Structural Disorder in Two-Dimensional Network-Forming Materials



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List of Notes By David

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1 | Modelling Bilayer Materials

A computationally tractable Monte Carlo method using triangle rafts is developed to generate bilayers of SiO₂ and related materials. The method allows defect free networks of any given shape to be grown with both tuneable ring statistics and topologies, controlled by a combination of the "allowed" rings and the effective growth "temperature". Configurations are generated with Aboav-Weaire parameters commensurate with those obtained from an analysis of experimental configurations, improving significantly on previous methods. The ability to efficiently grow configurations allows exploration of the structural basis of Lemaître's law, where the commonly observed value of $p_6 \approx 0.4$ is presented as a balance between entropic and enthalpic factors. The deviations of ring areas from the ideal values are discussed and the relative insensitivity of the ring area to relatively strong distortions is highlighted.

1.1 Bilayer Materials

An important class of two-dimensional materials which have emerged in the 21st century are bilayers of silica, SiO₂, and related species [11]. These can be prepared experimentally by chemical vapour deposition on metal and graphene supports [4, 5]. As in the three-dimensional glass, the basic building blocks of silica bilayers are vertex sharing SiO₄ tetrahedra, maintaining full coordination for all atoms in the bulk [22]. These are arranged such that three of the vertices are connected to tetrahedra in the same layer, with the final vertex being shared between layers acting as a "bridge" (figure 1.1a). A consequence of these bridging oxygen atoms is to enforce a symmetry plane between the upper and lower layers.

Topologically, the symmetry plane means that these materials can be viewed as effective two-dimensional networks. Taking one of the layers, without the bridging

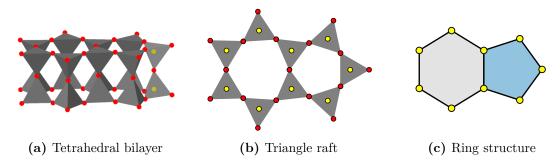


Figure 1.1: Silica bilayers of vertex sharing tetrahedra in (a) can be represented as a two-dimensional triangle raft in (b). Silicon and oxygen atoms are coloured yellow and red respectively. The ring structure then emerges from the three-coordinate network comprising the silicon atoms, (c).

oxygens, and projecting the atoms onto the horizontal plane reveals a representation of vertex sharing triangles, referred to as a triangle raft (figure 1.1b). The ring structure then emerges from the three-coordinate network formed by connecting the silicon atoms of adjacent triangles as in figure 1.1c. Indeed, scanning tunnelling microscopy (STM) has been used to directly visualise the ring structure in silica bilayers, revealing both crystalline and glassy arrangements and even the interface between the two [128, 129].

More recently experimentalists have also succeeded in synthesising bilayers of germania, GeO₂ [7, 8]. Add a bit more experimental context here, discuss with Mark

1.2 Review of Existing Methods

As mentioned in the introduction, both *ab initio* methods and classical molecular dynamics have been used in computational studies of silica bilayers, which often require a starting atomistic configuration [20–22, 28]. One approach is to simply take an experimental sample as the starting structure. Whilst this may be on the surface the best solution, the experimental configurations may contain defects or areas where the image is corrupted *i.e.* the configuration may not be "pristine". Additionally, the location of each atom has an associated uncertainty which leads to discrepancies in the observed bond lengths and angles, which can be compounded by any out-of-plane distortions. Whilst computational refinement can attenuate these problems [23, 130], there remains the more fundamental question of how

"typical" the available images are from experiment, as STM provides exceptional information but only on relatively small sample sizes. Computational techniques can therefore prove a valuable tool for generating a large number of high-quality configurations and corroborating experimental information.

One current approach is to transform amorphous graphene configurations [22]. Here amorphous samples of carbon are generated using a bond switching method (as outlined in section ??), before the carbon atoms are swapped from silicon and decorated with oxygens. Whilst this is a valid approach, the method assumes that the two materials are topologically equivalent. This is likely an oversimplification, as the presence of the bridging oxygens in silica afford the structure increased flexibility when compared to the carbon analogue. This likely explains why this method has struggled to mirror experimentally observed values of the ring statistics and Aboav-Weaire parameter, with small and large ring proportions being under-estimated [29].

An alternative approach is to use molecular dynamics coupled with an effective pair potential to obtain viable configurations [28]. Such methods are relatively common, having been employed previously to study amorphous graphene [104]. Such methods offer the potential for generating realistic configurations but are difficult to control as the cooling rates which must be applied are necessarily huge compared to experimental rates. A potential artefact of the high cooling rates is the effectively freezing in of defect states, either in terms of local coordination environments or highly-strained (three-membered) rings. In addition, as with the method above, such methods appears to systematically underestimate the Aboav-Weaire parameter, indicative of too little structural ordering.

1.3 Triangle Raft Method

The motivation of this work was to develop a construction algorithm to generate samples of silica bilayers which can capture the full two-dimensional network topology; both the ring distribution and correlations. The model should be able to explore all phases from crystalline to amorphous yet computationally efficient enough to produce configurations suitable for further high throughput calculations.

To achieve this a grow-from-seed Monte Carlo algorithm has been developed, where rings are individually added to build a triangle raft. This approach takes inspiration from the first hand-built models, which have been noted to bear close resemblance to experimental structures [131, 132]. Such models were superseded by computational techniques designed to generate periodic configurations. However, the recent development in techniques to simulate aperiodic samples, such as sliding boundary conditions for molecular dynamics [133], makes this constraint no longer essential, and benefit may be gained from the added freedom of an aperiodic model.

1.3.1 Potential Model

As explained in figure 1.1 it is possible to capture the full topology of silica bilayers with a simplified representation consisting of a network of vertex-sharing SiO₃ triangles. As the focus of this chapter is on generating a large number of samples with varying ring statistics, working with this reduced representation is sufficient, as it provides a computationally efficient way to produce networks with the required topology. The precise geometry of the bilayer can be refined with advanced optimisation techniques if required [134].

In order to simulate bilayer systems in two dimensions, a suitable potential model is needed which captures the essential physics of the system: the local triangular environment of the SiO₃ units and the relative energies of rings of different sizes. The model used here is modified from a relatively simple potential used in all-atom bilayer calculations [22, 23], a schematic for which is given in figure 1.2. Each SiO₃ unit has a harmonic potential acting between all three Si–O pairs, and the three nearest-neighbour O–O pairs, given by:

$$\mathcal{U}_{ij} = \frac{K}{2} \left(r_{ij} - r_{ij}^0 \right)^2, \tag{1.1}$$

where K is a constant, r_{ij} is the interatomic separation and r_{ij}^0 the equilibrium interatomic separation between i, j. The spring constant, K, is set to be very stiff, whilst the equilibrium separations are set according to elemental species such that $r_{\text{OO}}^0 = \sqrt{3} \, r_{\text{SiO}}^0$, maintaining a set of ideal SiO₃ triangles.

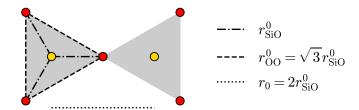


Figure 1.2: Schematic of the potential model in triangle rafts. Stiff harmonic springs (dashed and dashed-dotted lines) preserve the triangular subunits, whilst the shifted and cut 24-12 potential (dotted line) maintains an equilibrium angle of 120° between neighbouring subunits.

The Si–O–Si angle, which determines the strain associated with different ring sizes, is controlled by a shifted and cut 24-12 potential of the form:

$$\mathcal{U}_{ij} = \begin{cases} \epsilon \left[\left(\frac{r_0}{r_{ij}} \right)^{24} - 2 \left(\frac{r_0}{r_{ij}} \right)^{12} \right] + \epsilon & r_{ij} \le r_0 \\ 0 & \text{otherwise} \end{cases}$$
 (1.2)

where ϵ is a constant and r_{ij} is now the Si–Si separation between atoms in adjacent triangles. It is the value of r_0 which sets the Si–O–Si angle at which strain begins to be felt and therefore the relative ring energies. Taking the hexagonal lattice as being the zero in energy it follows that $r_0 = 2r_{\rm SiO}$. Rings which deviates increasingly from the ideal hexagon will therefore incur an increasingly energetic penalty.

To summarise, the primary aim here is to generate topologies suitable for later investigation using more detailed (and hence more accurate but more computationally-demanding) potential models. As a result, the harmonic springs simply control the local (triangular) geometries whilst the 24-12 potential controls the repulsion between these local polyhedra. These functions are chosen as deliberately simple to improve computational efficiency and achieve high throughput of idealised networks. Furthermore, the parameters K and ϵ need have no direct physical meaning, simply controlling the meaning of the system "temperature" as discussed below. The only requirement is that they generate energies of the same magnitude to allow for efficient structural evolution.

1.3.2 Algorithmic Details

Using the model detailed above, a Monte Carlo construction algorithm has been developed which allows two-dimensional networks to be built ring by ring in the

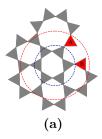
shape of a specified function. The main steps of the algorithm are outlined below:

- 1. Take a starting seed, such as a single ring or experimental configuration
- 2. Select triangles on which to build the next ring (see figure 1.3)
 - (a) Overlay a function on the network (e.g. circle, square)
 - (b) Check for atoms with dangling bonds lying inside the function region
 - (c) If no such atoms exist, systematically increase the function size until an atom is found
 - (d) Find the next nearest atoms which also have a dangling bonds
 - (e) Choose the two triangles that correspond to the largest starting ring size
- 3. Determine the probability of constructing rings of different sizes
 - (a) Build trial rings in the range k_{\min} to k_{\max} (see figure 1.4)
 - (b) Geometry optimise the local structure and calculate minimised potential energy (as explained in section ??)
 - (c) Calculate the probabilities of each ring occurring, P_k , equation (1.3)
- 4. Accept single trial ring according to the probability distribution
- 5. Repeat steps $2 \to 4$ until the target number of rings is reached

The probability of a ring of size k being accepted, P_k , is given by the equation:

$$P_{k} = \frac{\exp\left[-\left(\mathcal{U}_{k} - \mathcal{U}_{k}\right)/T\right]}{\sum_{k} \exp\left[-\left(\mathcal{U}_{k} - \mathcal{U}_{0}\right)/T\right]},$$
(1.3)

where \mathcal{U}_k and \mathcal{U}_0 correspond to the energy of the trial structure and lowest energy of all trial structures respectively, and T is a "temperature". The parameter T controls how easily the potential energy landscape can be explored, and therefore how accessible strained rings become. In the low T limit, the acceptance probabilities are dominated by the energy term, and the rings which are selected will be those with the lowest energy. Note that this is not necessarily the 6-ring, but rather is



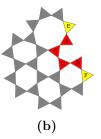


Figure 1.3: Panel (a) shows how triangles used to construct a ring are initially selected. There are no atoms with dangling bonds within the first search region (blue dashed line), and so the search area is extended (red dashed line), where triangles A and B are found. Panel (b) gives the three possibilities for the triangles that will form part of the constructed ring: A–C–D–B, A–E, B–F. As A–C–D–B corresponds to the largest starting ring size this is selected.

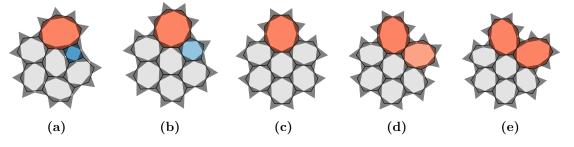


Figure 1.4: Geometry optimised structures for trial rings in the range k = 4 - 8. The ring structure is shown along with the SiO₃ triangle

Table 1.1: Variation of acceptance probabilities with temperature for the configurations in figure 1.4.

P_k	4	5	6	7	8
$T = 10^{-4}$	0.0000	1.0000	0.0000	0.0000	0.0000
$T = 10^{-3}$	0.0000	0.8837	0.1162	0.0001	0.0000
$T = 10^{-2}$	0.0336	0.4104	0.3351	0.1659	0.0550
$T = 10^{-1}$	0.1734	0.2227	0.2183	0.2034	0.1822
$T = 10^{0}$	0.1973	0.2023	0.2018	0.2004	0.1982

dependent on the local environment. On the other hand, in the high T limit, the acceptance probabilities are approximately equal, and rings are selected on a more random basis. This is demonstrated in table 1.1, using the example configurations from figure 1.4. The "temperature" parameter is therefore the primary method for controlling the distribution of ring sizes in constructed networks.

1.4 Properties of Triangle Rafts

The triangle raft method is evaluated in terms of its effectiveness in producing configurations which accurately replicate the network properties of experimental silica bilayers *i.e.* the ring statistics and Aboav-Weaire parameter. It is also compared against the existing methods introduced in section 1.2, namely generation from amorphous graphene or molecular dynamics. This is performed in the wider context of systematically varying the model parameters to explore the behaviour of generic networks of this type.

1.4.1 Network Growth

The triangle raft method is robust and controllable, and is able to generate configurations with tuneable ring statistics and topologies. Results will largely focus on the system where k=4-10, denoted $\{4,10\}$, mimicking the experimentally observed range for silica bilayers. Six example configurations are given in figure 1.5, which are generated with a range of temperatures and growth geometries. Figures 1.5a-1.5d provide a good qualitative analysis of the effect of temperature on the ring structure. At low temperature a phase boundary can be seen separating crystalline and amorphous regions, as seen in experimental silica bilayers [129]. In these samples although the proportion of small and large rings is low, their positions are highly correlated and chain structures of alternating rings sizes are clearly present. These motifs are reminiscent of defects found in a wide range of materials, including amorphous graphene and thin silicon and germanium oxides [3, 7, 11, 20]. The increase in temperature is coupled with the emergence of rings of more extreme sizes and regions which could be viewed as nano-crystalline are dispersed. The high temperature limit reveals a fully amorphous structure.

Figures 1.5e and 1.5f give examples of the diverse geometries in which samples may be constructed. It is interesting to note that even "difficult" shapes, such as those containing concave regions and cusps, do not prevent growth. Although the shape does not affect the network topology and is in a sense arbitrary, certain calculations may benefit from the different configurational shapes. For instance,

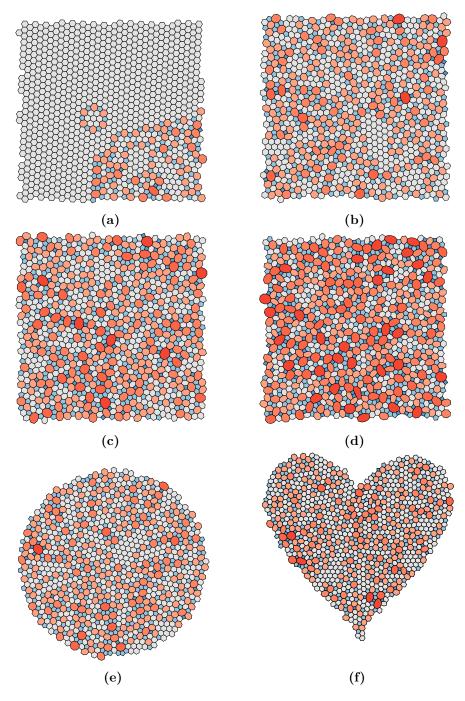


Figure 1.5: Example 1,000 ring configurations generated with different temperatures and shapes. Panels (a) through (d) show square lattices grown at $T=10^{-4.0}$, $10^{-3.0}$, $10^{-2.5}$, $10^{-2.0}$ respectively. The samples show the increasing diversity in ring structure as temperature is increased. Panels (e), (f) show configurations with alternative lattice shapes at $T=10^{-3.0}$, demonstrating the flexibility of the method in growing samples with variable geometries. Rings are coloured according to size with k<6 as blue, k=6 as grey and k>6 as red.

molecular dynamics with sliding boundary conditions requires fitting of a smooth function to the sample perimeter, which is facilitated by having a near-circular form. Other areas such as percolation problems may benefit from square samples.

1.4.2 Network Properties

The quantitative relationship between temperature and ring structure was investigated for three systems of varying ring size ranges; $\{5,7\}$, $\{4,8\}$ and $\{4,10\}$. For each system, 100 samples consisting of 1000 rings were grown at temperatures between $T = 10^{-4.5} \rightarrow 10^{-1.5}$. The evolution of the combined ring statistics with temperature is presented in figure 1.6. Figures 1.6a-1.6c give bar representations of the ring size distributions for the three systems, which show different behaviours. For $\{5,7\}$ the individual p_k are all monotonically increasing $(k \neq 6)$ or decreasing (k = 6) functions, but both $\{4,8\}$ and $\{4,10\}$ have p_k containing maxima. Additionally, both $\{5,7\}$ and $\{4,8\}$ achieve uniform distributions in the high temperature limit but $\{4,10\}$ does not.

This disparity in behaviour can largely be traced back to the constraint of Euler's theorem. As $\{5,7\}$ comprises of just three ring sizes, Euler's formula demands that $p_5 = p_7 = (1 - p_6)/2$ and so the system is relatively well defined. Hence, as the 5 and 7-rings are more strained than the 6-ring, p_5 and p_7 show a systematic increase with temperature. Furthermore, the uniform equilibrium distribution can only satisfy Euler's formula when the ring size range is symmetric about 6, as is observed for $\{5,7\}$ and $\{4,8\}$. The form of the ring statistics at intermediate temperatures and for $\{4,10\}$ follow the maximum entropy solutions according to Lemaître's law, discussed in section ?? and later in this section.

The ring distribution for $\{4, 10\}$ is also shown as a function of temperature in figure 1.6d, along with the value of the Aboav-Weaire parameter, α , allowing for more facile comparison with experiment. The temperature which gives the best agreement between our model and amorphous experimental samples is highlighted by the vertical dashed line. The values of p_k and α are provided in table 1.2, alongside results from two experimental samples. It is evident that the model can

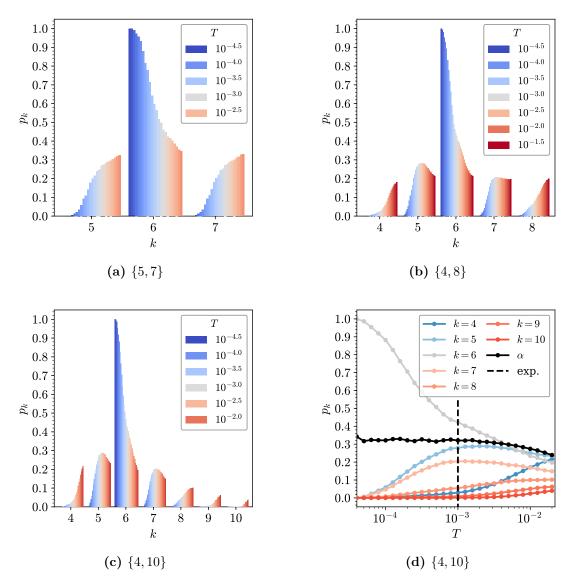


Figure 1.6: Variation in ring statistics with temperature over a given allowable k-range. Panels (a)-(c) show bar graph representations of the ring statistics, coloured by temperature, for the $\{5,7\}$, $\{4,8\}$ and $\{4,10\}$ systems, respectively. Panel (d) gives an alternative line graph representation of the ring statistics for $\{4,10\}$, coloured by ring size, along with the Aboav-Weaire parameter. The temperature which gives the best match to the experimentally observed amorphous region is also highlighted (vertical black dashed line).

be successfully tuned to match the topology of the experimental system. Not only are the ring distributions in very good accordance, but also the ring correlations, which have until now proved difficult to capture. This provides confidence that this simplified but physically motivated triangle raft model is able to reproduce the behaviour of real systems.

1.4. Properties of Triangle Rafts

Table 1.2: Comparison of silica bilayer samples from experiment, computational modelling and theory.

	Experiment			Computation			Theory
	Ru(0001) [132]	Graphene [4]	$\mathrm{MC^{a}}[29]$	$\mathrm{MC^{a}}[29]$	$MD^{b}[28]$	TR^{c}	Lemaître [62]
\overline{N}	317	444	216	418	16×85000	1000×100	_
p_3	0.0000	0.0000	0.00	0.00	0.0038	0.0000	0.0000
p_4	0.0379	0.0383	0.02	0.00	0.0537	0.0295	0.0280
p_5	0.2744	0.2725	0.33	0.37	0.2686	0.2786	0.2834
p_6	0.4448	0.4189	0.37	0.32	0.3773	0.4234	0.4200
p_7	0.1609	0.2117	0.21	0.25	0.2224	0.2034	0.2077
p_8	0.0757	0.0495	0.07	0.06	0.0602	0.0544	0.0518
p_9	0.0063	0.0068	< 0.01	0.00	0.0118	0.0097	0.0082
p_{10}	0.0000	0.0023	0.00	0.00	0.0018	0.0010	0.0009
$p_{>10}$	0.0000	0.0000	0.00	0.00	0.0004	0.0000	0.0000
μ_2	0.9460	0.9333	0.94	0.86	1.1302	0.9208	0.8985
α	0.32	0.33	0.18	0.23	0.25	0.32	-

Note: Each method is given alongside the number of rings in the sample, N, followed by the ring statistics, p_k , the second moment of the ring statistics, μ_2 , and the Aboav-Weaire parameter, α

^a Bond switching Monte Carlo (graphene potential)

^b Molecular dynamics ^c Triangle rafts, this work, $T = 10^{-3}$

Table 1.2 also lists the ring statistics obtained from previous computational studies which used both Monte Carlo and molecular dynamics methods. As mentioned in the review of these methods above, neither fully succeeds in accurately capturing the topology of silica bilayers. Kumar et al. attempted to transform an amorphous graphene structure generated from bond switching Monte Carlo into a silica bilayer. The ring statistics of the resulting structure were approximately correct, but the proportion of 5- and 6- rings over- and under-estimated respectively. In addition the Aboav-Weaire parameter was substantially lower than experiment, indicating a relative lack of structure in the ring ordering. The origin of these discrepancies is likely the use of a graphene potential model. The increased stiffness of the carbon network (which unlike silica lacks bridging oxygens) means a high temperature must be used to obtain an amorphous structure with the required disorder. This leads to heavily distorted rings (as noted in the original paper) which reduces the requirement for small rings to be adjacent to large.

Roy et al. have an alternative approach of generating configurations with an effective pair potential and molecular dynamics. As can be seen the ring statistics are closer to the experimental values, but now contain artefacts, with a significant fraction of highly strained 3-membered rings and large rings up to k = 14. These manifest as a result of the artificially high cooling rates in the computational studies which trap defect states in the configurations. Once again the final Aboav-Weaire parameter, α , is underestimated.

It is worth re-emphasising here that the triangle raft method is able to replicate experimental values of both p_k and α , due to its tuneable approach and "organic" growth mechanism, where sample formation is not influenced by enforced periodicity. Beyond this, the controllable nature of the method also allows insight into key questions about silica bilayers, for instance the form of the ring distribution in this amorphous phase. As detailed in section ??, the maximum entropy ring distribution can be calculated numerically given the value of p_6 . For example, table 1.2 gives the maximum entropy solution for $p_6 = 0.42$, which agrees very well with the results from

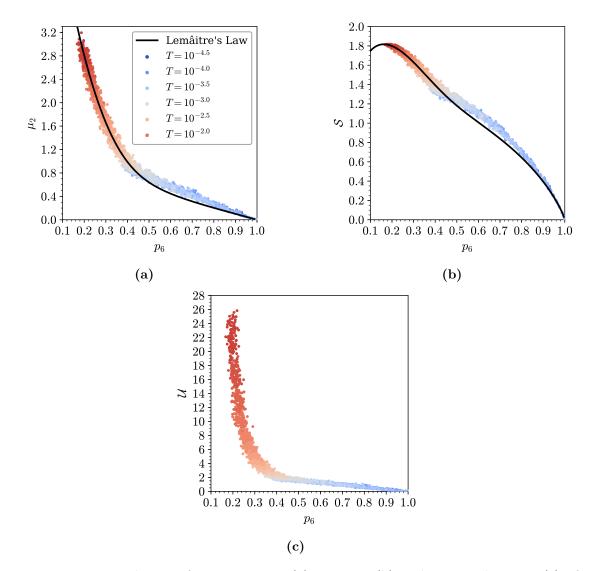


Figure 1.7: Evolution of ring statistics (a), entropy (b) and potential energy (c) of triangle rafts with temperature. The experimental value of $p_6 \approx 0.4$ occurs just before the exponential increase in potential energy, reflecting the balance of energetic and entropic factors.

triangle rafts and experiment. This second moment of the distribution, μ_2 , is then uniquely related to p_6 via Lemaître's law, shown as the black line in figure 1.7a.

However, Lemaître's law gives no information on why a particular maximum entropy distribution is found for a given system. The triangle raft method allows systematic generation of configurations with different p_6 values by tuning the temperature parameter. The resulting configurations follow Lemaître's law across the entire temperature range. Figures 1.7 gives the results from the individual 1000 ring samples, coloured by temperature. Figures 1.7a and 1.7b compare the

observed μ_2 and S (entropy) of the generated configurations to those expected from Lemaître's law, showing the law provides a good fit, with only a small deviation observed for $p_k > 0.5$.

Figure 1.7c plots the geometry optimised potential energy of the samples against p_6 , which increases as the ring sizes become more diverse. The curve is split into two regimes, with gradual increase in energy from $p_6 = 1.0 \rightarrow 0.4$ followed by exponential increase for $p_6 < 0.4$. This is consistent with the information in figure 1.6d which shows that below $p_6 \approx 0.4$, not only does the number of extreme ring sizes increase rapidly, but they become less correlated with a lower α , decreasing the number of favourable small-large ring pairings.

It can now be proposed why the experimental amorphous distributions are found with a value of $p_6 \approx 0.4$. The system aims to maximise entropy by obtaining a ring distribution along the Lemaître curve with the minimum p_6 possible. However, for $p_6 < 0.4$ the energetic cost becomes prohibitively large, as higher entropy distributions can only be achieved by increasing the proportion of extreme ring sizes at the expense of relatively low strain 5- and 7- rings. Interestingly it is also evident why no configurations are present below $p_6 \approx 0.16$, even at the highest temperature. Below this point, the entropy of the $\{4, 10\}$ system decreases whilst the energy continues to rise and so there is no driving force to sample this area of phase space.

1.4.3 Physical Properties

As an additional check that the developed triangle raft model behaves physically, the angle distribution between adjacent SiO₃ units, $f(\theta)$, was calculated for the $\{4, 10\}$ system across the range of temperatures studied. The results are summarised in figure 1.8a. The angle distributions are necessarily symmetric about 120°, as each triangle pair contributes two complementary angles. At lower temperatures the distribution is dominated by angles close to 120°, as a consequence of the large proportion of near strainless six membered rings. Furthermore, at the temperature corresponding to the amorphous experimental region, $T = 10^{-3}$, the distribution has a similar extent to the angle distribution found in experimental samples (see

for example figure 7 reference [28]). However, as the temperature increases, the form of $f(\theta)$ does not simply broaden as might be expected, but becomes bimodal. This can be rationalised by considering the angles that would be present in regular polygons of different sizes, marked by vertical lines in figure 1.8a. These ideal angles are clustered away from the mean value of 120°, and hence increasing the diversity of ring sizes through temperature acts to shift the most commonly observed angles from the central value of 120°. It is therefore interesting to note that increasing structure in the angle distribution does not necessarily translate to increased order in the atomic configurations.

A final check comes from examining the ring areas in the generated configurations. Inspection of amorphous experimental samples reveals that the rings appear highly regular in shape. This can be quantified by determining the average dimensionless area for each ring size, A_k , and comparing it to the area of the corresponding regular polygon, A_k^0 , where:

$$A_k = \frac{\langle Area(k) \rangle}{(r_{SiSi}^0)^2},\tag{1.4}$$

$$A_k^0 = \frac{k}{4\tan(\pi/k)}. (1.5)$$

As the regular polygon has the maximum achievable area for a given ring size, the ratio A_k/A_k^0 is expected to lie in the range $0 \to 1$, with a lower value corresponding to increased deviation from regularity, and assuming r_{SiSi}^0 to be fixed.

The study by Kumar et al. found that whereas for experimental configurations, $A_k/A_k^0 \approx 1$, configurations generated using thier bond switching method generally displayed ratios much less than unity [104], indicative of large distortions in the ring structure. For larger rings, a value of $A_k/A_k^0 > 1$ was also found, which can only be achieved if there is appreciable bond stretching (see equations (1.4), (1.5)).

The analogous results for the method presented in this chapter can be found in figure 1.8b, for $T=10^{-3}$. This figure demonstrates that there is now good agreement between experimental and computational results. In both cases the deviation from regularity increases with ring size, as the flexibility of the rings

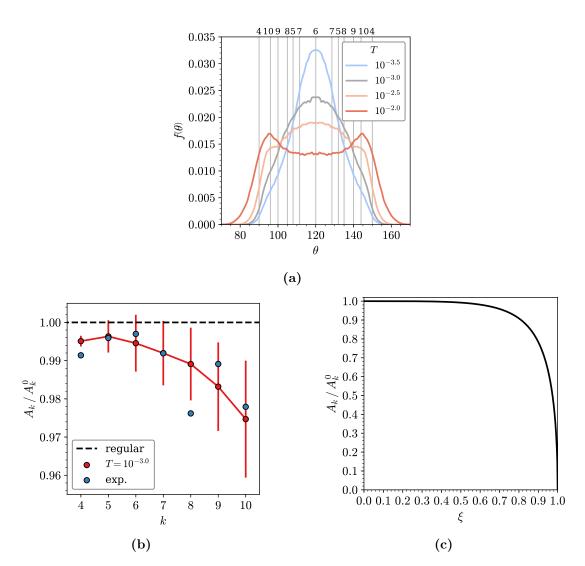


Figure 1.8: Panel (a) gives the ring angle distribution function for triangle rafts formed at different temperatures. Panel (b) compares the regularity of rings in computational and experimental amorphous configurations, with points indicating the mean value and bars corresponding to the standard deviation. Experimental data is taken from ref. [29]. Panel (c) shows the effect on the area when distorting a circle to an ellipse whilst maintaining a constant perimeter length.

increases. Again it can be proposed that the difference between current and previous methods could be due to the lack of enforced periodicity on the system. By allowing the network to grow relatively freely, the system can avoid a build up of strain associated with maintaining periodic boundaries.

Even with this analysis, an argument can be made that by visual inspection the rings in the experimental configurations are still more regular than those generated from computational samples. Therefore one can consider if deformation of a ring should be expected to lead to significant reduction in area. This can be explored by considering the distortion of a circle to an ellipse. The degree of distortion can be described by the eccentricity of the ellipse,

$$\xi = \left(1 - \frac{b^2}{a^2}\right)^{1/2},\tag{1.6}$$

where a, b are the major and minor axis radii respectively. This change in area with distortion is shown in figure 1.8c, the calculation of which can be found in appendix ??. As can be seen, a large degree of eccentricity is needed for a significant change in the observable area. For example, if a = 1.5b, the area is still $\approx 0.94\%$ of the area of the corresponding circle.

For silica networks the Si–Si distances lie in a narrow range because of the covalent nature of the atomic bonding and the near-linear Si-O-Si bridges which join the two layers. Hence we would expect similar behaviour to occur, with ring areas relatively invariant to distortions in the ring shape (this same analysis would not be expected to hold for foams for example, where the length of the boundary is much more flexible). This suggests that the ring area is not the most suitable metric for quantifying the regularity of rings in systems such as this, and could explain any disagreement between the seemingly near ideal ring areas and the visual evidence. As previously stated, although the potential model used is physically motivated, it is lightweight in order to facilitate generation of a large number of configurations with the correct network topology. In future it would be informative to see if the required regularity can be achieved by geometry optimising the resulting bilayer configurations with a more accurate potential, such as the TS potential which includes potentially significant electrostatic interactions including many-body polarisation effects [134].

1.5 Chapter Conclusions

A method has been developed for effective modelling of silica bilayers and related materials. Bilayers are represented as triangle rafts, which are sequentially constructed from a seed using a stochastic growth algorithm. The algorithm is flexible, allowing control over the ring size distribution and overall system topology. The success of triangle rafts in modelling silica bilayers has been demonstrated by the values of the Aboav-Weaire parameter, α , which are are more commensurate with those obtained from experimental imaging than configurations generated by previous methods. Moreover, consideration of the ring areas shows that triangle raft configurations contain highly regular polygons - another experimental observation that has proved challenging to previously capture.

The real advantage of the method is that it enables a computationally tractable and systematic exploration of bilayer systems at increasing levels of disorder. This has been employed in this chapter for a detailed analysis of Lemaître's law, which rationalised why the fraction of six-membered rings observed in real systems is often ≈ 0.4 . However, it will also be used in chapter ?? to investigate the use of persistent homology in amorphous materials. The ability to build a triangle raft from any user-defined starting seed also opens further possibilities for the method. In particular, it would be interesting to see how network growth is affected by the presence of a template, which could be for example a very large ring. This could lead to insight into how to control pore geometries in these materials.

Appendices

A | Calculation of Forces

The calculation of forces from a given potential model is central to the geometry optimisation routines (see section ??) employed in many algorithms in this thesis. This appendix derives the equations used to calculate the forces for the potentials covered throughout this work. The notation used will aim to be consistent, and is outlined as follows. A position vector is denoted by $\mathbf{r} = [x \ y]^T$, with:

$$\mathbf{r}_{ij} = \mathbf{r}_j - \mathbf{r}_i \,, \tag{A.1}$$

$$r_{ij} = |\mathbf{r}_{ij}|, \tag{A.2}$$

$$\hat{\mathbf{r}}_{ij} = \frac{\mathbf{r}_{ij}}{r_{ij}} \,. \tag{A.3}$$

The derivative of a function with respect to \mathbf{r} is then given by $\frac{\partial f(\mathbf{r})}{\partial \mathbf{r}} = \left[\frac{\partial f(\mathbf{r})}{\partial x} \frac{\partial f(\mathbf{r})}{\partial y}\right]^T$. It therefore follows that:

$$\frac{\partial r_{ij}}{\partial \mathbf{r}_{ij}} = \mathbf{\hat{r}}_{ij} , \quad \frac{\partial r_{ij}}{\partial \mathbf{r}_{i}} = -\mathbf{\hat{r}}_{ij} , \quad \frac{\partial r_{ij}}{\partial \mathbf{r}_{j}} = \mathbf{\hat{r}}_{ij} . \tag{A.4}$$

Angles are denoted by θ_{ijk} , representing the angle between \mathbf{r}_{ij} and \mathbf{r}_{ik} . It will also be useful to determine the derivative of the cosine of angles with respect to a position vector:

$$\frac{\partial \cos \theta_{ijk}}{\partial \mathbf{r}_{j}} = \frac{\partial}{\partial \mathbf{r}_{j}} \left(\frac{\mathbf{r}_{ij} \cdot \mathbf{r}_{ik}}{r_{ij} r_{ik}} \right)
= \frac{r_{ij} r_{ik} \mathbf{r}_{ik} - \mathbf{r}_{ij} \cdot \mathbf{r}_{ik} \hat{\mathbf{r}}_{ij} \mathbf{r}_{ik}}{r_{ij}^{2} r_{ik}^{2}}
= \frac{1}{r_{ij}} \left(\hat{\mathbf{r}}_{ik} - \hat{\mathbf{r}}_{ij} \cdot \hat{\mathbf{r}}_{ik} \hat{\mathbf{r}}_{ij} \right)
= \frac{1}{r_{ij}} \left(\hat{\mathbf{r}}_{ik} - \cos \theta_{ijk} \hat{\mathbf{r}}_{ij} \right) ,$$
(A.5)

and similarly

$$\frac{\partial \cos \theta_{ijk}}{\partial \mathbf{r}_k} = \frac{1}{r_{ik}} \left(\hat{\mathbf{r}}_{ij} - \cos \theta_{ijk} \hat{\mathbf{r}}_{ik} \right) . \tag{A.6}$$

These relationships form the basis to derive the forces for the various stretching and angular potentials used in this thesis.

The force on a given particle at \mathbf{r}_i is given by the negative derivative of the potential:

$$\mathbf{F}_i = -\frac{\partial \mathcal{U}}{\partial \mathbf{r}_i}.\tag{A.7}$$

As forces are conservative, the sum of the forces on all particles must be zero *i.e.* for stretching and angular terms respectively:

$$\mathbf{F}_i = -\mathbf{F}_i \,, \tag{A.8}$$

$$\mathbf{F}_i = -\mathbf{F}_j - \mathbf{F}_k. \tag{A.9}$$

In the following sections K denotes a force constant and a subscript zero an equilibrium value.

A.1 Harmonic Stretching Potential

The harmonic stretching potential is a simple bonding potential that approximates many atomic potentials at small displacements. The interaction between two particles at separation \mathbf{r}_{ij} is given by:

$$\mathcal{U} = \frac{K}{2} (r_{ij} - r_0)^2 . (A.10)$$

The forces are therefore:

$$\mathbf{F}_{j} = -\frac{\partial \mathcal{U}}{\partial r_{ij}} \frac{\partial r_{ij}}{\partial \mathbf{r}_{j}} = -K \left(r_{ij} - r_{0} \right) \hat{\mathbf{r}}_{ij}, \qquad (A.11)$$

with \mathbf{F}_i given by equation (A.8).

A.2 Quartic Stretching Potential

The quartic stretching potential is related to the harmonic potential, but is even more computationally efficient as there is no square root operations are required. The interaction between two particles at separation \mathbf{r}_{ij} is given by:

$$\mathcal{U} = \frac{K}{4} \left(r_{ij}^2 - r_0^2 \right)^2 \,. \tag{A.12}$$

The forces are therefore:

$$\mathbf{F}_{j} = -\frac{\partial \mathcal{U}}{\partial r_{ij}} \frac{\partial r_{ij}}{\partial \mathbf{r}_{j}} = -K \left(r_{ij}^{2} - r_{0}^{2} \right) \mathbf{r}_{ij}, \qquad (A.13)$$

with \mathbf{F}_i given by equation (A.8).

A.3 Shifted and Cut 24-12 Potential

The shifted and cut 24-12 potential is a computationally efficient way to imitate short range hard repulsions, whilst maintaining continuous forces. The interaction between two particles at separation \mathbf{r}_{ij} is given by:

$$\mathcal{U} = \begin{cases} \epsilon \left[\left(\frac{r_0}{r_{ij}} \right)^{24} - 2 \left(\frac{r_0}{r_{ij}} \right)^{12} \right] + \epsilon & r_{ij} \le r_0 \\ 0 & \text{otherwise} \end{cases}$$
(A.14)

The forces are therefore:

$$\mathbf{F}_{j} = -\frac{\partial \mathcal{U}}{\partial r_{ij}} \frac{\partial r_{ij}}{\partial \mathbf{r}_{j}} = \begin{cases} \frac{24\epsilon}{r} \left[\left(\frac{r_{0}}{r_{ij}} \right)^{24} - \left(\frac{r_{0}}{r_{ij}} \right)^{12} \right] & r_{ij} \leq r_{0} \\ 0 & \text{otherwise} \end{cases}, \tag{A.15}$$

with \mathbf{F}_i given by equation (A.8).

A.4 Harmonic Cosine Angle Potential

In analogue with the stretching potential, the harmonic cosine angle is a elegant yet simple form angular potential, utilising the cosine function to reduce overheads when calculating angles. The interaction between three particles with angle θ_{ijk} is given by:

$$\mathcal{U} = \frac{K}{2} \left(\cos \theta_{ijk} - \cos \theta_0 \right)^2. \tag{A.16}$$

The forces are therefore:

$$\mathbf{F}_{j} = -\frac{\partial \mathcal{U}}{\partial \cos \theta_{ijk}} \frac{\partial \cos \theta_{ijk}}{\partial \mathbf{r}_{j}}$$

$$= -\frac{K}{r_{ij}} \left(\cos \theta_{ijk} - \cos \theta_{0}\right) \left(\mathbf{\hat{r}}_{ik} - \cos \theta_{ijk} \mathbf{\hat{r}}_{ij}\right) , \qquad (A.17)$$

with \mathbf{F}_k having an analogous form and \mathbf{F}_i given by equation (A.9).

A.5 Restricted Bending Potential

The restricted bending (ReB) potential is a modification on the harmonic cosine angle potential which diverges at $\theta_{ijk} = 0, \pi$, ensuing angles cannot become reflex. The interaction between three particles with angle θ_{ijk} is given by:

$$\mathcal{U} = \frac{K}{2} \frac{\left(\cos \theta_{ijk} - \cos \theta_0\right)^2}{\sin^2 \theta_{ijk}}.$$
 (A.18)

The forces are therefore:

$$\mathbf{F}_{j} = -\frac{\partial \mathcal{U}}{\partial \cos \theta_{ijk}} \frac{\partial \cos \theta_{ijk}}{\partial \mathbf{r}_{j}}$$

$$= -\frac{K}{r_{ij} \sin^{4} \theta_{ijk}} \left(\cos \theta_{ijk} - \cos \theta_{0}\right) \left(1 - \cos \theta_{ijk} \cos \theta_{0}\right) \left(\mathbf{\hat{r}}_{ik} - \cos \theta_{ijk} \mathbf{\hat{r}}_{ij}\right) , \quad (A.19)$$

with \mathbf{F}_k having an analogous form and \mathbf{F}_i given by equation (A.9).

A.6 Keating Potential

The Keating potential combines the quartic stretching potential with a computationally efficient angle potential of the form:

$$\mathcal{U} = \frac{K}{2} \left(\mathbf{r}_{ij} \cdot \mathbf{r}_{ik} - r_0^2 \cos \theta_0 \right)^2 \tag{A.20}$$

for three particles with an angle given by \mathbf{r}_{ij} and \mathbf{r}_{ik} . The forces are therefore:

$$\mathbf{F}_{j} = -\frac{\partial \mathcal{U}}{\partial \mathbf{r}_{j}}$$

$$= -K \left(\mathbf{r}_{ij} \cdot \mathbf{r}_{ik} - r_{0}^{2} \cos \theta_{0} \right) \mathbf{r}_{ik}, \qquad (A.21)$$

with \mathbf{F}_k having an analogous form and \mathbf{F}_i given by equation (A.9).

A.7 Proper Line Intersection

Some potential models have an additional term to prevent overlap of edges in a two-dimensional network, termed proper line intersection. This can be detected using standard computational geometry algorithms [241]. The signed area of a triangle, A, is given by:

$$A(\mathbf{r}_0, \mathbf{r}_1, \mathbf{r}_2) = \frac{1}{2} \sum_{i=0}^{2} (x_i y_{i+1} - y_i x_{i+1}).$$
 (A.22)

A point can then be designated "left" of a line segment if A>0 and "right" otherwise. Overlap of two line segments can be detected if one point of one segment is "left" and the other point "right" with respect to the other segment, and no three points are collinear.

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