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Chemistry Principles and Practice

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Summary: 1. Introduction to Chemistry. 2. Atoms, Molecules, and Ions. 3. Equations, the Mole, and Chemical Formulas. 4. Chemical Reactions in Solution Thermochemistry. 6. The Gaseous State. 7. Electronic Structure. 8. The Periodic Table: Structure and Trends. 9. Chemical Bonds. 10. Molecular Structure and Bonding Theories. 11. Liquids and Solids. 12. Solutions. 13. Chemical Kinetics. 14. Chemical Equilibrium. 15. Solutions of Acids and Bases. 16. Reactions Between Acids and Bases. 17. Chemical Thermodynamics. 18. Electrochemistry. 19. Metals, Coordination Chemistry, and Metallurgy. 20. The Chemistry of Hydrogen, Elements in Groups 3A through 6A, and the Noble Gases. 21. Nuclear Chemistry. 22. Organic Chemistry and Biochemistry. Appendix A: Math Procedures. Appendix B: Selected Physical Constants. Appendix C: Unit Conversion Factors. Appendix D: Names of Ions. Appendix E: Properties of Water. Appendix F: Solubility Products, Acids, and Bases. Appendix G: Thermodynamic Constants for Selected Compounds. Appendix H: Standard Reduction Potentials at 25.'C Appendix I: Glossary. Appendix J: Answers to Selected Exercises.



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