

## ***Holt Chemistry Stoichiometry Additional Problems Answers***

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**Holt Chemistry Stoichiometry Additional Problems**

Holt ChemFile: Problem-Solving Workbook 99 Stoichiometry Name Class Date Problem Solving continued Sample Problem 1 Ammonia is made industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is  $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$ . If 4.0 mol of  $\text{H}_2$  react, how many moles of  $\text{NH}_3$  will ...

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Holt ChemFile: Problem-Solving Workbook 97 Stoichiometry Stoichiometry So far in your chemistry course, you have learned that chemists count quantities of elements and compounds in terms of moles and that they relate moles of a substance to mass by using the molar mass. In addition, you have learned to

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Holt ChemFile: Problem-Solving Workbook 51 Mole Concept Name Class Date Problem Solving continued Sample Problem 2 A student needs 0.366 mol of zinc for a reaction. What mass of zinc in grams should the student obtain? Solution ANALYZE What is given in the problem? amount of zinc needed in moles What are you asked to find? mass of zinc in grams ...

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Honors Chemistry Extra Stoichiometry Problems 1. Silver nitrate reacts with barium chloride to form silver chloride and barium nitrate. a. Write and balance the chemical equation.  $2\text{AgNO}_3 + \text{BaCl}_2 \rightarrow 2\text{AgCl} + \text{Ba}(\text{NO}_3)_2$  b. If 39.02 grams of barium chloride are reacted in an excess of silver nitrate, how many

**Honors Chemistry Extra Stoichiometry Problems**

Set up a stoichiometry calculation to find the amount of product that can be ... Holt ChemFile: Problem-Solving Workbook 130 Percentage Yield Name Class Date Problem Solving continued Additional Problems 1. Ethyl acetate is a sweet-smelling solvent used in varnishes and fingernail-polish remover. It is produced industrially by heating acetic ...

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Determine the amount (in moles) of a product from a given amount of one reactant.

### **Ideal stoichiometry (practice) | Khan Academy**

Stoichiometry: Mixed Problems (KEY) 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  What volume of  $\text{NH}_3$  at STP is produced if 25.0 of  $\text{N}_2$  is reacted with an excess of  $\text{H}_2$ ? 3 3 3 2 3 2 2 40.0L  $\text{NH}_3$  1mol  $\text{NH}_3$  22.4L  $\text{NH}_3$  1mol N 2mol  $\text{NH}_3$  28.0g N 25.0g N 1mol N  $\times \times \times =$  2)  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  If 5.0g of  $\text{KClO}_3$  is decomposed, what volume of  $\text{O}_2$  is produced at STP? 2

### **Stoichiometry: Mixed Problems (KEY)**

Additional Problems 1. Ethyl acetate is a sweet-smelling solvent used in varnishes and fingernail-polish remover. It is produced industrially by heating acetic acid and ethanol together in the presence of sulfuric acid, which is added to speed up the ... Holt Chemistry 44 Stoichiometry .

### **Additional Problems 1. - Wiggins Elementary School**

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