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I Concept Review continued. 21. Any mixture that is heterogeneous on a microscopic level is a. 22. A S L/Q WA IIIQ [1 is a. mixture in which particles of the mixture are more or. less evenly dispersed throughout a liquid or gas. ' 23. C C9 I 0 are intermediate between suspensions and solutions. 24.

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Holt Chemistry 6 Stoichiometry Section: Limiting Reactants and Percentage Yield Complete each statement below by choosing a term from the following list. ... Concept Review: Stoichiometry and Cars 1. c 2. b 3. d 4. b 5. d 6. If there is too much oxygen and not enough gasoline, the engine will stall.

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Holt Chemistry 5 Covalent Compounds Section: Drawing and Naming Molecules Complete each statement below by choosing a term from the following list. Terms may be used more than once. ... Concept Review: Covalent Bonds 1. A covalent bond forms when two or more valence electrons are attracted

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Holt Chemistry 4 Acids and Bases Name Class Date Concept Review continued 3. Table 1 shows that  $[H^+]$  and pH are related. As  $[H^+]$  decreases from  $10^{-8} M$  to  $10^{-10} M$ , pH from 8 to 10. In general, as the concentration decreases, the pH increases. In other words, the pH, the less acidic (more basic) the solution. 4. A solution of pH 6 has a(n)  $[H^+]$  ...

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Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of

measure-ments of the same quantity made in the same way agree with one another. 2. Answers may vary. Look at the defense of the answer. 3. a. 7 b ...

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