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### Gpb Chemistry 3 Answer Key

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### Chemistry & Physics | Georgia Public Broadcasting

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### Chemistry Matters | Georgia Public Broadcasting

Review Sheet: Unit 3 Name \_\_\_\_\_ CHEMISTRY: A Study of Matter © 2004, GPB 3.23a KEY I. Short Answer and Fill in the Blanks: 1. \_\_\_\_\_ are a familiar idea used to ...

### I. Short Answer and Fill in the Blanks

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The answer you get will be equal to the number of bonding electrons in the ... Chemistry Standard Set 2. Key Vocabulary ... CHEMISTRY: A Study of Matter© 2004, GPB, 5.1.

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$3 \times 6 \text{ mol F} + 2 \times 4 \text{ mol AlF}_3 = 30 \text{ mol F}$  2 b. \_\_\_\_\_ moles of  $\text{AlF}_3$  will react with 0.6 moles of  $\text{O}_2$ . ? mol  $\text{AlF}_3 = 0.6 \text{ mol O}_2 \times 4 \text{ mol AlF}_3 / 3 \text{ mol O}_2 = 0.8 \text{ mol AlF}_3$  4. \_\_\_\_\_  $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  a. How many moles of oxygen react with 11 moles of  $\text{C}_3\text{H}_8$ ? ? mol  $\text{O}_2 = 11 \text{ mol C}_3\text{H}_8 \times 5 \text{ mol O}_2 / 1 \text{ mol C}_3\text{H}_8 = 55 \text{ mol O}_2$  b. How many moles ...

### CHEMISTRY: A Study of Matter - Weebly

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### (3)CHEM-note taking/ episode 301 Flashcards | Quizlet

Given a mass of 24 grams and a volume of 3 milliliters, calculate the density.  $D = m/V = 24 \text{ g} / 3 \text{ mL} = 8 \text{ g/mL}$  D What is the mass of 32 milliliters of water?  $m = D \times V = 1.0 \text{ g/mL} \times 32 \text{ mL} = 32 \text{ g}$  Given that the density of iron is 7.9 grams per centimeters cubed, what would be the volume of a 3.5 gram piece of iron?  $V = m/D = 3.5 \text{ g} / 7.9 \text{ g/cm}^3 = 0.44 \text{ cm}^3$

### **CHEMISTRY: A Study of Matter - Thomas County Schools**

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Taking...

### **Honors Chem - Unit 6B - Stoichiometry - Spring 2015 - Mrs ...**

= 3.7!1024 atoms Al 2. Convert 5.3 x 1025 molecules of CO 2 to moles. ? moles CO 2 = 5.3!1025  
molecules CO 2! 1 mole CO 2 6.02!103moleculs CO 2 = 88 mol CO 2 3. How many formula units of  
sodium acetate are in 0.87 moles of sodium acetate? ? f.u. NaC 2 H 3 O 2 = 0.87 mol NaC 2 H 3 O  
2! 6.02!1023 f.u. NaC 2 H 3 O 2 1 mol NaC 2 H 3 O 2 = 5.2!103 f ...

### **2 3 2 3 2 1 2H3O2 3 2 ? f.u. NaCl = 3.55 mol NaCl 6.02 H ...**

3 = 256.3 g Na 2 CO 3! 1 mol Na 2 CO 3 106.0 g Na 2 CO 3! 6.02!1023 f.u. Na 2 CO 3 1 mol Na 2  
CO 3 = 1.46!1024 f.u. Na 2 CO 3 3. What is the mass of 12.4 molecules of carbon tetrachloride? ? g  
CCl 4 = 12.4 molecules CCl 4! 1 mol CCl 4 6.02!1023moleculs CCl! 154.0 g CCl 4 1 mol CCl 4 =  
3.17!10-21 g CCl 4 4. How many moles are contained in 0.43 g ...

### **Worksheet: More Mole Problems Name**

3 is a compound used to manufacture pesticides. A reaction requires that 96.7 g of PCI 3 be raised  
from 31.7 oC to 69.2oC. How much energy will this require given that the specific heat of PCI 3 is  
0.874 J/g oC? 10. A quantity of water is heated from 25.0 oC to 36.4 oC by absorbing 325 J of heat  
energy. What is the mass of the water? 325 J=(m ...

### **13-06a,b,c Heat and Heat Calculations wkst-Key**

3. Choose the coefficients which, in order, would correctly balance the following equation: HgO Hg  
+ O 2 a. 1, 1, 1 b. 2, 1, 1 c. 2, 2, 1 d. 2, 2, 2 4. When balancing a chemical equation, one is allowed  
to adjust the \_\_\_\_\_. a. formulas of the reactants. b. formulas of the products. c. coefficients d. both A  
and B

### **Chemistry Quiz: Episode 604 Name K EY - PC\|MAC**

CHEMISTRY: A study of Matter 0 2004, GPB 13.2 When heat (Q) is absorbed by a system, part of it  
(C) goes into storage as energy and art of it is used to make the molecules move around , raising  
the (At). Why does sand get hotter in the day and colder at night than the water? CHEMISTRY: A  
study of Matter e 2004, GPB 13.1

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