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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain about 1/5 of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

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This reference is a must for students who need extra help, reteaching, or extra practice. The guide moves students through the same concepts as the text, but at a slower pace. More descriptive detail, along with visual algorithms, provides a more structured approach. Each chapter closes with a large bank of practice problems. Book jacket.

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5. Determine the percent yield if the theoretical yield is 20 g and the actual yield is 15 g. If the calculated value of B is larger than the amount of A, then B is the limiting reactant. You compare the calculated amount of B to the actual amount available. If more is required, then B is the limiting

reactant.

Holt Chemistry Chapter 9: Stoichiometry - Study.com

Is the answer reasonable? Yes; the volume of N₂ should be 2/3 the volume of O₂. Practice 1. In one method of manufacturing nitric acid, ammonia is oxidized to nitrogen monoxide and water. 4NH₃(g) + 5O₂(g) → 4NO(g) + 6H₂O(g) ... Holt ChemFile: Problem-Solving Workbook 190 Stoichiometry of Gases Name Class Date

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