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Ch 6 Study Guide answers

Section 6.3 Periodic Trends 171 Group Trends in Atomic Size In the Figure 6.14 graph, atomic radius is plotted versus atomic number. Look at the data for the alkali metals and noble gases. The atomic radius within these groups increases as the atomic number increases. This increase is an example of a trend.

6.3 Periodic Trends - Henry County Schools / Overview

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Section 3.4 Assessment page 90 24. Compare and contrast elements and compounds. Elements cannot be broken down into simpler substances by ordinary chemical means, whereas compounds can. 25. Describe the basic organizational feature of the periodic table of elements. The periodic table is organized by rows (called

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Periodic Assessment 2 Chemistry Answers Periodic Assessment 2 Chemistry Answer "The chemistry of an atom depends only on the number of electrons, which equals the number of protons and is called the atomic number. Chemistry is simply numbers, an idea Pythagoras would have liked.

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

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