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1 Which organic compounds provide most of the energy required ... C4 plants are more efficient in taking up carbon dioxide and have a higher rate of ...

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644 CHAPTER 21 Organic Compounds Section Assessment 1.Explain why carbon can form so many organic compounds? 2.Compare and contrast ethane, ethene, and ethyne. 3.How is an unsaturated hydrocarbon different from a saturated hydrocarbon? 4.How did organic compounds get this name? 5.Think Critically Cyclopropane is a satu-rated hydrocarbon ...

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Best Answer: an organic compound is any member of a large class of chemical compounds whose molecules contain carbon. For historical reasons discussed below, a few types of compounds such as carbonates, simple oxides of carbon and cyanides, as well as the allotropes of carbon, are considered inorganic.

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