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Holt Chemistry 6 Covalent Compounds Name Class Date Concept Review continued 10. Draw the Lewis structure for water, H 20. 11. Draw the Lewis structure for carbon monoxide, CO. 12. Draw the Lewis structure for the rocket propellant nitryl fluoride, NO 2F. 13. Draw the Lewis structures for SO 2 3 2. Answer the following item in the space ...

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Holt Chemistry 6 Stoichiometry Section: Limiting Reactants and Percentage Yield ... Answer the following items in the space provided. 11. ... Concept Review: Stoichiometry and Cars 1. c 2. b 3. d 4. b 5. d 6. If there is too much oxygen and not enough gasoline, the engine will stall. If, on the other hand, gasoline is in

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Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of measure-ments of the same quantity made in the same way agree with one another. 2. Answers may vary. Look at the defense of the answer. 3. a. 7 b ...

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Concept Review: Trends in the Periodic Table 1. ionization energy 2. bond radius 3. electron affinity 4. electronegativity 5. increases, decreasing 6. d 7. a 8. c 9. The electron cloud model is based on the probability of finding an electron at a specific location. As you move far-ther out from the nucleus, the proba-bility of finding electrons ...

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Holt Chemistry 1 Atoms and Moles Section: Substances Are Made of Atoms In the blank at the left of each word or phrase, write the letter of the expression ... Concept Review continued Answer the following items in the space provided. 12. In Thomson's cathode-ray experiment, what evidence led him to believe that

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Answer the following questions in the space provided. 1. Why do Li, Na, K, Rb, Cs, and Fr all react with Cl in a 1:1 ratio forming sub- ... Holt Chemistry 25 The Periodic Table Section: Tour of the Periodic Table ... Concept Review continued 14. Why are the main-group elements called the representative elements? 15. Why are Group 2 elements ...

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