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A chemical change that may take place in either direction depending upon the conditions. Dynamic equilibrium: (1 mark) A state in which the rate of dissociation of the  $\text{N}_2\text{O}_4(\text{g})$  into  $2\text{NO}_2(\text{g})$  exactly balances the rate at which  $2\text{NO}_2(\text{g})$  combines to form  $\text{N}_2\text{O}_4(\text{g})$ . (ii) Write an expression, and state the units, for  $K_p$  and  $K_c$ .  $K_p$ : Expression:  $P_2 \text{ NO}_2$  ...

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PEARSON CHEMISTRY: The law of conservation of mass is explained in Lesson 2.4 (p. 50). Students are introduced to balancing chemical equations in Lesson 11.1 (pp. 346- 354). Students learn to write and balance the five different types of chemical reactions in Lesson 11.2 (pp. 356-367).

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11.2 Types of Chemical Reactions> 13 A decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products. • Decomposition reactions involve only one reactant and two or more products. • The products can be any combination of elements and compounds. • Most decomposition reactions require ...

### **11.2 Types of Chemical Reactions> - Useful Advice**

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© 2015 Pearson Education, Inc. Chapter 3 Chemical Reactions and Reaction Stoichiometry Prepared by John N. Beauregard Based on a presentation by James F. Kirby

### **Chapter 3 Chemical Reactions and Reaction Stoichiometry**

school chemistry, and Small-Scale Chemistry Laboratory, also published by Pearson. In addition, he has published numerous professional papers in peer-reviewed journals including the Journal of the American Chemical Society, the Journal of Organic Chemistry, and the Journal of Chemical Education.

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Thomas J. Greenbowe is a Professor of Chemistry at Iowa State University. He served as Chair of the Division of Chemical Education of the American Chemical Society, Chair of the 18th Biennial Conference on Chemical Education and Chair of the 2009 Gordon Research Conference on Chemical Education Research and Practice.

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