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Equilibrium chemistry is concerned with systems in chemical equilibrium. The unifying principle is that the free energy of a system at equilibrium is the minimum possible, so that the slope of the free energy with respect to the reaction coordinate is zero. This principle, applied to mixtures at equilibrium provides a definition of an equilibrium constant.

Equilibrium chemistry - Wikipedia

What exactly do we mean by 'salt' in chemistry? Learn more about the definition, chemical, and physical properties and how we can easily figure out the chemical formula of many salts.

What is Salt in Chemistry? - Definition & Formula - Study.com

The timeline of chemistry lists important works, discoveries, ideas, inventions, and experiments that significantly changed humanity's understanding of the modern science known as chemistry, defined as the scientific study of the composition of matter and of its interactions. The history of chemistry in its modern form arguably began with the Irish scientist Robert Boyle, though its roots can ...

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