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How do strong and weak acids differ? Use lab tools on your computer to find out! Dip the paper or the probe into solution to measure the pH, or put in the electrodes to measure the conductivity. Then see how concentration and strength affect pH. Can a weak acid solution have the same pH as a strong acid solution?

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pHET Simulations: Acid-Base Solutions and pH Scale Objectives: 1. Describe molecular representations for acid or base solutions 2. Provide relative amounts of particles in acid or base solutions to estimate strength and/or concentration 3. Estimate strength and/or concentration 4. Determine if a solution is acidic or basic Assignment:

pHET Simulations: Acid-Base Solutions and pH Scale

Procedure: PhET Simulations Play With Sims Chemistry Acid-Base Solutions The concentration of the acids and bases used in the at 0.010 (1x 10-2) Molar. Begin with a strong acid and lower the pH probe into the beaker. What is the pH of this solution? Test this strong acid with both pH paper and

the conductivity probe.

Strong and Weak Acids and Bases PhET Lab

How do strong and weak acids differ? Use lab tools on your computer to find out! Dip the paper or the probe into solution to measure the pH, or put in the electrodes to measure the conductivity. Then see how concentration and strength affect pH. Can a weak acid solution have the same pH as a strong acid solution?

Acid-Base Solutions - pH, Concentration, Conductivity - PhET

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Chemistry II Name _____ Unit 11 PhET Lab – Strength vs. Concentration of Acids and Bases Problem: What is the difference between a strong or weak acid and base? How does the concentration of an acid or base change its properties?

Chemistry II Name Unit 11 PhET Lab - Bloomer High School

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