SBMLsqueezer: Differential Equation System "untitled"

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1 Rate Laws

1.1 Reaction: re1

$$v_1 = \mathtt{vmaf_re1} \cdot \frac{\left(\left[\mathtt{s1} \right] \cdot \mathrm{vol} \left(\mathtt{c1} \right) \right)^{\mathtt{hic_re1_s1}}}{\left(\left[\mathtt{s1} \right] \cdot \mathrm{vol} \left(\mathtt{c1} \right) \right)^{\mathtt{hic_re1_s1}} + \mathtt{ksp_re1_s1}^{\mathtt{hic_re1_s1}}} \tag{1}$$

1.2 Reaction: re2

$$v_2 = [\texttt{s3}] \cdot \text{vol} (\texttt{c1}) \cdot \frac{\text{kcrf_re2_s3} \cdot \frac{[\texttt{s4}] \cdot \text{vol}(\texttt{c1})}{\text{kmc_re2_s4_s3}} \cdot \frac{[\texttt{s5}] \cdot \text{vol}(\texttt{c1})}{\text{kmc_re2_s5_s3}} - \text{kcrr_re2_s3} \cdot \frac{[\texttt{s6}] \cdot \text{vol}(\texttt{c1})}{\text{kmc_re2_s6_s3}}}{\left(1 + \frac{[\texttt{s4}] \cdot \text{vol}(\texttt{c1})}{\text{kmc_re2_s4_s3}}\right) \cdot \left(1 + \frac{[\texttt{s5}] \cdot \text{vol}(\texttt{c1})}{\text{kmc_re2_s5_s3}}\right) + \frac{[\texttt{s6}] \cdot \text{vol}(\texttt{c1})}{\text{kmc_re2_s6_s3}}}$$
 (2)

1.3 Reaction: re3

$$v_{3} = [\mathtt{s8}] \cdot \mathrm{vol}\left(\mathtt{c1}\right) \cdot \frac{\frac{\mathtt{kcrf}.\mathtt{re3}.\mathtt{s8}}{\mathtt{kmc}.\mathtt{re3}.\mathtt{s6}.\mathtt{s8}} \cdot [\mathtt{s6}] \cdot \mathrm{vol}\left(\mathtt{c1}\right) - \frac{\mathtt{kcrr}.\mathtt{re3}.\mathtt{s8}}{\mathtt{kmc}.\mathtt{re3}.\mathtt{s7}.\mathtt{s8}} \cdot [\mathtt{s7}] \cdot \mathrm{vol}\left(\mathtt{c1}\right)}{1 + \frac{[\mathtt{s6}] \cdot \mathrm{vol}\left(\mathtt{c1}\right)}{\mathtt{kmc}.\mathtt{re3}.\mathtt{s6}.\mathtt{s8}} + \frac{[\mathtt{s7}] \cdot \mathrm{vol}\left(\mathtt{c1}\right)}{\mathtt{kmc}.\mathtt{re3}.\mathtt{s7}.\mathtt{s8}}}$$

$$(3)$$

1.4 Reaction: re4

$$v_{4} = \mathtt{vmaf_re4} \cdot \frac{\left([\mathtt{s2}] \cdot \mathtt{vol} \left(\mathtt{c1} \right) \right)^{\mathtt{hic_re4_s2}}}{\left([\mathtt{s2}] \cdot \mathtt{vol} \left(\mathtt{c1} \right) \right)^{\mathtt{hic_re4_s2}} + \mathtt{ksp_re4_s2^{\mathtt{hic_re4_s2}}}} \tag{4}$$

1.5 Reaction: re5

$$v_5 = zkass_re5 - kdiss_re5$$
 (5)

1.6 Reaction: re6

$$v_6 = zkass_re6 - kdiss_re6$$
 (6)

2 Equations

2.1 Species: s11

$$\frac{\mathrm{d}[\mathbf{s}_{11}]}{\mathrm{dt}} = -v_1 \tag{7}$$

2.2 Species: s2

$$\frac{\mathrm{d}[\mathbf{s}_2]}{\mathrm{d}t} = 0 \tag{8}$$

2.3 Species: s1

$$\frac{\mathrm{d}[\mathbf{s}_1]}{\mathrm{d}t} = 0 \tag{9}$$

2.4 Species: s4

$$\frac{\mathrm{d}[\mathbf{s}_4]}{\mathrm{dt}} = -v_2 \tag{10}$$

2.5 Species: s5

$$\frac{\mathrm{d}[\mathbf{s}_5]}{\mathrm{dt}} = -v_2 \tag{11}$$

2.6 Species: s6

$$\frac{\mathrm{d}[\mathbf{s}_6]}{\mathrm{dt}} = -v_3 \tag{12}$$

2.7 Species: s3

$$\frac{\mathrm{d}[\mathbf{s}_3]}{\mathrm{dt}} = -v_5 \tag{13}$$

2.8 Species: s7

$$\frac{\mathrm{d}[\mathbf{s}_7]}{\mathrm{d}t} = 0 \tag{14}$$

2.9 Species: s8

$$\frac{\mathrm{d}[\mathbf{s}_8]}{\mathrm{dt}} = -v_6 \tag{15}$$

2.10 Species: s12

$$\frac{\mathrm{d}[\mathbf{s}_{12}]}{\mathrm{dt}} = -v_4 \tag{16}$$

2.11 Species: s9 (sa3_degraded)

$$\frac{\mathrm{d[s_9]}}{\mathrm{d}t} = 0\tag{17}$$

2.12 Species: s10 (sa8_degraded)

$$\frac{\mathrm{d}[\mathbf{s}_{10}]}{\mathrm{d}t} = 0 \tag{18}$$

3 Species

Species	Initial concentration	compartment
$\overline{s_{11}}$	NaN	c1
s_2	NaN	c1
s_1	NaN	c1
s_4	NaN	c1
s_5	NaN	c1
s_6°	NaN	c1
s_3	NaN	c1
s_7	NaN	c1
s_8	NaN	c1
s_{12}	NaN	c1
s_9	NaN	c1
s_{10}	NaN	c1

4 Compartments

Compartment	Volume
$\overline{c_1}$	1.0

For a more comprehensive \LaTeX export, see

http://www.ra.cs.uni-tuebingen.de/software/SBML2LaTeX