced from 2.5 moles of solute if a 2.0 M solution is needed? $2.0 \text{ M} = \frac{2.5 \text{ moles}}{\text{liters of solution}}$ liters of solution = $1.25 \text{ L} = 1.3 \text{ L}$ 8. What would be the concentration of a solution formed when 1.00 g of NaCl are dissolved in water to make 100.0 mL of solution? $? \text{ mol} = \frac{1.00 \text{ g NaCl}}{58.5 \text{ g/mol}}$ $\times 1 \text{ mol NaCl}$ 58.5 g ...

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