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What percentage of water was in the hydrate? %H 2O = (5.0 g - 4.3 g) / 5.0 g \* 100 = (0.7 g) / 5.0 g \* 100 = 14% 3. A 2.5 g sample of a hydrate of Ca(NO 3) 2 was heated, and only 1.7 g of the anhydrous salt remained. What percentage of water was in the hydrate? ... Microsoft Word - KEY - Composition of Hydrates.doc Author:

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Percentage Composition Worksheet ... Percentage Composition Worksheet – Answer Key Give the % composition of all elements in these compounds. Show all work! \_\_\_\_ 1) ammonium sulfite % N 24.1~% H 6.9~% S 27.6~% O 41.3~...

# **Percentage Composition Worksheet - Auburn School District**

Percent Composition Worksheet II Fir	nd the percent compositions of all of the elem	ents in the
following compounds: 1) CuBr 2 Cu:	Br:	

#### **Percent Composition Worksheet II**

Chemistry IF8766 Page 27 Instructional Fair, Inc. Title: Microsoft Word - pg 27 - Atomic Structure.doc Author: Nathan Created Date: 3/28/2007 10:37:14 PM ... Percent Composition

#### **Instructional Fair Inc Chemistry If8766 27**

PERCENTAGE COMPOSITION Name ine the percentage composition of each of the compounds below. Det 1. Mnoa 2, 4. (NH4)3P04 (7 16e 77563 Z Al Solve the following problems. 6. How many grams of oxygen can be produced from the decomposition of 100. g '100 JS/7( 1.7 elf Y/ 7. How much iron can be recovered f om 25.0 g of Fe O?

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