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Holt ChemFile: Problem-Solving Workbook 187 Stoichiometry of Gases Name Class Date Problem Solving continued Sample Problem 2 Ethylene gas burns in air according to the following equation.

C 2H 4(g) 3O 2(g) 3 2CO 2(g) 2H 2O(l) If 13.8 L of C 2H 4 measured at 21°C and 1.038 atm burns completely with oxygen, calculate the volume of CO 2 produced ...

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