

## *Periodicity Of Elements Pogil Answer Key*

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4. In general, what is the trend in atomic radius as you go across a period (left to right) in Model 1? Support your answer, using examples from two periods. 5. Using your knowledge of Coulombic attraction and the structure of the atom, explain the trend in atomic radius that you identified in Question 4.

### **15 Periodic Trends-S - Weebly**

Periodicity of Elements POGIL and Answer Key Assigned as CW on 12/19/18 Review Book Question Answers- Q #1-21 on pages 83 and 84- Assigned as HW on 1/8/18 1) 1.

### **Piersa, Amanda / Unit 6: Periodic Table**

series, the second and third transition elements in each period have nearly the same size. For example, both Ru and Os have a radius of 130 pm. Key Questions 1. Using only a periodic table, predict the order of increasing atomic radius for each of the following sets of elements: Ca, Mg, Be Ga, Br, Ge Al, Tl, Si Zn, Cd, Fe 2.

### **Chem 115 POGIL Worksheet - Week 10 Periodic Trends Why ...**

Periodicity of Elements 5. Within a group, which period, top or bottom is likely to contain atoms with the larger radius? Explain your answer. 6. In terms of the number of valence electrons, describe one cyclic property that can be observed in the model. 7. Refer to a Periodic Table and write the number of protons in the nucleus for the period 2 elements listed: 8.

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