

## *Pearson Education Atoms And Bonding Answers*

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### **PEARSON EDUCATION ATOMS AND BONDING ANSWERS**

Metallic Bonds and Metallic Properties 1. Is the following sentence true or false? Metals are made up of cations and valence electrons, not neutral atoms. 2. What are metallic bonds? 3. Name three properties of metals that can be explained by metallic bonding. a. b. c. 4. What happens to an ionic crystal when a force is applied to it? 5.

### **BONDING AND INTERACTIONS**

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a. metal atoms held together by covalent bonds b. metal atoms held together by ionic bonds c. positive metal ions surrounded by freely moving valence electrons d. neutral metal atoms surrounded by freely moving valence electrons Completion Fill in the line to complete each statement. 13.

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Atoms and Bonding Chapter Test Atoms and Bonding Multiple Choice Write the letter of the correct answer on the line at the left. \_\_\_\_ 1. A(n) \_\_\_\_ is an atom or group of atoms that has an electric charge. a. compound b. ion c. crystal d. neutron \_\_\_\_ 2. A covalent bond is called \_\_\_\_ if the atoms equally

### **Atoms and Bonding Chapter Test Atoms and Bonding**

Atoms and Bonding Chapter Test A Atoms and Bonding \_\_\_\_ 9. \_\_\_\_ ions are composed of more than one atom. a. Covalent b. Molecular c. Polar d. Polyatomic \_\_\_\_ 10. In a double bond, \_\_\_\_ electrons are shared. a. two b. four c. six d. eight Completion Fill in the line to complete each statement. 11.

### **Atoms and Bonding Chapter Test A Atoms and Bonding**

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Atoms and Bonding Review and Reinforce Covalent Bonds Understanding Main Ideas Answer the following questions in the spaces provided. 1. Circle all the covalent bonds in the electron dot diagrams. 2. Which bond(s) shown are double bonds? \_\_\_\_ 3. Which bond(s) shown are triple bonds? \_\_\_\_ 4. What makes the bond in HF a polar bond?

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This section describes how covalently bonded atoms are held together. It explains how the properties of molecular compounds differ from those of ionic compounds.

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Atoms and Bonding. Atoms and Bonding Guided Reading and Study. Ionic Bonds. This section explains how an atom becomes electrically charged. It also describes the characteristic properties of bonds formed by the attraction of electrically charged atoms. Use Target Reading Skills. Before you read, preview Figure 17. Then write two questions that ...

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Name Date Class Atoms and Bonding Guided Reading and Study Ionic Bonds (continued) 4. Ions that are made of more than one atom are called 5. Use the table in the textbook to complete the table below.

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solutions for today and a forward-thinking plan for tomorrow connect teacher education and development, curriculum content and instruction, assessment, and information and school design and ... Atoms, Bonding, and the Periodic Table 130-137, Ionic Bonds 132, Figure 3 - Formation of an ... Pearson Interactive Science, 12 Module Series ...

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