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Periodic Table and their properties. Compare/contrast properties of elements within a period or group 1, 2, 13-18 on periodic table. Predict group of an element given chemical formula. Predict group of an element by knowing its electron configuration. Explain placement of an unknown element in Periodic Table based on properties.

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Elements and the Periodic Table 1. A substance that cannot be broken down into simpler substances by chemical or physical means is called a(n). 2. The document in which elements are organized by their properties is known as the . 3. Circle the letter of the name for the columns within the periodic table. a. periods b. groups c. metals d ...

Chapter 2 Minerals Section 2.1 Matter

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