

① Given: $BOD_u = 4 \text{ mg/L} = L_0$

$$k = 0.3 \text{ day}^{-1}$$

$$t = 5 \text{ days}$$

$$y(5) = L_0(1 - e^{-kt})$$

$$y(5) = 4 \text{ mg/L} (1 - e^{-0.3 \times 5}) = 3.1 \text{ mg/L}$$

12 pt

6 pt Oxygen used = 3.1 mg/L

$$BOD \text{ remaining, } L(5) = L_0 - y(5) = 4 - 3.1 = 0.9 \text{ mg/L}$$

6 pt BOD remaining = 0.9 mg/L

②

$$BOD_5 = 10 \text{ mg/L (full strength)} \quad DO_{in} = 8 \text{ mg/L}$$

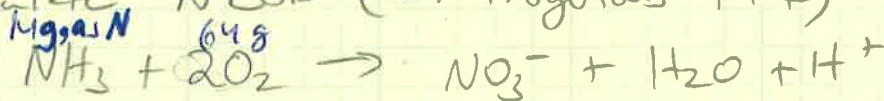
$$NH_3 - N = 2 \text{ mg/L (full strength)}$$

Diluted by 10

$$BOD_5 = \frac{10}{10} = 1 \text{ mg/L (diluted, in bottle)}$$

$$NH_3 - N = \frac{2}{10} = 0.2 \text{ mg/L (diluted, in bottle)}$$

③ Calculate NBOD (= nitrogenous ThOD)



$$NBOD = 0.2 \text{ mg/L } NH_3 - N \times \frac{64g O_2}{14g NH_3 - N} = 0.9 \text{ mg/L (4 pt)}$$

Tot

(b) Calculate BOD_u or L_0 in bottle

(option 1) $BOD_5 = 1 \text{ mg/L} = y(5)$

$$y = L_0 (1 - e^{-kt}) , \text{ Assume } k = 0.25 \text{ day}^{-1}$$

$$L_0 = \frac{y}{1 - e^{-kt}} = \frac{1 \text{ mg/L}}{1 - e^{-0.25 \times 5}} = 1.4 \text{ mg/L} \quad (4 \text{ pt})$$

After a very long time, $L_0(BOD_u)$ & NBOD will be consumed, so

$$O_2 \text{ consumed} = L_0 + NBOD = 1.4 + 0.9 = 2.3 \text{ mg/L}$$

$$\begin{aligned} DO_{\text{final}} &= DO_{\text{in}} - O_2 \text{ consumed} = \\ &= 8 \text{ mg/L} - 2.3 \text{ mg/L} = 5.7 \text{ mg/L} \end{aligned}$$

12 pt

$$DO_{\text{final}} = 5.7 \text{ mg/L}$$

for $k = 0.2 - 0.3 \text{ day}^{-1}$

$$DO_{\text{final}} = 5.5 - 5.8$$

(option 2)

Calculate NBOD, same as in option 1:

$$NBOD = 0.9 \text{ mg/L}$$

To calculate (estimate) BOD_u (L_0), assume

$$BOD_5 = \frac{2}{3} BOD_u$$

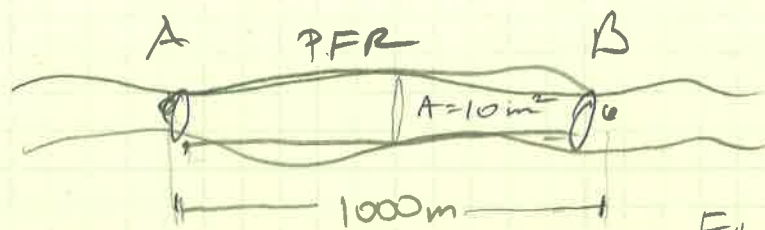
$$BOD_u = 1.5 BOD_5 = 1.5 \times 1 = 1.5 \text{ mg/L}$$

$$DO \text{ consumed} = 1.5 + 0.9 = 2.4 \text{ mg/L}$$

$$DO_{\text{final}} = DO_{\text{in}} - DO_{\text{cons}} = 8 - 2.4 = 5.6$$

$$DO_{\text{final}} = 5.6 \text{ mg/L}$$

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For each sample
 $Y_5 = 2(1 - e^{-k_{1.65} \times 5})$

Find k_r

At point A

$\text{BOD}_5 = 7.2 \text{ mg/L}$

$$L_{0A} = \frac{Y_5}{1 - e^{-k_{1.65} \times 5}}$$

$$L_A = \frac{7.2}{1 - e^{-k_{1.65} \times 5}}$$

In the Lab

At point B

$\text{BOD}_5 = 3.5 \text{ mg/L}$

$$L_{0B} = \frac{Y_5}{1 - e^{-k_{1.65} \times 5}}$$

$$L_B = \frac{3.9}{1 - e^{-k_{1.65} \times 5}}$$

$$\frac{L_B}{L_A} = \frac{\frac{3.9}{1 - e^{-k_{1.65} \times 5}}}{\frac{7.2}{1 - e^{-k_{1.65} \times 5}}} = \frac{3.9}{7.2}$$

In the river

PFR $\Rightarrow L_B = L_A e^{-k_r t}$, $t = \theta = \frac{V}{Q} = \frac{10,000 \text{ m}^3}{\frac{100 \text{ m}^3}{\text{h}}} = 100 \text{ h}$

$$e^{-k_r t} = \frac{L_B}{L_A}$$

$$-k_r t = \ln\left(\frac{L_B}{L_A}\right)$$

Mpt

$$k_r = \frac{-\ln\left(\frac{L_B}{L_A}\right)}{t} = \frac{-\ln\left(\frac{3.9}{7.2}\right)}{100 \text{ h} \times \frac{1 \text{ day}}{24 \text{ h}}}$$

$$k_r = 0.15 \text{ day}^{-1}$$

4

@ 20°C

$$y_5 = 4.15 \text{ mg/L}$$

$$L_0 = \frac{y_5^{20}}{1 - e^{-k_{20} \times 5}}$$

@ 35°C

$$y_5 = 6.56 \text{ mg/L}$$

$$L_0 = \frac{y_5^{35}}{1 - e^{-k_{35} \times 5}}$$

* L_0 not dependent on temperature

$$k_T = k_{20} \Theta^{\frac{T-20}{35-20}}$$

$$k_{35} = k_{20} \Theta$$

$$k_{35} = k_{20} \cdot 1.05^{15}$$

$$k_{35} = 2.08 k_{20}$$

$$L_0 = \frac{4.15}{1 - e^{-5k}}$$

$$L_0 = \frac{6.56}{1 - e^{-5 \times 2.08k}}$$

$$\frac{4.15}{1 - e^{-5k}} = \frac{6.56}{1 - e^{-5 \times 2.08k}}$$

10 pt if get this far!

Solve for k using solver

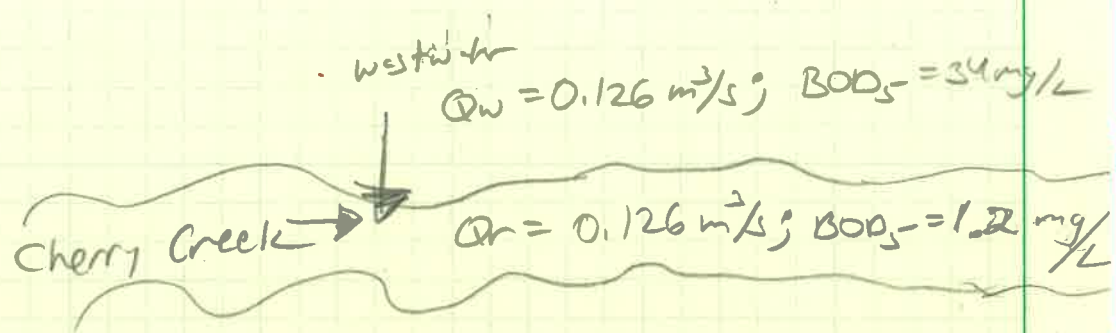
$$k = 0.12 \text{ day}^{-1}$$

$$L_0 = 9.2 \text{ mg/L}$$

12 pt

$$\boxed{\text{BOD}_u = 9.2 \text{ mg/L}}$$

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Calculate L_o after mixing $\equiv L_m$

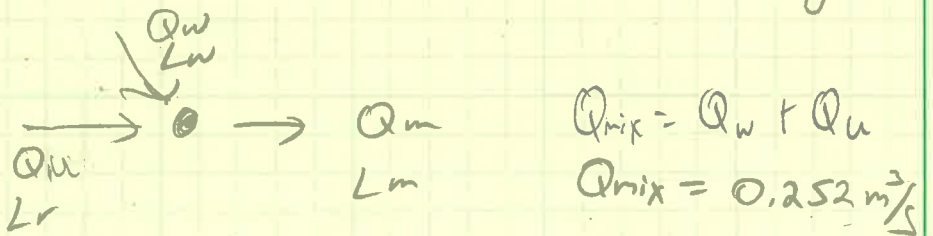
First Calculate L_o for the river $\equiv L_r$ $k_r = 0.09 \text{ d}^{-1}$

$$L_r = \frac{Y_5}{1 - e^{-5k_r}} = \frac{1.2}{1 - e^{-5 \times 0.09}} = 3.31 \text{ mg/L } 4 \text{ pt}$$

Similarly, calculate L_o for the wastewater $\equiv L_w$ $k_w = 0.222$

$$L_w = \frac{Y_5}{1 - e^{-5k_w}} = \frac{34}{1 - e^{-5 \times 0.222}} = 50.7 \text{ mg/L } 4 \text{ pt}$$

Conduct Mass Balance at point of mixing



$$Q_m = Q_w + Q_r$$

4 pt

$$Q_m \times L_m = Q_r L_r + Q_w L_w$$

$$L_m = \frac{Q_r L_r + Q_w L_w}{Q_w} = \frac{0.126 \times 3.31 + 0.126 \times 50.7}{2 \times 0.126}$$

$$L_m = 27 \text{ mg/L}$$

12 pt

⑥ ① Wastewater Meat $k_{20} = 0.2 \text{ dy}^{-1}$
 $\text{BOD}_5 = 2100 \text{ g/m}^3$

$$y(t) = L_0(1 - e^{-kt})$$

$$\theta = 1.05$$

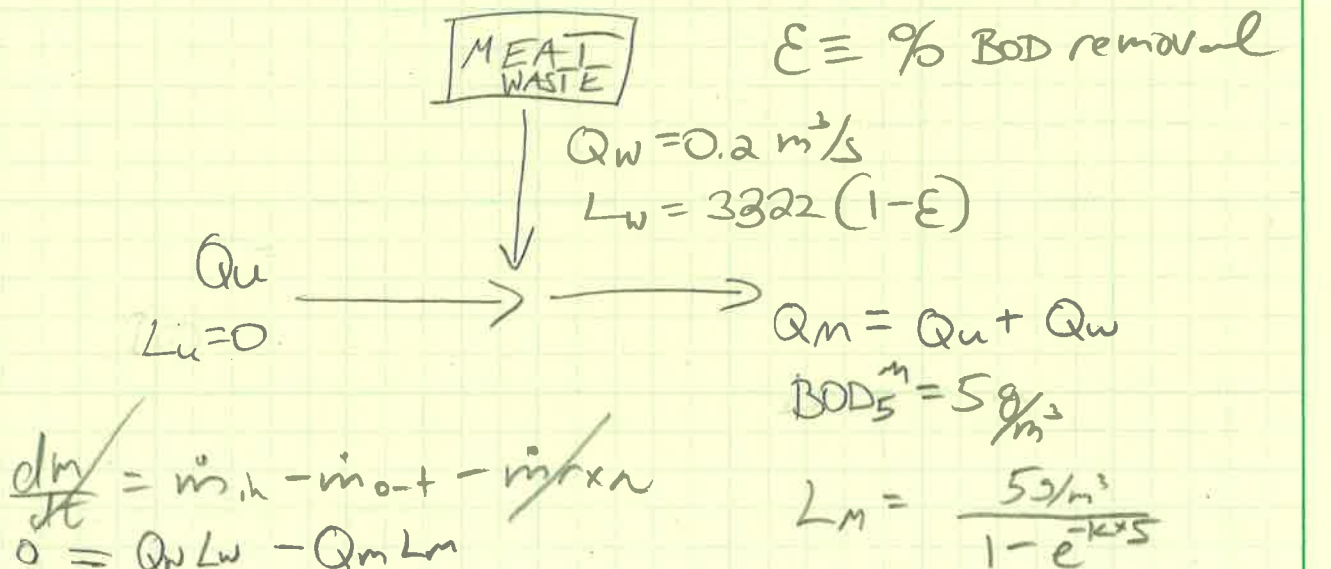
$$L_0 = \frac{2100}{1 - e^{-0.2 \times 5}} = 3322 \text{ g/m}^3$$

② Calculate k values for summer & winter =

$$\text{Summer} \rightarrow k = k_{20} \theta^{\frac{26-20}{10}} = 0.2 \times 1.05^6 = 0.268 \text{ dy}^{-1}$$

$$\text{Winter} \rightarrow k = k_{20} \theta^{\frac{6-20}{10}} = 0.2 \times 1.05^{-14} = 0.101 \text{ dy}^{-1}$$

③ Mass Balance



④ Summer $Q_m = Q_u + Q_w = 175 + 0.2 = 175.2 \text{ m}^3/\text{s}$

$$L_m = \frac{0.5}{1 - e^{-0.268 \times 5}} = 0.68 \text{ mg/L}$$

$$0 = Q_w L_w - Q_m L_m$$

$$L_w = \frac{Q_m L_m}{Q_w} = \frac{175.2 \times 0.68}{0.2} = 593 \text{ g/m}^3$$

$$L_w = 593 = 3322(1-E)$$

$$E = \frac{3322 - 593}{3322} \times 100\% = \underline{82\%}$$

⑤ Winter

$$Q_m = Q_u + Q_w = 95 + 0.2 = 95.2 \text{ m}^3/\text{s}$$

$$L_m = \frac{0.5}{1 - e^{-0.1 \times 5}} = 1.3 \text{ g/m}^3$$

$$L_w = \frac{Q_m L_m}{Q_w} = \frac{95.2 \times 1.3}{0.2} = 605 \text{ g/m}^3$$

$$L_w = 605 = 3322(1-E)$$

$$E = \frac{3322 - 605}{3322} \times 100\% = 82\%$$

$$\text{Percent Removal} = 82\%$$

14.5

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DAF only : Every day - 33m^3 , Solids conc 3.8%

w/ Belt Filter Press : V_{new} , Solids : 24%

Mass of solids stays the same.

$$C_1 V_1 = C_2 V_2$$

$$3.8\% \times 33\text{m}^3 = 24\% \times V_{\text{new}}$$

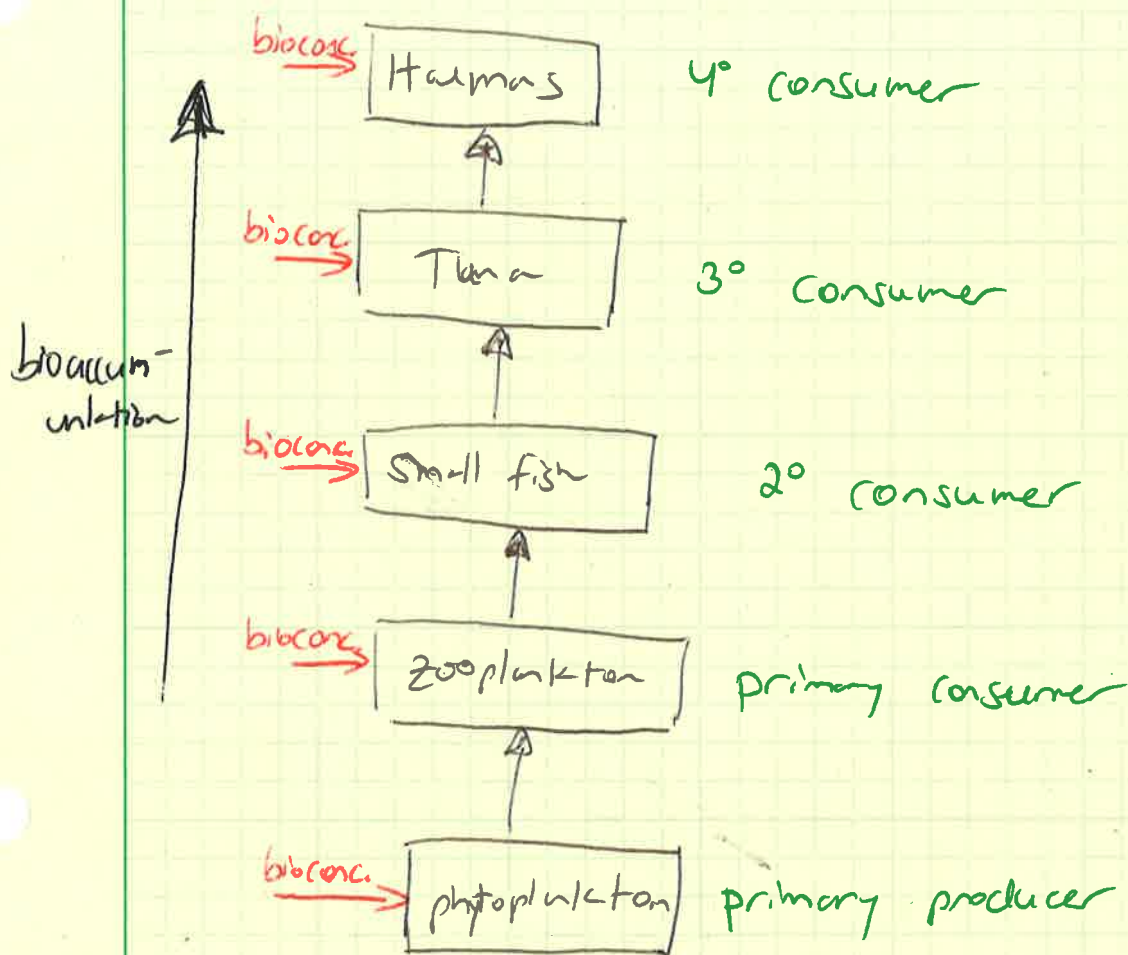
$$V_{\text{new}} = 5.22 \text{ m}^3 \text{ per day}$$

$$V_{\text{old}} = 33\text{m}^3/\text{day} \times \frac{365 \text{ days}}{1} = 12,045 \text{ m}^3/\text{y}$$

$$V_{\text{new}} = 5.2 \text{ m}^3/\text{day} \times \frac{365 \text{ days}}{1} = 1907 \text{ m}^3/\text{y}$$

$$V_{\text{sludge savings per year}} = 10,138 \text{ m}^3$$

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Biocentration is common to all species. Bioaccumulation increases as you move up the food chain.