

Enabling the Net-Zero Transition: Integrating Complex Multiphysics Efficiently at the Design Optimisation Level

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1. Executive Summary

The role of the chemical industry in modern society is rapidly evolving; on the one hand, it must develop novel technologies based on the new requirements for the energy transition, and on the other hand, it must decarbonise, transform and modernise an existing asset base of chemical process. These existing assets, including hydrocarbon cracking units, hydrogen production plants, ammonia production facilities, etc., rely on fossil-fuel-combustion-based furnaces to supply energy to the chemical process. Radically new technologies are necessary to decarbonise these processes efficiently while maintaining profitability. On top of this, the energy transition unlocks a wide variety of opportunities for the industry, such as carbon capture, electrochemical process (*e.g.*, electrolyzers), synthesising sustainable feedstocks & fuels, and many more. Given the rapidly urgent climate emergency, these developments need to be accelerated.

However, a key bottleneck in designing, developing and optimising these new technologies and chemical processes is aerochemically coupled numerical simulations using detailed, comprehensive (and accurate) chemical kinetic models. Large kinetic models are often required for accuracy (consisting of $\mathcal{O}\{1000\}$ species and $\mathcal{O}\{10,000\}$ reactions). But the price paid is an acute increase in computational overhead, as CPU time scales cubically with the number of species $\mathcal{O}(N_{\text{species}}^3)$. Therefore, this places a heavy burden on accurate, but time-consuming and costly, physical experiments in order to reduce uncertainty and risk.

To address this, this work proposes a new methodology to mitigate this computational bottleneck. We will demonstrate that it is now possible to accelerate reacting flow numerical simulations by two to three orders of magnitude, while accurately capturing the response dynamics of a *detailed* chemical kinetic model and maintaining a high-fidelity 3-D viscous aerothermal fluid flow model. This new methodology is an automated approach that leverages machine learning to uncover preexisting low-dimensional latent spaces hidden within the governing physics.

In this report, this methodology is explained in the context of a new turbomachinery concept that can decarbonise endothermic chemical reaction processes such as steam cracking. This is an ideal test bed because (1) the flowfield is highly complex and there are unique aerochemical interactions, and (2) this is the first time turbomachines have had to be designed to control and optimise chemical reaction dynamics along the bladed flow path; therefore, there is limited experimental data to guide the design process. By accelerating reacting flow simulations (100 to 1000 times) using this new methodology, it will be possible to perform aerodynamic optimisation early in the design cycle to maximise reaction efficiency and minimise coke formation rate (which are key design metrics of the process). To put into context the importance of tuning the chemical reaction through careful aerodynamic design, for a typical hydrocarbon cracking plant (with capacity = 1 million tonnes per year), a 10% improvement in reaction efficiency can increase the annual profits of the plant operator by €90 – €110 million (as of 2024).

To further advance this methodology, we intend to generalise and modularise this numerical tool into a user-friendly platform that can accelerate chemical kinetic modelling for a wide variety of industry-relevant chemical reaction processes ranging from hydrogen production to electrochemical reactions. A numerical tool that can accurately but highly efficiently predict yield and coking distributions in chemical processes will help produce more cost-efficient designs, reduce dependence on costly physical experiments, and ultimately contribute to accelerating the net-zero transition.

2. Introduction and Motivation

Decarbonising high-temperature industrial heat within fossil-fuel-fired furnaces is the “grand challenge” of energy engineering, and is a critical hurdle in the development of a net-zero society. Greenhouse gas emissions from high-temperature heat ($T > 400^{\circ}\text{C}$) alone account for more than

10% of the global total. For low-temperature heat, various mature low-carbon technologies already exist. However, high-temperature sectors have been plagued with techno-economic hurdles. To address this, Fig. 1 showcases the revolutionary new turbomachine called the turbo-reactor

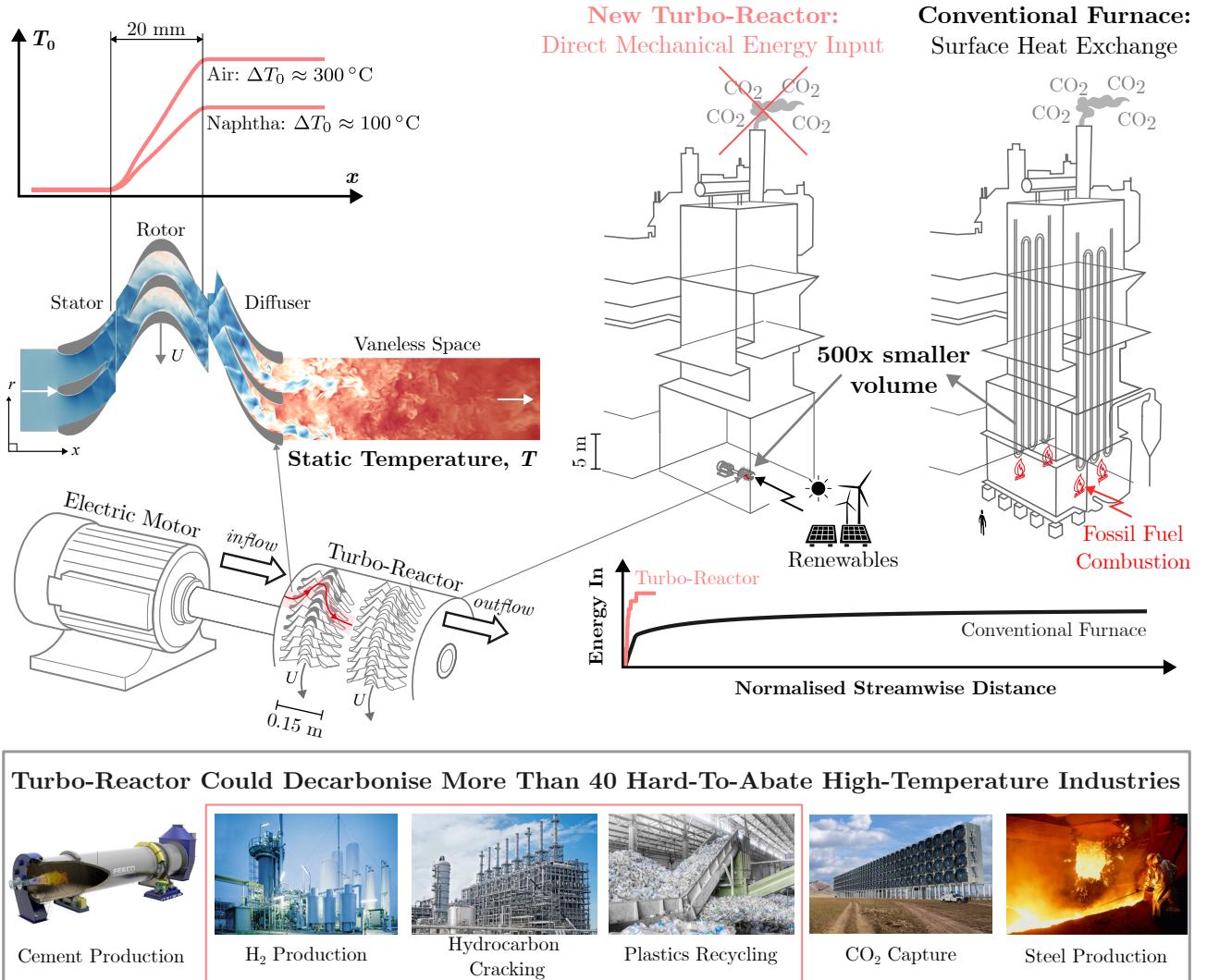


Figure 1: Electrified turbo-reactor could decarbonise many hard-to-abate industrial processes

proposed as an efficient, effective and financially attractive solution for decarbonising over 40 hard-to-abate sectors. By replacing the surface heat exchange energy transfer mechanism within a furnace with direct mechanical energy transfer to the fluid within the turbo-reactor, the power density can increase by 50 to 500 times (Fig. 1), while simultaneously improving the quality of the process. This is enabled by a new turbomachinery principle, in which all mechanical energy transferred to the fluid is converted into kinetic energy and subsequently into internal energy of the fluid, rather than pressure.

This research focuses on endothermic chemical reaction applications of the turbo-reactor

(*e.g.*, hydrocarbon cracking, seen in Fig. 1) as they pose a unique set of new requirements for the turbomachinery design. For the first time in turbomachinery, the working fluid chemically reacts along most of the bladed path. Therefore, to design the aerothermal bladed path to optimise the chemical reaction dynamics, an accelerated aerochemically coupled numerical solver is required. This work introduces a multi-fidelity, Efficiently Coupled Aerochemistry Model (ECAM) guided by machine learning (ML) to routinely bring reacting flow simulations into the early design phases, rather than being “one-off” simulations.

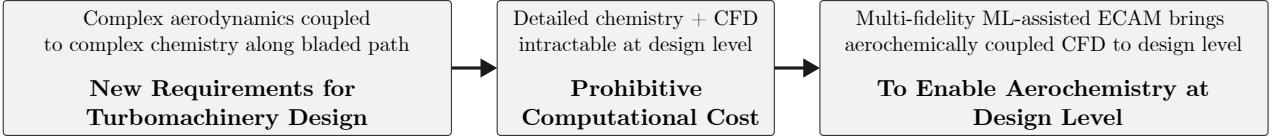
3. Data-Driven Efficiently Coupled Aerochemistry Flow Solver

3.1. Importance

For applications of the turbo-reactor where the fluid undergoes a chemical reaction, in addition to power input requirements, the driving design metric is reaction efficiency rather than aerodynamic efficiency. The reaction efficiency is strongly affected by aerochemical interactions due to the complexity of the flow physics and the similarity of the characteristic fluid and chemical timescales. However, up to now, these rich multiphysics interactions have been ignored due to the prohibitive computational cost of fully-reacting flow simulations, making it challenging to perform design optimisation of the machine and thus meet reaction performance targets. To address this, a highly efficient multi-fidelity ML-assisted ECAM is introduced to mitigate the computational burden of aerochemically coupled computational fluid dynamics (CFD) without sacrificing accuracy (see Fig. 2). Consequently, for the first time in turbomachinery, aerochemically guided optimisation can be incorporated at an early stage of the design, potentially facilitating a step-change improvement in reaction performance.

3.2. Methodology

Figures 2 & 3 outline the new multi-fidelity ML-assisted ECAM methodology to accelerate multiphysics numerical analysis of the turbo-reactor. First, non-reacting CFD of the turbo-reactor establishes a realistic envelope of energy supply profiles into the chemical reaction (see Fig. 2(*a*)). From this envelope, thousands of energy supply profiles are generated and applied to a detailed chemical kinetic model in a 1-D plug-flow reactor (PFR) environment (see Fig. 2(*b*)). These ran-



Training Database Generation for Multi-Fidelity ML-Assisted ROM

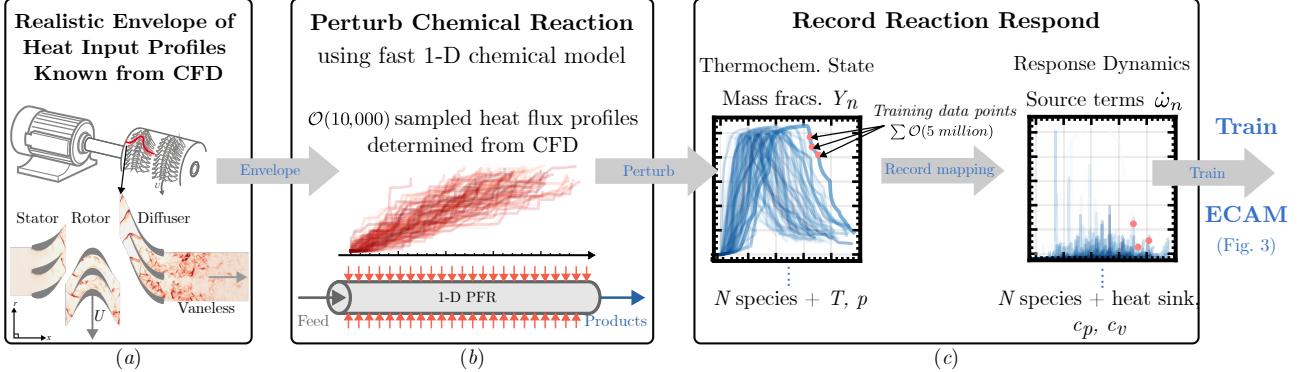


Figure 2: Multi-fidelity ML-assisted ECAM training database generation workflow

domly generated energy supply profiles (within a realistic envelope) lead to a broad spectrum of reaction perturbations for which the thermochemical state (*e.g.*, Y_n & T) and response dynamics (*e.g.*, $\dot{\omega}_n$) can be recorded. This forms a large database for training the ECAM (see Fig. 2(c)) with over 5 million samples. By simplifying the fluid dynamics (*i.e.*, by using a PFR) while retaining detailed kinetics, this methodology provides an elegant solution to rapidly generate a training database that densely and widely covers the relevant region of the thermochemical space in only a few hours.

Subsequently, as shown in Fig. 3, a ML-assisted ECAM is developed using an autoencoder neural network (NN). This NN architecture compresses the high-dimensional thermochemical space ($\mathcal{O}\{200 - 1000\}$ chemical species and $\mathcal{O}\{1000 - 10,000\}$ reactions) into a substantially lower-dimensional latent space ($\mathcal{O}\{3\}$), where a reduced set of “meta” species \tilde{y}_m are mapped to their corresponding source terms $\tilde{\omega}_m$. These source terms represent the dynamic response in the latent space. In effect, the training procedure described above produces a “custom compression algorithm” that uncovers preexisting low-dimensional latent spaces hidden within the governing physics.

One-dimensional verification of the ML-assisted ECAM, presented in Fig. 3, demonstrates accurate predictions compared to full chemical kinetics for both 0-D single-time-step and 1-D multi-time-step scenarios. The accumulation of errors for major and minor species is minimal, ensuring accurate forecasts even over extended time intervals.

Multi-Fidelity ML-Assisted ECAM Methodology

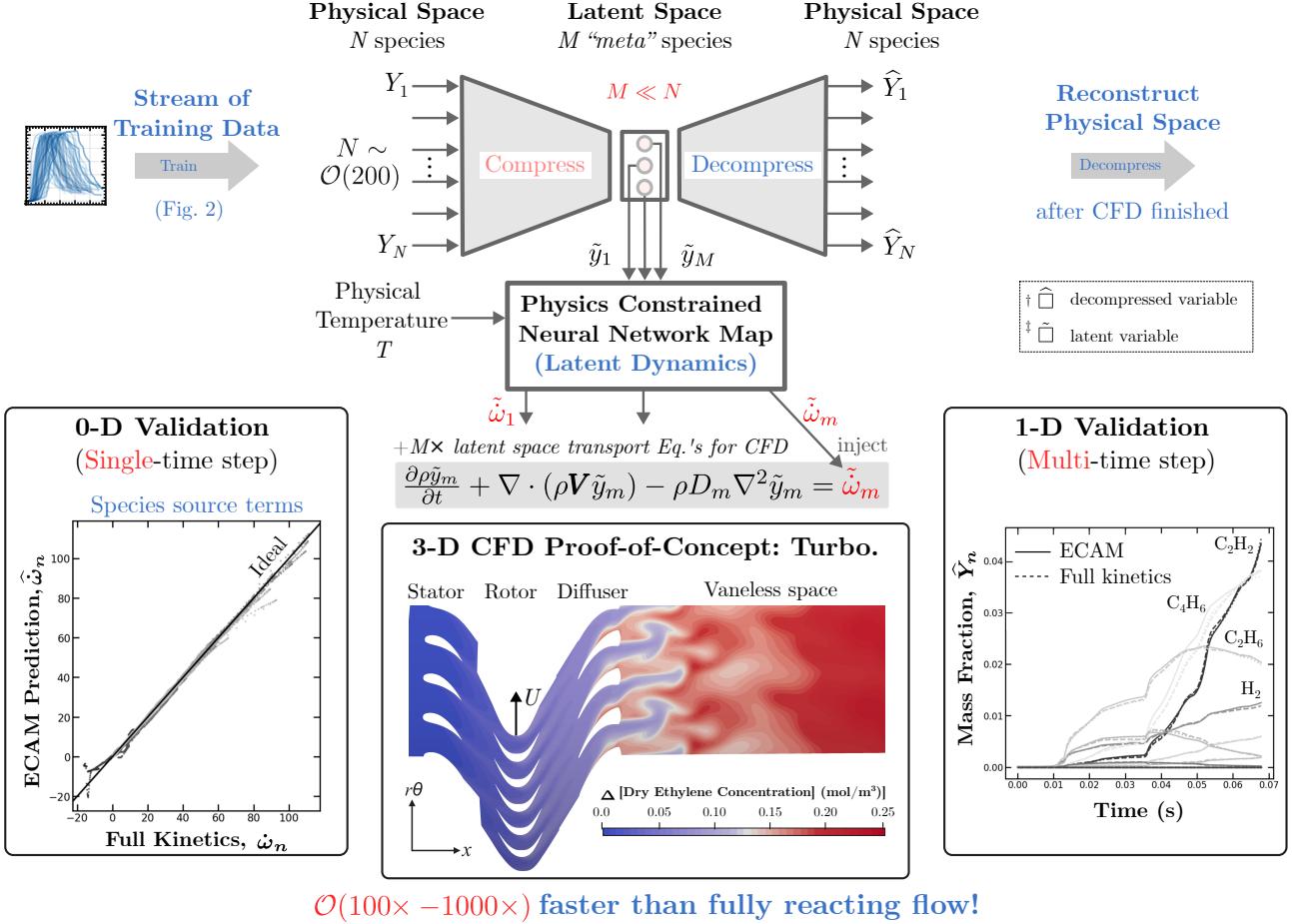


Figure 3: Multi-fidelity ML-assisted ECAM methodology overview

Coupling the ECAM with a 3-D viscous CFD solver entails adding only three additional “meta” species scalar transport equations, where the dynamics are captured through source terms $\tilde{\omega}_m$ mapped by the pre-trained NN (see Fig. 3). A key feature of this methodology is that on the fluid flow modelling side, no new modelling assumptions are made, and its fidelity can range from RANS to high-fidelity LES modelling.

Once the CFD has converged, the high-dimensional physical space, consisting of $\mathcal{O}\{200 - 1000\}$ species mass fractions, can be reconstructed (*i.e.*, decompressed) as shown in Fig. 3. This powerful methodology enables a speed-up of 100 to 1000 times relative to reacting flow CFD with detailed chemical kinetics, while maintaining high-fidelity aerothermal modelling and accurately capturing the dynamics of a detailed kinetic model. This will enable aerochemical optimisation to be brought to the design level for the first time.

3.3. 3-D Viscous Verification and Proof-of-Concept

To demonstrate the feasibility, success, and future potential of this new methodology, numerical predictions of the ECAM are compared with an industry-standard 3-D viscous reacting flow CFD solver FLUENT with a detailed chemical kinetic mechanism. A 5 m long heated square duct is used as a verification test case (see Fig. 5). Crucially, Fig. 4 illustrates that the ECAM can achieve a two-order-of-magnitude speedup relative to a commercial solver with detailed kinetics.

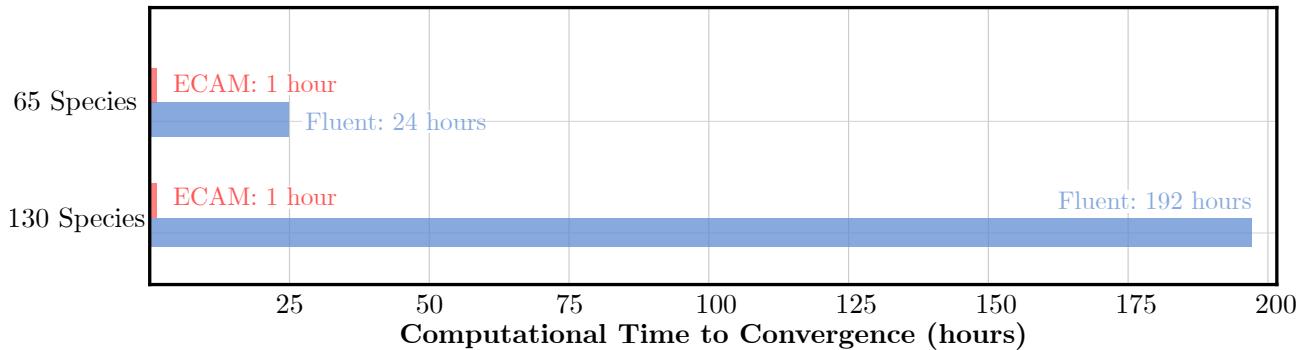
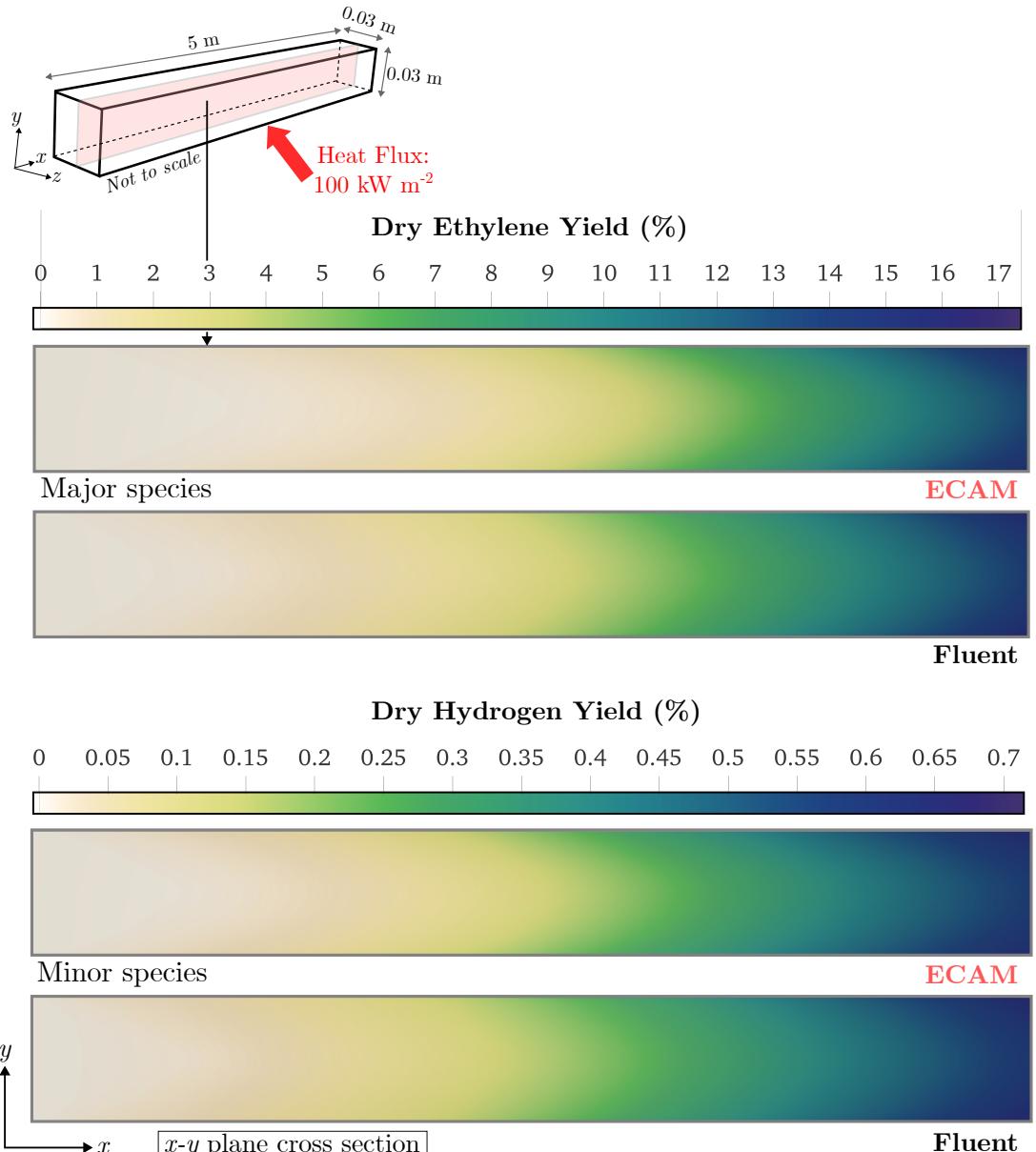


Figure 4: Computational speed of the ECAM relative to a commercial reacting flow solver FLUENT with a detailed chemical mechanism containing 65 species and 130 species

This sizeable computational acceleration is achievable whilst maintaining a sufficiently good level of accuracy such that reliable design optimisation can be performed in the future. Figure 5 demonstrates that the ECAM can predict dry yields for both major and minor species within a few percent relative error compared to FLUENT. Despite not being trained on any three-dimensional test cases (*i.e.*, all of the training data is from 1-D PFR simulations), the ECAM accurately predicts gradients in the yield perpendicular to the flow direction due to 3-D temperature non-uniformities (due to heated walls) and mass diffusion. Furthermore, Figs. 5 & 6 suggest that chemistry-related modelling errors do not accumulate and the yield distribution is accurate over the full 5 m flow path length.

The toughest test of the ECAM's capabilities is whether it can predict the spatial distribution of reaction rates, which are highly non-linear. Figure 6 shows that the ECAM is indeed successful in capturing spatial variability in reaction rates (induced primarily by temperature gradients). These factors support the generalisability and robustness of the ECAM given new and substantially different test data not seen during training. This strongly supports the feasibility of this new methodology.



Good prediction of yield distribution for both major and minor species in x and y directions

Figure 5: Verification of the ECAM against industry-standard commercial reacting flow solver FLUENT using a heated duct test case with a propane feedstock (65 species, 800 reactions): major and minor species mass fractions

As a proof-of-concept, the ECAM is now applied to an engine-relevant multistage axial turbo-reactor architecture. Figure 7 shows the ethylene yield & production rate, as well as the steady-state coke formation rate on the surfaces of the diffuser blade. Despite the highly complex nature of the flow in relation to the training data provided, the reaction responds as expected to sharp and complex gradients in the temperature and reaction timescale. This is the first-ever numerical simulation performed of the turbo-reactor capturing aerochemical interactions. This

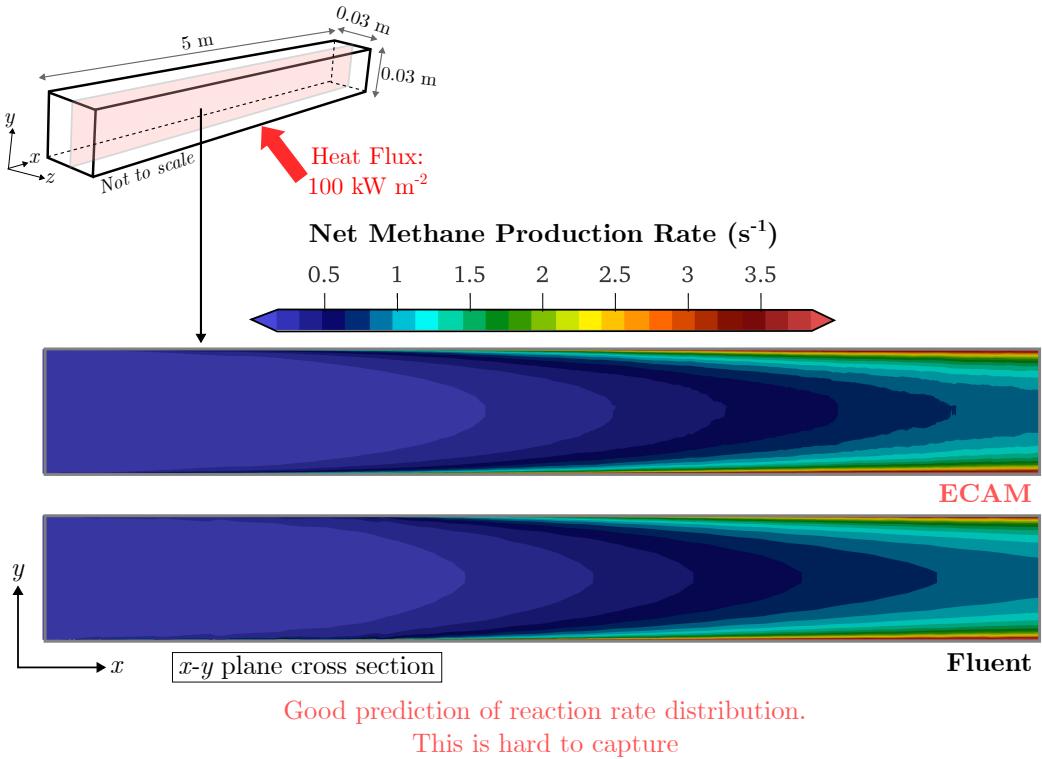


Figure 6: Verification of the ECAM against industry-standard commercial reacting flow solver FLUENT using a heated duct test case with a propane feedstock (65 species, 800 reactions): methane net production rates

will inform design decisions, facilitating aerodynamic design optimisation to improve reaction efficiency and mitigate coking.

3.4. Proposed Next Development Phase

In this next phase of methodology development and advancement, we intend to generalise, modularise and package this tool to be used universally across a range of industry-relevant chemical reactions and chemical processes. A first step towards this goal is to extend this multi-fidelity ML-assisted ECAM methodology not only to gas-phase reactions (achieved during my Doctoral research) but also to heterogeneous catalytic surface-phase reactions. This will substantially expand the applicability since many chemical processes feature catalytic reactions. This advancement in the ML-assisted ECAM methodology would involve

1. incorporating catalytic reactions in the training database generation
2. training an autoencoder neural network similar to that developed in Sec. 3.2 to capture the dynamics of surface-phase reactions

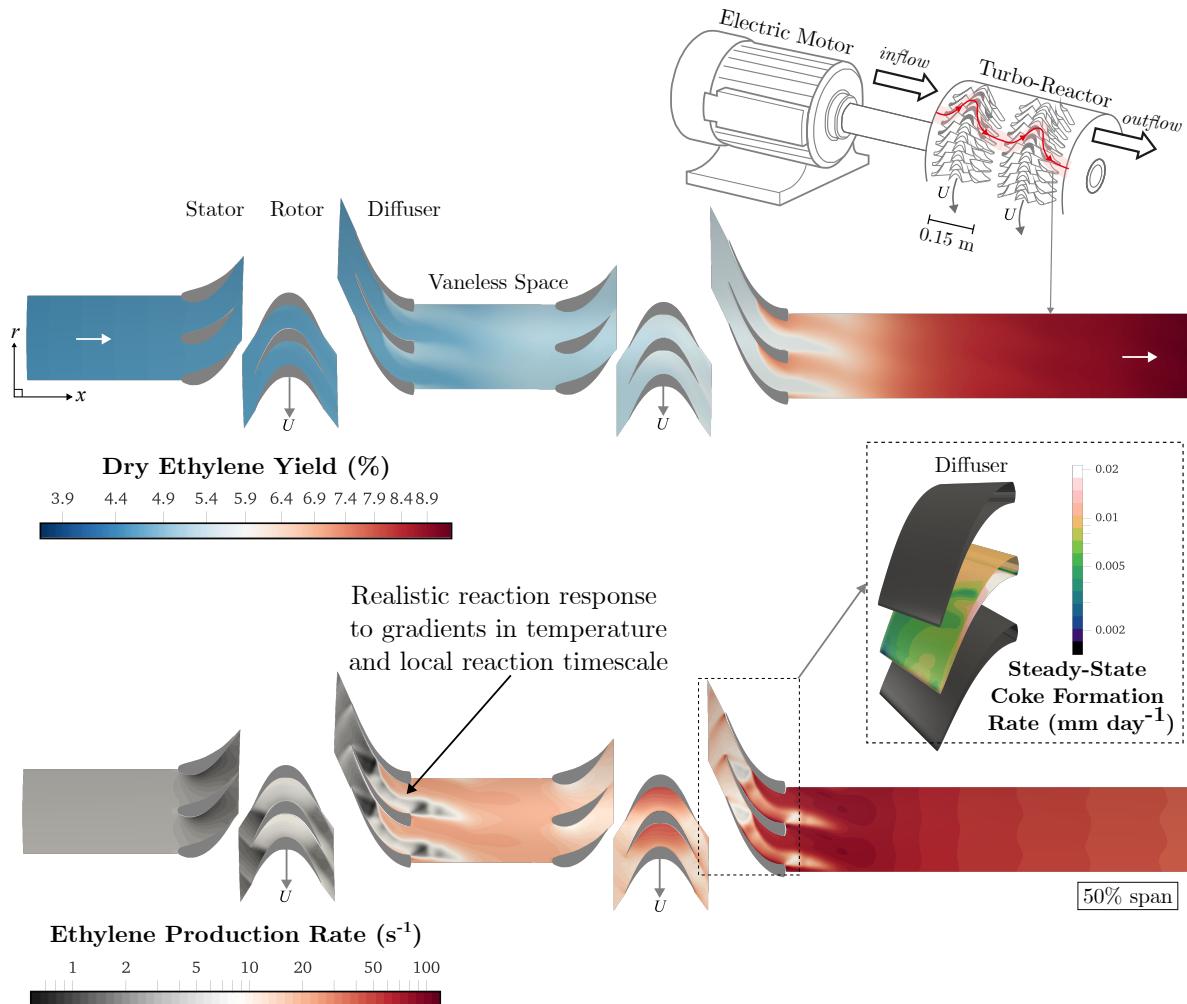


Figure 7: A proof-of-concept: first-ever numerical prediction of the chemical reaction yield, dynamics and coke formation rate predicted using the ML-assisted ECAM for a multistage axial turbo-reactor architecture

3. developing an interface treatment to couple the surface- and gas-phase reactions
4. implementing additional transport equations in our in-house CFD solver for the evolution of active surface sites

Once this has been achieved, this methodology can be translated into a user-friendly platform that can be universally applied to industry-relevant reaction processes.

Commercialising a numerical tool developed in a research setting is always challenging. However, working in close collaboration with an industrial partner will help to efficiently guide this development process, maximising the user-friendliness and modularity of the tool, and ensuring the requirements of industry are embedded in the platform early in the tool development cycle. Moreover, industry can provide unique validation data from chemical plants, which will help mitigate the risk and uncertainty of the methodology.

3.5. Summary

In summary, this work describes the development of a new multi-fidelity ML-assisted ECAM to provide an efficient solution to incorporate aerochemical interactions early in the design cycle. This will enable detailed and comprehensive chemical kinetic models that accurately capture realistic reaction dynamics (but contain $\mathcal{O}\{1000\}$ species and $\mathcal{O}\{10,000\}$ reactions) to be compressed into a reduced-dimensionality space, and coupled with a high-fidelity fluid solver at a computational cost 100 – 1000 times lower than FLUENT. This is a vital step in designing a new class of turbomachines with the unique requirement of controlling reactions along the bladed flow path.

This step-change advancement in the field of computational science brings multiphysics from being “one-off” simulations to being exploited routinely at the design optimisation level. Whilst historically, this development cycle would require many physical experiments, now, by integrating complex multiphysics early in the design cycle, the burden on experiments can be reduced and costs saved, accelerating the net-zero transition.

We hope to collaborate closely with the chemical industry to develop a modular, production-ready platform to automatically accelerate high-dimensional reacting flow numerical simulations. Iterative feedback with industry will be vital for rapidly developing a user-friendly tool that meets the user specifications.

Nomenclature

List of Symbols

$\dot{\omega}$	Net species production rate (s^{-1})
\mathcal{O}	Order of magnitude (–)
\tilde{y}	<i>Meta</i> species mass fraction (%)
M	Total number of <i>meta</i> species (–)
N	Total number of physical species (–)
T	Static temperature (K)
Y	Physical species mass fraction (%)

Abbreviations & Acronyms

ECAM	Efficiently coupled aerochemistry model
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CFD Computational fluid dynamics

LES Large eddy simulation

ML Machine learning

NN Neural network

PFR Plug-flow reactor

RANS Reynolds-averaged Navier-Stokes

Subscripts

\square_m Species index of M total *meta* species

\square_n Species index of N total physical species

Superscripts

$\hat{\square}$ Reconstructed quantity

$\tilde{\square}$ Latent space quantity