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Concept Review: Covalent Bonds. 1. A covalent bond forms when two or more valence electrons are attracted by the positively charged nuclei of two atoms and thus are shared between both atoms. 2. The H₂ molecule is stable because each hydrogen atom now has a shared pair of electrons and has achieved a stable noble gas configuration.

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resource to supplement your studies.Holt Chemistry Bonds Compounds Concept Review Answers Holt Chemistry NY: Chapter 5 - Ions and Ionic Compounds. an ionic compound composed of 3 or more elements, one part of the formula is a metallic cation, the other part is a non-metallic

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Property of ionic compounds when they are in the liquid phase or dissolved in water. The act of the cations and anions in an ionic compound moving apart when dissolved into water. all ionic compounds will do this to some extent. The overall charge that an ion carries.

Chemistry - Chapter 5 - Ions and Ionic Compounds ...

If a bond's character is more than 50% ionic, then the bond is called a(n) ionic bond. 12. A bond's character is more than 50% ionic if the electronegativity difference between the two atoms is greater than 1.7. 13. Write the formula for an example of each of the following compounds: Answers will vary.

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