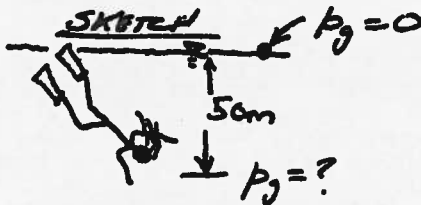


CE 3305 Water Systems Design  
Quiz 4  
Spring 2014

1. Some divers can go as deep as 50 meters on a breath-hold dive<sup>1</sup>. What is the gage pressure at this depth in fresh water? Assume the temperature is 20°C.



KNOWN

$$\Delta h = 50 \text{ m}$$

$$\text{FRESH WATER } \rho = 9800 \text{ N/m}^3$$

$$p_0 = 0 \text{ gage}$$

GOVERNING EQUATIONS

$$p = \rho g h$$

UNKNOWN

$$p_{g0} = ?$$

SOLUTION

APPLY  $p = \rho g h$ ; CONVERT UNITS TO Pa OR kPa

$$p = \underbrace{p_0}_{0 \text{ gage}} + \rho g h = \frac{9800 \text{ N}}{\text{m}^2} (50 \text{ m}) = 490,000 \frac{\text{N}}{\text{m}^2} = 490 \text{ kPa}$$

<sup>1</sup> A breath-hold dive is when the diver takes a breath at the surface and then holds their breath during the dive. Well trained divers can stay submerged nearly 5 minutes, more if they hyperventilate before the dive. To get to 50 meters, they usually use a weighted sled. The pressure in their lungs is at atmospheric, unlike SCUBA where the pressure is the same as the surrounding water.