

## Problem 12-15

```

• # physical data
• begin
•     Ea = 4e4 # cal/mol
•     R = 1.987 # cal/mol/K
•     k0 = 6.6e-3 # 1/min
•     Fa0 = 80. # mol/min
•     cp = 50 # cal/mol
•     ΔH = -7500. # cal/mol
•     Ua = 8e3 # cal/min/K
•     Ta = 300 # K
•     τ = 100 # min
• end;

```

(a)

Heat generation:

$$G = r_A \Delta H$$

Heat removal:

$$R = U_A(T_a - T)$$

Rate law:

$$-r_A = kC_{A0}(1 - X_A)$$

$$k = k^0 \exp \left[ \frac{-E_a}{R} \left( \frac{1}{T^0} - \frac{1}{T} \right) \right]$$

Performance equation:

$$k\tau = \frac{X_A}{1 - X_A}$$

```

• begin
•   T = [300:0.1:500...]
•
•    $r_a(T) = k(T) * F_{a0} * (1 - X(T))$ 
•    $X(T) = k(T) / (1 + k(T))$ 
•    $k(T) = k^0 * \exp(-E_a * (1/350 - 1/T) / R)$ 
•
•    $heat\_inp(T; T_0) = c_p * (T - T_0)$ 
•    $heat\_rem(T) = U_a * (T_a - T)$ 
•    $heat\_gen(T) = r_a(T) * \Delta H$ 
• end;

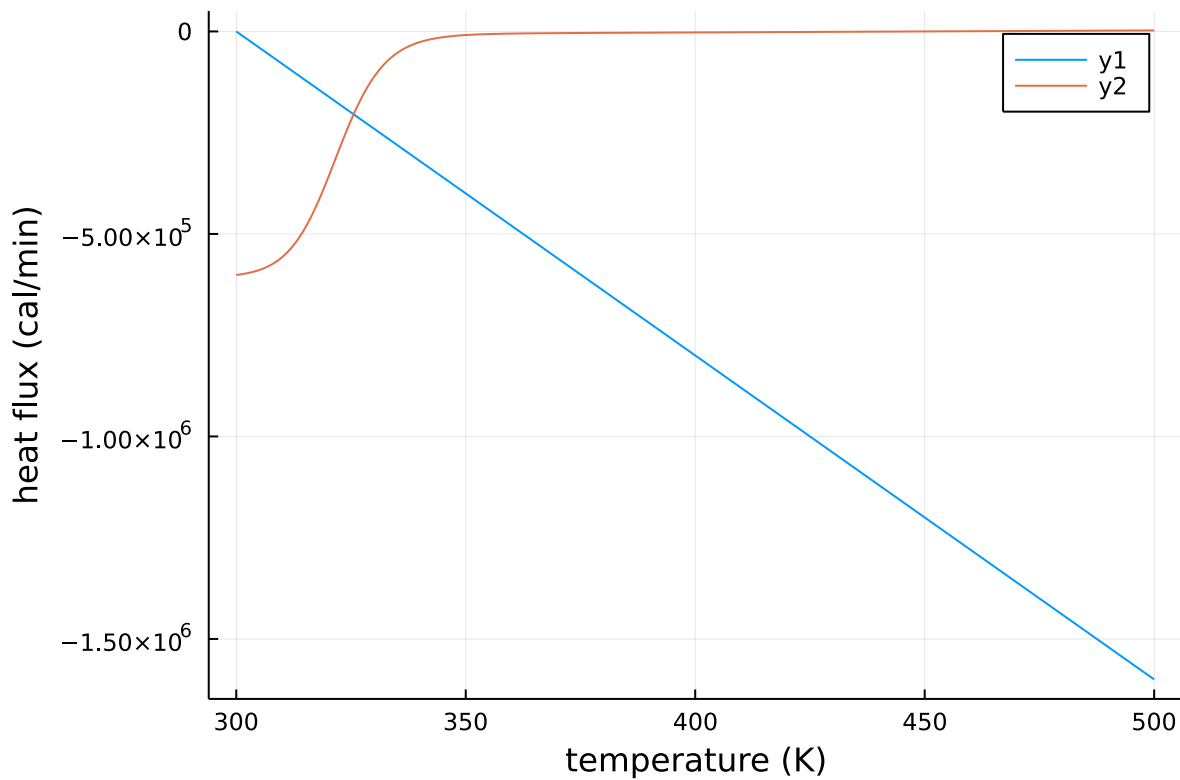
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The curves intersect at 326 K, corresponding to 33 % conversion.

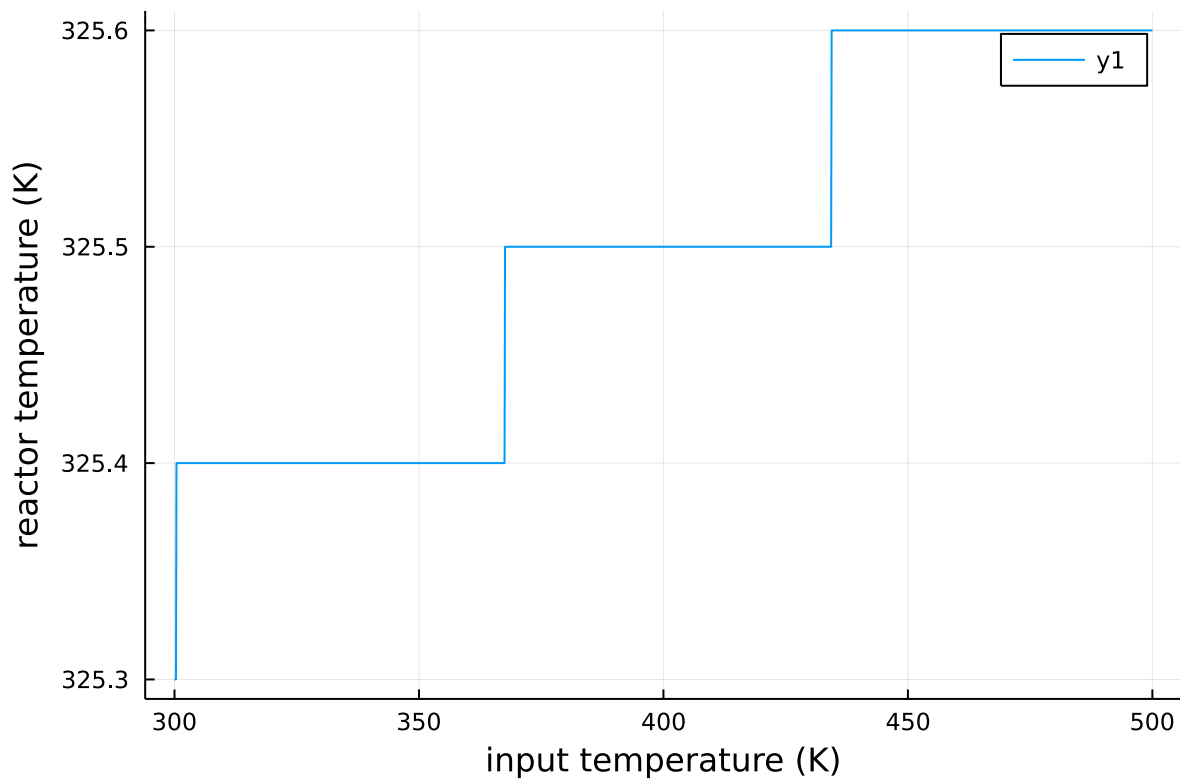
```

• begin
•   Rs = heat_rem.(T)
•   Gs = heat_gen.(T) + heat_inp.(T, T0=450.)
•   local i = argmin(abs.(Rs .- Gs))
•   local t = Int(round(T[i]))
•   local c = Int(round(X(T[i]) * 100))
•   md"The curves intersect at $t K, corresponding to $c % conversion."
• end

```



(b)



```
• begin
•     function find_T(T0)
•         Rs = heat_rem.(T)
•         Gs = heat_gen.(T) + heat_inp.(T, T0=T0)
•         i = argmin(abs.(Rs .- Gs))
•         return T[i]
•     end
•
•     plot(T, find_T.(T), xlabel="input temperature (K)", ylabel="reactor temperature
•         (K)")
• end
```