

Nuclear and Particle Physics

Workshop 1

1. For heavy, instable atomic nuclei α -decay is a comparatively common phenomenon, but proton emission is virtually non-existent. What is the reason for this observation?

- a) Based on the semi-empirical mass formula, calculate the energy release for α -decay (the pairing term can be neglected $\delta = 0$)

$$E_\alpha = M(A, Z) - M(A - 4, Z - 2) - M(4, 2).$$

- b) Show that an approximate expression for E_α for $Z = A/2$ and $\frac{1}{A} \ll 1$ is given by

$$E_\alpha \approx -4a_V + \frac{8}{3} \frac{a_s}{A^{1/3}} + a_c \frac{5}{3} A^{2/3} + B(4, 2)$$

- c) Use the numerical values $B(4, 2) = 28.3 \text{ MeV}$ and

$$\begin{aligned} a_V &= 15.84 \text{ MeV}, & a_s &= 18.33 \text{ MeV}, \\ a_c &= 0.71 \text{ MeV}, & a_a &= 92.80 \text{ MeV}. \end{aligned}$$

to show that $E_\alpha \geq 0$ for $A \gtrsim 93$. Therefore, only for heavy nuclei the energy release is positive and α -decay is possible. Note that $Z = A/2$ is not a good approximation for heavy nuclei. The more realistic value derived in the homework exercise $Z = A/(2 + 0.015A^{2/3})$ yields $E_\alpha \geq 0$ for $A \gtrsim 150$.

- d) Calculate the energy release for proton emission

$$E_P = M(A, Z) - M(A - 1, Z - 1) - M(1, 1).$$

and show that in the same approximation as above, $Z = A/2$ and $\frac{1}{A} \ll 1$,

$$E_P = -a_V + \frac{2}{3} \frac{a_s}{A^{1/3}} + a_c \frac{11}{12} A^{2/3},$$

which is always smaller than E_α for nuclei with $A \gtrsim 93$. Proton emission is typically energetically less favourable than α -decay for nuclei with $E_\alpha \geq 0$.

2. A Uranium nucleus ${}^{236}_{92}\text{U}$ can break apart through spontaneous fission. Assume that it breaks in two roughly equal parts. Using that the electrostatic energy of a sphere with uniformly distributed charge Q is given by $E_{\text{stat}} = \frac{3}{5} \frac{Q^2}{4\pi R}$ and the radius of both the Uranium atom and the fission products can be described by $R = 1.2 \cdot 10^{-13} A^{1/3}$, calculate the energy released in ${}^{236}_{92}\text{U}$ fission.