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a force that acts on the surface of a liquid and that tends to minimize the area of the surface. evaporation. the process by which water changes from liquid form to an atmospheric gas. boiling point. the temperature at which a liquid boils at sea level. condensation. the process of changing from a gaseous to a liquid or solid state.

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Intermolecular forces are electrostatic in nature and include van der Waals forces and hydrogen bonds. Molecules in liquids are held to other molecules by intermolecular interactions, which are weaker than the intramolecular interactions that hold the atoms together within molecules and polyatomic ions.

12.6: Intermolecular Forces: Dispersion, Dipole-Dipole ...

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