

***Holt Chemfile Problem Solving Workbook Limiting Reactants  
Answers***

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Holt ChemFile: Problem-Solving Workbook 58 Mole Concept Name Class Date Problem Solving continued CONVERTING NUMBER OF ATOMS OF AN ELEMENT TO MASS Sample Problem 4 uses the progression of steps 1 → 2 → 3 to convert from the mass of an element to the number of atoms.

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Holt ChemFile: Problem-Solving Workbook 191 Stoichiometry of Gases Name Class Date Problem Solving continued Rearrange the ideal-gas-law equation to solve for the unknown quantity, V. PV nRT V nR P T COMPUTE EVALUATE Are the units correct? Yes; units canceled to give liters of SO<sub>2</sub>. Is the number of significant figures correct?

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Problem Solving continued Sample Problem 1 Ammonia is made industrially by reacting nitrogen

and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The ... Holt ChemFile: Problem-Solving Workbook 102 Stoichiometry Name Class Date Problem Solving continued

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Holt ChemFile: Problem-Solving Workbook 318 Electrochemistry Name Class Date Problem Solving continued Sample Problem 2 A cell is constructed in which the following two half-reactions can occur in either direction.  $\text{Zn}^{2+} + 2\text{e}^{-} \rightleftharpoons \text{Zn}$   $\text{Br}_2 + 2\text{e}^{-} \rightleftharpoons 2\text{Br}^{-}$  Write the full ionic equation for the cell in the spontaneous direction,

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Holt ChemFile: Problem-Solving Workbook 280 Titrations Name Class Date Problem Solving continued Sample Problem 3 A supply of NaOH is known to contain the contaminants NaCl and MgCl<sub>2</sub>. A 4.955 g sample of this material is dissolved and diluted to 500.00 mL with water. A 20.00 mL sample of this solution is titrated with 22.26 mL

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Holt ChemFile: Problem-Solving Workbook 5 Conversions Name Class Date Problem Solving continued Sample Problem 2 A metallurgist is going to make an experimental alloy that requires adding 325 g of bismuth to 2.500 kg of molten lead. What is the total mass of the mixture in kilograms? Solution ANALYZE

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**Ideal Gas Law Answers Holt Chemfile - pdfsdocuments2.com**

Holt ChemFile: Problem-Solving Workbook 127 Percentage Yield Name Class Date Problem Solving continued Sample Problem 2 Acetylene, C<sub>2</sub>H<sub>2</sub>, can be used as an industrial starting material for the production of many organic compounds. Sometimes, it is first brominated to form 1,1,2,2-tetrabromoethane, C<sub>2</sub>H<sub>2</sub>Br<sub>4</sub>, which can then be

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Holt ChemFile: Problem-Solving Workbook 21 Significant Figures Name Class Date Problem Solving continued Sample Problem 2 In an experiment to identify an unknown gas, it is found that 1.82 L of the gas has a mass of 5.430 g. What is the density of the gas in g/L? Solution ANALYZE What is given in the problem? the measured mass and volume of the gas

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