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the electron. \_\_\_\_\_ represents the sublevel of the electron, which depends on the type of Pauli's Exclusion Principle states that within an atom, no two electrons can have

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Practice: Write electron distributions and do the orbital notation for the following. P 3s 3? CNe]3s Bp Lts? 20 Ortlý do the electron distributions for the following 2? les 2 5sl CHEMISTRY: n study Matter

An E level can contain s orbital, making up the sublevel". shaped An E level can contain p up the "p sublevel". Shaped like double dumbbells

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