

**Thermodynamics – Weekly Problem, Th. 1****Publication: 11<sup>th</sup> Oct****Completion: 21<sup>st</sup> Oct**

- a) The first part of the problem gives you chance to look at the properties of functions which may be either exact or inexact differentials.

- i) A gas has equation of state given by

$$pV = RT + \frac{aT^2}{V},$$

where  $a$  is some constant. Determine the total derivative of the pressure,  $dp$ .

- ii) Show that the following function is in fact exact,

$$z = x^9 y^3 + 5x^{-3} y^4.$$

- iii) Determine whether the differentials given below are exact or inexact.

$$da = \frac{4}{3} b^3 \exp(5c) db + \frac{5b^4}{3} \exp(5c) dc,$$

$$dl = 12p^2 T \sin T dp - 4p^3 T \cos T dT.$$

[6 marks]

- b) Two bodies having temperature independent heat capacities  $C_1$  and  $C_2$  are initially at temperatures  $T_1$  and  $T_2$ . They are brought into contact through a diathermal wall and allowed to reach an equilibrium state.

- i) Show that the final temperature is in general given by

$$T_f = \frac{C_1 T_1 + C_2 T_2}{C_1 + C_2}.$$

- ii) If  $C_1 \gg C_2$ , show that the final temperature can be approximated by

$$T_f \approx T_1 + \frac{C_2}{C_1} (T_2 - T_1).$$

Hence explain why the material of heat capacity  $C_2$  would be appropriate to use to make a thermometer.

[4 marks]