

ONE HOUR THIRTY MINUTES

A list of constants is enclosed.

UNIVERSITY OF MANCHESTER

Introduction to Quantum Mechanics

99th January 2022, ?..? p.m. - ?..? p.m.

Answer **ALL** parts of question 1 and **TWO** other questions

Electronic calculators may be used, provided that they cannot store text.

The numbers are given as a guide to the relative weights of the different parts of each question.

Formula Sheet

In plane polar coordinates, (r, ϕ) , the Laplacian operator is given by

$$\nabla^2 = \frac{\partial^2}{\partial r^2} + \frac{1}{r} \frac{\partial}{\partial r} + \frac{1}{r^2} \frac{\partial^2}{\partial \phi^2},$$

and the angular momentum operator is given by

$$\hat{L}_z = -i\hbar \frac{\partial}{\partial \phi}.$$

In spherical polar coordinates, (r, θ, ϕ) , the Laplacian operator is given by

$$\nabla^2 = \frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} + \frac{1}{r^2} \frac{\partial^2}{\partial \theta^2} + \frac{1}{r^2} \frac{\cos \theta}{\sin \theta} \frac{\partial}{\partial \theta} + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \phi^2},$$

and the angular momentum operators are given by

$$\hat{L}^2 = -\hbar^2 \left(\frac{\partial^2}{\partial \theta^2} + \frac{\cos \theta}{\sin \theta} \frac{\partial}{\partial \theta} + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right),$$

$$\hat{L}_z = -i\hbar \frac{\partial}{\partial \phi}.$$

1. a) i) The normalized wavefunction of a particle moving in one dimension is Ψ . Consider an operator \hat{A} associated with a physical measurement, A .
Write down the definition of the expectation value of \hat{A} .
Briefly describe the physical meaning of this value and how it is related to the outcomes of particular measurements of A .
[5 marks]
- ii) Write down the definition of the Hermitian conjugate, \hat{A}^\dagger , of the operator \hat{A} .
Show that this definition results in the relation $(\hat{A}\hat{B})^\dagger = \hat{B}^\dagger\hat{A}^\dagger$.
[5 marks]
- b) A quantum rotor has a moment of inertia of $5 \times 10^{-48} \text{ kg m}^2$. What are the values of the three lowest rotational energy eigenvalues, in eV? What is the degeneracy of each of these three energy levels?
[5 marks]
- c) A helium atom has states in which each of the two electrons can be in $1s$, $2s$ or $2p$ orbitals.
Explain how the occupation of these states is affected by the spins of the two electrons.
Explain why the ground state of helium has a total electron spin of $S = 0$.
Explain why the first excited state is $(1s)(2s)$ rather than $(1s)(2p)$.
[5 marks]
- d) Write down the atomic ground-state electronic configurations of titanium ($Z = 22$) and nickel ($Z = 28$). Determine the total S , L and J of each of their ground states and write down the corresponding spectroscopic term symbols.
[5 marks]

2. a) Consider a 1D simple quantum harmonic oscillator of classical angular frequency ω .

i) Draw on one sketch, the potential energy function, energy levels, and wave-functions of the three lowest energy states.

[8 marks]

ii) Calculate the value of the zero-point energy of the oscillator if its classical angular frequency is $2.0 \times 10^{13} \text{ s}^{-1}$.

[2 marks]

iii) If the particle in the quantum harmonic oscillator is charged, transitions between energy levels can involve the absorption or emission of a single photon.

Explain how the frequency of this photon is related to the classical frequency of the oscillator.

[3 marks]

iv) In a carbon monoxide (CO) molecule, the effective spring constant of the covalent bond has a value of 1860 N m^{-1} . Estimate the first excitation energy for vibrational motion of a CO molecule.

You may take the atomic masses of carbon and oxygen to be 12.0 and 16.0 respectively.

[5 marks]

b) Now consider a 2D simple quantum harmonic oscillator of classical angular frequency ω .

i) Write down the general form of a single-valued eigenfunction, $\Phi(\phi)$, of the angular momentum operator \hat{L}_z ,

$$\hat{L}_z \Phi(\phi) = L_z \Phi(\phi),$$

where ϕ is the polar angle, and state the corresponding value of L_z .

[2 marks]

ii) The following are all energy eigenfunctions of the 2D quantum harmonic oscillator with the same energy, $3\hbar\omega$:

$$\psi_{2,0} = A \left(2 \frac{x^2}{a^2} - 1 \right) e^{-(x^2+y^2)/2a^2}, \quad \psi_{0,2} = A \left(2 \frac{y^2}{a^2} - 1 \right) e^{-(x^2+y^2)/2a^2},$$

$$\psi_{1,1} = B \frac{xy}{a^2} e^{-(x^2+y^2)/2a^2},$$

where A , B and a are constants. By writing the Cartesian coordinates x and y in plane polar coordinates, show that these states are not eigenfunctions of angular momentum.

Show that they can all be written as superpositions of states with $L_z = 2\hbar$, 0 and $-2\hbar$.

[5 marks]

3. a) For a particle of mass m moving in one dimension, write down the definition of the momentum operator \hat{p}_x and derive the kinetic energy operator \hat{T} .
The particle is moving in a region of free space with a wavefunction

$$\psi(x) = A \cos kx,$$

where A and k are constants. Show that $\psi(x)$ is not an eigenfunction of \hat{p}_x but is an eigenfunction of \hat{T} .

What are the possible outcomes of a measurement of the particle's momentum?

[9 marks]

- b) The magnetic energy operator for an electron in a magnetic field B , pointing in the z direction, is given by

$$\hat{V}_{mag} = \frac{eB}{2m_e} (\hat{L}_z + g\hat{S}_z),$$

where g is the gyromagnetic ratio of the electron, which we will take to be equal to 2, and the other symbols have their usual meanings.

- i) When the magnetic field is strong, the quantum numbers m_ℓ and m_s can be assumed to be good quantum numbers. Use this to calculate how many energy levels the $2p$ orbital of hydrogen is split into, in a strong magnetic field.

Calculate the spacing between these energy levels. You may state your answer in terms of Bohr magnetons, or electronvolts.

[4 marks]

- ii) When the magnetic field is weak, the quantum numbers m_ℓ and m_s are not good quantum numbers. Explain what this means and why it is the case.

What are the good quantum numbers in this case?

[4 marks]

- iii) When the magnetic field is weak, the magnetic energy operator is equivalent to

$$\hat{V}_{mag} = \frac{eB}{2m_e} g_L \hat{J}_z,$$

where g_L is the Landé factor,

$$g_L = 1 + \frac{j(j+1) + s(s+1) - \ell(\ell+1)}{2j(j+1)}.$$

For the $2p$ orbital of hydrogen, what values of j are allowed?

For each of these j values, calculate the spacing between the magnetic energy levels. You may state your answer in terms of Bohr magnetons, or electronvolts.

[6 marks]

- iv) How would you decide if a given value of magnetic field would be considered “strong” or “weak” in this context?

You do not need to provide a quantitative calculation.

[2 marks]

4. a) Consider the following wavefunctions of a hydrogen atom

$$\begin{aligned}\psi_A(r, \theta, \phi) &= A e^{-r/a_0}, \\ \psi_B(r, \theta, \phi) &= B \left(1 - \frac{r}{2a_0}\right) e^{-r/2a_0}, \\ \psi_C(r, \theta, \phi) &= C \left(\frac{r}{a_0}\right)^3 e^{-r/4a_0} \sin^2 \theta \cos \theta e^{2i\phi},\end{aligned}$$

where A , B , C and a_0 are constants.

- i) Deduce by inspection the values of the quantum numbers (n, ℓ, m_ℓ) for each of these eigenfunctions. [4 marks]
- ii) Sketch the radial probability distribution $P(r)$ corresponding to the wavefunction ψ_B . [4 marks]
- iii) Show that the states ψ_A , ψ_B and ψ_C are all orthogonal to one another. *You may use the result $\int_0^\infty x^n e^{-x} dx = n!$ for integer $n \geq 0$.* [4 marks]

- b) i) Write down the quantum operators for the angular momentum components \hat{L}_x , \hat{L}_y and \hat{L}_z , in terms of \hat{x} and \hat{p}_x , etc. [3 marks]

- ii) Given the following commutation relation,

$$[\hat{L}_x, \hat{L}_y] = i\hbar \hat{L}_z,$$

use the cyclic rule to write down the other two commutation relations between the angular momentum component operators. Hence show that

$$[\hat{L}_z, \hat{L}^+] = \hbar \hat{L}^+, \quad (1)$$

where $\hat{L}^+ = \hat{L}_x + i\hat{L}_y$.

[5 marks]

- iii) Use Eq. (1) to show that the wavefunction defined by

$$\hat{L}^+ \psi_C,$$

where ψ_C is given in part (a), is also an eigenfunction of \hat{L}_z . Find its eigenvalue. [5 marks]

END OF EXAMINATION PAPER