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3. Indicate the oxidation number of phosphorus in each of the following compounds: a. HPO_3 ____ d. H_3PO_4 ____ b. H_3PO_2 ____ e. $\text{H}_4\text{P}_2\text{O}_7$ ____ c. H_3PO_3 ____ f. $\text{H}_5\text{P}_3\text{O}_{10}$ ____ 4. Give oxidation numbers for the underlined atoms in these molecules and ions: a. Cs_2O b. PtCl_6^{2-} c. CaI_2 d. SnF_2 e. Al_2O_3 f. ClF_3 g. H_3AsO_3 h. SbF_6^- ...

Worksheet: Oxidation Numbers Name - MARRIC

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Worksheet Oxidation Numbers 14 4 Answer Key Quantum numbers worksheet key 1. write the 14 sets of quantum numbers that describe the 14 electrons of silicon, si. $1, 0, 0, +\frac{1}{2}$ and $-\frac{1}{2}$ $2, 0, 0, +\frac{1}{2}$ and $-\frac{1}{2}$ pretense - splurge. two step math problems for 2nd grade worksheets grade 4 terra nova math pratice

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CHEMISTRY: A Study of Matter © 2004, GPB 8.2 Problem Set One How many moles of water will be produced when _____ grams of hydrogen gas react with the oxygen in the air?

8-01,02,03 Note Taking Guide Ep 801 - Welcome to Mole Cafe

7. In compounds, the elements of groups 1 and 2 as well as aluminum have oxidation numbers of +1, +2, and +3 respectively. 8. The sum of the oxidation numbers of all atoms in a neutral compound is 0. 9. The sum of the oxidation numbers of all atoms in a polyatomic ion equals the charge of the ion. Answer Key 1. Cl:0 7. Al:3+ 13. N:3- H:1+ 19.

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