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On March 16, 2014, in Chapter 05: Work and Energy, by Mastering Physics Solutions Part A = 3062 J
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Mastering Physics Solutions

Pearson Physics Level 30 Unit VIII Atomic Physics: Chapter 16 Solutions Student Book page 791
Example 16.1 Practice Problems 1. Given $Z = 12$ $A = 24$ Required neutron number (N) Analysis and Solution Since $A = Z + N$, $24 = 12 + N$ $N = 12$ Paraphrase There are 12 neutrons in a nucleus of $^{24}_{12}\text{Mg}$. 2. Given $Z = 92$ $N = 146$ Required atomic mass number ...

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