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What causes a gas to exert pressure? a. collisions b. density c. temperature d. elevation ____ 2. The SI unit for pressure is a. newton. b. mm Hg. c. pascal. d. liter. ____ 3. The pressure exerted by a gas does not depend on a. temperature. b. volume. c. number of moles present. d. the identity of the gas. ____ 4. At sea level, the average ...

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Concept Review j j. Gas particles may move apart to fill any container they occupy. This behavior occurs because gas particles are [close together, far apart] and are [attracted, not strongly attracted] to one another. 6. [Physical, Chemical] changes are changes in which the identity of a substance does not change.

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The common unit of pressure, symbolized mm Hg. Pressure (P) Defined as the force per unit area on a surface. Newton (N) The force that will increase the speed of a one-kilogram mass... The amount of force exerted per unit area of a surface A unit of pressure equal to the force Earth's atmosphere exert... An instrument...

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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

•Most metals react with the Group 17 elements, the halogens, to form either ionic or covalent compounds. Section 2 Types of Chemical. Chapter 8 Reactions. •Group 1 metals react with halogens to form ionic compounds with the formula MX, where M is the metal and X is the halogen.

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