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The answer you get will be equal to the number of bonding electrons in the ... Chemistry Standard Set 2. Key Vocabulary ... CHEMISTRY: A Study of Matter© 2004, GPB, 5.1.

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the electron. \_\_\_\_\_ represents the sublevel of the electron, which depends on the type of Pauli's Exclusion Principle states that within an atom, no two electrons can have

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Practice: Write electron distributions and do the orbital notation for the following. P 3s 3? CNe]3s Bp Lts? 20 Ortlly do the electron distributions for the following 2? les 2 5sl CHEMISTRY: n study Matter An E leve) con ont in s orbital, making up the sublevel". shaped An E level con contain p up the "p

sublevel". Shaped like double dumbbells

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