

Palm Matlab Solutions Manual

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$P_{ideal} = 1.0004$, $P_1 = nRT/(V - nb)$ $P_1 = 1.0029$, $P_2 = (a \cdot n^2)/V^2$ $P_2 = 0.0129$, $P_{waals} = P_1 + P_2$ $P_{waals} = 1.0158$ The ideal gas law predicts a pressure of 1.0004 atmospheres, while the van der Waals model predicts 1.0158 atmospheres.

Instructor's Manual c Introduction to Matlab 6 for ...

10.2-1 Simulink solution of 10.2-2 Exporting to the MATLAB workspace 10.2-3 Simulink model for ...
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end disp(x) T4.5-1 The script le is angle = input('Enter an angle in degrees.').switch angle case 45 disp('Angle is in first quadrant')case -45 disp('Angle is in second quadrant')case 135 disp('Angle is in third quadrant')case -135 disp('Angle is in fourth quadrant')otherwise disp('Quadrant is unknown.').end T4.7-1 The 0.75 in the matrix should be replaced with 0.70.

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