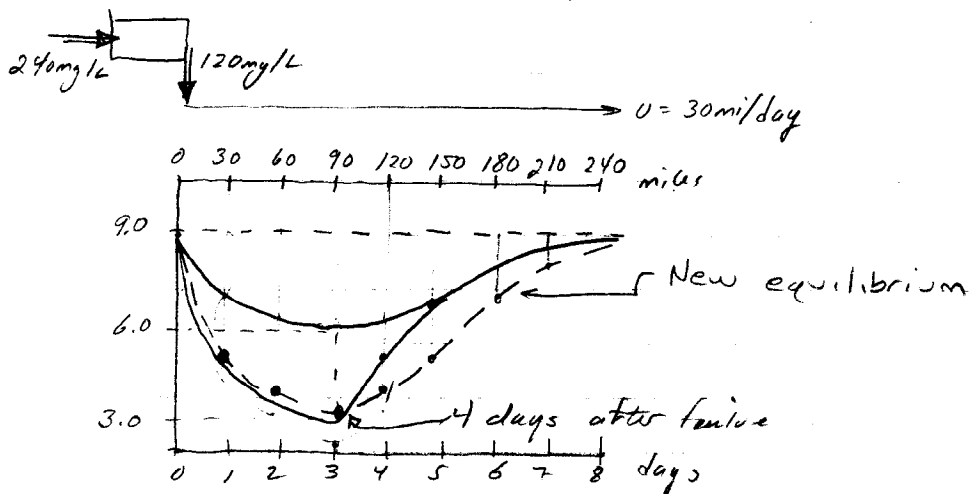


CIVE 3331

PROBLEM 5-26

Raw sewage BOD = 240 mg/L sent to plant that achieves 50% removal. Effluent produces Oxygen sag curve shown.



- a) Assume plant fails, draw new DO sag curve after long time (equilibrium)

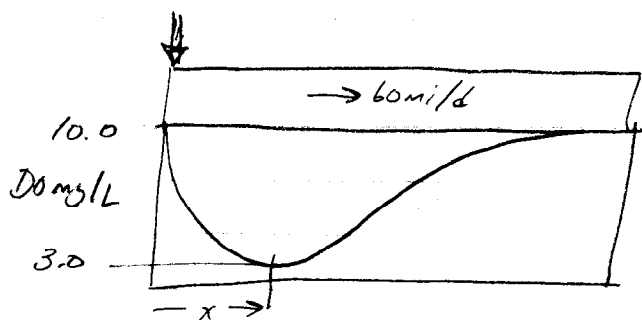
$$D(x) = \frac{k_d(L_0)}{k_r - k_d} (e^{-k_d x/u} - e^{-k_r x/u}) + D_0 e^{-k_r x/u}$$

Only change in equation is value of L_0 , the ultimate BOD. It essentially doubles. Therefore $D(x)$ is doubled.

- b) sketch DO sag curve after only 4 days.

- day 5 & beyond no change (upset has not arrived yet)
- days 1 - 3 are probably near new equilibrium.
- day 4 is between old & new equilibrium.

Suppose only source of BOD is untreated waste from food processing plant. DO sag 3mg/L occurs downstream of plant as shown



$$k_r = 0.8/d \quad k_d = 0.2/d$$

a) by what % should BOD of wastes be reduced to achieve $DO = 5.0 \text{ mg/L}$ everywhere?

$$D = \frac{k_d L_0}{k_r - k_d} (e^{-k_d x/u} - e^{-k_r x/u}) + D_0 e^{-k_r x/u}$$

from sketch we observe that $D_0 = 0$

$$D = \frac{k_d L_0}{k_r - k_d} (e^{-k_d x/u} - e^{-k_r x/u})$$

$$\frac{x_c}{u} = \frac{1}{k_r - k_d} \ln \left\{ \frac{k_r}{k_d} \left(1 - \frac{D_0 (k_r - k_d)}{k_d L_0} \right) \right\} = \frac{1}{0.8 - 0.2} \ln \left(\frac{0.8}{0.2} \right) = 2.31 \text{ day}$$

$$D(x_c) = 7.0 \text{ mg/L} = \frac{(0.2)(L_0)}{(0.6)} (e^{-0.2(2.31)} - e^{-0.8(2.31)})$$

$$7.0 \text{ mg/L} = 0.1575 L_0$$

$$L_0 = \frac{7.0}{0.1575} = 44.4 \text{ mg/L}$$

Goal is to reduce $D(x_c)$ from 7.0 mg/L to 5.0 mg/L

$$L_0 = \frac{5.0}{0.1575} = 31.7 \text{ mg/L}$$

$$\therefore \frac{44.4 - 31.7}{44.4} \times 100\% = 28.4 \quad \% \text{ reduction required}$$

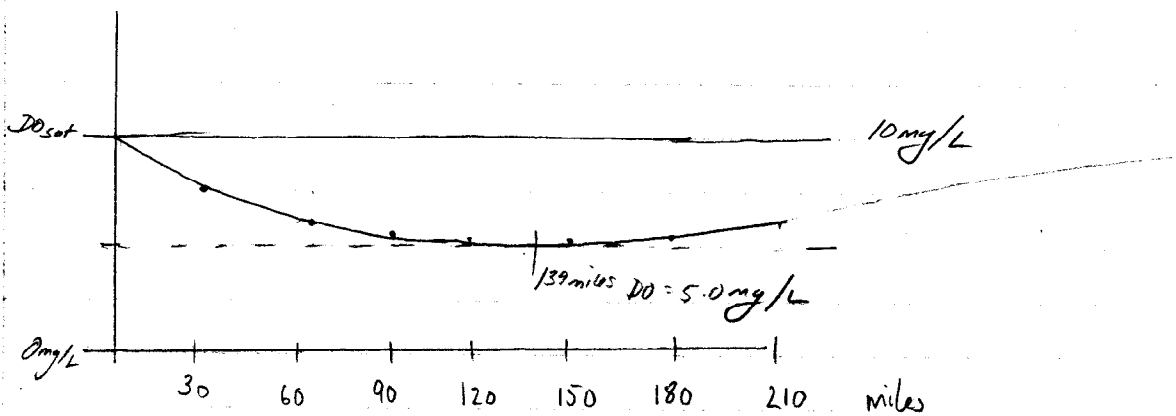
Will a primary plant achieve this reduction? Cannot answer based on information in text.

b) Where does lowest DO occur?

$$\frac{V_0}{U} = 2.31 \text{ days} \quad X_c = U \cdot 2.31 \text{ day} = 60 \text{ mi/day} \cdot 2.31 \text{ day} = 138.6 \text{ miles}$$

c) Already calculated $l_0 = 31.7 \text{ mg/L}$ for 5.0 mg/L DO .

d) Sketch the DO curve



CIVE 3331

PROBLEM 5.31

Find min DO expected.

	Waste	River	
Q	0.3	0.9	m ³ /s
L_0	6.4	7.0	mg/L
DO_0	1.0	6.0	mg/L
k_d		0.2	day ⁻¹
k_r		0.37	day ⁻¹
U		0.65	m/s
DO_{sat}	8.0	8.0	

$$D = \frac{k_d L_0}{k_r - k_d} \left(e^{-k_d x/U} - e^{-k_r x/U} \right) + D_0 e^{-k_r x/U}$$

Need L_0 of mixture

$$L_0 = \frac{Q_W L_{W0} + Q_R L_{R0}}{Q_W + Q_R} = \frac{(0.3)(6.4) + (0.9)(7.0)}{1.2} = 6.85 \text{ mg/L}$$

Need D_0 (initial deficit) of mixture

$$D_0 = DO_{sat} - \frac{Q_W DO_W + Q_R DO_R}{Q_W + Q_R} = 8.0 - \frac{(0.3)(1.0) + (0.9)(6.0)}{1.2} = 3.25 \text{ mg/L}$$

Now find $\frac{x_e}{U}$ from

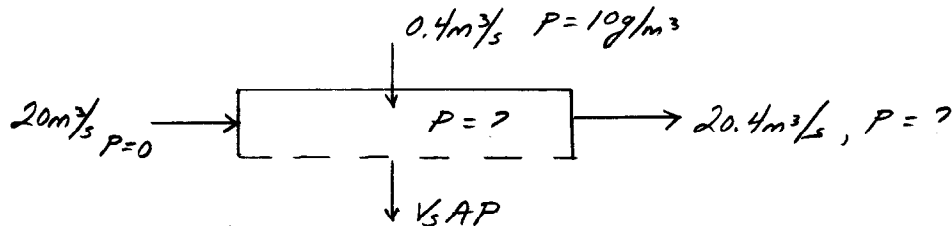
$$\begin{aligned} \frac{x_e}{U} &= \frac{1}{k_r - k_d} \ln \left(\frac{k_r}{k_d} \left(1 - \frac{D_0 (k_r - k_d)}{k_d L_0} \right) \right) \\ &= \frac{1}{(0.37 - 0.2)} \ln \left(\frac{0.37}{0.2} \left(1 - \frac{3.25(0.37 - 0.2)}{0.2(6.85)} \right) \right) = 0.581 \text{ days} \end{aligned}$$

$$x_e = (0.65 \text{ m/s})(86400 \text{ s/d})(0.581 \text{ day}) = 32662 \text{ meters} = 32.6 \text{ km}$$

$$\begin{aligned} \text{Now find } D(x_e) &= \frac{(0.2)(6.85)}{(0.17)} \left[e^{-0.2(0.581 \text{ day})} - e^{-0.37(0.581 \text{ day})} \right] + 3.25 e^{-0.37(0.581)} \\ &= 3.29 \text{ mg/L} \end{aligned}$$

LAKE S.A. = $100 \cdot 10^6 \text{ m}^2$ P SOURCE IS $0.4 \text{ m}^3/\text{s}$ @ 10 mg/L (10 g/m^3)STREAM FLOW $Q = 20 \text{ m}^3/\text{s}$, $P = 0$. $V_s \approx 10 \text{ m/yr}$.

ESTIMATE P CONC.



INFLOW	OUTFLOW	ACCUMULATION
$20(0) + 0.4(10)$	$-(10)(100 \cdot 10^6)P$ $(20.4)P$	$= 0$

$$4 \text{ g/sec} - (10 \text{ m/yr}) \left(\frac{1 \text{ yr}}{365 \text{ d}} \right) \left(\frac{1 \text{ d}}{86400 \text{ sec}} \right) (100 \cdot 10^6 \text{ m}^2) P - 20.4 P = 0$$

$$4 \text{ g/sec} - 31.7 \frac{\text{m}^3}{\text{sec}} P - 20.4 \frac{\text{m}^3}{\text{sec}} P = 0$$

Solve for P $P = \frac{4 \text{ g/sec}}{52.1 \text{ m}^3/\text{sec}} = 0.0767 \text{ g/m}^3$

$$0.0767 \text{ g/m}^3 = \frac{76.7 \text{ mg}}{\text{m}^3} \cdot \frac{1 \text{ m}^3}{1000 \text{ L}} = \underline{\underline{0.0767 \text{ mg/L}}}$$

HOW MUCH TREATMENT TO GET $P = 0.01 \text{ mg/L}$?

$$\frac{x}{52.1} = 0.01 \quad x = 0.521$$

$$0.4(y) = 0.521 \quad y = 1.302 \text{ g/m}^3$$

$$\% \text{ removal} = \frac{10 - 1.302}{10} \times 100 = 87\%$$