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The molecular weight of O2 is 32 Pogil the mole answer key. 00 u, so a mole of O2 would have a mass of 32. 00 g and would contain $6.022 \times 1023 \text{ O}$. 2 molecules. Each O2 molecule is composed of two oxygen atoms, so one mole of O2 contains two moles of oxygen atoms Pogil the mole answer key. 3. The mole is sometimes described as "the chemist's dozen".

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Using complete sentences and working as a group, compare the new mole map in Model 4 to the original mole map on your cheat sheet. $_$ H2 (g) + $_$ O2 (g) ($_$ H2O (g) Given the equation above, determine the number of moles of water produced when 5.2 g O2 are reacted. Make sure to show the dimensional analysis in your work.

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The molecular weight of O2 is 32.00 u, so a mole of O2 would have a mass of 32.00 g and would contain 6.022×1023 O. 2 molecules. Each O2 molecule is composed of two oxygen atoms, so one mole of O2 contains two moles of oxygen atoms. 3. The mole is sometimes described as "the chemist's dozen".

Moles & Stoichiometry Answers Key Questions & Exercises

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Key Questions 1. The atomic weight of carbon is 12.0107 u, so a mole of carbon has a mass of 12.0107 g. Why doesn't a mole of carbon weigh 12 g? 2. The atomic weight of oxygen is 16.00 u. What is the mass of a mole of O2(g)? How many O2 molecules does a mole of O2(g) contain? How

many moles of oxygen atoms does a mole of O2(g) contain? 3.

Chem 115 POGIL Worksheet - Week 4 Moles & Stoichiometry

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This is called Stoichiometry- (stoi-key-ah-meh-tree) Another way of looking at it is using the mole ratio from the balanced equation and information about one compound in the reaction to determine information about another compound in the equation.

POGIL- Stoichiometry

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All of the answers can be found online for almost every single POGIL..... Currently, there is a significant amount of discussion on teaching list serves about the frustration of people posting answer keys online and students checking the internet instead of doing the work. Many teachers, myself included, experience this frustration.

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Flinn Scientific and The POGIL Project[™] have collaborated to publish this series of student-centered learning activities for high school chemistry. Create an interactive learning environment with 36 specially designed guided-inquiry learning activities in 13 major topic areas. 290 pages, 8½" x 11".

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