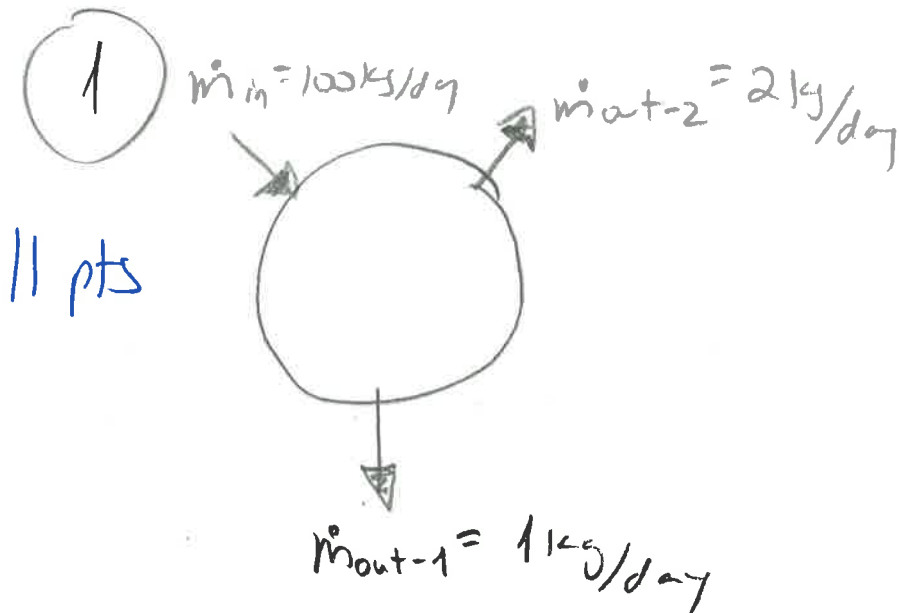


PSet 4 - KEY

CENG 310
Deborah Sills



$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} \pm \dot{m}_{rxn} \quad \text{no rxn}$$

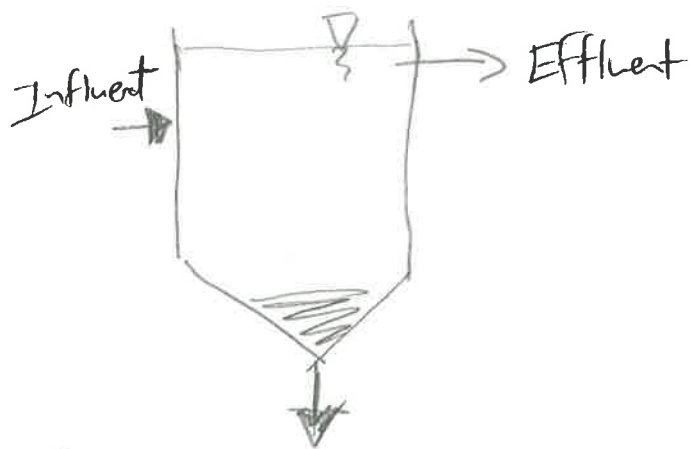
$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out-1} - \dot{m}_{out-2}$$

$$\frac{dm}{dt} = 100 \text{ kg/day} - 1 \text{ kg/day} - 2 \text{ kg/day}$$

$$\boxed{\frac{dm}{dt} = 97 \text{ kg/day}}$$

2

11 pts



Assume that $Q_{in} = Q_{out} = Q$
 b.c. $Q_{sludge} \ll Q_{out}$

Sludge production = $\frac{dm}{dt}$

MASS Balance on Sludge no rxn

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} \pm \dot{m}_{rxn}$$

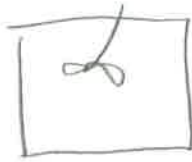
$$\frac{dm}{dt} = Q_{in} \times C_{in} - Q_{out} \times C_{out} = Q (C_{in} - C_{out})$$

$$\frac{dm}{dt} = 3780 \frac{m^3}{day} \times (220 \frac{mg}{L} - 5 \frac{mg}{L}) \times \frac{1000 L}{m^3} \times \frac{1 kg}{10^6 mg}$$

Sludge production = $\frac{dm}{dt} = \underline{\underline{813 \text{ kg/day}}}$

3

BATCH REACTOR



Mass Balance of chlorine

$$\frac{dm}{dt} = \cancel{\dot{m}_{in}} - \cancel{\dot{m}_{out}} - \dot{m}_{rxn}$$

no flow

11 pt

$$V \frac{dC}{dt} = -V \left(\frac{dC}{dt} \right)_{rxn}$$

$$\frac{dC}{dt} = -kC$$

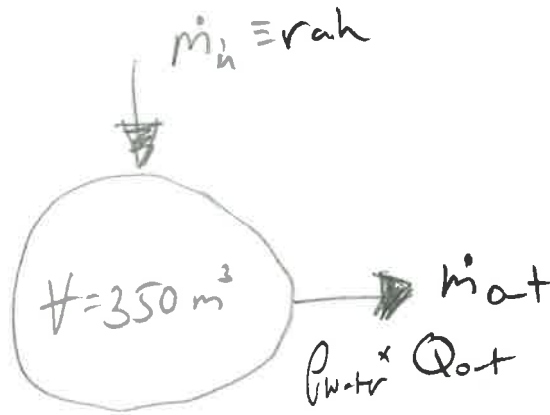
$$\int_{C_0}^C \frac{dC}{C} = \int_0^t -k dt$$

$$C = C_0 e^{-kt} = 1 \text{ mg/L} e^{-0.360 \frac{1}{\text{day}} \times 1 \text{ day}}$$

$$C = 0.698 \text{ mg/L}$$

4

11 pt



$\rho_{H_2O} \equiv$ density of water

$$V = 350 \text{ m}^3$$

$$t = 6 \text{ h}$$

$$Q_{out} = 320 \text{ L/min}$$

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out}$$

$$\cancel{\rho_{H_2O}} \frac{dV}{dt} = \cancel{\rho_{H_2O}} Q_{in} - \cancel{\rho_{H_2O}} Q_{out}$$

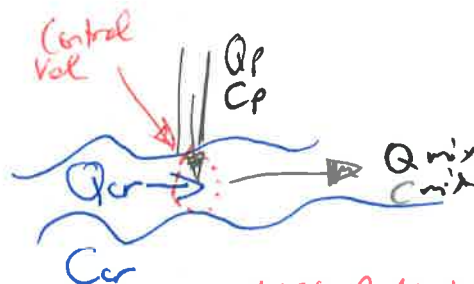
$$\frac{dV}{dt} = Q_{in} - Q_{out}$$

$$Q_{in} = \frac{dV}{dt} + Q_{out} = \frac{350 \text{ m}^3}{6 \text{ h}} \times \frac{1000 \text{ L}}{1 \text{ m}^3} \times \frac{1 \text{ h}}{60 \text{ min}} + 320 \text{ L/min}$$

$$Q_{in} = 1290 \text{ L/min}$$

5

11 pts



MASS BALANCE ON SALT

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} \pm \dot{m}_{rxn}$$

$$\frac{dm}{dt} = 0 \quad \text{@ junction}$$

⑤ cont...

no rxn

$$0 = \dot{m}_{in} - \dot{m}_{out} \pm \dot{m}_{rxn}$$

$$0 = Q_{cr} C_{cr} + Q_p C_p - Q_{mix} C_{mix}$$

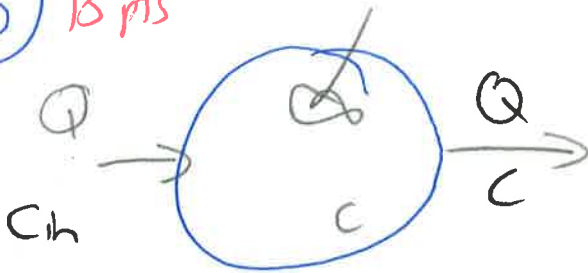
$$C_{mix} = \frac{Q_{cr} C_{cr} + Q_p C_p}{Q_{mix}}$$

but $Q_{mix} = Q_{cr} + Q_p = 0.5 + 2.8 = 3.3 \text{ MGD}$

$$C_{mix} = \frac{Q_{cr} C_{cr} + Q_p C_p}{Q_{cr} + Q_p} = \frac{2.8 \text{ MGD} \times 175 \text{ mg/L} + 0.5 \text{ MGD} \times 35,000 \text{ mg/L}}{3.3 \text{ MGD}}$$

$$C_{mix} = 5450 \text{ mg/L} = \underline{5.45 \times 10^3 \text{ mg/L}}$$

⑥ 15 pts



$r = 0.001 \frac{\text{mg}}{\text{L} \cdot \text{day}}$ zero order
 $C_{in} = 3.4 \text{ mg/L}$

7.5 pts
 a

Steady state

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} \pm \dot{m}_{rxn}$$

$$0 = Q C_{in} - Q C - rV$$

$$C = \frac{Q C_{in} - rV}{Q} = \frac{103 \text{ m}^3/\text{day} \times 3.4 \text{ mg/L} - 0.001 \frac{\text{mg}}{\text{L} \cdot \text{day}} \times 106 \text{ m}^3}{103 \text{ m}^3/\text{day}}$$

$C = 3.4 \text{ mg/L}$ ← nothing happens

6b

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} - \dot{m}_{rxn}$$

steady state

$$0 = QC_{in} - QC - KCV$$

$$C = \frac{QC_{in}}{Q + KV} = \frac{103 \text{ m}^3/\text{day} \times 3.4 \text{ mg/L}}{103 \text{ m}^3/\text{day} + 0.01 \frac{\text{m}}{\text{day}} \times 106 \text{ m}^3}$$

7.5 pts

$$C = 3.36 \text{ mg/L}$$

noting happened again

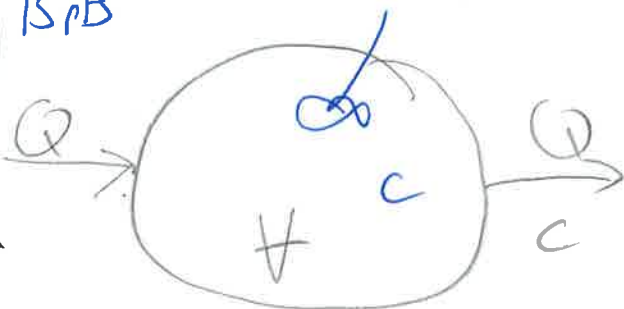
WHY?

Find out in lecture on Friday, 4 Oct!

7 15pts

9

C_{in}



$$\text{Area} = 10 \text{ m} \times 10000 \frac{\text{m}^2}{\text{m}} = 100,000 \text{ m}^2$$

$$V = \text{Area} \times d = 100,000 \text{ m}^3$$

7.5pts

9

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out} \pm \dot{m}_{rxn}$$

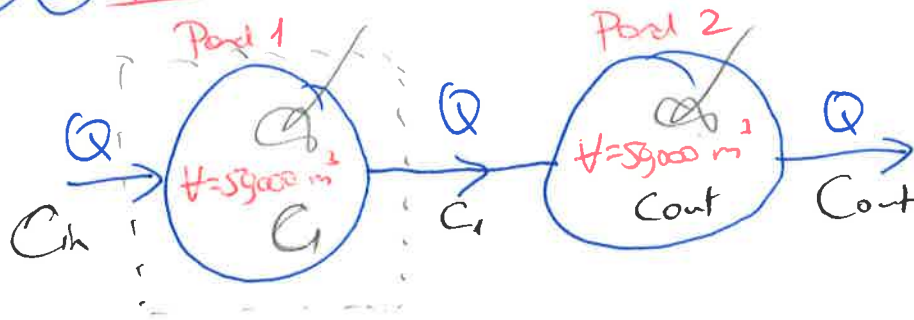
steady state

$$0 = QC_{in} - QC - KCV$$

$$K = \frac{Q(C_{in} - C)}{CV} = \frac{8640 \frac{\text{m}^3}{\text{day}} (100 - 20) \text{ mg/L}}{20 \text{ mg/L} \times 100,000 \text{ m}^3}$$

$$K = 0.35 \text{ day}^{-1}$$

⑦ ⑥ 5 pts



Mass Balance on
Pond 1

$$\frac{dn}{dt} = m_{in} - m_{out} - m_{rxn}$$

$$0 = QC_{in} - QC_1 - kC_1V$$

Eq. 1 $C_1 = \frac{QC_{in}}{Q + kV}$

Mass Balance on
Pond 2

$$\frac{dn}{dt} = m_{in} - m_{out} - m_{rxn}$$

$$0 = QC_1 - QC_{out} - kC_{out}V$$

Eq. 2
 $C_{out} = \frac{QC_1}{Q + kV}$

Substitute C_1 from Eq. 1 into Eq. 2

$$C_{out} = \left(\frac{Q}{Q + kV} \right)^2 C_{in}$$

$$\left(\frac{Q}{Q + kV} \right)^2 = \frac{C_{out}}{C_{in}}$$

$$\frac{Q}{Q + kV} = \sqrt{\frac{C_{out}}{C_{in}}}$$

$$Q = (Q + kV) \sqrt{\frac{C_{out}}{C_{in}}}$$

$$k = \frac{Q(1 - \sqrt{\frac{C_{out}}{C_{in}}})}{V \sqrt{\frac{C_{out}}{C_{in}}}} = \frac{8640(1 - \sqrt{\frac{20}{100}})}{5000 \sqrt{\frac{20}{100}}}$$

$$k = 0.21 \text{ day}^{-1}$$

When Sewage Stops Flowing - Non-Steady State Problem

$$\frac{dm}{dt} = m_{in} - m_{out} - m_{rxn}$$

$$V \frac{dC}{dt} = Q(C_{in} - C) - KC V$$

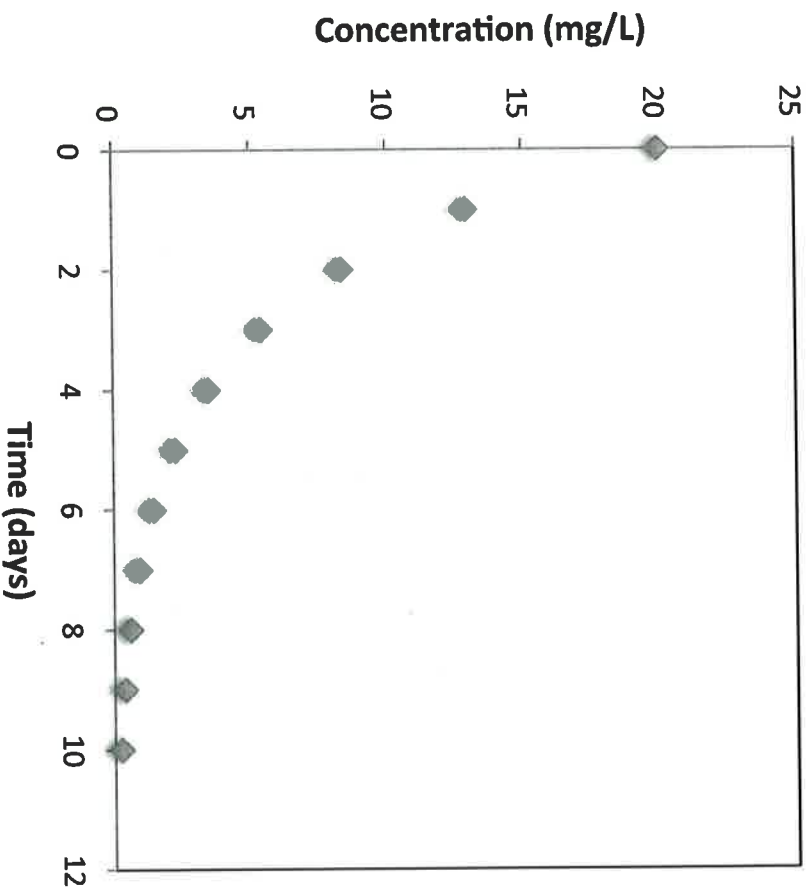
$$\int_{C(0)}^C \frac{dC}{C} = \int_0^t -(Q/V + K) dt$$

$$C = C_0 e^{-(Q/V + K)t} \quad \text{--- plot}$$

$$C(0) = 20 \text{ mg/L}$$

$k = 0.35 \text{ day}^{-1}$
 $Q = 8640 \text{ m}^3/\text{day}$
 $V = 100000 \text{ m}^3$

t days	Cout
0	20
1	12.9
2	8.4
3	5.4
4	3.5
5	2.3
6	1.5
7	0.9
8	0.6
9	0.4
10	0.3

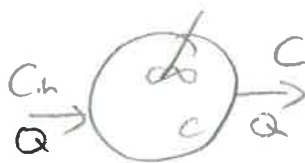


7 c
Sps

15pts



7.5pts

Steady State
CMFR

96% removal

$$C = 0.04 C_{in}$$

$$k = 0.8 \text{ /day}$$

$$\frac{dm}{dt} = m_{in} - m_{out} - m_{rxn}$$

$$0 = QC_{in} - QC - kCV$$

$$0 = C_{in} - C - kC \frac{V}{Q}$$

$$\text{HRT} \rightarrow \frac{V}{Q} = \theta$$

$$\theta = \frac{C_{in} - C}{kC} = \frac{(100 - 4) \text{ mg/L}}{0.8 \cdot 4 \text{ /day}} = 30 \text{ days}$$

$$\boxed{\text{HRT} = 30 \text{ days}}$$

7.5pts

Steady
State PFR $-k\theta$

$$C_{out} = C_{in} \cdot e^{-k\theta}$$

$$e^{-k\theta} = \frac{C_{out}}{C_{in}}$$

$$\theta = \frac{-\ln\left(\frac{C_{out}}{C_{in}}\right)}{k} = \frac{-\ln\left(\frac{4}{100}\right)}{0.8 \text{ /day}} = 4 \text{ day}$$

$$\boxed{\text{HRT} = 4 \text{ days}}$$

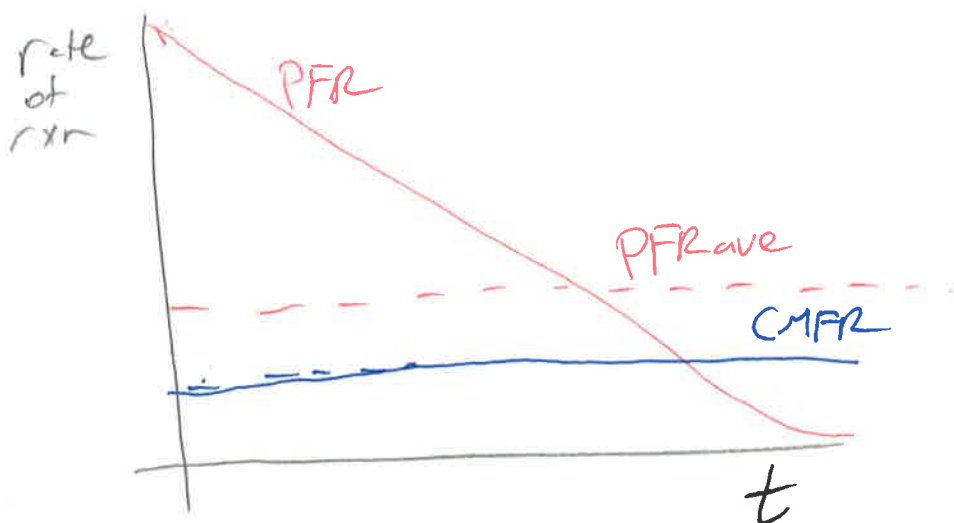
⑧ © For 1st order rxns

5 pts extra credit.

$$\text{HRT}_{\text{PFR}} < \text{HRT}_{\text{CMFR}}$$

Since the ~~concentration~~^{influent} of contaminant mixes throughout the CMFR as soon as the contaminant enters the reactor, the average rate of reaction in the CMFR is lower than in a PFR.

The influent contaminant in a PFR, on the other hand, does not mix with the fluid in the reactor, resulting in a higher reaction rate (compared to the CMFR) throughout more than half of the reactor.



If the rxn rate in PFR is higher, this results in a lower HRT (compared to CMFR) needed for the same performance! $\rightarrow \text{HRT}_{\text{PFR}} < \text{HRT}_{\text{CMFR}}$