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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Modern Chemistry 133 Chapter Test Name Class Date Chapter Test B, continued PART IV Write the answers to the following questions in the space provided. 35. Explain the difference between strong acids and weak acids. 36. Explain how the production of sulfur trioxide, SO 3, in industrial processes can result in acid rain. Write an equation for ...

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Modern Chemistry 36 Chapter Test Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. ... Modern Chemistry 39 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_\_16. Which of the following elements have full outer energy levels when

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# Assessment Chapter Test A - mrbijlani.weebly.com

CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

# 6 Chemical Bonding - Effingham County Schools / Overview

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

## 10 States of Matter - Ms. Agostine's Chemistry Page

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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