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The molecular weight of O₂ is 32.00 u, so a mole of O₂ would have a mass of 32.00 g and would contain 6.022 x 10²³ O₂ molecules. Each O₂ molecule is composed of two oxygen atoms, so one mole of O₂ contains two moles of oxygen atoms. 3. The mole is sometimes described as “the chemist’s dozen”.

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