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What are the answers to Holt chemistry's stoichiometry ...

Holt ChemFile: Problem-Solving Workbook 103 Stoichiometry Name Class Date Problem Solving continued Is the number of significant figures correct? Yes; two significant figures is correct. Is the answer reasonable? Yes; 41 g is about $\frac{1}{3}$ of the molar mass of KClO_3 , and $0.5 \frac{2}{3} \frac{1}{3}$. Practice 1.

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Holt ChemFile: Problem-Solving Workbook 184 Stoichiometry of Gases. Name Class Date. Problem Solving continued. Mass of substance A — solid, liquid, or gas Convert using the mole ratio between A and B. The volume ratio is the same as the mole ratio if the two gases are under the same set of conditions.

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Holt ChemFile: Problem-Solving Workbook 55 Mole Concept. Name Class Date. Problem Solving continued. CONVERTING THE AMOUNT OF AN ELEMENT IN MOLES TO THE NUMBER OF ATOMS In Sample Problem 3, you were asked to determine the number of moles in 1.204 10²⁴ atoms of lithium.

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Holt Chemistry 6 Stoichiometry Section: Limiting Reactants and Percentage Yield Complete each statement below by choosing a term from the following list. Terms may be used more than once. excess product limiting stoichiometric percentage actual theoretical 1.

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Holt ChemFile: Problem-Solving Workbook 130 Percentage Yield Name Class Date Problem Solving continued Additional Problems 1. Ethyl acetate is a sweet-smelling solvent used in varnishes and fingernail-polish remover. It is produced industrially by heating acetic acid and ethanol together in the presence of sulfuric acid, which is added to speed ...

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2. d Which of the following would not be studied within the topic of stoichiometry? (a) the mole ratio of Al to Cl in the compound aluminum chloride (b) the mass of carbon produced when a known mass of sucrose decomposes

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