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The molecular weight of O<sub>2</sub> is 32.00 u, so a mole of O<sub>2</sub> would have a mass of 32.00 g and would contain 6.022 x 10<sup>23</sup> O<sub>2</sub> molecules. Each O<sub>2</sub> molecule is composed of two oxygen atoms, so one mole of O<sub>2</sub> contains two moles of oxygen atoms. 3. The mole is sometimes described as “the chemist’s dozen”.

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