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### **10 States of Matter - Ms. Agostine's Chemistry Page**

Holt Chemistry 12 Chemical Equations and Reactions ... Balancing Chemical Equations In the space provided, write the letter of the term or phrase that best answers the question. Questions 1-6 refer to the following unbalanced formula equation.  $\text{KBr} + \text{Cl}_2 \rightarrow \text{KCl} + \text{Br}_2$  ... Quiz—Section: Balancing Chemical Equations 1. c6. 2. d7. 3. a 8. b 4. b 9 ...

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**mc06se cFMsr i-vi**

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Modern Chemistry 145 Chemical Equilibrium CHAPTER 18 REVIEW Chemical Equilibrium SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ Raising the temperature of any equilibrium system always (a) favors the forward reaction. (b) favors the reverse reaction. (c) favors the exothermic reaction.

**CHAPTER 18 REVIEW Chemical Equilibrium**

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equilibrium. Chemical Equilibria Are Dynamic If you drop a small ball into a bowl, the ball will bounce around and then come to rest in the center of the bowl. The ball has reached static equilibrium. Static equilibrium is a state in which nothing changes. Chemical equilibrium is different from static equilibrium because it is dynamic.

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