

Periodic Table Basics Cards Answers

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Periodic Table Basics Step 1: Complete the card for each element. Complete the top section for each element by adding the element's atomic number, name, and atomic mass. Determine the number of protons, neutrons, and electrons in each element. Darken the correct circle to show if the element is a solid

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Periodic Table Basics – Answer Key 1. How many elements in your table were: (a) solids: 10 (b) liquids 0 (c) gases 8 2. Which elements had complete outer shells? Give the name and symbol for

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*There are up to 7 of these according to the modern model of the atom; Electrons found in the one closest to the nucleus have the lowest energy while those in the outermost one have the highest energy; The row # on the periodic table is the number of these an atom has.

Periodic Table basics Flashcards | Quizlet

Step 6: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper. Step 7: Answer the questions on the back of this worksheet using the information on your Periodic Table. T. Trimpe 2002

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