

## Statistical Physics: Workshop Problems 6

- (1) (a) Explain in what circumstances Maxwell-Boltzmann, Fermi-Dirac and Bose-Einstein statistics are appropriate making reference to the number of particles that can occupy energy states and their degeneracies.
- (b) Give a physical discussion on why differences in the three statistics become unimportant at high temperatures.
- (c) At what temperature (order of magnitude) will quantum statistics have to be used for a system containing  $10^{18}/\text{cm}^3$  of neutrons?
- (2) Consider a material where each of the constituent atoms may occupy one of two sites. For each atom the sites have energy  $\epsilon_i$  and  $-\epsilon_i$ , where  $i$  labels the atom.
  - (a) If the energy levels are the same two values for all of the atoms ( $\epsilon$  and  $-\epsilon$  say) calculate the contribution of the atoms to the heat capacity of the material.
  - (b) Now return to the case where the energies are atom dependent ( $\epsilon_i$  and  $-\epsilon_i$ ). Find the behaviour of the low temperature heat capacity (i.e.  $k_B T \ll \epsilon$ ). Express the result as both a sum and an integral (do not attempt to evaluate either).

- (3) The quantum energy levels of a rigid rotor of mass  $m$  and length  $a$  are

$$\epsilon_j = j(j+1) \frac{h^2}{8\pi^2 m a^2}$$

where  $j = 0, 1, 2, \dots$ . The degeneracy of each level is  $g_j = 2j + 1$ .

- (a) Find the general expression for the partition function and show that at high temperatures it can be approximated by an integral (and evaluate the integral).
- (b) Evaluate the high temperature energy and heat capacity.
- (c) Find a low temperature approximation for the partition function and hence the low temperature internal energy and heat capacity.
- (4) The electronic energy levels of a hydrogen atom are given by  $E_n = -E_0/n^2$  where  $E_0$  is a constant and  $n = 1, 2, 3, \dots$  and the degeneracy of the states are  $g_n = 2n^2$ .
  - (a) Write down the expression for the electronic partition function for a single isolated hydrogen atom at temperature  $T$  (do not try to evaluate the sum). Does the expression diverge for  $T = 0$  and what about for  $T \neq 0$ ?
  - (b) Is this divergence caused by the chosen zero of energy? [Add some constant  $E'$  onto the expressions for energy levels].
  - (c) Calculate average thermal energy,

$$\langle E \rangle = \frac{\sum_i E_i e^{-E_i/k_B T}}{Z},$$

of the system.

- (d) We seem to have non-physical results. Can you explain what is going on? Consider what happens when an atom is confined to a large finite volume  $L^3$  in a quantum calculation of the full partition function (you don't need this full calculation, just think degeneracies and energy levels).