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Holt Chemistry Chapter 10 Causes of Change. When a compound forms it usually is more stable than the individual elements, which means it is at a lower energy state. The difference in energy would be negative and so the standard enthalpy (total energy) would be lower.

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Chapter 10: Causes of Change. SI unit of temperature; indicates number of units above absolute zero. The measurement of heat-related constants, such as specific heat or latent heat. The law that states the amount of heat released or absorbed in a chemical reaction does not depend on the

number of steps in the reaction.

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Chapter 10-Holt causes of change 1. Energy as Heat • A sample can transfer energy to another sample. • One of the simplest ways energy is transferred is as heat. ... • Both the total and average kinetic energies of a substance's particles are important to chemistry, because these quantities account for every particle's kinetic energy.

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