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80 Chemistry: Matter and Change • Chapter 14 Study Guide for Content Mastery Section 14.2 The Combined Gas Law and Avogadro's Principle In your textbook, read about the combined gas law. Fill in the following table. State what gas law is derived from the combined gas law

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Chemistry: Matter and Change Study Guide for Content Mastery Answer Key Name Date Class 74 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery In your textbook, read about gas pressure. Circle the letter of the choice that best completes the statement or answers the question. 13. Pressure is defined as force per unit

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### **Study Guide for Content Mastery SE**

Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 8 43 Ionic Compounds Ionic Compounds Section 8.1 Forming Chemical Bonds In your textbook, read about chemical bonds and formation of ions. Use each of the terms below just once to complete the passage. The force that holds two atoms together is called a(n) (1).

### **Study Guide for Content Mastery**

noble gas formation. A CF ion is an example of a cation. The ending -ide is used to designate an anion. Nonmetals form a stable outer electron configuration by losing electrons and becoming anions. 43 Chemistry: Matter and Change Chapter 8 Study Guide for Content Mastery

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### **Chemistry: Matter And Change, Study Guide For Content ...**

Physical Science Packet Chapter 16: Kinetic Theory of Matter Name: \_\_\_\_\_ Due: Date of Chapter 16 Test . 2 Kinetic Theory of Matter Study Guide Major topics on the test will include: A. Kinetic Theory of Atoms a. What is an Atom? ... Directed Reading for Content Mastery: Overview - Solids, Liquids, and Gases . 13

### **Physical Science Packet Chapter 16: Kinetic Theory of Matter**

STUDY GUIDE FOR CONTENT MASTERY CHAPTER States of Matter Section 13.1 Gases In your textbook, read about the kinetic-molecular theory. Complete each statement. I. The kinetic molecular theory describes the behavior of gases in terms of particles in 2. The kinetic-molecular theory makes the following assumptions. a.

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